

# **PRETREATMENT TECHNOLOGY DEVELOPMENT**

**Presented to:**

**Nuclear Waste Technical Review Board**

**W. C. Miller**

**Richland, Washington  
May 11, 1992**

# **Pretreatment Technology Development**

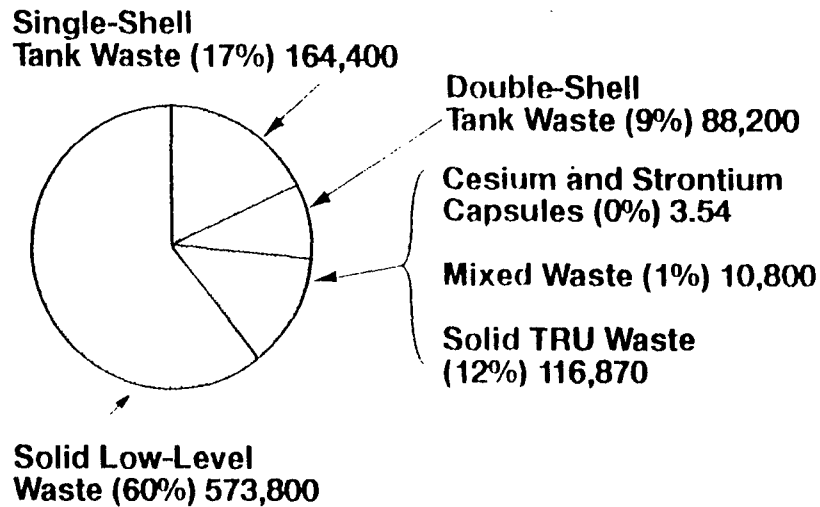
---

## **OVERVIEW**

- **Pretreatment to separate high- and low-level fractions of tank wastes presents some of the greatest technological challenges in the cleanup of Hanford**
- **Ability to separate radioactive constituents and destroy selected hazardous materials will have greatest single impact to total cost of tank waste disposal**
- **Recent program decisions have significantly changed the focus and approach to pretreatment**

# Hanford Site Waste

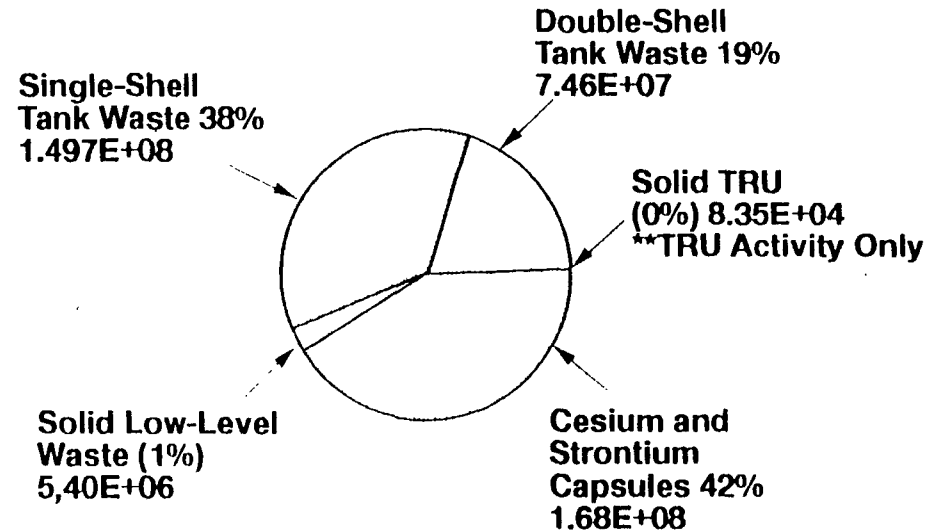
## Volume



Total Volume: \*954,073 m<sup>3</sup>

\*Does not include past practice units and surplus facilities  
Information per 1991 Integrated Data Base submittal

## Radioactivity



Total Radioactivity: \*3.98E+08 Ci

## Key Hazardous Materials

- Nitrates
- Nitrites
- Chromium
- Cobalt
- Heavy Metals
- Cyanide
- Carbon Tetrachloride
- Selenium
- Trichlorethylene

# Pretreatment Technology Development

---

## DST AND SST HIGH-LEVEL WASTE

<u>TYPE</u>	<u>INVENTORY AND SOURCE</u>	<u>CHARACTERISTICS</u>
<b>Single-Shell Tanks</b> <ul style="list-style-type: none"> <li>● Saltcake and Sludge</li> </ul>	164,000 m <sup>3</sup> (37 million gal) Fuel processing	The saltcake is almost entirely NaNO <sub>3</sub> , NaNO <sub>2</sub> , Na <sub>3</sub> PO <sub>4</sub> , NaAlO <sub>2</sub> , and NaOH. The Sludge consists of metal oxides and hydroxides precipitated during neutralization.
<b>Double-Shell Tanks</b> <ul style="list-style-type: none"> <li>● Neutralized Current Acid Waste (NCAW)</li> </ul>	5,300 m <sup>3</sup> (1.4 million gal) PUREX, 1st Cycle Extraction	High heat. Sludge is principally iron hydroxide (20 vol%) contaminated with <sup>90</sup> Sr and actinides. The alkaline supernate contains Na and Al salts, and is contaminated with <sup>137</sup> Cs.
<ul style="list-style-type: none"> <li>● Neutralized Cladding Removal Waste (NCRW)</li> </ul>	3,400 m <sup>3</sup> (0.9 million gal) PUREX, fuel cladding dissolution	Low heat. The sludge contains Zr contaminated with TRU. The alkaline supernate is LLW.
<ul style="list-style-type: none"> <li>● Plutonium Finishing Plant (PFP)</li> </ul>	400 m <sup>3</sup> (100,000 gal) Plutonium Finishing Plant	Low heat. Sludge is principally metallic compounds contaminated with insoluble TRU compounds. The alkaline supernate is LLW.
<ul style="list-style-type: none"> <li>● Complexant Concentrate (CC)</li> </ul>	16,300 m <sup>3</sup> (4.3 million gal) Prior Sr and Cs Recovery Operations	The sludge contains metal compounds, degraded complexants and precipitated TRU elements. The alkaline supernate contains <sup>137</sup> Cs and TRU solubilized with complexants.

# Pretreatment Technology Development

---

## HANFORD DOUBLE-SHELL TANK WASTE COMPOSITIONS

<u>Chemical Components (M)</u>	<u>NCAW Slurry</u>	<u>NCRW Sludge</u>	<u>PFP Sludge</u>	<u>CC Sludge</u>	<u>CC Supernate</u>
A1	0.50	0.20	0.85	2.50	0.40
Cr <sup>(b)</sup>	0.012	0.016-04	0.40	0.001-15	.004-008
Fe	0.07	0.02	0.20	0.04	0.04
K	0.12	0.35	0.08	0.10	0.05
Na	5.00	6.50	2.20	15.00	10.00
PO	0.025	NA	0.3	0.06-12	.02-05
Zr	0.04	1.04	NA	0.001	0.001
OH	1.00	0.80	1.00	2.00	0.10
TOC	0.14	0.50	NA	3.30	3.30
<b>Radioactive Components (uCi/L)</b>					
Cs-137	2E6	7E4	1E5	6E5	4E5
Sr-90	2E6	NA	1E4	1E5	1E5
Pu-239	160	700	1E6	100	100
Am-241	8900	300	3E6	800	800
Volume <sup>(a)</sup> (10 <sup>6</sup> Gal)	1.4	0.8	0.25	0.6	4.1

(a) No Future PUREX or PFP operations assumed.

(b) Shaded areas identify chemical components which limit waste loading in glass.

# Pretreatment Technology Development

---

## ESTIMATED TOTAL SINGLE-SHELL TANK WASTE CONTENTS DISTRIBUTION

<u>Compound or Nuclide</u>	<u>Total Weight Mg</u>	<u>wt% of Total Waste</u>	<u>wt% of Total Waste (water free basis)</u>
<b>Non-Radioactive</b>			
NaNO <sub>3</sub>	132,500	58.3	72.6
Na <sub>3</sub> PO <sub>4</sub>	14,800	6.6	8.1
NaNO <sub>2</sub>	7,200	3.1	3.9
NaOH	6,940	3.0	3.8
NaAlO <sub>2</sub>	4,350	1.9	2.4
Na <sub>2</sub> CO <sub>3</sub>	2,400	1.1	1.4
Na <sub>2</sub> SO <sub>4</sub>	2,440	1.1	1.3
Other Solids	10,390	4.6	5.7
Water	<u>44,800</u>	<u>19.7</u>	<u>---</u>
<b>Total Non-Radioactive</b>	<b>226,000</b>	<b>99.4</b>	<b>99.2</b>

# Pretreatment Technology Development

---

## ESTIMATED TOTAL SINGLE-SHELL TANK WASTE CONTENTS DISTRIBUTION (CONT)

<u>Compound or Nuclide</u>	<u>Total Weight</u> <u>Mg</u>	<u>wt%</u> <u>of Total Waste</u>	<u>wt%</u> <u>of Total Waste</u> <u>(water free basis)</u>
<b>Radioactive</b>			
<b>Radionuclides</b>			
<sup>238</sup> U	1,400.	0.6	0.8
<sup>235</sup> U	9.40	--	--
<sup>93</sup> Zr	1.60	--	--
<sup>99</sup> Tc	0.93	--	--
<sup>239</sup> PU	0.36	--	--
<sup>90</sup> Sr	0.33	--	--
<sup>135</sup> Cs and <sup>137</sup> Cs	0.22	--	--
Other TRU	0.06	--	--
Other Radionuclides	<u>0.20</u>	<u>--</u>	<u>--</u>
<b>Total Radionuclides</b>	<u><u>1,413.</u></u>	<u><u>0.6</u></u>	<u><u>0.8</u></u>
<b>Total Compounds and Nuclides</b>	227,413.	100.0	100.0

# **Pretreatment Technology Development**

---

## **BACKGROUND**

- **Degree of pretreatment dramatically impacts quantity of glass canisters produced**
  - **> 200,000 canisters with no separations**
  - **> 40,000 canisters with removal of soluble salts only**
  - **10,000-15,000 canisters with selected radionuclide separation**
- **Current estimate of canister disposal costs (approximately \$350K/canister) indicates need for aggressive pretreatment strategy**



# **Pretreatment Technology Development**

---

## **PURPOSE / OBJECTIVE**

- **Resolve tank safety issues by destroying organics and ferrocyanides**
- **Minimize glass volume and disposal costs through efficient separation of radionuclides and inert material**
- **Provide feed for vitrification and grout within allowable feed specifications**

# **Pretreatment Technology Development**

---

## **PROCESS FUNCTIONS**

- **Soluble Salts Removal**
- **Soluble Fission Product Removal**
- **Insoluble Fission Product Removal**
- **Actinide Separations**
- **Organic/Ferrocyanide Destruction**
- **Nitrate/Nitrite Destruction (TBD)**

# Pretreatment Technology Development

---

## PROCESS OPTIONS

- **Soluble Salt Removal**
  - Liquid/solids separation
  - In-tank sludge washing
  
- **Soluble Fission Product Removal**
  - Ion exchange
  - Solvent extraction
  - Precipitation
  
- **Insoluble Fission Product Removal**
  - Sr extraction
  - Precipitation
  
- **Actinide Separations**
  - TRUEX
  - Solvent extraction other than TRUEX
  - Selective leaching
  - Ion exchange/solid sorbants
  - Precipitation
  - Salt splitting

# **Pretreatment Technology Development**

---

## **PROCESS OPTIONS (CONT)**

- **Organic/Ferrocyanide Destruction**
  - Thermal oxidation
  - Chemical oxidation
  
- **Nitrate/Nitrite Destruction**
  - Thermal oxidation

# **Pretreatment Technology Development**

---

## **THREE PHASED APPROACH FOR PRETREATMENT**

- **Near Term Processes -- generally proven technologies implemented mostly in-tank (or "modules")**
  - **Organic/ferrocyanide destruction**
  - **Sludge wash**
  - **Ion exchange/cesium removal**
- **Intermediate Term Processes -- more complex, evolving technologies implemented in-tank (or "modules")**
  - **Chrome leaching**
  - **Aluminum leaching**
  - **Transuranic leaching**

# **Pretreatment Technology Development**

---

## **THREE PHASED APPROACH FOR PRETREATMENT (CONT)**

- **Long Term Processes -- aggressive new technologies to minimize disposal costs, likely implemented in new facilities**
  - **Sludge dissolution**
  - **Transuranic extraction**
  - **Strontium extraction**
  - **Technetium removal - ion exchange**
  - **Organic destruction**

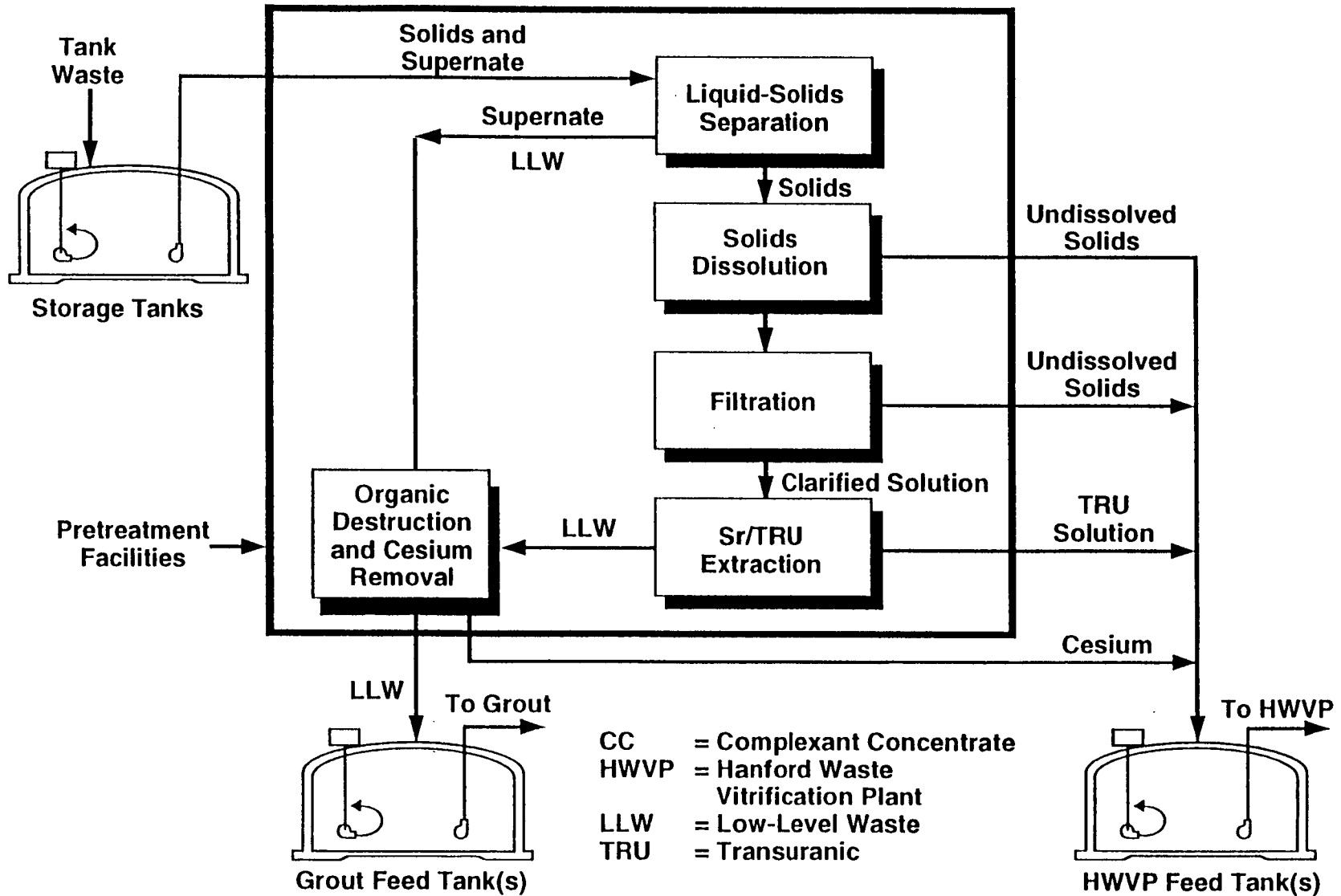
# **Pretreatment Technology Development**

---

## **ALTERNATIVE PROCESSES**

- **Innovative technologies being pursued which have performance and/or cost incentives; e.g.:**
  - **HLW/LLW splitting by salt crystallization**
  - **Solid sorbants**
  - **Nitrate destruction**
  - **Leaching/dissolution**
  - **Calcination/leaching**

# Typical Pretreatment Flow Diagram





# **Pretreatment Technology Development**

---

## **PRETREATMENT FOR TANK SAFETY ISSUES**

### **Problem**

- **Hydrogen, ferrocyanide, and organic salts tanks are all on the safety watch list due to the combination of a fuel (hydrogen, ferrocyanide, or organic salts) and an oxidizer (nitrate) in the tanks**

### **Solution**

- **If remediation is required after mitigation is complete, all three categories would be treated by some form of oxidation process**

# **Pretreatment Technology Development**

---

## **POTENTIAL OXIDATION PROCESSES**

- **Ozonization**
- **High temperature/high pressure wet oxidation**
- **Electrochemical oxidation**
- **Calcination**
- **Other**

# **Pretreatment Technology Development**

---

## **SLUDGE WASHING**

- **Simple process**
  - **Gravity settling of solids**
  - **Water washing**
- **Not a new concept**
- **Similar to processing at Savannah River, West Valley, and previous Hanford HLW**
- **In-tank washing planned for NCAW**
- **Other candidates include 101-AY, 106-C, and 105-C**
  - **Additional candidates under consideration**

# **Pretreatment Technology Development**

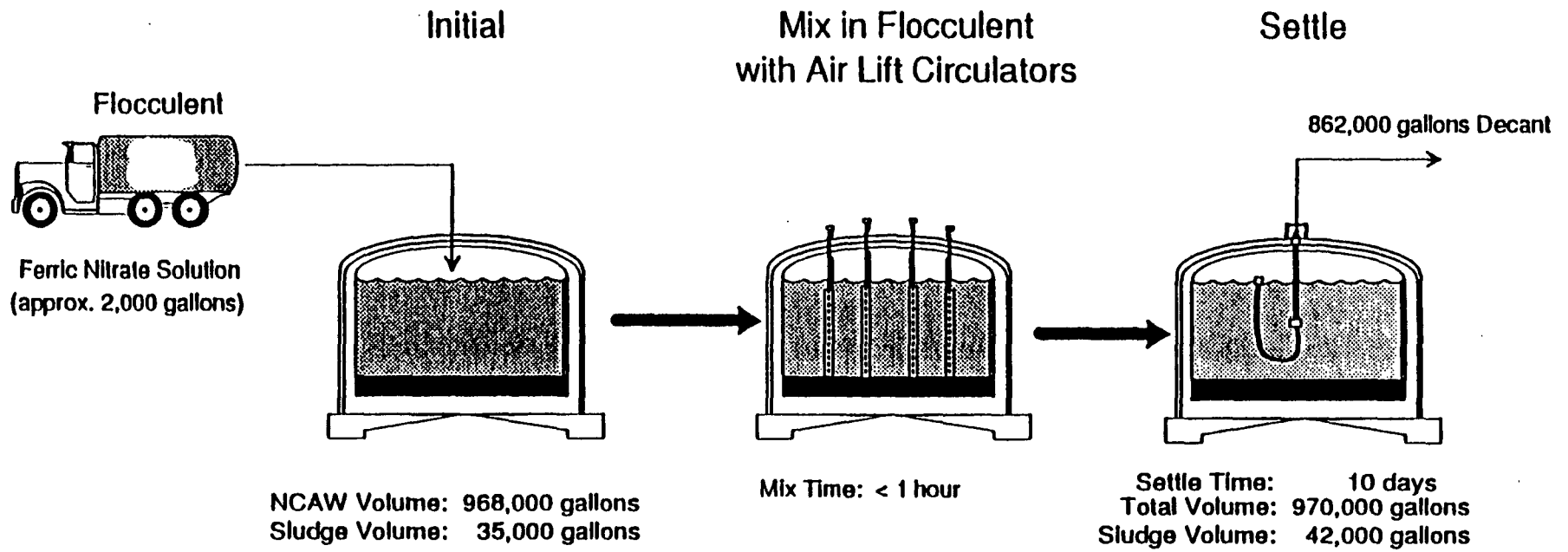
---

## **SLUDGE WASHING (CONT)**

- **Significant development work completed**
  - **Pilot plant test with synthetic NCAW**
  - **Lab test with actual NCAW**
  - **Process testing with actual NCAW at B Plant**

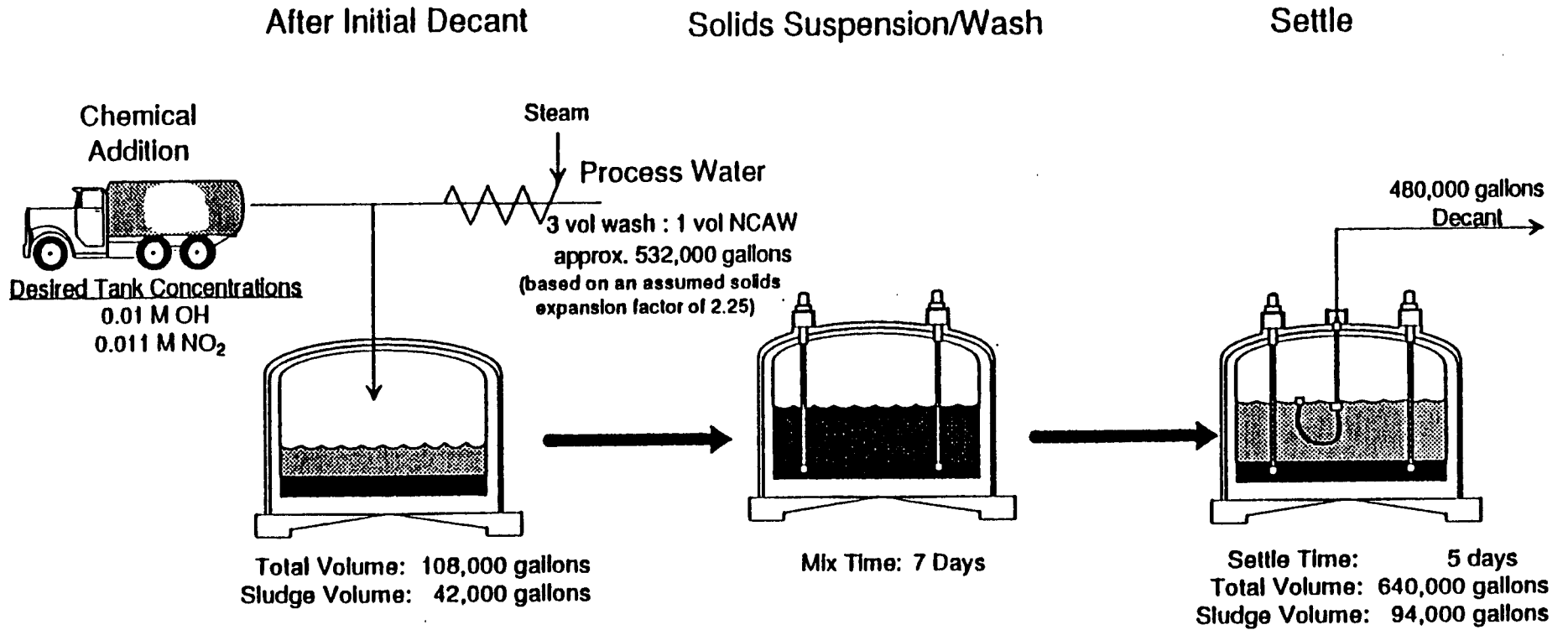
# NCAW In-Tank Sludge Washing

## Initial Treatment & Decant of Tank 241-AZ-101



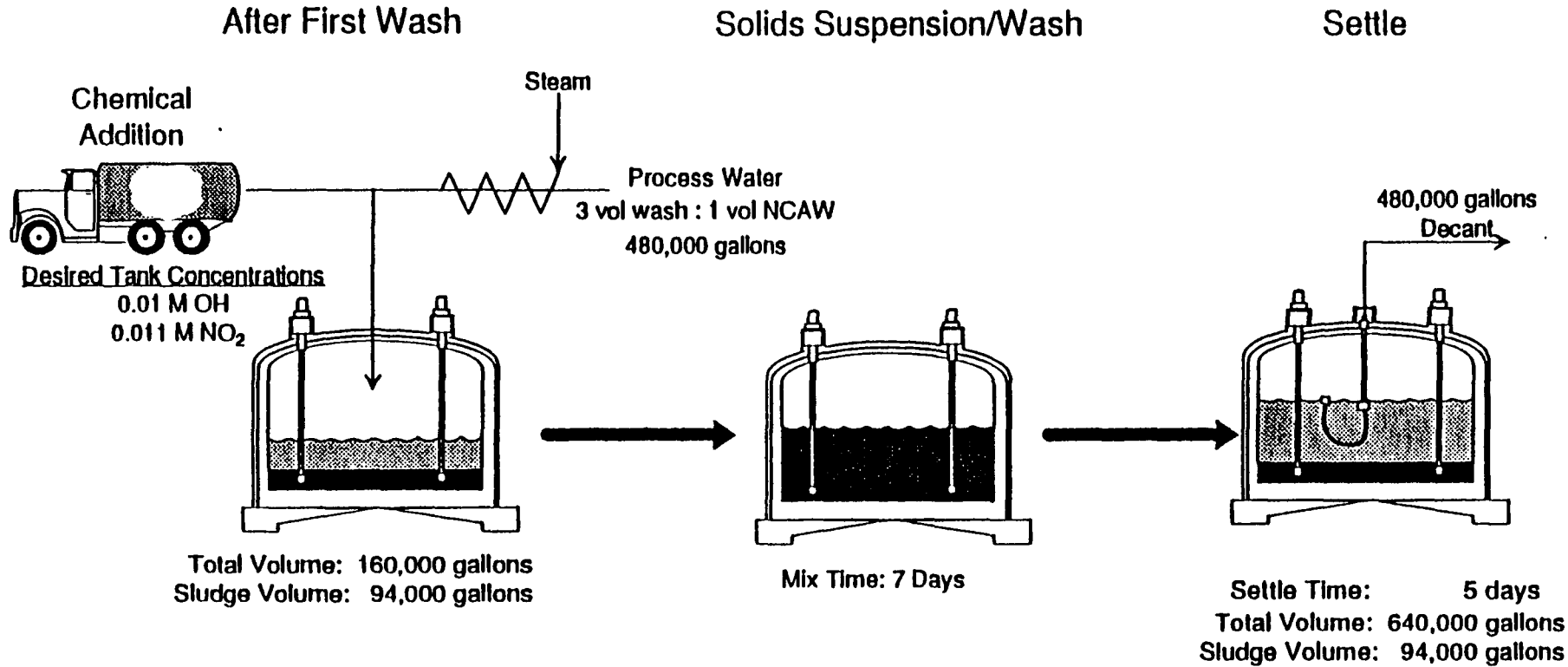
# NCAW In-Tank Sludge Washing

## First Wash Cycle



# NCAW In-Tank Sludge Washing

## Second Wash Cycle



# **Pretreatment Technology Development**

---

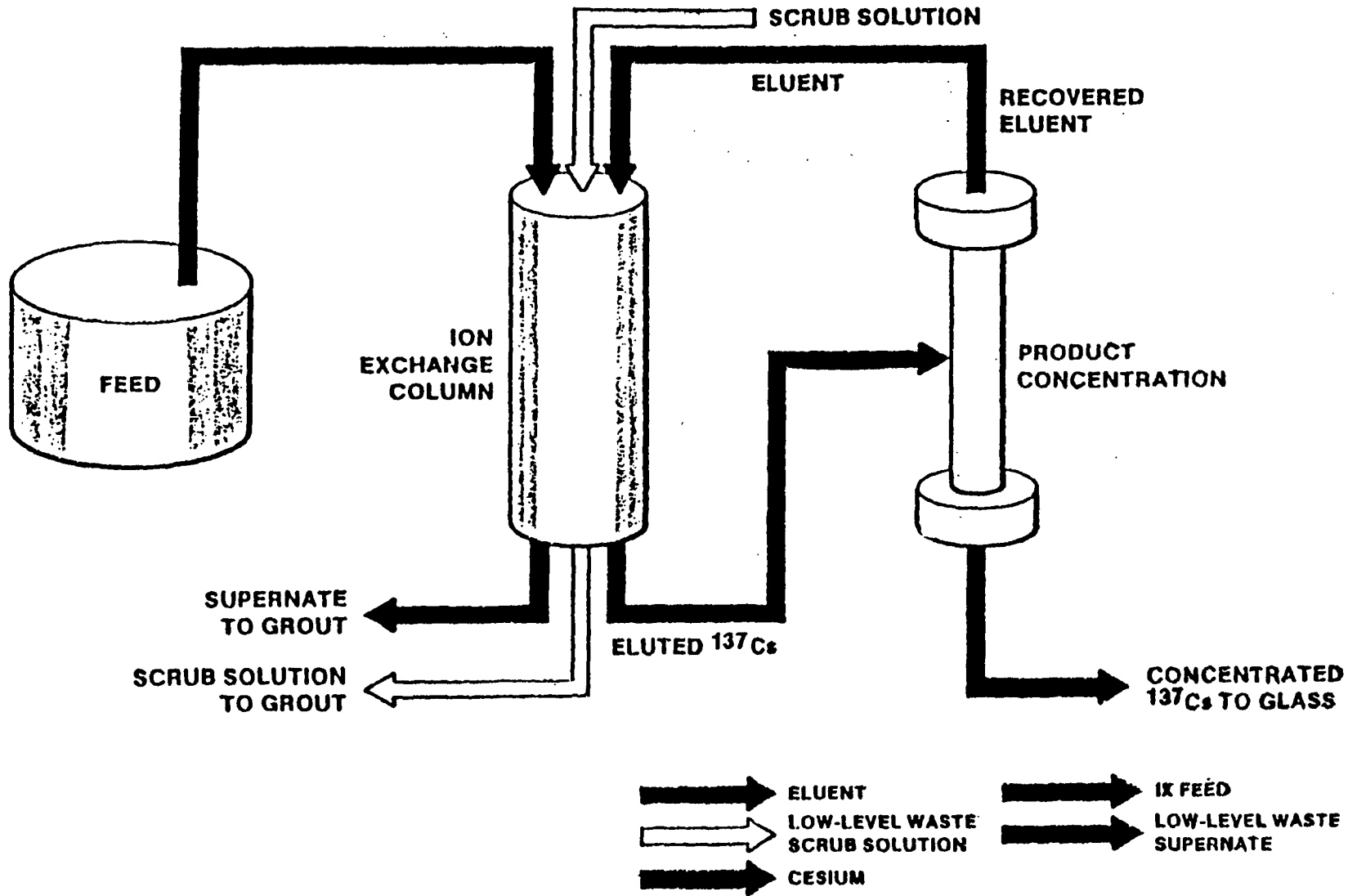
## **ION EXCHANGE**

- **Batch screening tests of eight organic and inorganic ion exchange media**
  - **Resorcinol/formaldehyde resin (SRL-DJ) has highest Cs capacity**
  - **"Duolite CS-100"<sup>TM</sup> is next best**
- **Radiation effect**
  - **Hanford data shows CS-100 resists capacity degradation better than resorcinol/formaldehyde resin**
  - **Recent SRL data shows resorcinol/formaldehyde resists degradation better than CS-100**
  - **Jury is still out on radiation stability**

**"Duolite CS-100 is a trademark of Rohm and Haas Company."**



# ION EXCHANGE SCHEMATIC



# **Pretreatment Technology Development**

---

## **ION EXCHANGE (CONT)**

- **Column loading and elution tests with CS-100**
  - As high as 45 column volumes before significant breakthrough
  - Cs recovery efficiency of the total cycle was 95%
- **Aging/cycling tests with CS-100**
  - After 55 elution/regeneration cycles, CS-100 had 60% of initial Cs capacity as determined by a batch contact
  - Particle attrition was apparent
- **Current column loading tests with CS-100**
  - NCAW supernates and wash waters

# **Pretreatment Technology Development**

## **STATUS OF ION EXCHANGE PROGRAM**

- **Column loading experiments are in process**
- **Cs removal alternatives study completed**
- **Some engineering studies have commenced**
- **Parametric batch equilibrium studies have commenced**

# **Pretreatment Technology Development**

## **OBJECTIVES OF INTERMEDIATE-TERM WASTE PRETREATMENT PROCESSES**

- **Pretreat wastes to reduce canister production compared to sludge-wash only**
  - **Facility requirements similar to sludge-washing/ion exchange processing**
- **Provide suitable feeds to HWVP until long-term pretreatment approaches (sludge dissolution - Actinide separation) are available**
  - **Ensure cost effective, viable Tank Waste Disposal Program**

# **Pretreatment Technology Development**

---

## **INTERMEDIATE TERM PROCESS DEVELOPMENT**

- **Review waste characterization data to identify potential candidate tanks for intermediate processing**
  - **Look for high fission product/TRU tanks that would produce relatively low volumes of glass if selected components such as chrome or aluminum were removed from waste**
  - **Look for low fission product, high TRU wastes that would benefit from leaching TRU from waste**
- **Continue lab testing on actual PFP waste to remove chrome from waste**
- **Initiate lab tests on potential candidate wastes for those tanks that have been sampled**

# **Pretreatment Technology Development**

---

## **LONG-TERM PROCESS DEVELOPMENT**

- **Focus of program had been on solvent extraction (TRUEX) process for double-shell tank wastes**
- **Systems engineering study for disposal of single-shell tank (SST) wastes identified solvent extraction as a prime candidate for SST wastes**
- **Recent reviews have raised technical concerns with TRUEX process and lack of fallback options**
- **Program was revised to investigate other alternatives prior to selecting long-term processes**
  - **Three phased strategy provides the needed time to perform these investigations**
- **Alternatives will be evaluated by Pretreatment Technology Working Group and processes selected for investigation will be documented in Pretreatment Technology Plan**

# **Pretreatment Technology Development**

---

## **SOLVENT EXTRACTION PROCESS STATUS**

- **PNL and ANL have done extensive batch contacts on a variety of synthetic wastes including the following Hanford specific wastes:**
  - **NCAW**
  - **PFP Acid Waste**
  - **CC Waste**
- **ANL has done some continuous tests on synthetic waste including the PFP acid waste**
- **ANL has done significant work on the effect of radiation on the stability of CMPO; no significant degradation found**

# **Pretreatment Technology Development**

---

## **REMAINING SOLVENT EXTRACTION DEVELOPMENT**

- **Material of Construction**
  - A corrosion program has been initiated
- **Effect of feed composition variations to dissolution and solvent extraction processes**
  - A lab program is planned which includes investigating feed variations
- **Waste minimization**
  - A denitrification process to reduce nitrate levels in waste is being pursued
- **Confirmation of a continuous process with actual waste**
  - Continuous lab tests are planned to support confirmation
  - A pilot plant is proposed to confirm the process



# **Pretreatment Technology Development**

---

## **CURRENT ACTIVITIES IN ALTERNATE TECHNOLOGIES**

- **Leaching/dissolution of SST sludges**
  - **Strong caustic for aluminum removal**
  - **Carbonate leaching for TRU reduction**
  - **Nitric acid leaching and dissolution**
- **HLW/LLW splitting by salt crystallization**
- **Solid sorbant for actinide separation**
- **Salt splitting/recycle with bipolar electrodyatic membrane (LLNL)**
- **Destruction of organics by steam reforming (SANDIA)**
- **Nitrate destruction in aqueous solutions (ORNL)**
- **TRUEX model development (ANL)**

# **Pretreatment Technology Development**

---

## **ADDITIONAL ALTERNATE PROCESSES BEING CONSIDERED**

- **In-tank dissolution and precipitation**
- **Solid/solid separations**
- **Calcination/leaching**
- **In-tank processing**
- **Electro-chemical separations**

# **Pretreatment Technology Development**

---

## **ADDITIONAL ALTERNATE PROCESSES BEING CONSIDERED (CONT)**

- **Magnetic separations**
- **Nitrate destruction/conversion**
- **Solvent extraction alternatives**
  - **Initiate batch tests of DST waste using French diamide solvent**
- **Other**
  - **Initiate foreign technology exchange**
  - **Complete alternatives study to examine other processes**

# **Pretreatment Technology Development**

---

## **SUMMARY**

- **Near-term pretreatment processes are mature and will provide several years of feed to HWVP**
- **Substantial development is required to implement cost-effective long-term processes which will minimize total costs of Hanford Tank Waste Disposal Program**
- **Three-phased approach to pretreatment provides time to complete this development**