

SENSITIVITY ANALYSIS OF WASTE PACKAGE

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OUTLINE

- Previous Analysis
- Current Analysis and Basis
- Long-term Passivity
- Conclusions

PREVIOUS NRC ANALYSIS TOTAL-SYSTEM PERFORMANCE ASSESSMENT (TPA) CODE

- All corrosion parameters were from electrochemical tests in pure sodium chloride solutions.
- Deliquescent salt mixture or inhibitors were not considered.
- Drip Shield Life Time:
 Sampled from lognormal distribution of [3700, 27300] years
- No corrosion failure of waste packages in 10,000 years
- TPA Results: ~ 0.03 mrem/yr at 10,000 years

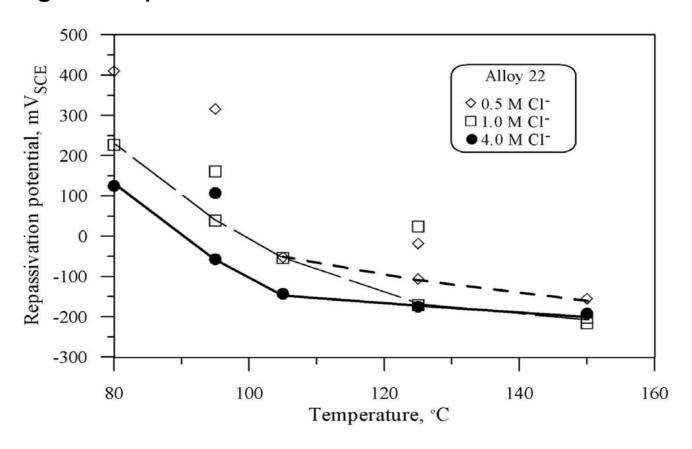
CURRENT ANALYSIS

Effects of Deliquescent Salts

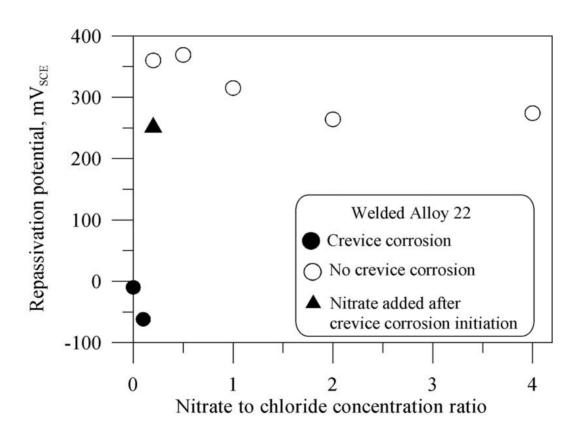
Effects of Inhibitors

Effects of Perforations

Repassivation Potentials for Deliquescence at High Temperature



Inhibitors may reduce susceptibility for localized corrosion.

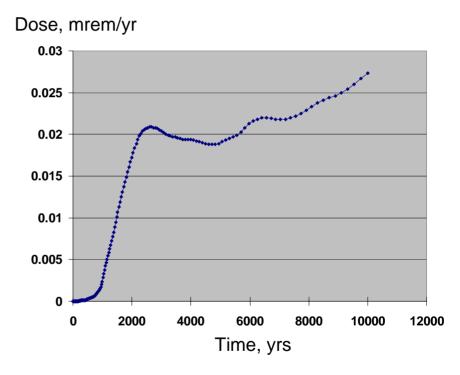


Analysis using current information on repassivation potential equation

Dose, mrem/yr 4 3.5 3 2.5 2 1.5 0 0 2000 4000 6000 8000 10000 12000 Time, yrs

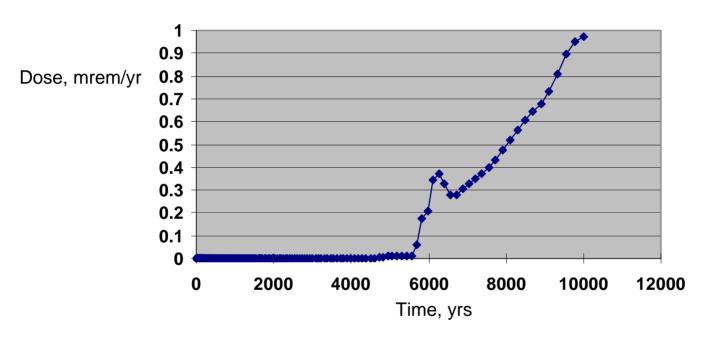
- About 87 % waste package failure
- Sodium chloride solution

Analysis considering inhibitors in groundwater contacting waste package



- No corrosion failure of waste package
- Sodium chloride solution with abundant nitrate; No drip shields (however, availability of fluorides can limit drip shield corrosion.)

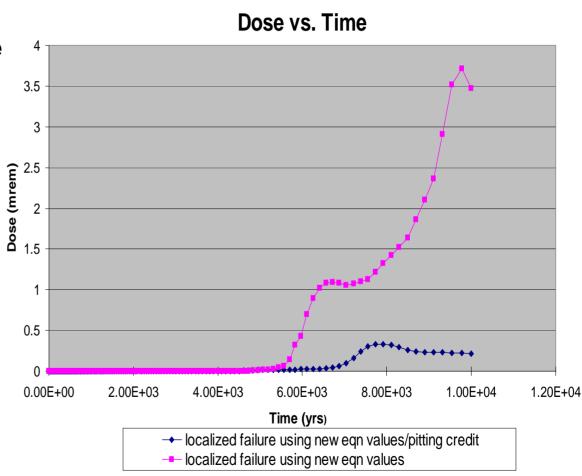
Probabilistic approach of the evaluation of high temperature deliquescence and inhibitors by sampling critical relative humidity (RH): normal distribution [0.35,0.60]



- About 17 % waste packages were failed from the distribution.
 - reference: no waste package failure without deliquescence in the previous NRC analysis, ~ 0.03 mrem/yr at 10,000 years)
- Detailed distributions are under study.

Effects of Perforation Size and Frequency

- Modified inputs to estimate the effects of the exposed surface area from size and frequency of perforations
- Pits could be stifled under open-circuit corrosion conditions.
- Crevice area is likely to be restricted.



UNIFORM CORROSION

Potential Effects of High Temperature and High Acidity

- Na-K-Cl-NO₃ High Temperature Deliquescence:
 - Corrosion rates decrease with time (e.g., weight loss measurements).
- Ca/MgCl₂ High Temperature Deliquescence:
 - pH may go down leading to enhanced uniform corrosion.
 - The fraction of Ca/MgCl₂ is low. Also, Ca/MgCl₂ is likely to decompose and the resulting acids will evaporate.

LONG-TERM PASSIVITY

Time and extent of waste package corrosion is important.

 Given no localized corrosion condition with passivity from laboratory testing, assess the stability of passive film over a geological time period.

 Use inference from modeling and analogue study, emphasizing potential long-term latent effects.

LONG-TERM PASSIVITY

Modeling

- Void Formation: mechanical and chemical stability of passive film
- Anodic Sulfur Segregation at Metal-Oxide Interface: mechanical stability and enhanced transient current on unstable passive film
- Anion Selective Sorption in Crevice: latent initiation of localized corrosion
- Development of Large Cathodic Surface Area of Corrosion Products: enhanced transient current on unstable passive film
- Analogue Study
 - Investigations of responsible mechanisms for the long-term survivability of analogue (e.g., josephinite, Ni₃Fe)
- Modeling and analogue studies give better technical bases as to the long-term stability of passive film.

SUMMARY

- Evaluation of corrosion of Alloy 22 needs to consider both deleterious and beneficial conditions.
 - The high temperature deliquescence environment could occur under specific chemical conditions such as Ca/MgCl₂ or mixture of Na-K-Cl-NO₃ salts.
 - Waste packages could be passivated by the effects of inhibitors.
 - The release could be restricted by the limited amount of deleterious Ca/MgCl₂ and limited exposed surface area from deleterious localized corrosion.
- Performance assessment provides tools to evaluate the impacts of the high temperature deliquescence on the dose and the associated uncertainties. Detailed evaluation of the uncertainties continues.
- Understanding of the stability of passive film over a geological period assisted by analogues and modeling

Disclaimer: The NRC staff views expressed herein are preliminary and do not constitute a final judgment or determination of the matters addressed or of the acceptability for a geologic repository Yucca Mountain