



U.S. DEPARTMENT OF ENERGY OFFICE OF CIVILIAN RADIOACTIVE WASTE MANAGEMENT

PRESENTATION TO THE NUCLEAR WASTE TECHNICAL REVIEW BOARD

SUBJECT: RETARDATION OF GAS PHASE RADIONUCLIDES

PRESENTER: AREND MEIJER

PRESENTER'S TITLE: GEOCHEMIST ORGANIZATION: LANL/GCX

PRESENTER'S TELEPHONE NUMBER: (505) 256-3769

JULY 12, 1994

RETARDATION OF GAS-PHASE RADIONUCIDES IN UZ

PRESENTATION OUTLINE

- GAS-PHASE SOURCE TERM
- GAS FLOW IN UZ DATA AND MODELS
- RETARDATION MECHANISMS
- DATA AND MODELING NEEDS
- CONCLUSIONS

GAS-PHASE SOURCE TERM

- RADIONUCLIDES OF GREATEST CONCERN
 - 14C, 129I, 99Tc, 79Se
- IMPORTANT CHEMICAL SPECIES IN GAS-PHASE
 - CO2, I2, Tc2O7, SeO2
- 14CO₂ MOST IMPORTANT GAS-PHASE RADIONUCLIDE
- APPROXIMATELY 2% OF 14CO₂ PRESENT IN LABILE FRACTION AVAILABLE FOR QUICK RELEASE AFTER FAILURE OF A WASTE PACKAGE
- TOTAL 14CO₂ INVENTORY APPROX. 1.3X10³ MOLES (VAN KONYNENBURG, 1989, 1992, 1993 and PARK, 1991)

MODELS OF GAS FLOW IN UZ

- BUOYANCY-DRIVEN ADVECTION (ROSS ET AL., 1992)
 - 14CO₂ TRANSPORTED TO SURFACE IN 0.5-2 X 10³ YR. DEPENDING ON RETARDATION AND RELEASE TIMES
 - MODEL USED IN TSPA 93
 - PREDICTED RELEASES EXCEEDING EPA REGULATIONS
- BAROMETRIC PUMPING (NILSON ET AL., 1994)
 - 14CO₂ TRANSPORTED TO SURFACE IN 1-2X10² YR. DEPENDING ON BULK PERMEABILITY (1-50 DARCIES), RETARDATION AND RELEASE TIMES
- HOT REPOSITORY WILL INCREASE TRANSPORT RATE

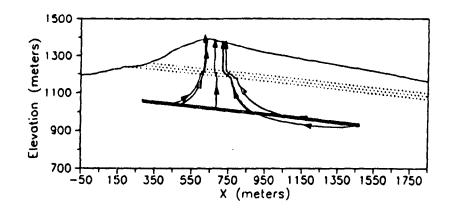


Figure 6a. Path lines with ambient temperature, 0.1× permeability contrast.

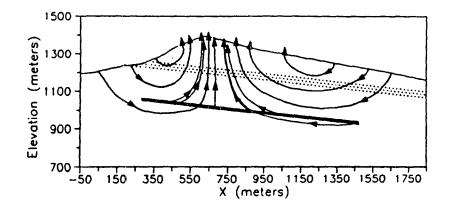
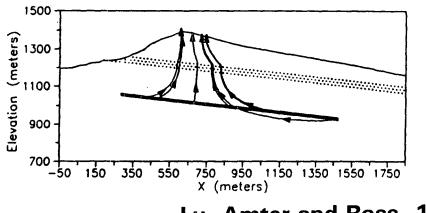


Figure 6b. Path lines with ambient temperature, no permeability contrast.



Lu, Amter and Ross, 1991

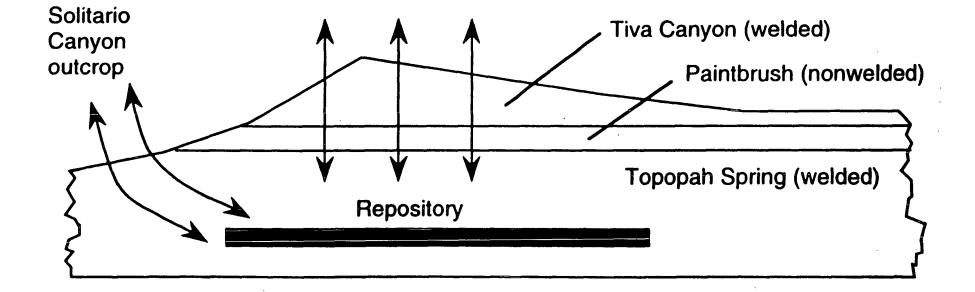


Figure 21-2. Schematic of proposed Yucca Mountain repository. Repository may breathe vertically through the paintbrush or horizontally through the Solitario Canyon outcrop.

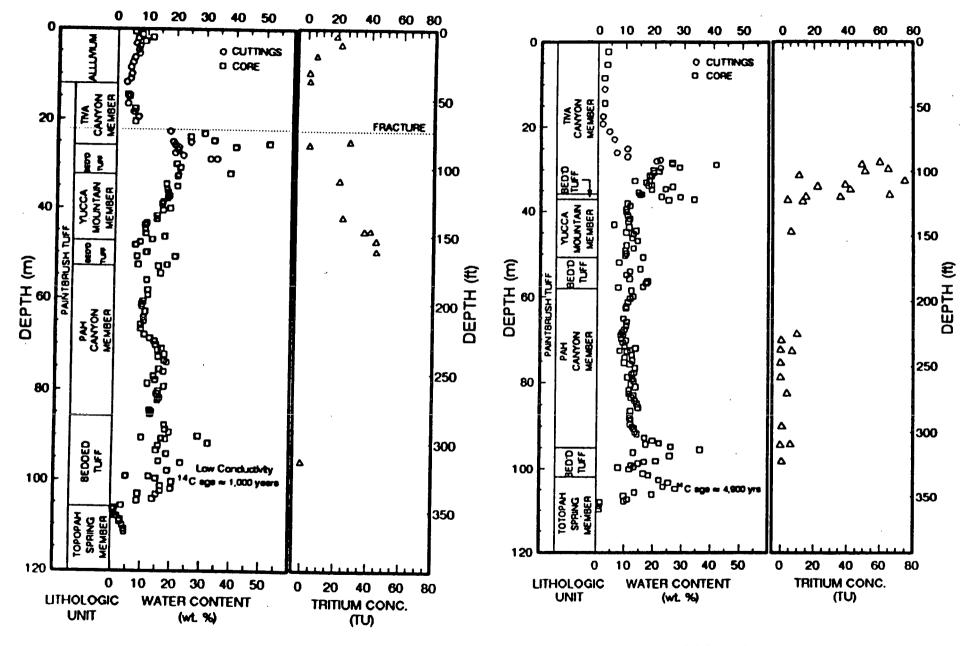
NILSON ET AL. (1994)

GAS-FLOW IN UZ - ISOTOPIC EVIDENCE

- ISOTOPIC DATA SUGGESTS 2 FLOW REGIMES
 - "Shallow" (<50m) and "Deeper" (>50m)
- "SHALLOW" REGIME SHOWS BOMB-PULSE 3H AND 36CI ABOVE PAH CANYON MEMBER
- "DEEPER" REGIME SHOWS LINEAR DECREASE IN 14CO2 WITH INCREASING DEPTH SUGGESTING DOWNWARD TRANSPORT
- GAS-PHASE AND AQUEOUS-PHASE 14CO₂ "AGES" (1-10X10³ YR) ARE ON AVERAGE MUCH YOUNGER THAN AQUEOUS-PHASE 36CI "AGES" (<50-750X10⁵ YR)

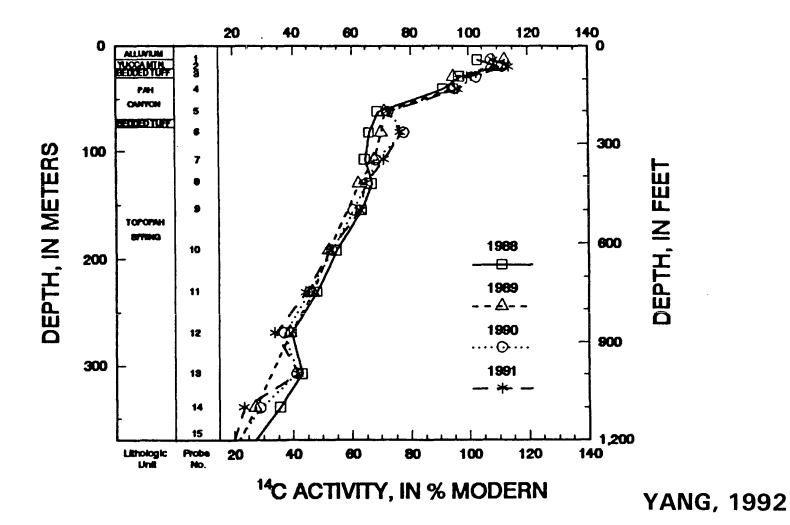






YANG ET AL., 1993





POTENTIAL 14CO2 RETARDATION MECHANISMS

- LOW REDOX POTENTIAL IN ENGINEERED BARRIER
- <u>SORPTION</u> ONTO METAL OXIDES AND OXYHYDROXIDES
- <u>COPRECIPITATION</u> AS CaCO₃ IN CEMENT/CONCRETE
- <u>COPRECIPITATION</u> AS CaCO₃ DURING DRYOUT PHASE
- ISOTOPE EXCHANGE WITH HCO3- IN UZ PORE WATER
- ISOTOPE EXCHANGE WITH Ca12CO3 IN CEMENT
- ISOTOPE EXCHANGE WITH Ca12CO3 IN DRY-OUT ZONE

LOW REDOX POTENTIAL IN ENGINEERED BARRIER

- UZ GAS PHASE AT POTENTIAL REPOSITORY HORIZON LEVEL CURRENTLY HAS ATMOSPHERIC OXYGEN CONTENT
- RADIOLYSIS TENDS TO PRODUCE OXIDIZING AGENTS (E.G., H₂O₂, HNO₃, ETC)
- EVEN TRACE AMOUNTS OF O₂ RESULT IN THE PRODUCTION OF CO₂ IN FUEL MAINTAINED IN ARGON
- EXISTENCE OF SUFFICIENTLY REDUCING CONDITIONS TO PREVENT THE FORMATION OF CO₂ AFTER WASTE PACKAGE FAILURE CONSIDERED UNLIKELY SCENARIO
- OXIDATION RATE OF FUEL WILL RETARD RELEASE OF ALL BUT INITIAL "QUICK-RELEASE" FRACTION (CODELL AND WESCOTT, 1992)

SORPTION ONTO METAL OXIDES

- WASTE PACKAGES TO CONTAIN APPROX. 5 m³ CARBON STEEL EACH. 2,000 WASTE PACKAGES = $1.0 \times 104 \text{ m}^3$.
- ASSUME OXIDATION RESULTS IN OXIDE PHASE WITH 10m²/g SURFACE AREA
- ACCORDING TO van GEEN ET AL. (1994), CO₂ SURFACE COVERAGE ON GOETHITE = $3.8 \times 10^{-6} \text{ M/m}^2$
- TOTAL CO₂ RESERVOIR ON OXIDIZED CORROSSION ALLOWANCE LAYER = $3 \times 10^6 \text{ M}$
- TOTAL 14CO₂ INVENTORY IN REPOSITORY = $1.3 \times 10^3 \text{ M}$

COPRECIPITATION WITH Ca¹²CO₃ IN CONCRETE AND DRY-OUT ZONE

- CARBONATION OF CEMENT/CONCRETE BY ATMOSPHERIC CO₂ WILL LIKELY BE NEAR MAXIMUM AT TIME REPOSITORY IS SEALED
- COPRECIPITATION OF Ca¹⁴CO₃ WITH Ca¹²CO₃ IN ROCK UNITS SURROUNDING REPOSITORY DURING DRY-OUT IS POSSIBLE BUT VERY DEPENDENT ON WASTE PACKAGE FAILURE TIMES
- COPRECIPITATION OF 14CO₃ IN CARBONATE MINERALS PROBABLY NOT A SIGNIFICANT RETARDATION MECHANISM

MAGNITUDES OF 12C AND 14C RESERVOIRS

- TOTAL 14CO₂ INVENTORY IN WASTE IS 1.3X10³ MOLES
- AQUEOUS 12C RESERVOIR ON ORDER OF 1.0 X 109 MOLES IN THE UZ ABOVE THE POTENTIAL REPOSITORY. THIS PROVIDES A RETARDATION FACTOR RANGING FROM 40-180 AS A FUNCTION OF TEMP. INCLUDED IN TSPA 93.
- CARBONATE MINERALS PRECIPITATED AS A RESULT OF NEAR-FIELD DRY-OUT WILL PROVIDE A SMALLER (MINERAL SURFACE) EXCHANGE RESERVOIR COMPARED TO THE DISPLACED AQUEOUS RESERVOIR
- TOTAL 12C RESERVOIR PRESENT IN CEMENT/CONCRETE IS ON ORDER OF 1-2X10⁸ MOLES BUT ONLY A PORTION OF THIS (i.e., SURFACES) AVAILABLE FOR EXCHANGE

CARBON ISOTOPE EXCHANGE KINETICS ON SOLID CARBONATE MINERALS

- EXPERIMENTS BY MOZETO ET AL. (1984) INDICATE CARBON ISOTOPE EXCHANGE ON CALCITE IS A 2-STEP PROCESS - ADSORPTION (HRS) AND ISOTOPE EXCHANGE
- ISOTOPE EXCHANGE MECHANISM DOMINATED BY RECRYSTALLIZATION OF CALCITE SURFACE LAYERS
- AQUEOUS PHASE REQUIRED FOR RECRYSTALLIZATION PROCESS TO OPERATE. EXTENDED DRY-OUT PERIOD WILL INHIBIT EXCHANGE PROCESS ON MINERAL SURFACES
- FRACTION OF TOTAL (SOLID) CARBONATE 12C RESERVOIRS AVAILABLE FOR ISOTOPE EXCHANGE UNCERTAIN

DATA AND MODELING NEEDS

- NEED EXPERIMENTAL DETERMINATION OF CARBON ISOTOPE EXCHANGE RATES AND CAPACITIES ON CARBONATES IN CONCRETE TO BE USED IN REPOSITORY AND POSSIBLY ON OXIDIZED CORROSION ALLOWANCE LAYER TO BE USED IN WASTE PACKAGES
- NEED ADDITIONAL EXPERIMENTAL DATA ON WATER/GAS CARBON ISOTOPE EXCHANGE KINETICS
- NEED MORE DETAILED GAS FLOW MODELS THAT EXPLAIN SITE CHARACTERIZATION DATA AND BETTER QUANTIFY CARBON ISOTOPE EXCHANGE REACTIONS

CONCLUSIONS

- UZ AQUEOUS PHASE LARGEST POTENTIAL 14CO2 EXCHANGE RESERVOIR
- DRY-OUT ASSOCIATED WITH "HOT REPOSITORY" WILL SUBSTANTIALLY DECREASE SIZE OF AQUEOUS PHASE RESERVOIR FOR THOUSANDS OF YEARS
- ISOTOPE EXCHANGE OF 14CO₂ WITH 12CO₂ ON SURFACES OF CARBONATE MINERALS WILL PROVIDE ADDITIONAL EXCHANGE RESERVOIR ALTHOUGH PROBABLY SMALLER THAN "DISPLACED" AQUEOUS RESERVOIR

CONCLUSIONS (cont'd)

- TSPA93 USED CURRENT AQUEOUS RESERVOIR FOR 14CO₂ RETARDATION. RELEASE RATE CALCULATED EXCEEDED EPA REGULATIONS.
- "HOT REPOSITORY" MAY LEAD TO LARGER 14CO₂ RELEASES BECAUSE OF POSSIBLE DECREASES IN SIZES OF EXCHANGE RESERVOIRS.
- EXPERIMENTAL DATA REQUIRED TO EVALUATE ALTERNATE EXCHANGE RESERVOIRS
- ADDITIONAL MODELING REQUIRED TO IMPROVE CORRES-PONDENCE WITH SITE CHARACTERIZATION DATA AND TO EVALUATE CAPACITY OF NON-AQUEOUS RESERVOIRS

U.S. DEPARTMENT OF ENERGY OFFICE OF CIVILIAN RADIOACTIVE WASTE MANAGEMENT PRESENTATION TO THE NUCLEAR WASTE TECHNICAL REVIEW BOARD SUBJECT: REVIEW OF GASEOUS **ISOTOPES PRESENTER: RICHARD A. VAN KONYNENBURG PRESENTER'S TITLE** AND ORGANIZATION: ENGINEER UNIVERSITY OF CALIFORNIA, LAWRENCE LIVERMORE NATIONAL LABORATORY PRESENTER'S **TELEPHONE NUMBER:** (415) 422-0456 **DECEMBER 11-12, 1989**

U.S. DEPARTMENT OF ENERGY OFFICE OF CIVILIAN RADIOACTIVE WASTE MANAGEMENT PRESENTATION TO THE NUCLEAR WASTE TECHNICAL REVIEW BOARD SUBJECT: GASEOUS AND SEMI-VOLATILE RADIONUCLIDES **PRESENTER: DR. U-SUN PARK PRESENTER'S TITLE** AND ORGANIZATION: SENIOR STAFF ENGINEER SCIENCE APPLICATIONS INTERNATIONAL CORPORATION LAS VEGAS, NEVADA PRESENTER'S **TELEPHONE NUMBER:** (702) 794-7643 **REGISTRY HOTEL, DENVER, COLORADO**

JUNE 25-27, 1991

OFFICE OF (U.S. DEPARTMENT OF ENERGY CIVILIAN RADIOACTIVE WASTE MANAGEMENT
	ASTE TECHNICAL REVIEW BOARD FULL BOARD MEETING
SUBJECT:	CARBON-14 RELEASES
PRESENTER:	DR. RICHARD A. VAN KONYNENBURG
PRESENTER'S TITLE AND ORGANIZATION:	ENGINEER LAWRENCE LIVERMORE NATIONAL LABORATORY (LLNL LIVERMORE, CALIFORNIA
PRESENTER'S TELEPHONE NUMBER:	(510) 422-0456



ENVIRONMENTAL TRANSPORT OF GASEOUS RELEASES OF RADIONUCLIDES

Richard A. Van Konynenburg Lawrence Livermore National Laboratory

Meeting of the National Academy of Sciences' Committee on the Technical Bases for Yucca Mountain Standards

Las Vegas, Nevada

August 26, 1993

NRC Analysis of C-14 Releases at the Yucca Mountain site for Iterative Performance Assessment, Phase 2

by Richard Codell and Rex Wescott U.S. Nuclear Regulatory Commission

for the High-Level Waste/Carbon-14 Subcommittee of the SAB Radiation Advisory Committee June 16-17, 1992, Arlington VA

