

UNITED STATES
NUCLEAR WASTE TECHNICAL REVIEW BOARD

TRANSCRIPT
SPRING 2013 BOARD MEETING

Tuesday
April 16, 2013

Marriott Courtyard
480 Columbia Point Drive
Richland, WA 99353

NWTRB BOARD MEMBERS PRESENT

Rodney C. Ewing, Ph.D., Chairman
Steven M. Becker, Ph.D.
Susan L. Brantley, Ph.D.
Sue B. Clark, Ph.D.
Efi Foufoula-Georgiou, Ph.D.
Gerald S. Frankel, Sc.D.
Linda K. Nozick, Ph.D.
K. L. (Lee) Peddicord, Ph.D.
Paul J. Turinsky, Ph.D.
Mary Lou Zoback, Ph.D.

NWTRB EXECUTIVE STAFF

Nigel Mote, Executive Director
Debra L. Dickson, Director of Administration

NWTRB SENIOR PROFESSIONAL STAFF

Bruce E. Kirstein
Bret W. Leslie
Daniel S. Metlay
Gene W. Rowe
Karyn D. Severson

NWTRB ADMINISTRATION STAFF

Linda J. Coultry, Meeting Planner
William D. Harrison, Systems Administrator

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1 A few words about the Board. The Board is an
2 independent agency in the Executive Branch. We are not part
3 of DOE or any other federal agency. The Board was created in
4 the 1987 amendments to the Nuclear Waste Policy Act to
5 perform ongoing evaluation of the technical and scientific
6 validity of DOE activities related to implementing the
7 Nuclear Waste Policy Act. These activities include
8 transporting, packaging, and disposing of spent nuclear fuel
9 and high-level radioactive waste. The Board reports its
10 findings and conclusions, recommendations to Congress and to
11 the Secretary of Energy. And I will call your attention to a
12 one-page handout that you can get off the table outside,
13 which summarizes the charge to the Board, and on the back
14 there is a list of the Board members.

15 The second point is to discuss for just a moment
16 what the Board hopes to learn from this visit to the Hanford
17 site. Relevant to today's meeting is to emphasize that the
18 Board's technical and scientific purview does not include the
19 safety or operation of DOE-owned facilities. This is the
20 responsibility of the Defense Nuclear Facility Safety Board.
21 We also do not review DOE management or disposal of low-level
22 radioactive waste. Our technical and scientific review of
23 DOE activities at the Hanford facility is focused primarily
24 on vitrified high-level waste, which will require disposal in
25 a deep mine geologic repository, and any other waste stream

1 that would be considered as high-level waste and, of course,
2 spent nuclear fuel.

3 Appropriate issues for the Board include: What is
4 the inventory and the state of the spent nuclear fuel and the
5 high-level radioactive waste at Hanford? What is the
6 technical impact of the delay in opening a geologic
7 repository? And what are the volumes and compositions of the
8 waste at Hanford that will require geologic disposal?

9 I also want to say a few words about the Board
10 holding a meeting during the time of budgetary constraint.
11 Of course, during our tour of the Hanford site we heard of
12 people who weren't on the tour because of furloughs and so.
13 Like all federal agencies, our budget is and was affected by
14 sequestration, but the Board does have some flexibility to
15 address the most important and timely issues. Thus, we
16 allocate our resources to address those timely and critical
17 issues. And, for the Board, the visit to the Hanford site is
18 particularly important, because we need to understand the
19 origin and form and the volumes of the high-level waste that
20 will require geologic disposal, as well as the DOE-owned
21 spent nuclear fuel.

22 The second reason that we pushed ahead with this
23 meeting is, we wanted to provide the opportunity for members
24 of the interested public to comment on the issues and to
25 directly interact with Board members and representatives from

1 DOE and other federal agencies. We know, from more than 20
2 years of experience, that holding face-to-face meetings is
3 the best and most efficient way to accomplish these
4 objectives, and in a moment I'll discuss a new approach
5 toward enhancing those interactions.

6 And then, finally, I should point out, the Board
7 will follow up this meeting, as it does with all meetings,
8 with letters to the relevant DOE offices, which convey our
9 observations and recommendations; and these letters are
10 posted on the Board's website.

11 Now let me introduce the members of the Board. As
12 I mention their name, I'd ask them to simply raise their hand
13 so that they can be identified. As I said earlier, my name
14 is Rod Ewing, so I'll raise my hand; and I'm a professor at
15 the University Michigan.

16 Steven Becker is a Professor of Community and
17 Environmental Health in the College of Health Sciences at Old
18 Dominion University in Norfolk, Virginia.

19 Susan Brantley is a Distinguished Professor of
20 Geosciences in the College of Earth and Mineral Sciences at
21 Pennsylvania State University, where she is also the Director
22 of the Earth and Environmental Systems Institute, and she is
23 a member of the National Academy of Sciences.

24 Sue Clark is Regents Distinguished Professor of
25 Chemistry at Washington State University.

1 Gerald Frankel is Professor of Material Science and
2 Engineering and Director of the Fontana Corrosion Center at
3 Ohio State University.

4 Efi Foufoula is the Distinguished McKnight
5 University Professor of Civil Engineering and Director of the
6 National Center for Earth Surface Dynamics at the University
7 of Minnesota.

8 Linda Nozick is a Professor in the School of Civil
9 and Environmental Engineering and Director of the College
10 Program in Systems Engineering at Cornell University.

11 Lee Peddicord has served as Director of the Nuclear
12 Power Institute at Texas A&M University since 2007, and he is
13 Professor of Nuclear Engineering at Texas A&M.

14 Paul Turinsky is Professor of Nuclear Engineering
15 at North Carolina State University. Since 2010 he has served
16 as Chief Scientist for the Department of Energy's Innovation
17 Hub for Modeling and Simulation of Nuclear Reactors.

18 Mary Lou Zoback is Consulting Professor in
19 Environmental Earth System Science Department at Stanford
20 University. She is a seismologist and a member of the
21 National Academy of Sciences.

22 One of our members, Jean Bahr, is attending a
23 workshop on behalf of the Board that, unfortunately,
24 conflicts with today's meeting. Jean is a Professor of
25 Geosciences at the University of Wisconsin-Madison. She is

1 also a member of the Geological Engineering Program and is
2 faculty affiliate to the Nelson Institute for Environmental
3 Studies. Jean will join the Board tomorrow for our business
4 meeting.

5 All of the Board members serve part-time, but we
6 have a full-time staff that provide tremendous intellectual
7 support as well as continuity in our efforts. The technical
8 staff are seated at the table against the wall.

9 I should mention that the procedure for how we
10 conduct the meetings is: Questions will be taken at the end
11 of the talks, first questions from members of the Board and
12 then members of the staff. We'd ask the public to hold their
13 questions till the end of the day where we we'll have a
14 public comment meeting and have those interactions.

15 Now let me briefly review the day's agenda. We'll
16 first be welcomed by a representative from the Office of
17 River Protection. Then we'll hear a presentation from DOE's
18 Office of Environmental Management about approaches taken
19 across the complex to manage DOE-owned waste, including
20 issues associated with the type and amounts of waste streams,
21 the disposition strategy for each, and how the management of
22 these wastes has been affected by the delay in the geologic
23 repository.

24 Next we'll learn about vitrification as a complex-
25 wide management practice for the disposition of high-level

1 radioactive waste. A glass waste form has been selected for
2 disposing of high-level radioactive waste; thus,
3 vitrification is the focus of the morning session.

4 We will have two panels. The first will present
5 the technical experiences with waste vitrification from
6 perspectives in France and various laboratories in the U.S.,
7 including Hanford, West Valley, and Savannah River. And then
8 the second panel is composed of experts from Savannah River,
9 Catholic University, and Pacific Northwest National
10 Laboratory, who will discuss DOE's waste form technology
11 development program, including new waste forms, synthesis of
12 waste forms, and the long-term durability of those waste
13 forms.

14 After lunch a panel composed of representatives of
15 tribal, state, and public organizations will present their
16 views on the most important technical issues associated with
17 the disposal of high-level waste and spent nuclear fuel
18 stored at the Hanford site. We understand how important
19 these issues are, in particular for those of you who live in
20 this region. And so we look forward to hearing your views
21 and the discussion that follows the panel.

22 The panel discussion will be followed by a non-
23 Hanford-related update on the analyses being performed by
24 DOE's Office of Nuclear Energy on the potential for the
25 direct disposal of the very large dry storage containers

1 currently in service at nuclear power plants.

2 The last presentation of the day will be on DOE's
3 recently-issued strategy for the management and disposal of
4 used nuclear fuel and high-level radioactive waste. This
5 strategy is the administration's response to the final
6 recommendations of the Blue Ribbon Commission on America's
7 Nuclear Future, and it includes DOE plans for moving forward
8 on the disposal of high-level radioactive waste and spent
9 nuclear fuel; hence, very relevant to issues at Hanford. We
10 are extremely pleased that Dr. Pete Lyons, the Assistant
11 Secretary for Nuclear Energy will join us today to make that
12 presentation.

13 As mentioned earlier, hearing the views of the
14 interested public is a very important part of these meetings,
15 so we have scheduled time for public comment at the end of
16 the day. We welcome your comments, particularly those
17 related to the Board's scientific and technical mandate.
18 Please enter your name on the sign-up sheet at the table near
19 the entrance to the room. If you prefer, written remarks or
20 comments can be submitted, and they will be made part of the
21 meeting record. Oral comments will appear in the transcripts
22 of the meeting. All transcripts and other meeting materials,
23 including submitted written comments or statements, are
24 posted on the Board's website.

25 Immediately following the meeting there will be a

1 special event that I hope you will find useful and
2 informative. We have arranged for a small poster session on
3 vitrification and other technical topics that will provide
4 you an opportunity to meet and talk to some of the scientists
5 and engineers who work on these important issues. The
6 posters will be just outside of the meeting room. Please let
7 us know whether you think this first-of-a-kind, for the
8 Board, event is useful.

9 Finally, I need to say that during the meeting
10 Board members freely express their own personal views and
11 opinions, or you might infer their views from the types of
12 questions they ask. We certainly encourage this, but we also
13 want you to know that the comments of individual Board
14 members during the meeting are not the official--not to be
15 taken as official Board statements. The Board's positions
16 are found in our reports and letters to Congress and the
17 Secretary of Energy.

18 Finally, at the end of this long introduction, a
19 few housekeeping details. Please turn off your cell phones.
20 When you speak, please identify yourself and your affiliation
21 so that we have it for the record; and please speak into the
22 microphone so that we can have a complete transcript of the
23 meeting.

24 So, because of my long-windedness, we are now
25 behind schedule.

1 So it's my pleasure to begin the meeting and turn
2 the podium over to Stacy Charboneau, who will welcome us on
3 behalf of the Office of River Protection.

4 CHARBONEAU: Good morning. I'm Stacy Charboneau, the
5 Deputy Manager for the Office of River Protection. And on
6 behalf of the Office of River Protection and the Richland
7 Operations Office, the two EM offices responsible for the
8 cleanup at the Hanford site, we are honored and welcome you
9 back to the Washington State and the Hanford site.

10 It's a beautiful day out there, and it's a
11 beautiful river out there; and it just underscores the
12 mission that we have here to clean up the Hanford site. I
13 hope you enjoyed your long day yesterday, the tour of the
14 Hanford site, and you even got to experience some of our
15 historically famous termination winds, I think. Back in the
16 early days of the Hanford site when the construction forces
17 were in full force, when strong winds like that would kick up
18 and kick dust across the Hanford site, many folks would
19 leave; so they've been termed the termination winds here at
20 Hanford.

21 As a part of the tour that you had yesterday, I
22 hope you had an opportunity to really see and underscore the
23 progress we've made with regard to the cleanup at the Hanford
24 site. Certainly that's more evident than ever along the
25 river corridor where hundreds of facilities, many nuclear

1 facilities, have been demolished and removed from the river
2 corridor. Additionally, hundreds of waste sites have been
3 cleaned up, soil contamination moved up to the central
4 plateau and disposed of in an environmental restoration
5 disposal facility.

6 You saw an opportunity to see the Waste Treatment
7 Plant and the progress made there with over 60 percent
8 construction complete with excellent progress being made on
9 the low activity waste facility, the analytical lab, and the
10 balance of facilities there while we continue to resolve
11 technical issues specific to high-level waste and
12 pretreatment.

13 Maybe not as evidence yesterday was the progress
14 that's being made on tank retrievals, given that the large
15 underground tanks are underground; but certainly progress is
16 being made there with over ten tanks emptied in the C farm
17 area. Three were underway retrieval today. And so as we
18 prepare the feed for the Waste Treatment Plant, we continue
19 retrievals out of our single-shell tanks. And most recently
20 you've probably heard some about the leaking tanks at
21 Hanford. Historically, we have identified 67 tanks of our
22 single-shell tanks to be assumed leakers. And just last fall
23 we identified one of our double-shell tank inner shells was
24 leaking into the annulus of that double-shell tank.

25 So this certainly underscores the urgency of the

1 cleanup mission we have here at Hanford, and we welcome your
2 insight into our path forward and the progress on our path
3 forward in retrieving this waste, vitrifying this waste for
4 permanent disposal.

5 We are fortunate to have Albert Kruger on the staff
6 at the Office of River Protection, and you will hear from him
7 later today specific to glass formulation and the waste forms
8 that we are looking at for the final treatment and
9 disposition of the waste for the 177 underground tanks and
10 the 56 million gallons of tank waste we have here at Hanford.
11 And we hope that you'll hear from many other research and
12 scientists today that also will have contributed to our
13 mission here at Hanford and are helping us along that mission
14 in resolving some of our technical issues in that waste form
15 and disposition of that waste.

16 So, again, welcome to Hanford, welcome to
17 Washington State, and we hope that you have a very enjoyable
18 and fruitful meeting here today. Thanks.

19 EWING: All right. Thanks very much.

20 And so we'll continue with the program. The next
21 speaker is Ken Picha, and the topic will be The Complex Wide
22 Overview of the Department of Energy Office of Environmental
23 Management Program.

24 PICHA: Good morning, everybody. Dave sends his
25 apologies for not being able to be here, so you get the

1 B team here.

2 The last time I was in the Hanford area doing a
3 public meeting, it was on basically discussing the
4 Department's alternatives for dispositioning scrap metals
5 that were either in a--had been in a radiological area or had
6 been potentially contaminated. And I'll tell you, I much
7 prefer this topic. I think, although there is not unanimity
8 on all the details for how we proceed with tank wastes and
9 disposition of some of our materials, I think we can all
10 agree that getting waste out of the aging tanks, some of
11 which had exceeded their design life, and solidify into a
12 solid stable form is a good thing.

13 So this is just a topic that we're going to talk
14 about. The NWTRB had provided some information about some of
15 the topics that they wanted us to talk about. We're actually
16 going to talk about some of the radioactive waste derived
17 from tanks wastes, talk about our interactions with the
18 Office of Nuclear Energy, and Dr. Lyons will be here--if he's
19 not here already--to talk about some of the nuclear energy
20 activities--and we have some role in working with them to
21 support what they are doing in terms of implementing some of
22 the BRC recommendations--and then talk a little bit about the
23 impact of a delay in opening a geologic repository.

24 So, first, a little bit of discussion about the EM
25 program. EM started in 1989. We had over a hundred sites

1 that we worked on. As you can see now, we've actually done
2 quite a bit of cleanup. We're down to about 16, 17 sites.
3 Of course, the hard ones are the ones still to go: Hanford,
4 Savannah River, Idaho, Oak Ridge, some of our larger sites.
5 And this, by the way, is a slide that was in Dave Huizenga's
6 budget roll-out presentation.

7 We have immobilized over 5,000,000 gallons of
8 radioactive liquid waste. That includes a little over
9 600,000 to 700,000 gallons in a couple tanks at the West
10 Valley Demonstration Project, and then also the bulk of it
11 represents the waste we've solidified at Savannah River at
12 the Defense Waste Processing Facility. And then we've
13 basically taken all the plutonium that we were responsible
14 for managing in the complex, and it's in safe storage at the
15 Savannah River site.

16 This is a slide that basically describes where we
17 are in our tank waste program. As you can see, the large
18 sites are Savannah River and Hanford. As Stacy said, we've
19 got a number of tanks at the Hanford site, 177 tanks; 149 of
20 those are single-shell tanks and 28 are double-shell tanks.
21 There's about 175 million curies of waste stored in those
22 tanks and then about 50--it varies when we do this--55, 56
23 million gallons. And we're projecting to have about 9,700
24 canisters. Of course, that number is subject to variability,
25 and the work that Albert is doing with some of you-all in the

1 audience here will help see if we can reduce those numbers.

2 At Idaho there were 15 tanks and the--I don't know
3 whether it was prudence or luck, but they decided to stay on
4 the acidic side at Idaho, and so they ended up with a little
5 bit of an easier mission to clean up their tanks. So they've
6 actually closed 11 of their 15 tanks. They have about
7 900,000 gallons remaining of liquid waste that they will be
8 targeting to treat perhaps early next year--starting to treat
9 early next year into a sodium bearing waste form using a
10 steam reforming process. And then the calcine, which is
11 where most of the high-level waste volume is, is in a number
12 of bin sets; and the preferred technology for solidifying the
13 calcine or treating it for final waste form is a hot
14 isostatic pressing process.

15 At Savannah River we had 51 large tanks. Four have
16 been closed, two are essentially clean, and we're hoping that
17 we can start closure activities later this year. There's
18 about 37 million gallons of tank waste there. And there is
19 where we actually have an operating vitrification plant. It
20 started up in 1996. We have about 3,600 canisters to date.
21 The melter that they have there just finished ten years, and
22 I think they have a lid heater that has failed, but they're
23 still able to get good throughput. And so we'll see how long
24 that melter lasts.

25 And then the last site is the West Valley

1 Demonstration Project. There the vitrification campaign, as
2 I said, only had about 600,000 or 700,000 gallons to process.
3 Most of the volume there was treated through a pretreatment
4 process and solidified as a cement into what they call
5 drums--they were actually square drums--stored on site and
6 eventually disposed of in Nevada. So they only ended up with
7 about 275 canisters, which are on site in the old chemical
8 process building where they actually did reprocessing. And
9 they're planning a storage pad to store those 275 canisters
10 until there is a final disposition place for those.

11 So this is basically our tank waste management
12 strategy in a nutshell. Obviously safe storage is a
13 priority. As Stacy indicated, we do have some tanks that
14 indicate at Hanford a decreasing loss of levels that they're
15 looking at and exploring fairly carefully to understand
16 what's going on in those tanks. The next step is retrieve
17 the waste from the tanks to prepare for some kind of
18 treatment process. And at Hanford, Savannah River, and West
19 Valley we basically are going to separate the waste into a
20 low-activity fraction, which is most of the volume--it's over
21 90 percent of the volume--but varying per site anywhere from,
22 let's say, two, three percent radioactivity of the curies to
23 as much as five percent.

24 And then at those sites also we are going to be
25 treating the high-level waste through vitrification. And at

1 Hanford here we also have the low-activity waste
2 vitrification facility, which will be treating a good portion
3 of the low-activity waste generated.

4 And at Idaho we'll retrieve and dispose calcine.
5 It says "directly" there. That's an old slide. I apologize.
6 That's not the case now. That's not the baseline. The
7 baseline is to HIP it for a final disposition. And then,
8 finally, to stabilize tank waste residues for in-place
9 closure at our four sites.

10 So this is just a slide that shows the DWPF, the
11 Defense Waste Processing Facility, at Savannah River and the
12 Integrated Waste Treatment Unit at Idaho. As the slide shows
13 there, we completed construction in 2012. During start-up
14 testing there was an issue with the off-gas system that has
15 caused us to go back and look at some of the design of some
16 of the components and do some redesign. And it's delayed--
17 our start-up--probably until sometime in about 2014.

18 The Department's radioactive waste management
19 activities are governed by a number of regulations and laws.
20 I'll point to the one in the upper left under our Atomic
21 Energy authority. We have a DOE order that implements that
22 for managing our radioactive waste, and the Department
23 actually has three radioactive waste categories: low-level
24 waste, transuranic waste, and high-level waste. Now, we'll
25 talk a little bit later about low-activity waste, which is

1 basically the separated portion of the tank waste that's
2 relatively low in radioactivity and high in volume, but it's
3 not a formal radioactive waste classification.

4 So this is basically just a slide that shows some
5 of the different--and I picked Hanford for several reasons.
6 One is, we're here obviously; and, two, it probably has all
7 the potential disposition pathways that we are looking at for
8 our tank waste program. And it's just meant to show a
9 pictorial representation of the different pathways of the
10 waste. You can see in the left-hand side--

11 (Pause.)

12 So the Department in, I believe, the '70s and '80s
13 ran a process to extract much of the cesium and strontium
14 from the tank waste, and that's in about 2,000 capsules that
15 are stored in a waste encapsulation storage facility here on
16 the Hanford site. But the bulk of the tank waste, the 56
17 million gallons, will go through a pretreatment process down
18 here and be separated into a high-activity and a low-activity
19 fraction, of which the bulk of the volume--as I said before,
20 the low-activity waste will be converted to a solid form and
21 disposed of as low-level waste into an integrated disposal
22 facility here on the Hanford site. The high-level waste
23 component will be treated via the high-level waste facility--
24 vitrification facility--and that represents the bulk of the
25 radionuclides and 5, 10 percent of the volume for ultimate

1 disposition in some kind of a repository.

2 And then last month the Department, through a NEPA
3 action, announced a preferred alternative to look at some
4 waste that's in 20 tanks and which was associated with things
5 like the PFP finishing process, that we believe we have--for
6 some of those tanks that we have a good process knowledge
7 that we can determine the wastes are transuranic waste and go
8 through all the permitting processes, make a determination
9 that it's transuranic waste, retrieve that waste, package it,
10 and send to WIPP. And, in fact, just recently the Department
11 issued a Department request to the State of New Mexico to
12 start down that path. So I think those are the main
13 disposition pathways for our tank wastes.

14 With respect to spent nuclear fuel, we primarily
15 manage those at the same sites as we do our tank waste, with
16 the exception of West Valley. The bulk of our spent nuclear
17 fuel that we manage here at Hanford, most of that's in MCOs
18 in the canister storage building out on the site. And you
19 may or may not have seen that on your tour. I'm not sure.
20 Okay. We have some spent nuclear fuel at our Idaho site,
21 defense and non-defense. We actually have title to some
22 waste from the gas-cooled commercial reactor at Fort St. Vrain
23 near the Fort St. Vrain site in Colorado. And then the rest
24 of our spent nuclear fuel is at Savannah River. And a lot of
25 that is domestic research reactor and foreign research

1 reactor fuel that we've retrieved as part of our Global
2 Threat Reduction Initiative and actions associated with that.

3 The Department, through the Office of Science,
4 operates the high flux intensity reactor at its Oak Ridge
5 site, and the cores from that are being stored at the
6 Savannah River site. We don't show that on here, but that's
7 part of our fuel that we manage and store it at Savannah
8 River.

9 EM is partners with the Office of Nuclear Energy.
10 Our main efforts have been to support them in looking at
11 different proposals for a repository. One of those that we
12 have some experience with through our operation of the waste
13 isolation pilot plant in New Mexico is certainly looking at
14 the viability of salt repositories. In 2012 NE and EM
15 jointly sponsored a workshop to look at a salt research and
16 development study plan. We have supported that activity,
17 including a potential underground research laboratory in the
18 WIPP area here, as show on this slide. We followed that up
19 with a workshop in March of this year with a focus of
20 identifying additional R&D activities and with an attempt to
21 try to get to an integrated path forward later this month.

22 Now, we are doing this in support of NE as they
23 look at--our work primarily in that arena has to do with
24 technical validation of data, analyses, and we are doing this
25 in accordance with a deliberate process as NE is implementing

1 some of the recommendations from the Blue Ribbon Commission
2 to look at consensus-based approaches for selection of a
3 repository. And we have continued to meet regularly to
4 coordinate our activities.

5 One of the things that WIPP has done is we started
6 with contact-handled transuranic waste. There wasn't much
7 heat generation. I want to say about five years ago we
8 started placing and disposing remote-handled waste that
9 certainly has some capability of heat generation. One of the
10 things we're looking to try to do this year or start--I'm
11 sorry, next year--is start some heater tests at WIPP that
12 could be used to look at how the salt formations behave under
13 more intensive heat sources. So that's something that we'll
14 be doing and certainly working in conjunction with Office of
15 Nuclear Energy.

16 One of the things we do each year or have done each
17 year is, the Department prepares an environmental liability
18 report. With a \$6 billion program, give or take, this is
19 going to go out for a number of years. We have fairly
20 significant liability for the Department, so one of the
21 things we did is we wanted to look at, well, what is the
22 liability associated with potentially a 20-year delay in
23 opening up some kind of a repository? I can remember several
24 years ago when we were thinking--it must have been several
25 years ago--2012 was a date that we were looking at. And then

1 it got progressively pushed off, and now we don't necessarily
2 have an identified place to put this or a specific time. So
3 we did just look at sort of a case of what would happen if we
4 delayed a place for putting our high-level waste and spent
5 nuclear fuel for 20 years. And the results were about a
6 \$1.1 billion liability, which basically continues our safe
7 storage of treated high-level wastes at our four sites and
8 spent nuclear fuel as well.

9 Certainly in terms of how we would store those
10 materials, there is not a significant difference. The big, I
11 guess, cost-intensive aspect of that is we might need more
12 storage facilities than we originally planned. Right now we
13 have two storage facilities for our canisters at Savannah
14 River, and we're looking at potentially starting to close on
15 completing and filling the second one sometime in the next
16 five years. So we're already thinking, okay, what do we need
17 to do to plan for additional facilities for the high-level
18 waste?

19 So, our impact of repository delays on agreements,
20 most of our agreements with states and other regulatory
21 bodies are basically having to do with tanks. There are
22 cease-use requirements. There are tank waste retrieval
23 milestones. Both, as Stacy indicated, were well underway to
24 meeting our C tank farm retrieval milestones. Savannah River
25 has similar milestones, both for bulk waste removal from

1 specific numbers of tanks as well as tank closures. We also,
2 in the form of site treatment plans, have regulatory
3 agreements to retrieve the waste and treat it to some kind of
4 a waste form. And then at Idaho, under an agreement with a
5 former governor, we actually have a date to have our high-
6 level tank waste road-ready.

7 We continue to review the impacts of delays in the
8 repository program with our regulators. They have been very
9 understanding in terms of that. But the biggest risk
10 reduction is to get the waste out of the tanks and treated.

11 And that's about all I have.

12 EWING: Okay, thank you very much.

13 And so now questions from the Board? Sue?

14 CLARK: Sue Clark, Board. I was wondering if you could
15 go back to Slide 10.

16 PICHA: Okay.

17 CLARK: And I'm curious, first of all, if you've got a
18 typo about Savannah River--oh, one too far. Should that be
19 30 metric tons heavy metal? Because on all of the other ones
20 the defense and the non-defense add up to--

21 PICHA: Yeah, yup, thank you.

22 CLARK: So that should be 30 tons. But then you also
23 went on to say something about fuel from Oak Ridge that's not
24 included in that number; right?

25 PICHA: It is included in that number. I'm sorry.

1 CLARK: Oh, it is included.

2 PICHA: Yes, yes.

3 CLARK: Okay. And so--

4 PICHA: It's not--I just didn't break it out separately
5 and show Oak Ridge on the (inaudible).

6 CLARK: And that fuel is now included, is that in the
7 defense part or the non-defense?

8 PICHA: Non-defense.

9 CLARK: Okay, great. Thank you.

10 EWING: Other questions? Paul

11 TURINSKY: Paul Turinsky, Board. Could you give me some
12 insight when you're allocating resources?

13 EWING: Paul, speak close to your microphone.

14 TURINSKY: Could you give me some insight when you're
15 doing resource allocation, budget allocation? What sort of
16 factors come in to deciding if we should be putting this
17 resource of this site to this resource at this site?

18 Obviously you--I think it's pretty obvious, you never have
19 enough resources to do everything you want to do. So what
20 sort of issues do you consider in doing that?

21 PICHA: Sure. Well, tank waste in nuclear materials is
22 the highest priority for the EM program, so that's sort of an
23 overlying principle. And then we look at what we've
24 historically allocated to the various sites in terms of,
25 okay, what's our nominal baseline? I don't want to use the

1 word baseline, but what have we typically been funded? And
2 then we also look at what's going on at the specific sites in
3 terms of what might be driving the need. For instance, at
4 the Savannah River site the real workhorse for their
5 pretreatment will be something called the Salt Waste
6 Processing Facility, and I didn't show a picture of that.
7 Construction is about 60 percent--seems to be a common
8 number--and we were hoping to have that on line about
9 mid-2014 with a late finish of 2015.

10 Well, turns out now that's not feasible. Various
11 things led to construction delays, and so we're looking now
12 at what is an optimum time frame to bring that facility on
13 line and start operations for the site, given that we don't
14 really want to have a--either produced sludge-only canisters
15 in the high-level waste program or salt-only. We're trying
16 to--we'd like to do that with the balance. And they do have
17 a prototype version of that technology that's currently
18 operating on the site, and they have some--they're looking
19 at, actually, capabilities to ramp that facility up and that
20 will maybe mitigate some of the impacts.

21 But the fact is, we hadn't budgeted for
22 construction dollars in either FY13, FY14, FY15; so we had to
23 do some accounting in that regard and say, okay, how are we
24 going to provide funds for construction on that facility, do
25 all our baseline activities in terms of safely managing tank

1 waste there at Savannah River or operate DWPF. And at
2 Hanford here, given some of the recent things that Stacy
3 talked about with the potential leaking from six tanks,
4 resolving the technical issues, and proceeding with a
5 different approach to addressing some of the more thorny
6 technical issues having to do with the pulse jet mixed
7 vessels that we are going to a large--I'm sorry--a full-scale
8 test and using some of the actual vessels, that's probably
9 going to be a bit more than we planned. So it's really a
10 consideration of all those aspects together.

11 I'm not sure I answered your question specifically,
12 but--

13 TURINSKY: How does public health risk enter decision
14 making?

15 PICHA: Well, certainly the leaks from the tanks is
16 something that is a consideration. So the folks here are
17 looking at some different alternatives, and we wanted to see
18 if we could--for instance, you'll see that in the '14 budget
19 the tank farms actually got a little bit of a boost up so
20 that we could consider that as part of our--in our overall
21 thinking.

22 TURINSKY: Okay, thank you.

23 EWING: Okay. Mary Lou.

24 ZOBACK: That was a really nice summary. Thank you.
25 And a couple of points just for clarification and one for my

1 own enlightenment. On Slide 8 where you summarize the
2 gallons--tank inventory and then the curies--

3 PICHA: I'm not sure I actually went over that slide,
4 did I?

5 ZOBACK: I think you skipped over it, but your tests
6 show the same material. Why are the curies so much larger at
7 Savannah River than at Hanford even though the volume is
8 less?

9 PICHA: It's because they pulled out the cesium and
10 strontium into the capsules.

11 ZOBACK: Okay. So that--

12 PICHA: If you consider them together, it's more
13 comparable.

14 ZOBACK: Okay, good. So that was a good thing to do.

15 The other question I have is on the liability. You
16 said the 20-year delay would result in a \$1.1 billion
17 liability. That seemed low. Is that \$1.1 billion a year or
18 \$1.1 billion over 20 years?

19 PICHA: No, it was--I don't have the details with me,
20 and I can get the details, but it was over that 20-year
21 period.

22 ZOBACK: Over the 20-year. Okay, thanks.

23 EWING: Okay. Jerry, did you have a question?

24 FRANKEL: Jerry Frankel. So you spoke about the tank
25 issues. And, of course, it impacts us through the pressure

1 that it puts onto a final repository. Given everything and
2 what we've heard previously, I'm just wondering what the
3 level of confidence is that exists now in the Department of
4 Energy in the ability of the tanks to contain the waste for
5 some decades now that they're needed.

6 PICHA: Well, certainly this has raised questions. Last
7 month Stacy came to Washington. We actually had some
8 discussions with some congressional folks and some other
9 folks. And I think they have a very robust program to look
10 at and do some health, if you will, of the system and the
11 various tanks to understand how viable the tanks are. We are
12 trying to get out of the single-shell tanks and retrieve
13 waste into the double-shell tanks. We are looking at some
14 alternatives that may be able to speed treatment up of some
15 of the tank waste that might open up some additional storage
16 capacity in the double-shell tanks. But it's certainly an
17 area of focus. We have a single-shell integrity program;
18 there's a double-shell tank integrity program. And I can
19 tell you that Stacy and her folks are more vigilant.

20 EWING: Okay. Efi.

21 FOUFOULA: Efi Foufoula, University of Minnesota. Will
22 you go to Slide 13?

23 PICHA: Okay.

24 FOUFOULA: You (inaudible) point out that continued
25 development--improved techniques for reducing costs and

1 schedule of treatment is an important element and is a very
2 important element, it seems to me, in the next five to ten
3 years. Can you give an insight on how much research and
4 development investment (inaudible) on that specific component
5 to reduce costs and schedules treatment?

6 PICHA: Sure. It turns out that the Office of
7 Environmental Management doesn't have much of a technology
8 development budget. We've been trying to push for that and
9 haven't been real successful, but some of the sites are doing
10 their own technology development activities.

11 For instance, we've looked at an improved solvent
12 that will go into the prototypical salt treatment facility at
13 Savannah River site. We also are looking at an at-tank
14 pretreatment or in-tank pretreatment facility--or
15 capability--I don't want to call it a facility--capability
16 that could provide some additional pretreatment capability at
17 Savannah River and provide an earlier pretreatment capability
18 here at Hanford.

19 I think you'll hear from Albert Kruger later today
20 that they're doing some glass formulation studies that may
21 look at reducing both the number of high-level waste
22 containers as well as low-activity waste containers through
23 understanding the glass composition and the waste
24 characteristics.

25 So it's a number of different things that we're

1 doing.

2 EWING: Yes, Lee.

3 PEDDICORD: On your Slide 3--

4 EWING: Please identify yourself for the record here.

5 PEDDICORD: I'm sorry. Lee Peddicord, Board. The
6 hearing focused primarily on the Hanford, Idaho, Savannah
7 River, and West Valley sites. Is it correct that, of the
8 other 13 sites on your map on the right, none of these will
9 be generating any high-level waste?

10 PICHA: I believe that's true. New York, the West
11 Valley site, already has high-level waste. Idaho, Savannah
12 River, and Hanford, that's correct.

13 PEDDICORD: So none of the other will?

14 PICHA: Correct.

15 PEDDICORD: Okay, thank you.

16 EWING: I have a few questions from the Chair, if I may.

17 PICHA: I mean, unless you're talking about high-level
18 waste including spent nuclear fuel, then we still have
19 (inaudible) cores coming to Savannah River.

20 PEDDICORD: Thank you.

21 EWING: Rod Ewing, Board. On Slide 9, what is the
22 disposition or the future path for the strontium and cesium
23 capsules? It looks like it goes back into the high-level
24 waste; is that--

25 PICHA: Well, I was just trying to show that they were

1 pulled out and separated.

2 EWING: Right.

3 PICHA: They were analyzed in the recently-approved
4 released Tank Closure and Waste Management EIS here at
5 Hanford, and I should know what we selected as the preferred
6 alternative for disposition for that. I don't know off the
7 top of my head, but I will get that information. I'm sure
8 folks here in the audience are aware of it now.

9 EWING: All right. I'd be very interested to know it,
10 because I think there has at least been discussion of mixing
11 it back into the high-level waste.

12 PICHA: Yes, yes, there's been discussion of that.
13 There's been discussion of straight disposal at other places.
14 Part of it's a waste classification issue as well. So--

15 EWING: All right. And then a second question. So as
16 the country moves forward with a strategy for a geologic
17 repository, probably we can now consider a variety of
18 different types of geologies. And it's clear the waste form
19 will be glass. Are there any research programs that ask the
20 question: What type of geology, what type of geochemical
21 environment, would enhance the performance of the glass?

22 PICHA: None that I am aware of, but I haven't been that
23 engaged in that part of it, so I can't answer that question.
24 It's probably better left to the folks that will be speaking
25 later today.

1 EWING: Right. Okay. Thank you.

2 Okay, Steve.

3 BECKER: Steven Becker, Board. In light of the events
4 in Japan in 2011, both the nuclear industry and many agencies
5 have been taking another look at seismic issues, and I'm just
6 wondering how that has affected your work.

7 PICHA: Good question. After the Fukushima accident,
8 our office--the Department's Office of Health, Safety and
9 Security initiated an action to look at beyond design basis
10 accidents across the DOE complex. And we looked at that for
11 all of our facilities. And it turns out that the facility
12 that probably poses amongst the highest risks is the waste
13 encapsulation storage facility here at Hanford. They're
14 being stored in liquid right now and water. So they're
15 looking at different approaches. I'm not specifically
16 involved in that program, but from previous involvement I
17 understand they're looking at different approaches to
18 mitigate those kinds of considerations.

19 But it turned out that we weren't that badly
20 positioned in terms for design basis activities and even in
21 the beyond design basis. That was the main facility of
22 concern. There was one other that's escaping me, but there
23 was a rigorous process to try to understand the impact, so I
24 can't say that. And those results are probably on the HSS
25 website. I can't say that for sure, but--

1 EWING: All right, let me turn to the Board.

2 Dan. To the staff. Sorry.

3 METLAY: Dan Metlay, Board staff. Thank you again for
4 your presentation. It was most illuminating. Let me
5 apologize at the front if I missed this in your presentation.

6 In the President's budget for FY 14, I think, is
7 there funds for the salt heater tests included in that
8 budget?

9 PICHA: We're trying to have flexibility to initiate
10 those tests. I'll just put it that way. We're looking at
11 some flexibility to do those tests.

12 EWING: Nigel.

13 MOTE: Nigel Mote, staff. Again, thanks, Ken, for the
14 presentation.

15 I'd like to come back to a clarification on the
16 point that Mary Lou Zoback asked. Could you tell us what the
17 environmental liability means in the environmental liability
18 report? Last week in Dr. Lyons' testimony before the Energy
19 and Water Development Subcommittee of the Senate
20 Appropriations Committee--I'm sorry, the House, I beg your
21 pardon--he said that the annual cost of the liability
22 payments to utilities for extended storage on the utilities
23 sites would be approximately half a billion dollars a year,
24 so in 20 years that would be \$10 billion. And that only
25 accounts for spent fuel.

1 Now, on the DOE sites, I think I've heard that
2 there is a potential for expanding the storage capacity at
3 Savannah River site for the vitrified waste because of the
4 inability to remove it from the site, so presumably there's
5 another liability there.

6 PICHA: Correct.

7 MOTE: So is this liability that you have here, \$1.1
8 billion, not the same issue as the total cost? For example,
9 the judgment fund is the source of funds for the liability
10 payments to utilities, and is that why this is lower--

11 PICHA: That's not included. It's not. This is
12 EM-managed materials only.

13 MOTE: Oh, it's only EM. Okay.

14 PICHA: Correct. I'm sorry.

15 MOTE: Okay.

16 PICHA: Yeah. So every year the Department does an
17 environmental liability audit to look at basically the EM
18 program and what the costs are for completion of the EM
19 program. So that's--I should have caveated that.

20 MOTE: Okay, all right, thanks.

21 EWING: Other questions from staff or Board?

22 BRANTLEY: Sue Brantley, Board. You gave a really nice
23 overview of the number of tanks, like a snapshot, and then
24 how many have been closed, etc. Any plans on the part of DOE
25 to make new tanks? What would make you make new tanks? Can

1 you just talk about, you know, the other side of the
2 equation?

3 PICHA: Sure. At Savannah River we certainly are not
4 planning to have new tanks, because, as you can see, we're
5 trying to close tanks; and we've got some success in terms of
6 closing four and hopefully another two this year. At Idaho,
7 same thing, we've closed eleven tanks, and we're hoping to
8 continue to close the rest. At West Valley, basically, there
9 we have not closed any tanks; in fact, the decision on how to
10 proceed with, I'll say, the disposition of the residues in
11 the tanks has been a bit deferred. What they have done there
12 is they have installed a tank drying system, and they keep
13 the relative humidity below some level to control liquids in
14 the tank. And they have an issue with intrusion into their
15 vaults that--in which the tanks are housed.

16 At Hanford that is a question that's come up,
17 particularly in light of the six tanks that have indicated
18 loss of levels. The site here and us at the headquarters
19 have been looking at different approaches for how we might be
20 able to come up with, I'll say, an approach to help to
21 mitigate that as well as be incorporated into a system-wide
22 approach for tank waste treatment in terms of things like
23 blending wastes and characterizing waste, sampling waste. So
24 we're looking at a few different options in that regard.

25 EWING: All right. To keep us on schedule, I think

1 we'll call an end to questions, but I want to thank you for a
2 very clear and useful presentation.

3 PICHA: Thank you.

4 EWING: So the next presentation is by Carol Jantzen
5 from Savannah River National Laboratory, and Carol will give
6 us an overview of Vitrification as a Complex-Wide Management
7 Practice.

8 JANTZEN: Well, good morning. My name is Carol Jantzen.
9 I've been at the Savannah River site since 1982, and I was
10 there for the groundbreaking for the Defense Waste Processing
11 Facility, so I've been there a long time and have a long
12 history in vitrification. And, of course, Rod knows me from
13 my previous days when I was doing ceramic waste forms.

14 I was asked to talk about vitrification as a
15 complex-wide management practice for high-level waste. And,
16 specifically, I was sent an e-mail, and I paraphrased these
17 bullets out of the e-mail that I was sent by Bruce as to what
18 I should be speaking about. I put in the timeline of how
19 high-level waste glass and glass-ceramics were developed.
20 There was a famous down-select between glass and ceramics in
21 Atlanta. I've got one slide on that, because people like Rod
22 and Werner and I lived through this.

23 And then I was asked to speak about what types of
24 glass will be produced and how much of each type and how many
25 waste canisters, how are the strategies different at the

1 different EM sites, how are they similar or how are they
2 different. Now, these last two bullets could be a talk all
3 unto themselves, so I had to kind of shoehorn that in at the
4 end of my talk.

5 And this talk runs long. I've got some
6 introductory slides for people who might not be as
7 knowledgeable about the history behind everything, how we got
8 here to where we are today. So I'm going to try and go
9 quickly through the introductory slides and make it all the
10 way to the end so that we can talk about what are the
11 technical and performance standards for glass as a waste form
12 and what kind of tests do we use for the determination of the
13 long-term performance of glass with respect to disposal in
14 different geologic environments.

15 My slide is older than Ken's, so my numbers don't
16 agree with his exactly, but the concept is still there in
17 terms of gallons and curies. I think Ken had 37 million
18 gallons at Savannah River; my slide says 32. And I think he
19 had 55 million gallons at Hanford, and mine says only 50.
20 And it's not that it's varying; it's just that people's
21 estimates (inaudible) at what's out there go up and down over
22 the years.

23 I think we all know that things like what are
24 colored in yellow are the elements found in waste. So if
25 you're going to make a glass out of this stuff, it's not

1 really simple, because you've got two-third of the periodic
2 table to deal with. On top of it being two-thirds of the
3 periodic table in the waste itself, we add additional
4 elements like lithium and boron to help flux the glass and
5 make it pourable, meltable at reasonable temperatures like
6 1150 so that you're not volatilizing too many of the--or the
7 radionuclides are not volatilizing some of them at all. And
8 then what I've done is I've circled the ones that are long-
9 lived radionuclides, the ones that kind of get to be
10 important when you're doing a performance assessment. You
11 know, if they come out of the glass at any significant rate,
12 they are the ones that are going to drive your long-term
13 performance assessment.

14 I know there are several geochemists on the Board,
15 and I have a background in geochemistry. And so I wanted to
16 highlight the fact that our glass, especially the high-ion
17 glass and the high-aluminum glasses, which the high-ion comes
18 from the PUREX process and the high-aluminum wastes come from
19 another process, if you take the boron out of it and
20 renormalize it, it's very, very similar to a tholeiitic or an
21 ijolitic basalt in terms of its aluminum content and in terms
22 of its iron content. And that makes it somewhat easier when
23 you are looking at the crystallization. I think you're going
24 to hear some talks this afternoon about the crystallization
25 of the high-level wastes. It simplifies down to looking at

1 either Nolan or Bailey and Schairer's basalt tetrahedron,
2 iron, silica, aluminum, and alkali or alkaline earth. You're
3 going to hear some about the crystallization of spinels and
4 the crystallization in nepheline probably in this afternoon's
5 talks. And that crystallization is actually driven by this
6 ternary inside the quadrilateral, and this ternary over in
7 here, you can actually define a pseudobinary across the
8 fields of nepheline and spinel in there. And, as I said, it
9 all comes out of the older literature around 1966 for basalt
10 magmas.

11 The other thing you're going to hear about, you're
12 going to hear about things like--you know, we've got sodium
13 nitrate in the waste at Savannah River. We add formic acid,
14 and we bring some of the salts down, the sodium down, as a
15 sodium formate, sodium oxalate, things like that. But what
16 happens when it goes into the melt obviously is that these
17 anions come off the formates, and the oxalates come off as
18 CO₂; the nitrates decompose, come off as NO_x. And so what
19 you get out the other end is just glass on an oxide basis
20 just like you would have for any geochemical or basalt or
21 rock that you analyze, so everything that comes out is
22 oxides. But you are going to hear some very, very complex
23 chemistry that goes on when you add these additions.

24 And I'm going to kind of go through this quickly,
25 but basically what this is is this is a timeline, this is

1 increasing melt temperature. The phosphate glasses tend to
2 melt at lower temperatures. The borosilicates are kind of in
3 mid-range, about 1100. The nepheline syenites, which the
4 Canadians were interested in, melted even higher; and the
5 Canadians and the Germans were also interested in looking at
6 glass-ceramics. And so the solid arrows indicate what
7 country was looking at these waste forms at what particular
8 times. I think everyone has now gone to borosilicate glass
9 except maybe Russia, who is still making aluminophosphate-
10 type glasses.

11 In the U.S. the borosilicate research actually
12 started at MIT back in the very, very late 1950s, a professor
13 by the name of Goldman. They were taking ceramic glazes and
14 trying to put nuclear waste into it. The glazes melted at
15 very high temperatures, so they put boron in to flux these
16 glazes, and thus we came up with borosilicate glass. A
17 gentleman who worked for Professor Goldman actually is one of
18 my neighbors in South Carolina.

19 The first borosilicate glass that was actually
20 poured was over in the U.K. in 1962. I've got a solid arrow
21 here, because the research and development was in the U.S.
22 That research and development then went over to the U.K. and
23 to Europe. We weren't doing very much here in this country.
24 And then it got picked up again in both the U.S. and in
25 Europe. In 1975 the Savannah River site decided to look at

1 borosilicate glass. This is the down-select, also known as
2 the Great Atlanta Shootout or the Hensch Panel, that occurred.
3 I've got some details on the next slide.

4 Our DWPF groundbreaking after that down-select was
5 in 1982 or 3. I can't read it from here. We went through
6 cold runs in 1994. We did non-radioactive runs in the
7 melter. We then went radioactive in 1996, and our second
8 melter was in 2003. And, as Ken said, we just passed our
9 tenth anniversary on our second melter.

10 The DOE Hensch Panel, it was a three-year study
11 comparing simulated high-level waste glasses and ceramics.
12 These are all the different types of waste forms that were
13 looked at. So basically we were looking either at glass down
14 here or at durable crystals up here. There is now a tendency
15 to look at things in between here at glass-ceramics to get
16 more waste into the glass. They did product scores, and they
17 did process scores. They were looking for something that was
18 continuous or semi-continuous, so glass got a very high score
19 for that. Whereas, you can see that SYNROC and some of the
20 tailored ceramics got higher product scores; they were more
21 durable-type waste forms. So, in the end, when they put them
22 together, borosilicate glass was chosen. It was recommended
23 for Savannah River and West Valley, and it was recommended
24 that ceramics continue to be studied.

25 In 20/20 hindsight, I think we've learned a lot

1 about why glass is actually more similar to ceramics than we
2 thought at the time at which this decision was made. There's
3 been a good deal of x-ray absorption fine structure work done
4 on the structure of glass. Everybody always thought that
5 glasses were completely random structures. What they found
6 is that they've actually got polymerized regions, these PR
7 things that I've labeled here, and they've got depolymerized
8 regions. And so your cations like aluminum and silicon help
9 the polymerization and some of your other modifiers break up
10 the network. This work was done in 1985.

11 So they found this short-range ordering and medium-
12 range ordering in alkali borosilicate glasses. They don't
13 have the long-range ordering that ceramics have, but they do
14 have very good bonding characteristics. They have also found
15 that things like uranium sometimes tend to cluster in groups,
16 not necessarily totally depolymerize, but more depolymerized
17 than the PR regions. They have also found that molybdenum,
18 which is of interest to the people in France, also forms in
19 these depolymerized regions, and so does sodium sulfate,
20 which is one of the reasons that sodium sulfate doesn't like
21 to be soluble in borosilicate glass, because it goes into
22 these depolymerized regions and can easily come out as a
23 secondary crystalline phase.

24 So we've learned a lot from actually looking at the
25 structure of glass about what makes it durable, what doesn't

1 make it durable, what makes it precipitate out something like
2 sodium sulfate or sodium chloride.

3 I was asked to talk about how much there is what
4 kinds of glasses. Right now everyone is making borosilicate
5 glass and/or intending to make borosilicate glass. At
6 Savannah River we've produced 6,350 metric tons of glass
7 between 1996 and March 2013. We've made 3,603 canisters. We
8 have that many more to go. West Valley had their 275. They
9 made about 500 metric tons between 1996 and 2002. And the
10 Hanford waste, if I've used the right projection here, which
11 I think I have, it's about 32,000 metric tons that will be
12 projected to be made and over 10,000 canisters. And the
13 document that I read said that the cesium/strontium capsules
14 would somehow be incorporated in those as an additional 120
15 canisters. And I wasn't clear whether that meant if it was
16 being put back in or whether it was a separate disposition
17 path. So that's just kind of a placeholder for it in that
18 table.

19 At Savannah River we've had 17 years of continuous
20 radioactive operation at the DWPF. I told you we did two
21 years--about a year and a half--of non-radioactive runs that
22 we call cold runs. This is the number of actual canisters we
23 made on a yearly basis. The first year we only operated five
24 months. We started in April, I think it was, and made 64
25 canisters. And you can see we progressively went up. As the

1 feeds became more difficult, the canister content went down.
2 Here we had a lid heater fail in, I think it was, late 2002
3 or early 2003; and so there was an outage while we changed
4 the melter out. When we got the new melter in, we changed
5 our liquidus, our crystallization model, which enabled us to
6 put more waste into each canister. We added a glass pump in
7 2004. It works like a coffee pot percolator. It sucked hot
8 glass up the stem and then splurged that hot glass out on the
9 top of the melt pool to help melt the cold feed that was
10 coming in. We slurry feed so the feed is cold, and it helps
11 melt that cold cap material that pours in.

12 And then we removed the glass pump and added argon
13 bubbling. We run a reducing flowsheet. We keep the total
14 iron at about .2 to minimize volatilization of technetium and
15 ruthenium, and so we didn't want to bubble air and disturb
16 that redox equilibrium, so we chose to bubble argon. And, as
17 I said, our first melter lasted eight and a half years; our
18 second melter just passed its tenth anniversary; and both of
19 them, actually, interesting, are failing because of the lid
20 heater. There is actually nothing wrong with the rest of the
21 melter.

22 SPEAKER: (Inaudible.)

23 JANTZEN: A lid heater. I'll point one out in a minute.

24 Now, I was asked to talk about similarities and
25 differences, and I've broken that into three parts. I'm

1 going to talk some about the hardware--I'm going to go over
2 the hardware pretty quickly--flowsheet designs--they are
3 complex flowsheets, but I'm going to have to go over them
4 quickly--and then the process control strategy. I think, you
5 know, where there are differences, it's like this group of
6 tomatoes. Some are green, some are yellow, some are red, but
7 they're all tomatoes. So they're all melters, and it's all
8 vitrification.

9 The similarities, they're all joule heated,
10 electrically heated. They all use Monofrax K-3 high-chrome
11 refractory, use Inconel® 690 electrodes. They're all slurry
12 fed. The canisters are all 304L stainless, and the nominal
13 melt temperature is 1150.

14 Differences are in the melter: size, shape,
15 primary type of melt pool convection. At Hanford--let me go
16 back over here. At West Valley they use--the melter was a
17 square configuration, 2.2 m² surface area, and it had only
18 natural convection.

19 At Savannah River our melter is round. We didn't
20 want to have--we originally started out with natural
21 convection. We didn't want cold corners in the melter where
22 things could crystallize, so we went with a round melter. In
23 2004 to 2010 we used this airlift pump that was like a coffee
24 pot thing, and then we switched to the argon bubblers, and we
25 have four bubblers in that melter.

1 The Hanford high-level waste melter, again, is--
2 I've got advanced joule heated. Advanced means it's bubbled,
3 so ours went from being joule heated to advanced. It's a
4 square configuration; it's air bubbled; it's larger than any
5 of the two before it; and there are six bubblers in that
6 melter--plan to be in that melter.

7 Here's the canisters. The canisters are West
8 Valley and DWPF's canisters, two foot in diameter by ten foot
9 tall, and the high-level Waste Treatment Plant canisters are
10 15 foot tall by two foot in diameter.

11 This is the DWPF melter at Savannah River. Right
12 there are the lid heaters, sir, right there, those two lid
13 heaters. When you start a melter up cold, you have to melt
14 the melt pool down; and so the lid heaters help you do that,
15 help keep the plenum hot. There are some electrodes. We
16 have an emergency canister to drain the melter if we have to.
17 This is the normal--normal pour goes up this teapot-like
18 thing, and we do what is called a differential pressure pour,
19 which makes it a semi-continuous flow of glass. The bottom
20 drain is for emergencies. The floor is slightly sloped.
21 There is an offset here between the floor and the pour spout
22 in case crystals or noble metals accumulate at the floor of
23 the melter. We have a complete dual jumpered off-gas system,
24 so if one system gets plugged, we use the other system, and
25 the off-gas plugs are actually water soluble so we can steam

1 clean the off-gas systems.

2 The size of the melter was limited by the crane
3 that could lift it. This is the crane actually lifting the
4 old melter out of its cell in 2003 to put the new melter in.
5 It was designed for 238 pounds of glass an hour, which is 2.6
6 metric tons per day. We actually did get those feed rates
7 with certain feeds; but now that we have the bubblers, we get
8 those kinds of rates routinely. And the off-gas is made of
9 Hastelloy® for resistance to acid gases, because you get
10 things like dilute sulfuric acid coming over or dilute
11 hydrochloric acid coming over from the halites that are in
12 the waste.

13 This is the West Valley melter. It had a shape
14 like an inverted prism, so very, very steeply sloped sides.
15 It used an airlift pour, so the glass comes over here, and
16 the airlift is in here; the sloped floor, again, for crystal
17 accumulation or noble metal accumulation. They didn't have
18 lid heaters, but on start-up they did hang some heaters in to
19 get the melt pool started and then took those back out. They
20 had a single off-gas system with certain spare components.
21 The size was 50 metric tons. The pour rate is--that should
22 be 45.5 pounds per hour, which is about one metric ton a day.
23 And the off-gas, again, was Hastelloy® and Inconel® for
24 resistance, same as DWPF.

25 And this is the high-level waste glass melter plan

1 here at WTP. They are planning to run two melters. They've
2 got a bubble rise overflow that is used for both normal and
3 emergency pours. They've got a flat floor, but there is a
4 very large offset of the pour spout to allow for noble metal
5 or crystal accumulation. No lid heaters; single off-gas
6 system for each melter with spare parts; almost 79 to 90
7 metric tons. The melter itself holds about 11 metric tons of
8 glass, and so replacement is not by a crane; it's by rail.
9 This is what it's designed to produce. And, again, the
10 off-gas is made out of acid-resistant alloys. Actually, let
11 me point that out. You can see the rail system down in
12 there.

13 This is the DWPF flowsheet, and this gets very
14 difficult, so I'm going to try not to go through all of it,
15 but kind of point you in the direction of what's important.
16 This is the waste out in the tanks that's separated. You've
17 got this sludge fraction, and then you've got this supernate
18 fraction. As Ken said, the supernate then goes through salt
19 processing, any one of these processes. Some of them are on
20 line, and we're waiting for that salt waste processing
21 facility to be completed, but we had these other technologies
22 in place. It removes the cesium and the strontium and the
23 actinides and brings them back over to this hold tank so that
24 they can be combined with the sludge waste that's coming in
25 from this stream over here. And then the decontaminated salt

1 supernate goes to grout that's poured into vaults at our
2 site.

3 The sludge then goes to a million-gallon tank, and
4 we do things like wash the sludge to get the soluble salts
5 off the sludge so that that can go back to the salt, and
6 thereby we're minimizing the amount of waste that we're
7 sending forward to the high-level waste glass melter. We're
8 getting rid of these soluble non-radionuclide-containing
9 components to here just to make sure that they're clean. We
10 again process them through here to get rid of cesium and any
11 actinides.

12 So we do our sludge washing. We do aluminum
13 dissolution. The aluminum dissolution, the aluminum comes
14 out as sodium aluminate, which is soluble. Again, that helps
15 us reduce how much sludge we have to send forward to the
16 melter. This is all done in a million-gallon tank. That
17 goes over here. This becomes our qualification tank, which
18 is another million-gallon. If I take a sample of that tank
19 and I take the uranium, I can actually take the uranium from
20 that tank and predict how much uranium there is going to be
21 in the glass all the way over here, because all of the
22 processes that happens here happens with this formic acid,
23 oxalic acid, those kinds of things, so that the cation
24 content of the material that I'm analyzing here does not
25 change in this box that's called the vitrification facility.

1 So the qualified sludge goes from here. It goes
2 into here. It gets acids added to it. The reducing acids,
3 they allow you to strip the mercury out--steam-strip the
4 mercury out. They bring the mercury down as metallic
5 mercury. When all that processing is finished, we go over to
6 the SME, we add the frit, the glass formers. This is our
7 hold point: Are we or aren't we making acceptable glass?
8 I'll show you when I get to the process control strategy how
9 we treat the melter's black box, which is why I've got it
10 colored in black. We meter in whatever we need to meter in
11 from this vessel to here. We go to the melter hold tank.
12 And basically what we've done during our cold runs was we cut
13 cans open, and we sliced doors in cans, and we sampled to
14 prove that we could predict what's in the canister from what
15 was in this vessel here. And that's why that's our hold
16 point.

17 I have written down here all the reasons we go
18 through this SRAT, sludge receipt and adjustment tank. This
19 is everything that happens in there. I don't have time to
20 talk about it all. But this tank is basically hydroxides,
21 nitrates, and carbonates when we're done.

22 I don't know as much about these, so I tried to
23 keep these as simple as I could. West Valley had one tank
24 with high-level waste sludge in it, and they had a secondary
25 small tank with THOREX waste in it. They had these zeolite

1 ion exchange columns. So they took the supernate, and they
2 ran it through the ion exchange columns. That all went out
3 to grout. They then washed their sludge two or three times
4 over, took that rinsate, let it set, took that decant off,
5 ran it through the zeolite columns, that went out to grout.

6 Now, when that was all finished, then they took the
7 zeolite columns, put them down into the sludge, and then they
8 mixed in the THOREX materials. And you'll notice they've got
9 stir bars in all these tanks, and we have stir bars in all
10 our Savannah River tanks.

11 They were then able to make smaller batches where
12 they came over to this tank, and they would choose a chemical
13 composition, a target, because they knew what the composition
14 of this tank was after they got all the zeolite and the
15 THOREX into it. And then they would--so that this tank
16 became their qualification and acceptable hold point.

17 And this is as simple as I could try to make the
18 Hanford flowsheet, the operations that happen at the tank
19 farm, operations that happen in pretreatment, and operations
20 that happen in the vitrification facility. These are your
21 million-gallon tanks in the tank farm, the kind of ones that
22 Savannah River uses to do all their mixing and pretreatment
23 in. In Hanford there's a separate pretreatment building that
24 handles all the pretreatment. And then this material goes
25 forward. And the hold point for the qualification are these

1 tanks over here. And there is a recycle loop, which is
2 almost off the edge of my picture, that comes back in here
3 for pretreatment.

4 So, in summary, the flowsheet differences are the
5 DWPF blend sludge in the tank farm to dampen composition
6 variation; we perform pretreatment in the tank farm; and we
7 qualify the sludge in million-gallon tanks in the tank farm.
8 A typical batch is only 300,000 to 800,000 gallons, and so
9 this minimizes how much analyses you have to do, because
10 you've got a constant 300,000-to-800,000-gallon batch. We
11 call it the macro batch concept. You don't have to take more
12 samples if you're still processing the same macro batch.

13 DWPF we use REDOX control. West Valley uses REDOX
14 control with sugar, and they did not bubble air. WTP uses
15 sugar also, but it's not really for REDOX control; it's to
16 reduce the nitrates. But they're bubbling air through their
17 bubblers, and so this re-equilibrates the melt pool to
18 oxidizing conditions. It works out that you get about 30 to
19 33 percent retention of the technetium in a single pass.
20 That's why they have that recycle loop in there to go back to
21 pretreatment. When you keep recycling, you can get up to 80
22 percent of the technetium retained in the glass.

23 All mix/blend transfer tanks are accessible due to
24 concerns about viscosity and erosion/corrosion from
25 crystalline sludge particles, and all of the tanks are

1 stirred mechanically. That was true for DWPF and West
2 Valley. WTP, some tanks are actually stirred, the ones
3 closest to the melter, and some are accessible, but many of
4 them, I'm sure everybody's heard, have either the pulse-jet
5 mixers and/or they're in "black cells".

6 We use a frit, a melted mixture of glass formers,
7 at Savannah River, chosen on the makeup of a large macro
8 batch. And this leads to only one transfer error and one
9 analytic error during batching when we get to the discussion
10 on process control. DWPF handles it this way. West Valley
11 and Waste Treatment Plant did not and will not.

12 Okay, so in terms of what do you have to know about
13 a glass to be able to process that glass, this list here--you
14 have to know certain things about the product. You have to
15 make sure it's durable; you have to make sure it's
16 homogeneous. And I'll talk about homogeneity and composition
17 in a minute. The regulatory, you can either test a range or
18 you can model TCLP. You need to know thermal stability; you
19 need to know mechanical stability. Process, you need to know
20 all these things to be able to get it into the melter and be
21 sure you're going to get it out of the melter.

22 Okay, I'm running really late. All right, so this
23 is a balancing act, and, if I have to, I'll just go right to
24 the end.

25 The important thing here is that if you're going to

1 allow crystals to form, either durable crystals or non-
2 durable crystals, you have to be careful with the non-durable
3 crystals. They often incorporate radionuclides; for example,
4 sodium sulfate incorporates cesium and strontium, just as an
5 example. If you allow crystals to form, you have to worry
6 about the durability vectors from the crystals.

7 One of the reasons we make homogeneous glass is we
8 don't have to worry about these durability vectors from
9 anything that's not homogeneous or crystals or the grain
10 boundaries. If you're going to allow crystals, you need to
11 know what the durability vectors are. This is a spinel
12 crystal. I have peeled away the leached layer on top of it
13 after I've leached it with a piece of Scotch tape, and you
14 can see the grain boundary dissolution underneath the spinel
15 crystals. So you have to know what these vectors are if
16 you're going to allow crystals to form.

17 At Savannah River we use what's called feedforward
18 statistical process control, because we've got these multiple
19 waste streams, because we have to be very, very confident
20 that these property constraints are met to the 95 percent.
21 And so what I want to talk about is how Savannah River
22 defined the process control region for DWPS and how we
23 qualified the DWPS process control region during non-
24 radioactive start-up and how we actually used this for waste
25 qualification.

1 This is our process control. You'll hear some
2 people talk about it a little bit more this afternoon. We
3 base this on the glass properties. So we use multivariate
4 theory to control limits within this multi-dimensional
5 composition space. So you can take any frit, any waste,
6 Waste 1, Waste 2. If it's got the right durability, right
7 along here, anything to the--let me see--that's your right of
8 this line makes durable glass--then all of these other
9 intersecting lines are the other properties, like you have to
10 know that if you put it into the melter it's going to have
11 the right viscosity back out of the melter.

12 And I know this is a lot of stuff to digest.

13 And so this particular process control system keeps
14 us in control 95 percent confident that we're going to be
15 able to make glass. We can target right down there at the
16 maximum waste loadings. The model accounts for model error,
17 analytic error, tank transfer error, and heels. When you do
18 tank transfer, you've always got to heel the previous
19 material in there. So, while these are the models, all the
20 inner little bands are all the error bands that help you
21 account for all these different sources of error.

22 I'm not going to go through this, but this is our
23 viscosity model. For example, we like to keep it really,
24 really simple; so we put a minimum number of components in
25 our models, and we try--if we have something else that we

1 think is going to be problematic like phosphate, sulfate,
2 titanium, we set a limit. At this point in time we are going
3 to be, actually, adding a titanium term to the model. But if
4 you don't need a titanium term, if that model with its seven
5 parameters defines your system at an R^2 of .97, then you
6 don't need as many terms in your models.

7 And all our models developed over very, very, very
8 wide ranges where 110 poise and 20 poise, for example, is the
9 limit of what the melter can actually do. So you've
10 developed your model over ranges that are wider than what
11 you're applying it. And I'm just going to use this as a very
12 quick example. The question came up: Do we need a uranium
13 term, do we need a thorium term in our viscosity model? But
14 we looked at what was out in the literature on uranium and
15 thorium bonding by Gordon Brown of Stanford and other people,
16 and it turned out that the Uranium⁺⁶ and ⁺⁴, they had two
17 bridging and four non-bridging oxygen bonds, and so they
18 cancelled each other out, and we didn't really need to put a
19 uranium term in our model. So, again, it's keep it as simple
20 as you can possibly keep it in terms of your sources of
21 error.

22 This was our cold runs. This is what I told you
23 about. We developed the process control system that I just
24 spoke about. We wanted to make sure that we could treat the
25 melter as a black box, so we cut open the canisters that were

1 sectioned. There were 30 of them. There were 56 that had
2 the walls removed. And we have this little glass sampler
3 that you can push into the stream of the glass and pull it
4 back out. And we analyzed that, and we analyzed the cans,
5 and we had our projections, and we proved with 106 different
6 types of samples that we could control. Not only could we
7 control if we had a constant tank full of, let's say, low-
8 viscosity high iron, but if we fed in high aluminum on top of
9 high iron, the feed would be slowly changing, because melters
10 act as continuously stirred tank reactors. So you have to be
11 sure when you're transforming from a high aluminum to a high
12 iron or back to a blend, you have to be sure that your
13 process control system actually can handle those transitions
14 in waste feeds.

15 This is the acceptance part. I hope I can do it in
16 five minutes or thereabouts. I've talked about the 95/95,
17 that you've got to be 95 percent confident that you've met
18 all these parameters, and the first and most important one is
19 this durability. So the way these models work is that we've
20 taken into account the interactions between components. I
21 showed you that on the previous slide. And we know that this
22 range of glass is processable. We've done it for the last 19
23 years at DWPF.

24 And so the process control was then used to
25 demonstrate the acceptable process by these kinds of linking

1 relationships. If you control the process in a given
2 composition range, then you've got composition control of
3 your glass, of your product. If you've got composition
4 control, then you've got dissolution rate control. And if
5 you've got dissolution rate control, then you've got
6 performance control and acceptable performance.

7 To prove this, we developed a thing called the
8 environmental assessment glass, which was a glass that was
9 used in the environmental assessment by the DWPF in 1981, and
10 this has become a standard glass around the DOE complex. And
11 so we've done lots and lots of tests on the environment
12 assessment glass. We know where its upper and lower 95
13 percent confidence bands are. So this is the lower 95
14 percent confidence band of the EA glass durability tested by
15 a lot of people. We did all kinds of round robins.

16 Then you have some kind of durability model where
17 glasses over here are less durable and glasses over here are
18 more durable. And you've got some kind of model; I don't
19 care if it's empirical or first principle, but you've got a
20 model. You've got an upper 95 percent confidence band in the
21 lower. And so where the upper 95 percent confidence band of
22 your model intersects the lower 95 percent confidence band of
23 your standard high-level waste glass, then you are 95/95
24 percent confident that you've made a good product.

25 What was done at West Valley--and I think that's

1 what's going to be done here at DWPT--is they had only that
2 one tank. They chose the target, and then they went 95
3 percent on either side of that target. This actually talks
4 about the models that are being used here at WTP. And I
5 don't have time to go through that whole slide.

6 This is actually the one I wanted. They chose a
7 target at West Valley; they analyzed the tank when they did
8 the tank transfer; they found the lower and upper 95 percent
9 confidence bands of that material; and then they made it into
10 glass. They harvested shards out of their canister, out of
11 the top of their canister, with a vacuum device. So here is
12 how they targeted their feeds that were going into the
13 melter. Here is the analysis of the shards from the
14 canister. And if the average and the upper and lower 95
15 percent confidence bands on what was in the can matched up--
16 the limits matched up to within what it was in the batch
17 before it went into the melter, then that was considered 95
18 percent acceptable.

19 And I'll let you read this for yourself, but a
20 production facility can't wait until a melt or a waste glass
21 has been made to assess whether it's acceptable or not. So
22 we made the acceptability decision on the upstream process
23 rather than on the downstream melt or glass. We used
24 mechanistic and semi-empirical models. The model is
25 mechanistic (inaudible) intercept, so they're really semi-

1 empirical.

2 The alternate methods are what's called statistical
3 quality control, and you've got the targeted middle of your
4 region. The glass product is sampled after the vitrification
5 is complete and then compared to what it was before it was
6 complete. It's sampling is analytic-intensive compared to a
7 minimal sampling of actual product if you do the statistical
8 process control.

9 This is another little bit of history. The
10 repository is kind of in the middle. And everybody--the
11 Environmental Protection Agency, the NRC, is talking to the
12 repository. The waste form producers--DWPF, West Valley--
13 were talking through the DOE Office of EM to the repository,
14 and we were also talking to the EPA. We provided the data
15 that made glass (inaudible) the best-developed available
16 technology for high-level waste (inaudible).

17 So here's the real issue, and this is the one that
18 could be a talk all unto itself. The waste form must be
19 acceptable to a repository yet to be sited and/or built. In
20 1982 we had the Nuclear Waste Policy Act Amendment; we had
21 multiple repositories, tuff salt, basalt. When it was
22 amended in 1987, we were told to just look at tuff in the
23 Nuclear Waste Policy Act Amendment. In 2009 the
24 administration cancelled Yucca Mountain.

25 But, you know, it doesn't matter if it's 1982,

1 1987, or 2009. The problem is still the same. You're making
2 glass, and you don't know when there's going to be a
3 repository ready, and you don't know what it's going to be.
4 So we had to come up with a strategy. One of the strategies
5 was to develop this glass durability standard, this EA glass.
6 It meets all the repository requirements, because all the
7 other glasses--if it meets all the requirements, you've
8 tested all your production glasses or a lot of your
9 production glasses; okay? And if the EA glass makes it and
10 all your glasses are better than the EA glass, then your
11 glass is going to be acceptable; all right?

12 What we did was also we related--we developed and
13 related a short-term test to measure how our glasses
14 performed against EA glass. This is ASTM C 1285, also known
15 as the Product Consistency Test, and its title says what it
16 is. You want to make sure that your product is consistent.
17 You want to make sure that the glass that you're making today
18 is as good as the glass that you made 18 years ago or 17
19 years ago; okay? So you want to make a consistent product.

20 And then we also did all these other things. We
21 performed long-term tests on high-level waste glass and
22 natural analogs. We performed repository relevant tests. We
23 made these rock cups out of basalt and salt and tuff. We
24 used various groundwaters. In the case of the basalt, we
25 actually equilibrated it in an argon glove box to make low EH

1 groundwater. And we performed materials interactions tests.
2 We related--we did some tests at WIPP with heaters. George
3 Wicks did that work in WIPP, in STRIPA and granite and in
4 Ballidon and clay in the United Kingdom, and we related long-
5 term and short-term testing.

6 What I didn't say up here is that we actually
7 always thought, well, if we had the results of this short-
8 term test, someday somebody will come along with a repository
9 relevant test, and we'll have to relate the response of our
10 short-term test to a repository relevant test. Well, what
11 happened was when they did the Yucca Mountain Total Systems
12 Performance Evaluation, they actually used this approach
13 here.

14 And this is kind of the time frame of how
15 everything happened, and I'll try to summarize this. I don't
16 want to read it all. Basically, back there around 1982 just
17 after glass was chosen as a waste form, some people did some
18 geologic repository modeling, and they said that fractional
19 dissolution rates between 10^{-4} and 10^{-6} parts per year would
20 be about the best that you could do with any waste form. And
21 this wording found its way into 10 CFR Part 60.113, which
22 specified those numbers.

23 And so when we made the EA glass, we chose a glass
24 that gave a durability in between those two numbers where
25 that would last anywhere from 10,000 to a million years. And

1 what's interesting about that is that it was for any
2 repository type geology. The modeling was done out at
3 Lawrence Livermore. It's in some very old documents. And it
4 didn't matter what the repository geology was if you could
5 guarantee that the glass would have this particular type of
6 durability.

7 Then both the MCC and the ASTM developed a whole
8 suite of tests--I list them all here--that could be used to
9 look at the mechanisms by which borosilicate glass dissolved.
10 Now, the important thing is that that glass standard, the EA
11 glass, is a borosilicate glass standard. And it's a
12 borosilicate glass standard because we used all these tests,
13 all these ASTM tests and MCC tests, to understand the
14 durability mechanism. We don't understand the durability
15 mechanism as well for other types of glass, for phosphate
16 glass, for example. So, theoretically, you would have to
17 develop another standard. But I've put here all the things
18 that are specific to borosilicate glass and kind of where we
19 are right now.

20 And I want to get to my last slide. These are the
21 waste acceptance product specifications. The ones in blue on
22 the left-hand side of the slide are the ones that have to do
23 with the glass itself. And you see right here that product
24 consistency, more durable than the EA glass by two standard
25 deviations, so that you can project the durability. And we

1 measured the production glasses. We still use that little
2 glass sample or put it in the neck and check and run the
3 durability standard in our high-level caves.

4 The rest of the standards that are over here have
5 to do with the canister itself and with the canistered waste
6 form. This is Rev. 3 of the WAPS, Waste Acceptance Product
7 Specifications. Our site has not adopted it yet, because
8 it's just fairly new out.

9 And then what I call the roadmap of predicting
10 long-term behavior is an ASTM standard that took a long,
11 long, long, long time to develop standard practice for the
12 prediction of the long-term behavior of the materials,
13 including waste forms used in engineered barrier systems for
14 geologic disposal of high-level waste. And it's a roadmap
15 for defining your problem, defining your repository
16 environment, testing, modeling, predicting, model
17 confirmation.

18 This is the prediction part of it. This is the
19 testing part of it. This is the modeling part of it. There
20 are loops that go back around through here. There's your
21 natural analogs. I wanted to point that out. And we have
22 managed to get all the way down to here, which is the
23 prediction for the repository when we got to the total
24 systems performance evaluation.

25 These are some of the submodules. How do you

1 develop an accelerated test? (Inaudible) my cartoon. Very
2 important when you do an accelerated test that you accelerate
3 the right mechanism. This came out of a--okay, everybody
4 gets the point of the slide. This actually came out of an
5 August 1998 workshop on developing test methods and models to
6 simulate accelerated aging of infrastructure of bridges and
7 buildings. So you've got to make sure that your test doesn't
8 overdo it.

9 And then these are all the tests that I talked a
10 little bit about. Right here I thought was kind of
11 interesting. MCC, from 1980, developed a lot of tests, and
12 most of those have become ASTM-type tests now. The ones that
13 they didn't develop were the repository interactions tests.
14 Those were never developed.

15 EWING: Carol, I want to be sure and leave time for
16 questions, so--

17 JANTZEN: Yup, yup, that's it, that's it. That's my
18 last slide.

19 EWING: All right.

20 JANTZEN: That's the roadmap.

21 EWING: Okay, thank you for covering really a huge
22 amount of history and information data. It's very, very
23 useful.

24 So questions from the Board? Jerry.

25 FRANKEL: Jerry Frankel, Board. Thank you again for

1 this talk. It was really educational for me. And,
2 unfortunately, it brought up a lot of questions that I'd like
3 to ask, but I'll try and limit it to one. I'd also like to
4 congratulate you and Savannah River for 17 years of
5 successful operation of the DWPF. Quite an accomplishment.
6 And I think that the experience is really, really valuable.

7 What I'd like to ask you about is the durability
8 and reliability, not of the glass but of the facility
9 components, say, rather than the melter. So how has it
10 performed over this time period, say, compared to
11 expectations of its performance and what lessons have been
12 learned that maybe are valuable to the WTP?

13 JANTZEN: It's interesting. I think the only failure
14 that we had on start-up was, in that teapot pour spout thing,
15 we had a disengagement problem. The molten glass would come
16 over a knife edge, and it would disengage too quickly and
17 then wander around before it found its way into the can. And
18 what we wound up doing was actually manufacturing a part that
19 fit into the existing knife edge that made another knife edge
20 further on down. So of all the things that could have gone
21 wrong during start-up, that was the only one that we had some
22 difficulty with.

23 FRANKEL: But over time how has it held up, the
24 durability over time?

25 JANTZEN: It's held up very well. I mean, we've learned

1 some things like we've decided to put a heated bellows in the
2 pour spout so that when you run higher concentrations of
3 waste in your feed, as the glass pours--it's about the size
4 of your pinky or a pencil--it starts to cool, and so
5 sometimes it can start to crystallize if the pour spout is
6 not kept very warm. So we made a heated bellows in the pour
7 spout so that we could handle higher waste loadings in the
8 glass. I mean, almost every one of the issues that we've--
9 and things like the heated bellows, they're in a very
10 corrosive environment and very hot environment, because
11 they're seeing the glass pour. So we have to keep spares; we
12 have to change them out.

13 FRANKEL: And the rest of the facility (inaudible)?

14 JANTZSEN: I think we've had one vessel that had
15 developed a hole in it, one of the three vessels that we mix
16 in with the stir bars. Occasionally we have to replace a
17 paddle. Occasionally we had to take--we had to go in with
18 the crane and weld a patch over the hole in this one vessel.
19 But every cell is accessible by a crane, and so when you have
20 to do an engineering fix, you have to have things be
21 accessible.

22 EWING: Mary Lou.

23 ZOBACK: Mary Lou Zoback, Board. I have a question
24 going back to very early in your talk when you showed the
25 output from the 17 years of operation, and in 2010 there was

1 a huge increase in output when the argon bubbling was added.
2 So where did the idea for the argon bubbling come from? Was
3 that R&D being done by the program, or was that a commercial
4 process that you adopted or what?

5 JANTZEN: That was R&D that had been done by Vitreous
6 State Laboratory.

7 ZOBACK: So it was funded by DOE?

8 JANTZEN: Yes, in support of WTP.

9 ZOBACK: Okay, thank you.

10 EWING: Other questions? Sue.

11 CLARK: Sue Clark, Board. And, actually, I have two
12 questions. But to build on what Jerry was asking about
13 earlier and thinking about what's planned here at Hanford,
14 based on your experience at Savannah River, does that provide
15 any confidence that might help with this issue of black cell?
16 You know, this idea that you would create a black cell that
17 needs no maintenance and you have no accessibility for many
18 years, is there anything that comes from Savannah River that
19 adds any confidence in that?

20 JANTZEN: We don't have black cells.

21 CLARK: Yeah, well, I mean, I guess where I'm going is,
22 if anything, your experience is that you've had to go in and
23 do maintenance.

24 JANTZEN: I think it's smart to assume that you'd have
25 to do maintenance in facilities that last this long.

1 CLARK: Okay. And then another--my second question has
2 to do with an early slide where you were talking about the
3 different types of glass. And so there was some early work
4 on phosphate glasses, but we never really--

5 JANTZEN: That was done at Brookhaven by a gentleman by
6 the name of Hatch.

7 CLARK: And was, in the whole Hensch Atlanta Shootout,
8 any consideration of these phosphate glasses, or had they
9 already been eliminated?

10 JANTZEN: I think there were phosphate glasses on that
11 list.

12 CLARK: Oh, okay. I didn't see it. It looked like it
13 was more of just a borosilicate (inaudible) versus--

14 JANTZEN: It was the--all right, how do I want to say
15 it--not all phosphate glasses are created equal. The
16 aluminum phosphate glasses seem to hold up very well. That's
17 why the Russians have continued using them for almost all of
18 their wastes. I didn't have time to go through it, but there
19 was lead-iron phosphate glass developed at Oak Ridge by a
20 gentleman by the name of Boatner. It had crystallization
21 issues. It was a lead-iron phosphate. A lot of the
22 crystallization issues and solubility issues, the
23 radionuclides didn't--for example, didn't want to go in it
24 very well due to the lead in it.

25 Now, the University of Missouri, you know, iron

1 phosphates where there's no lead in it, I've read some of
2 their work--I haven't read everything--but the phosphate
3 glasses could be corrosive to your melter materials of
4 construction. So I think the iron phosphate glass at
5 University of Missouri is an acceptable glass, but you would
6 have two issues. You'd probably have to go to a cold
7 crucible induction melter to get around the corrosion issues,
8 and you would probably need to prove that it--you'd have to
9 do one of two things: prove that it leaches the same as a
10 borosilicate glass or develop a different glass standard.

11 There has been difficulty--you know, you make a
12 glass standard, you make these tests, and people take them--
13 and I'm going to call them mix and match, you know, so people
14 say, well, you know, I've analyzed my sodium-iron phosphate
15 glass, and the sodium is lower than the sodium that comes out
16 of the EA glass, so I'm okay. Well, not necessarily, because
17 you haven't proved that it leaches by the same mechanism.

18 And while I'm on that soapbox, we did extensive
19 testing over about 15 years to prove, for example, that
20 sodium and technetium are in the same kinds of deep
21 polymerized regions in the glass, so they come out at similar
22 rates. So you can--if you measure the sodium, boron, and
23 lithium that comes out of your EA glass and compare that to
24 the sodium, lithium, boron that comes out of your production
25 glass that you made yesterday, okay, you can say, well, then

1 I know my--you know, nothing comes out faster than those.
2 And the only radionuclides that come out as fast are either
3 technetium or iodine-129, so you've got--but we had to do
4 those experiments and prove that the tech-99 and that the
5 iodine-129 came out as rapidly or congruently at the same
6 rate as the sodium, boron, and lithium.

7 CLARK: Great, thank you.

8 EWING: Other questions? Sue.

9 BRANTLEY: Sue Brantley of the Board. I actually have
10 two questions also. The first question is: Let's say you
11 were magically put in charge of the Hanford Waste and the Vit
12 Plant as a promotion.

13 SPEAKER: A promotion, huh? Okay.

14 BRANTLEY: One of the issues that we heard about
15 yesterday was not knowing what's in the waste that comes into
16 the plant, the composition. What would you be worried the
17 most about, based on your extensive knowledge? You know,
18 what elements or what compositional variety would you be the
19 most worried about?

20 JANTZEN: It's not so much the compositional variety.
21 What worries me when I delved into this for this talk, more
22 than I had in a lot of years, is the small batches that
23 they're trying to qualify. And while that strategy worked at
24 West Valley--

25 BRANTLEY: I'm not sure I know what that means. Can you

1 just tell me what that means, the small batches that they're
2 trying to qualify?

3 JANTZEN: Well, we've got to go back to that flowsheet
4 slide. I don't remember which one it was. Let me see--

5 BRANTLEY: Well, just conceptually, what are you saying?

6 JANTZEN: We qualify a 300,000-to-800,000-gallon batch.
7 If you're doing smaller batches, you've got all these
8 analyses that have to be run. And analyses take time. So if
9 you don't get--and both West Valley and WTP are--that green
10 octagon--they're qualifying on tanks that are much smaller
11 than the tanks that we're qualifying on; okay? So you've got
12 more analyses to do. And when you're--like I said, if you've
13 got a heel of a high iron waste and you're feeding in a high
14 aluminum waste, you've got some considerable variation in the
15 composition over time as the melter has less iron and more
16 aluminum into it.

17 So I think that can be problematic if you haven't
18 either blended off things in the tank farm, which we do a lot
19 of blending in the tank farm. We work very closely with our
20 tank farm people. They have our models. They actually come
21 up with a system plan for the (inaudible) years, and they
22 look at what tanks they can blend to try and, I'll call it,
23 help us, you know, help themselves. I mean, we're all in
24 this together. And I don't see that cooperation here.

25 BRANTLEY: Okay. Then my second question--

1 EWING: And last. I just don't want everyone--

2 JANTZEN: (Inaudible.)

3 EWING: Yeah, go ahead.

4 JANTZEN: Rod, I just got a promotion. Now, wait a
5 minute here.

6 BRANTLEY: I don't have that kind of power, so--this is
7 from your Slide 39 or something. But you talked about a rate
8 that would work--a rate of dissolution of your glass that
9 would work for any geological repository, you know, 10^{-5}
10 parts per year, and it came from some report. Can you just
11 talk a little bit more about that so I understand that? I
12 mean--

13 JANTZEN: It's actually in the National Academy report,
14 the waste forms report. Rod and I were on the National
15 Academy panel that wrote that document. And those two
16 slides, the ones with the blue--light blue, dark blue
17 stripes--they were numbered in the NAS report, and the
18 pertinent references are given there.

19 EWING: All right. Well, I know there's a lot more to
20 ask about and discuss, but it's my job to keep us on
21 schedule. So we'll take a break now, but I encourage people
22 to use the breaks to continue the discussions, and we'll
23 start promptly at 10:15 with the panel discussion. Thank
24 you.

25 Thank you, Carol.

1 (Whereupon, the meeting was adjourned for a brief
2 recess.)

3 EWING: So my first call is for panelists to come
4 forward and take their seat at the labeled places up front.
5 For the balance of the morning we'll have two panels, both
6 focused on vitrification, the technology and process, and the
7 second panel on waste forms, glass and alternative or other
8 waste forms.

9 So we'll change our procedure a little bit. The
10 moderator for both panels will be Professor Werner Lutze from
11 Catholic University. And so I will cede to Werner the power
12 to call on people for the panel members and staff, and he'll
13 run the panel discussions.

14 And I just ask that you not screw up, okay?

15 LUTZE: Thank you, Rod. I follow your model, you've
16 screwed up already. Six minutes behind schedule (inaudible).

17 Anyway, this morning we're going to have two
18 consecutive suites of presentations. One is focused on the
19 technical experience of vitrification, and the other one is
20 focused on the waste form. But I would like to say right
21 away that the presentations will probably cover both areas
22 more or less, because there is such a close relationship
23 between the waste form and the way the waste form is made.

24 We have, in the first part of this session, four
25 speakers. And, as you can see, we have asked them to make

1 presentations as they feel necessary. There are no titles
2 given here, but they will all address these basic issues that
3 I just spoke about. And we will have the discussion after
4 the fourth presentation, 30 minutes, and then we go to the
5 second suite of three presentations and have another 30
6 minutes of discussion.

7 So, to start, I would like to introduce Stéphane
8 Gin to come up and give us a presentation. Stéphane is a
9 visiting scientist from France at Pacific Northwest
10 Laboratory, and he is here for a year and works on the
11 understanding of (inaudible) corrosion glass. But he will
12 probably also address other issues.

13 Please.

14 GIN: Thank you, Werner. Good morning, everyone.

15 I have a very brief presentation of the French
16 experience in high-level waste vitrification today. I start
17 with a brief history. And my main message here is that we
18 have started in France research in vitrification following
19 research on glass formulation, glass properties, and
20 vitrification technology in parallel. And we started this
21 research at the end of the '50s with a first important
22 realization in '78 with the commissioning of the AVM facility
23 (inaudible) vitrification of Marcoule for the vitrification
24 of high-level waste from defense fuels and then about ten
25 years later the commissioning at la Hague of the R7 and then

1 T7 facilities that have six vitrification lines. Those six
2 vitrification lines were first hot crucible melters with
3 smaller size compared to the U.S. melters we have seen
4 previously. And in 2010 we started--we have replaced one of
5 the six hot crucible melters by a cold crucible melter for
6 the vitrification of more corrosive (inaudible) from the
7 different wastes.

8 What is also important to note is that we've
9 decided to massively invest in nuclear energy in France after
10 the first oil crisis. And we have also decided to reprocess
11 all of the fuels. We have currently 58 nuclear power plants
12 or 58 reactors in France, and it gives something like 1,230
13 tons of spent nuclear fuel that is reprocessed each year in
14 France. And all the (inaudible) waste coming from this
15 reprocessing are vitrified into borosilicate glass.

16 So, as I said, the research started at the end of
17 the '50s and continues at a constant effort. About a hundred
18 people are currently working in developing new--or improving
19 glass formulation, improving vitrification technologies, and
20 recently, since a couple of years, we have invested in
21 modeling the different processes we have to take into account
22 for improving the technology and the materials to cover what
23 happened in the melter, what are all the physical-chemical
24 properties of the materials, and also the prediction of the
25 long-term behavior in the future geological disposal. I want

1 to talk about that briefly just after.

2 So about one hundred people working in this field
3 in the frame of what is called a joint CEA-AREVA laboratory
4 at Marcoule with a lot of cold and hot facilities and
5 different scales from pilot to scale one process for
6 preparing vitrification in la Hague. The AVM facility has
7 been stopped at the end of last year, so the only remaining
8 facility for vitrification in France is in la Hague
9 (inaudible) industrial scale.

10 Okay. I will not skip this one, but I will be very
11 brief, because it has been explained. In fact, we
12 (inaudible) in parallel to improve the quality of the
13 material and the vitrification technology that allow the
14 fabrication of this material. So the quality criteria and
15 the qualification of the material take into account the
16 material's properties and the related technology with all the
17 parameters allowing to fabricate the material. So you have
18 the different properties we have to take into account for
19 this strategy.

20 These are some figures related to the
21 vitrification, so I have compared the situation in France
22 with three important figures. One is related to the fraction
23 of spent nuclear fuel reprocessed. In France (inaudible)
24 it's about 100 percent. It's very different in the other
25 countries that have developed or continue to develop

1 vitrification.

2 In terms of amount of glass produced at the end of
3 the last year, you can see that we have produced the same
4 amount of glass in France as in the U.S. About 7,000 tons of
5 glass have been already produced. But in terms of
6 radioactivity confined in glass, because the commercial fuels
7 that are reprocessed in France have a higher burnup than the
8 defense fuel you have in the U.S., the amount of
9 radionuclides confined in glass is much bigger in France. To
10 give you a comparison, we have already in our 7,000 tons of
11 glass one hundred times more radionuclides that you have in
12 the Hanford tanks, so it's a huge amount of radioactivity
13 that is already confined in borosilicate glass in France.

14 So in all these countries, even if the situations
15 are different, there is a common need of geological
16 repository with a need of designing a smart multi-barrier
17 system and a need of reliable prediction of the glass
18 durability but then on the fate of radionuclides over
19 something like a hundred thousand up to a million years.

20 I would quickly insist on the fact that, contrary
21 to what has been said previously, we think that the glass
22 durability strongly depends on the design of the multi-
23 barrier system. We have two examples here. One is the
24 current design in Belgium with a super-concrete container
25 surrounding the canister and the overpack, and this concrete

1 material will allow the solution to be very alkine and
2 prevent the corrosion of the iron overpack and delay the
3 beginning of the leaching by groundwater. And in the case of
4 the French design, we have only borehole dig in the clay host
5 rock and the canister directly placed in contact with the
6 clay. So we have no buffer in this case. And the very high
7 alkaline pH in contact with the glass will dramatically
8 decrease the glass lifetime once the leaching will start. So
9 that's a big difference.

10 The glass lifetime in this case is expected to be
11 around a thousand years; whereas, it's possible, if the
12 design is favorable to glass, to exceed millions of years.
13 So the concept is very, very important on the glass
14 durability. And I don't believe it's easy to say we have a
15 performance demonstration that is independent of the
16 scenario.

17 So two important milestones in France 2015, so very
18 soon we expect to have a license for opening a geological
19 repository in the (inaudible) of the Parisian Basin in clay
20 formation. And, if yes, we'll have about ten years to build
21 the site and start stirring first the intermediate-level
22 waste and then the high-level waste from 2025. That is a
23 demand of the 2006 Act on Waste Management in France.

24 And the last one, we believe that there is a
25 general need in the world of better understanding the glass

1 mechanism, the rate-limiting mechanism, to improve the
2 predictive model, and to be able to have reliable prediction
3 over this very long period of time. So there is a large
4 international corroboration on glass corrosion starting in
5 Seattle with the U.S. teams in 2009, and now six countries
6 are collaborating within this important topic that is glass
7 corrosion and improving the (inaudible) rate-limiting
8 mechanism in order to improve predictive models.

9 So I'm here at PNNL for working in this field for a
10 year and will be back to France with this new experience in
11 the next couple of months.

12 Thank you very much for your attention.

13 LUTZE: Thank you, Stéphane, for making up for some
14 time. We will go on directly to the next presentation, which
15 is presented by Mr. Hamel. Mr. Hamel is from the Waste
16 Management Plant, Assistant Manager at WTP for the Office of
17 River Protection and Project Director at the Department of
18 Energy.

19 And I think your original background is chemistry,
20 but you went heavily into engineering with many years of
21 experience in project management, among others project team
22 leader at DWPF. And you are going to present material on WTP
23 (inaudible).

24 HAMEL: Actually, I have a very focused presentation.
25 I'm going to talk about vitrification lessons learned from

1 West Valley, as applicable to the Waste Treatment Plant. And
2 I'm going to focus basically on the high-level waste melter
3 and melter design and operations very briefly.

4 This is the West Valley high-level waste processing
5 flowsheet. I think, as everybody is aware, West Valley
6 operated from 1996 to 2002, vitrifying 275 high-level waste
7 canisters in about 660,000 gallons of high-level waste; and
8 in the waste was about 23 million curies of activity. The
9 basic unit operations here for West Valley are the same as
10 WTP's, although on a much smaller scale; and, in addition,
11 the waste is much more complex out at Hanford.

12 As you can see, there's a pretreatment box in the
13 waste tank farm, which includes a separation of the low
14 activity from the high activity fractions, very similar to
15 WTP's design, and the de-ion exchange is done in the Tank
16 8D-1 labeled over there prior to the high-level waste being
17 sent over to the vitrification facility. Very similar to
18 WTP, there is basically a concentration step and then a glass
19 former addition step. Then it would head into the melter,
20 ultimately to be air-lifted into high-level waste canisters
21 and then ultimately to a high-level waste repository.

22 Moving on to the next slide--I've got that.

23 SPEAKER: (Inaudible.)

24 HAMEL: Sorry. I am trainable.

25 This is the high-level waste vitrification in cell.

1 And over in this section here, that's where the melter is.
2 That's where I'm going to be focusing. Over on this section
3 here, that is the melter off-gas system, and that's the SPS
4 scrubber and the high-efficiency mist eliminator, all the
5 basically off-gas trains focused there. Over on this side,
6 this is where the canister handling portion of it is, which
7 includes welding, cerium IV decon, and ultimately movement to
8 a transport cart that actually runs on rails right there and
9 then from there be taken to an interim storage cell.

10 This is the West Valley melter. Of note for this
11 melter is the fact that it ran for seven years. It actually
12 went hot in 1995 and was shut down in 2002. Of that time
13 period, in 1996 through 2002, it was actually processing
14 high-level waste. So it ran for seven years. It had some
15 minor difficulties, but it did not fail. It actually had
16 more projected life left at the end when it was shut down.

17 These are the specifics of the melter, as you can
18 see, that it's basically about one metric ton a day. I'm
19 going to put up a slide next that will show the HLW melter
20 from WTP as designed. One thing that you will note is that
21 the capacity of the WTP HLW melter is much higher. It's
22 about three metric tons a day. So one of the challenges that
23 is here is the scaling factor from going from West Valley to
24 WTP.

25 This is the WTP melter. Now, one thing you'll note

1 also is that one of the big changes in the design is based on
2 basically the electrode structure. There's two side plate
3 electrodes for this melter as opposed to the three electrodes
4 from the West Valley melter. That design change was based
5 off of lessons learned. What was seen in the West Valley
6 melter was that noble metal sludge actually accumulated in
7 the bottom of the melter; and while it didn't short the
8 melter out, it decreased the resistance of the melt, which
9 required more energy to keep that glass melt pool up and
10 going.

11 So, based on that, WTP has made a design change,
12 taking out basically a bottom electrode that was in there,
13 and now what it ends up doing is it would be harder if noble
14 metal sludge were to actually precipitate out of the glass
15 melt for it actually to basically short or decrease the
16 resistance. So that's one of the significant differences in
17 the design there. Similarly, you can see that the capacity
18 is a lot greater here. It's a three-metric-ton-per-day
19 melter. Also, you will see, if you look down at the design,
20 that there's actually two pour spouts as opposed to one pour
21 spout on the melter here, which gives you the increased
22 capacity.

23 Some of the key lessons learned from West Valley
24 that have been incorporated into operations and design at the
25 WTP. The electrical connectors that were inside that

1 actually went on the top of the melter and actually delivered
2 power to the electrodes basically were and had pathways to
3 the plenum of the melter, which actually resulted in
4 contamination being delivered ex-cell into the operating
5 aisles.

6 What was happening is that during the operations,
7 you were seeing pressure fluctuations in the plenum area, and
8 the contamination eventually migrated out and was detected in
9 the operating aisles. While there were no contamination
10 events, there were no uptakes or personnel contaminations,
11 this was a significant evolution, an event that needed to be
12 corrected. What they did at West Valley is they actually
13 opened up the electrical jumpers so that they vented to the
14 in-cell. And, in addition, on the ex-cell portion in the
15 operating aisles, they put HEPA filters so that they were
16 filtered.

17 Another lesson learned involved the actual pouring
18 of the glass in the melter discharge area. There are heaters
19 that hang from the melter top that are basically silicon
20 carbide ceramic melt tubes, if you will, to keep the glass
21 hot while it's pouring. Because they are very fragile and
22 they basically were originally cycled, they basically did not
23 last longer than twelve months, so they had to be replaced.
24 Throughout operational experience, it was found out that if
25 you actually connected them to backup power so they didn't

1 lose power, they were kept thermally hot, it decreased the
2 thermal cycling on that and actually increased their life of
3 service. So that was a change that was made during West
4 Valley and is something that we're looking at incorporating
5 into the WTP design.

6 On the melter plenum, the way they operate, because
7 of the way the feed is dropped onto the melter glass pool,
8 you actually get pressure surges and spikes in operation.
9 One of the things that was put in that was actually helpful
10 in terms of controlling the overall glass pour in the
11 pressure system was to put a quick reaction valve that
12 basically vented and allowed pressure to equilibrate in your
13 plenum area. So that actually was very helpful and actually
14 basically made operations much more stable.

15 During start-up we saw what we call basically angel
16 hair formation, and that was an excessive accumulation of
17 very thin glass fibers in the pour area. And the challenge
18 with that is, what happens with those glass fibers is they
19 actually get in the way of the pour stream and can basically
20 obstruct your pouring. That was actually solved with the
21 addition of an orifice that actually restricted air flow past
22 your glass pour. What was actually happening is, your air
23 was coming up and it was actually taking small fibers off
24 your pour and forming basically what looked like fiberglass,
25 and it was getting in the way of the glass stream.

1 In the off-gas system between the melter and the
2 plenum, we were seeing basically salts that were volatile
3 from the melt actually accumulate in the off-gas jumper, and
4 that was restricting your off-gas flow. One of the ways to
5 solve that was basically to inject water into your melter
6 plenum, which is operating at about 600 to 700C, and it would
7 volatilize obviously into steam, and that would flush
8 basically your off-gas jumper. So, because the salts were
9 water soluble, they'd end up in your SBS, and basically that
10 was a routine maintenance activity we did to keep the melter
11 operations stable also.

12 During melter start-up we had at West Valley an
13 initial melter dam failure. The melter dam is basically a
14 plate that holds your glass back from your pour area, and
15 it's incorporated into your refractory. And two things
16 happened there. During transposition of drawings, actually
17 two welds were actually left off, and they were not caught.
18 Obviously that would cause problems. And during the
19 expansion of the dam during your thermal heat-up, what
20 happened was that they basically popped; the welds that were
21 in place basically popped.

22 In addition to that, after reexamining the thermal
23 heat-up curves and the strategy for that, it was found out
24 that those curves were too quick, that they weren't allowing
25 for enough soak time to basically allow the surrounding

1 refractoring metal to basically come to equilibrium. So
2 during that heat-up you had basically undue stresses in the
3 wrong areas.

4 And those are some of the significant lessons
5 learned that happened at West Valley that are being
6 translated into the design and operations of the WTP high-
7 level waste melter.

8 LUTZE: Thank you very much. We're obviously much
9 faster than expected, and we'll therefore have more time for
10 discussion, which is great.

11 Our next presenter will then be Jonathan Bricker
12 right here.

13 I think you are a chemical engineer by training
14 with fluid dynamics as a specialty. You are now with DWPF,
15 and you are in charge of continuous improvements and advance
16 in the technology of (inaudible) and the process--

17 SPEAKER: Could you speak into the microphone?

18 LUTZE: --and the process itself. Yes. So I would like
19 you to make your presentation.

20 BRICKER: Thank you, Werner.

21 So this morning we heard an overview of the
22 Department of Energy Office of Environmental Management, as
23 well as an overview of vitrification as a management practice
24 for the high-level waste. What I'd like to do this morning
25 is to take the next ten minutes to talk a little bit more

1 specifically about the progress of the high-level waste
2 program at the Defense Waste Processing Facility. So this
3 presentation is a little bit different than the talks that
4 you heard this morning, that you'll likely hear this
5 afternoon, and it's less technical. And it provides a little
6 bit of an operational experience to today's talks.

7 And so, with that, since I come from an operating
8 facility, we usually begin these types of presentations with
9 a safety message. I'd like to do that since we're running a
10 little bit ahead of schedule.

11 So a lot of you are following the tragedy at the
12 Boston Marathon yesterday morning. And I woke up this
13 morning, I was watching the news, and I was watching a news
14 conference by the chief of police for Boston. And one of the
15 things that caught me or struck me is that he mentioned the
16 importance of the see-something-say-something mantra. I don't
17 know if any of you have heard of that. Well, that's
18 something that we use at the Savannah River Site and
19 something that we really care about. And if you ever have
20 the opportunity to visit the Savannah River Site, you'll see
21 that sign everywhere. So if something is out of place, it
22 probably is. So I encourage you to continue to use that in
23 the workplace as well as outside of the workplace.

24 So what we'll do today is, I'll provide you with
25 progress to date for the Defense Waste Processing Facility;

1 we'll provide an overview of the Defense Waste Processing
2 Facility for those of you unaware; and then we'll talk
3 briefly about some of the recent improvements made over the
4 last five years. We'll talk about future challenges for us
5 and work ongoing to face those challenges, and then I'll end
6 by talking about some lessons learned over our 18 years of
7 experience.

8 So currently progress at the Defense Waste
9 Processing Facility, we've processed as of last month 4
10 million gallons of high-level waste, and that translates to
11 about 14 million pounds of glass produced, representing 50
12 million curies, and that's out of an estimated 150 or 155
13 million curies associated with sludge waste. Note that we
14 have and we will continue to process by-products from salt
15 waste processing. These numbers do not account for that. We
16 have also produced 3,600 canisters. Actually, I checked this
17 morning, and we're at 3,618, for those of you keeping track.
18 And that's out of about 7,500 planned over the life cycle of
19 the facility. So that's just to give you an idea of where
20 we're at today.

21 Regarding the graph, here I show the number of
22 canisters produced as a function of fiscal year for the last
23 five years. Of importance to note is, Carol mentioned the
24 installation of bubblers to our melter in September of 2010.
25 So, prior to that, we were averaging about 200 canisters per

1 year. Since then, you can see the increase in production to
2 264 in fiscal year '11, 275 in fiscal year '12, which was a
3 production record for the facility. During that same fiscal
4 year we saw a single-month record of 30 canisters produced in
5 December. We also saw a twelve-month rolling average of 337
6 canisters, just to give you an idea of the capability the
7 facility has.

8 What's even more impressive is that not only are we
9 producing more canisters, but putting more waste into each
10 canister. As you can see, the waste loading trends here in
11 red. This graph, I think, in part speaks to the
12 environmental risk reductions, but it does not tell the whole
13 story. So one of the things that we mentioned is that we've
14 filled half our canisters, but we're at about a third of
15 dispositioning the curie count; right? One thing I'd like to
16 point out is, over the last four years of production at
17 Defense Waste Processing Facility, 25 million of the 50
18 million curies has been dispositioned, so we're starting to
19 attack that higher-risk material.

20 We're currently processing Sludge Batch 7. Sludge
21 Batch 8 will start in a few weeks here in May. There are 18
22 batches planned. And we will talk a little bit about, as I
23 go through the presentation, that production performance for
24 fiscal year '13 currently does not meet targets. And we'll
25 talk about that, and I'll roll some of that into some of the

1 lessons learned.

2 So, for those of you unfamiliar with our process, I
3 thought I'd take a few moments just to go through the process
4 very quickly. The process really starts in the tank farm in
5 the batch preparation and batch qualification portion. We
6 prepare material in one-million-gallon prep tanks. We
7 qualify the material in one-million-gallon feed tanks at the
8 Defense Waste Processing Facility. That qualification
9 process includes both simulant as well as real waste testing,
10 and really what we're getting at here is to determine the
11 processing windows with which DWPF can process within and
12 some of the additives, which we'll talk about the acids in
13 the frit additions.

14 The Facility receives sludge from the tank farm in
15 a batch process, so our sludge receipt and adjustment tank is
16 a 12-thousand-gallon tank, so nominally we take over about
17 7 to 8,000 gallons of sludge each batch. We also receive, as
18 I mentioned earlier, by-products from salt waste processing
19 that currently goes through the actinide removal process and
20 the modular caustic side solvent extraction unit, so we
21 receive a solids-rich stream from the actinide removal
22 product. We also receive a dilute nitric acid stream, which
23 contains cesium from the modular caustic side solvent
24 extraction unit. Those are added during the sludge receipt
25 and adjustment tank. The material is then adjusted with

1 acids, specifically nitric and formic acids. Carol talked
2 about some of the reasons with which we add those acids for
3 mercury reduction, manganese reduction, REDOX control.

4 Once we've added the acids, we then go through a
5 concentration step and a reflux step to be able to remove the
6 mercury through a steam stripping process. That material is
7 then moved on to the slurry mix evaporator. The sole
8 function of the slurry mix evaporator, just as its name
9 implies, is to be able to add frit and to concentrate. We
10 add frit in two different processes. Bill mentioned the
11 decontamination process at his facility. At our facility
12 what we do is we essentially sand blast the canisters, and so
13 we use a dilute slurry with frit. We recycle the frit, add
14 that to the SME, we blow off the water, then we also make up
15 the rest of it through a process of frit additions.

16 And then Carol mentioned in her talk, this is
17 really the crux of the facility, in which the slurry mix
18 evaporator is the hold point where we make sure that the
19 glass is acceptable for transfer to the melter feed tank.
20 The melter feed tank is a transition in the process from a
21 batch process to a continuous process as the melter feed tank
22 continuously feeds the melter. The melter, of course is a
23 joule-heated melter. We added bubblers again in September of
24 2010, which has drastically improved the throughput at the
25 Defense Waste Processing Facility.

1 And then what's not shown on here and what I won't
2 talk about today is canister handling, which is downstream.
3 That's not a limiting factor for us. And then something I
4 will talk about a little bit that's not represented here is
5 the recycle that's sent back to the tank farms, some of the
6 challenges that presents to us in the liquid waste
7 organization flowsheet as a whole. But, again, this process
8 works to produce highly durable borosilicate waste form.

9 I'd like to talk a little bit about some of the
10 improvements over the last couple of years. Really, we've
11 made extensive improvements to increase waste throughput.
12 We've worked really hard over the last couple of years to
13 address, really, two things, one of which is the new
14 processing demands producing at a higher rate; the other is
15 the new waste streams. So in 2007 we started receiving
16 by-products from salt waste processing, so trying to manage
17 all those new constraints.

18 I'd like, if I could, to skip to the second bullet
19 here, melter bubbler installation to increase the melt rate.
20 So, prior to the installation of the melter bubblers, the
21 capacity of the DWPF melter as well as DWPF batch prep was
22 similar. In fact, actually, the DWPF melter was the rate-
23 limiting step. Upon installation of the melter bubblers, you
24 can see now we have the rate-limiting step, the burden of the
25 facility moving from the melter to the batch prep. And

1 that's where a lot of our focus has been over the last couple
2 of years.

3 And one point I'd like to make that I don't think
4 was made earlier is, really, within the last five years the
5 only major change or alteration we've made to our flowsheet
6 is the melter bubbler installation. But even that, in and of
7 itself, was a relatively ingenious improvement in that we
8 retrofitted the existing melter with the bubbler, so there
9 was a lot of good work done there.

10 The other thing that we've worked really
11 extensively on for the recent improvements to the batch prep
12 is looking at reduction in cycle time, this particular step.
13 And a lot of what we'll see here is, we spent a lot of time
14 with the analytical improvements--I think Carol hit upon this
15 in her talk--in that it is very time-intensive to do the
16 analytical portion, so that's something we looked very hard
17 at.

18 The other thing is--something I want to point out
19 is, remember, it's not all about cycle time. It's a lot
20 about throughput. So another thing that we're looking at
21 very carefully is maximizing tank volumes, maximizing the
22 amount of material that's produced, irrespective of the cycle
23 time, for each batch. So that's something we're looking very
24 closely at.

25 And then something I mentioned in the previous

1 slide or two slides ago is a lot of work we've been doing to
2 increase the waste loading in our canisters. There's a lot
3 of work along the lines of tailoring the frit to each
4 specific sludge batch, and we can do this because of our
5 qualification program. Remember, we're qualifying these
6 batches in a million-gallon tank, and so this allows DWPF,
7 based upon that one qualification effort, to process for 12
8 to 18 months. So, really, the goal here is to maximize waste
9 throughput to reduce environmental risk.

10 So some challenges and future work, of course, I
11 think there was a question earlier, there is a growing need
12 to provide flexibility to accommodate variability in the SRR
13 System Plan. This increases our ability to address things
14 like waste feed compositions, differing waste feed
15 compositions, as well as input streams.

16 A couple of things that we're interested in doing
17 here is, one is really, as we go through the 18 years of
18 processing, we've failed to fully understand, I think, our
19 operating windows. We'd like to better understand those.
20 And then once we understand, then to be able to expand those
21 operating windows to create more flexibility in the system.

22 A couple of examples of that are an alternate
23 reductant project that we're working on. That's a project
24 that we've been working on for a couple of years with several
25 different R&D entities. Specifically, what we're looking at

1 is--I mentioned that we add formic acid to our system. The
2 problem with formic acid is that it decomposes catalytically
3 and produces catalytic hydrogen generation. So that's
4 something we had to deal with in our off-gas system. And
5 that really creates a small processing window in terms of the
6 amount of acid that you can add. So what we're looking at is
7 the ability to replace the formic acid with some other
8 reductant that doesn't reduce--or doesn't result in catalytic
9 hydrogen generation. So that's one example.

10 The other thing that I'll mention in the lessons
11 learned is that volume management is very important to us.
12 What you don't want to do is operate a vitrification facility
13 as an evaporator. A lot of water and a lot of time is spent
14 in our process getting rid of the water. So one of the other
15 things, as an example, just to open up these operating
16 windows is the dry frit project where currently the frit is
17 fed as a slurry, so we're looking at feeding it as a dry
18 material to reduce some of that water.

19 The other thing that we're working on is also
20 addressing the operating windows for future waste
21 compositions in terms of the glass formulation and then
22 addressing the demand for higher processing equipment
23 reliability, and this is due to the constraints I mentioned
24 earlier. And, really, the goal here is to position the
25 facility for continuous success.

1 And then, lastly, I want to end up with some
2 lessons learned, and hopefully these were apparent throughout
3 the talk; but, really, Carol talked to some of these. The
4 first two, the efficiency of the waste qualification program
5 and the success of the statistical process control, that kind
6 of goes into the analytical piece, which can be very
7 cumbersome if you're not careful. We also talked a little
8 bit about earlier the ability for us to perform hands-on work
9 on failed equipment. As we're an aging facility, that
10 becomes more and more important. And then our interaction
11 with our R&D facilities, that continuing interaction, to be
12 able to help us with short-term as well as long-term
13 improvements.

14 And then, lastly, there is--again, as we're an
15 aging facility, continuous improvement is something that's
16 very, very important. We're constantly looking at that to be
17 able to accommodate change in the SRR System Plan. And then,
18 lastly, I mentioned the importance of volume management in a
19 vitrification facility. Thank you.

20 LUTZE: Thank you very much for this presentation.

21 And we are coming to the last one in this session;
22 and, as I said, we will have lots of time to discuss. And
23 the last presentation is made by Albert Kruger, and his
24 background is in physical chemistry and material science.
25 And he worked for several companies in this country,

1 including Saint-Gobain in France, the glass company. And
2 today he is with the Office of Federal Protection, in charge
3 of everything that has the word "glass" to it, we could say.
4 And not only that, but with various aspects of the WTP.

5 So, please, your presentation.

6 KRUGER: Thank you. And welcome to Hanford. I used to
7 be with the Engineering Division as the Acting Director, and
8 now I'm a Glass Scientist, thanks to the Secretary of Energy.
9 So I have only one job, and that is the glass.

10 So I've got quite a number of slides here. I'll
11 leave the electronic copy. Rather than having produced hard
12 copies, the electronic copy is available. There is a lot
13 there for you in terms of the history of the site and how we
14 got to where we are. So I would ask, rather than having
15 generated the paper, you pick up the (inaudible).

16 (Pause.)

17 So a quick overview of my outline is that later you
18 will hear about a lot of the work that's come out of my
19 office since I joined the Department of Energy in 2007. One
20 of the things that you won't hear a lot about (inaudible) was
21 the introduction of the bubblers into the DWPF melter. Back
22 in 2008 I was told by one of Ken Picha's predecessors that I
23 would do something of value for one of the other sites. And
24 so with the help of Ian Pegg at The Catholic University, we
25 pursued adding the bubbler into the DWPF melter. That

1 success is clearly evident by the increase in throughput, and
2 it's quite logical to understand that when you have highly
3 refractory materials and a limited amount of flux and only a
4 certain amount of joules that you can drop or currents you
5 can drop between electrodes, anything you can do for heat
6 distribution is going to help you (inaudible). And, indeed,
7 that was borne out by the recent increases in production.

8 So, with that, we'll begin to discuss what it is
9 that we're doing here at Hanford, where the program has taken
10 off in 2007 when I joined. You'll hear later from John
11 Vienna. You'll hear from a variety of speakers and poster
12 presenters as to the work that's funded from my office into
13 the National Lab here in town, as well as to Catholic
14 University, the Vitreous State Laboratory, where they were
15 the recipients of the government deciding that a scale melter
16 was an important thing. And so the DM 1200 at Catholic
17 University is a one-third scale melter with the prototypic
18 off-gas system built in. So we can really do experiments
19 that are very meaningful and really help us understand in
20 what we'll do in operating our facility. In years past there
21 was a DM 3300, which was the equivalent on the LAW side, and
22 it's since been retired and is very likely part of
23 shipbuilding in Korea, the steel having been sold off.

24 The background of the Hanford site, unlike the
25 French experience, unlike many of the other experiences,

1 Hanford had nine reactors, they had four different fuel
2 reprocessing flowsheets, 100,000 metric tons. As we get into
3 it later, you will see that there are differences in canister
4 counts from the facility. The license application for Yucca
5 Mountain limited us to a certain allotment of the capacity at
6 Yucca. The high-level defense waste glass was going to have
7 a certain portion. We were allotted roughly 9,700 canisters
8 going into the facility, and the assumption was that there
9 was half a metric ton of heavy metal for each one of those
10 canisters. Those may or may not be a reality, but those were
11 the imposition of requirements from Yucca Mountain.

12 So when you see 9,700, that's from (inaudible).
13 When you see other such numbers, those are from projections
14 based on modeling efforts either by the BNI folks, who use a
15 G2 model, which is very different than the tank farm folks
16 and the system plan using H2's model, different constraints,
17 different assumptions. And so, as an example, you may see
18 numbers for the WTP in the nearly 20,000, and that's based on
19 taking the minimum waste loading that BNI achieved and
20 projecting it out through the entire mission. So those kinds
21 of subtleties may give rise to great confusion as one
22 considers.

23 So that is our problem. We have a lot of aluminum,
24 because we used aluminum clad. We don't have the problems of
25 Savannah River, because we didn't use a mercury stripping

1 process. And so those are some of the big differences in our
2 base chemistry.

3 Cesium/strontium was taken out for beneficial use.
4 Unfortunately, there were a couple of leaky wells, and those
5 canisters for irradiation were brought back, and they now sit
6 in WESF decaying. The plant does have the capability, should
7 it be decided that vitrification of that cesium is a good
8 thing, that in the pretreatment facility there are six stub-
9 ins on the north wall that would allow for (inaudible)
10 facility (inaudible) the capsules, solvating the cesium and
11 strontium, which are halite-based--they're chloride salts--
12 and so you would have to drain those into your process very
13 slowly because of the halites. The halites are not
14 particularly desirable in an off-gas, and they certainly
15 aren't desirable to the lifetime of the pipe from the
16 refractory (inaudible) melting. So, typically, we get a
17 safety question as to what would the impact to a dose be or
18 the consequence of adding minimal to none, because you can't
19 add very much of the halite at the same time that you bring
20 in the cesium and strontium.

21 The reactor as it was back in the day. The reactor
22 is the second reactor in our country after the University of
23 Chicago. It was interesting that, as an undergraduate
24 chemist, the chairman of the chemistry department was
25 actually in Chicago when the pile went critical, and one of

1 my other professors in New York was out at Brookhaven as the
2 Director of Chemistry. And I had absolutely nothing to do
3 with radioactive materials or an interest in it until I came
4 here to Hanford. It was mentioned that I worked for Saint-
5 Gobain. I subsequently learned that there was a branch of
6 Saint-Gobain called Saint-Gobain Nucleaire; and when the
7 French decided they were going to vitrify their waste, they
8 simply took a piece of the company for which I worked and
9 said, "You're now in the nuclear waste vitrification."

10 From that we--you've seen these kinds of slides
11 before, anywhere from 53 to 56 million gallons here, about
12 176 million curies of radioactivity. Again, the difference
13 between Hanford and Savannah River is, we took out roughly
14 half of the activity and have them in capsules.

15 We have a mix of double-shell and single-shell
16 tanks. Here is, I believe, S farm being constructed way back
17 when, and those are single-shell tanks. This, I believe, to
18 be the AX farm, based on the evaporator in the background
19 there, and that's one of the double-shell tanks under
20 construction.

21 One of the biggest challenges that the contractors
22 have and the Department has to live with is the ability to
23 mobilize the solids, the sludges and the salts that are in
24 these tanks, and to deliver them for treatment. Right now,
25 unfortunately, the operating contractor in the tank farm is

1 limited to about a 7 weight percent slurry. The first
2 receipt vessel within the high-level waste treatment facility
3 was actually spec'd out to have a 15 weight percent slurry
4 with the glass formers added. And so there's a challenge to
5 begin to mobilize the materials and pump it and deliver it to
6 the facility that needs to be worked on.

7 Groundbreaking for the WTP, there's a picture. I
8 happened to be out there on the day on which the power shovel
9 went out back over a decade ago, and so I snapped a picture
10 of them breaking ground for the WTP. And you've seen the
11 difference since the groundbreaking. That did need a
12 requirement to start construction by having that power shovel
13 out there.

14 Some of the initial--or some of the troubles you
15 hear about, the secretarial team with Milt Levenson and Tom
16 Hunter and Monica, who is here, they began to address some of
17 the rumblings of troubles in terms of mixing. There were a
18 concept that was brought over from England with black cell.
19 The English black cell, however, has access ports--ours
20 didn't--and what the BNFL left us with. These are some of
21 the initial receipt vessels, and the English also left us
22 with the pulse jet mixer. Those are currently viewed as
23 troublesome and problematic in deciding can you operate the
24 pretreatment facility without access, without maintenance,
25 without anything with pulse jet mixers and these black cells

1 as part of the baseline.

2 These are some of the vessels going in during
3 construction, 375,000-gallon capacity. There are four of
4 them as the head end of the facility.

5 The baseline. You'll see reference to the system
6 plan, Rev. 6. They say we'll produce 10,586 (inaudible)
7 won't be one canister above or below that, because they're so
8 accurately (inaudible) to five significant figures. I would
9 not do that. We do know that we have about 60 thousand
10 metric tons of sodium to treat, and we divide that between
11 the two facilities, the LAW glass and the HLW glass. The
12 current waste loadings that folks love to--when they come to
13 the Department to sell us new technologies or a new panacea
14 or a quickening of the mission or lessening of the mission
15 duration is that they will project based on the absolute
16 minimum waste loading in throughput, as defined in the
17 contract with the BNI Corporation.

18 BNI, in their contract, were given minima for waste
19 loadings of materials and performance. That was not a
20 requirement stated as a minimum and the only thing that
21 should be done; but contractors, being in business to add
22 additional business, BNI went out and succeeded in meeting
23 those minima and never really said, well, we could do better
24 without any additional funds being added. So in 2007 I was
25 brought on by the then-project director, and I was challenged

1 to do better than what the contractor had delivered to
2 demonstrate the full capability of the facility.

3 So that gave rise to the advanced glass
4 formulations work that we're doing. And the thing that's
5 most important to note is that the mission can significantly
6 demonstrate improvements without any new capital projects and
7 without any changes to the baseline design. I'm a materials
8 person. I'm a chemist. I'm not going to ask the engineers,
9 with the concrete, structural steel, and piping in place, to
10 go and change things. I'm going to use the existing
11 facility. I'm going to get better performance out of the
12 existing melter and hopefully not at the cost of accelerating
13 aging of components as we do it.

14 So the experience that we have and the opportunity
15 we have to do work at Catholic University with Ian is
16 incredibly wonderful in that we can measure these lifetimes,
17 we can look at corrosion rates, and we typically do this in
18 every set of runs that we do. We look at changes and how
19 they might impact the Inconel steels, how they might impact
20 the K3. And that's all in our reports.

21 I'll let you go through these. Performance, again,
22 enhancements through improved glass formulations are
23 transparent to the engineered facility.

24 What are our major accomplishments? We're
25 demonstrating much greater flexibility for what could

1 possibly come in the door for treatment. We are looking at
2 advances in glass science that are allowing significant
3 reductions in container and canister counts, container of the
4 LAW sides, and, most importantly, is by increasing aluminum
5 waste loading in the HLW glass. We no longer have to do
6 caustic leaching in a vessel where we then have the
7 secretary's team wondering about stress crack corrosion of
8 that very same vessel because of the temperature variations.
9 We can get more chrome, significantly more chrome, into the
10 HLW waste form than the contractor demonstrated.

11 That means we no longer have to do oxidative
12 leaching in some of the vessels. By simply removing the
13 inconvenience of the aluminum, the caustic leach, we now free
14 up pretreatment flowsheet by not having to address as much
15 as--over 90 percent of the batches of waste coming from the
16 farms would have to have caustic leaching in order to have
17 met the BNI glass models. Now it looks like that won't have
18 to be done. A third of the waste batches coming in would had
19 to have had oxidative leaching for chromium. Those can be
20 completely obviated by the new glass chemistry and
21 throughputs.

22 I'll let you go through that and some of the stuff
23 I've said. For HLW we have troubling ions, the aluminum, the
24 aluminum with sodium because of the naphthalene--you'll hear
25 more about that later--bismuth and chrome, again, crystal

1 formation and precipitation within melts. Those we've
2 addressed. Some here we're now looking at, in lieu of the
3 contractor having about a 11-to-12 weight percent aluminum
4 load in the waste form, I am well over 25 weight percent.
5 The throughput is now--by reformulating the glass in lieu of
6 having 800 or a thousand kilogram per square meter per day
7 scaling factor for the melter, we're now looking at 1,600,
8 2,000 without really stressing the rest of the system. Ian
9 has been up to over 3,000 kilogram, but there you're talking
10 about playing with additional temperature and additional
11 bubbling. Don't need to do those unless you have to.

12 And then on the LAW side, we won't talk about that.
13 As I said here, we have about three times the commissioning
14 targets now in our toolbox. Waste loading, we've been as
15 high as 55 weight percent with high aluminum waste materials.

16 Results for LAW, go right through them.
17 Robustness, things that we have to do is to revise the glass
18 models so as to demonstrate what our real mission or what a
19 different mission might look like once the new glass
20 experience is added to everyone else's experience, we
21 (inaudible) upon what they've done.

22 Again, models. Here are some visual
23 representations of where the work started as to high-
24 aluminum. Here is a progression of the aluminum increase and
25 throughputs. The understanding we get of the waste form

1 comes from PNNL. We'd like to add SRNL to our work. But, as
2 you can see, there is a very big difference between
3 formulating for glass where you have the same amount of
4 aluminum in your simulant to deal with, but in one case
5 you're very rapidly melting just by reformulating versus
6 sitting there.

7 And that was the Savannah River experience. They
8 had very refractory materials, and they couldn't get heat,
9 plus they were dealing with a certain frit. And so they
10 would have these periods of time where it appeared no melting
11 was going on and then sudden surges into the off-gas, and
12 they had a large volume of molten glass. Here we're actually
13 doing the work to really understand what are the melt
14 dynamics, what's the chemistry of the cold cap, and even
15 trying to understand how we can retain certain troubling ions
16 from LAW, as an example, by crystal growth within HLW to make
17 refractory type of mineral forms that would be inclusions in
18 the glass and not impact their performance.

19 Results for the bismuth, very rapidly go through.
20 As you can see, improvements. It wasn't so successful with
21 iron in terms of waste loading, but overall it's a 56 percent
22 improvement between the waste loading and the throughput in
23 the facility. The big message is sulfur. The contractor was
24 originally challenged at a half-weight percent sulfur in HLW
25 feed. We've demonstrated tolerances up to 2 percent. Very

1 little difference in going from 1-1/2 to 2 percent; not
2 really necessary in terms of economy of mission to go much
3 above 1-1/2.

4 Here again, the kinds of canister counts based on--
5 there used to be an anecdote that the sulfur would cause an
6 increase in the number of HLW canisters. Clearly, we have
7 demonstrated that you can get quite the opposite by
8 formulating appropriately.

9 Here is overall what our mission looks like based
10 on the glass improvements that we have been able to
11 demonstrate up through a one-third scale. Very significant
12 reductions and at least a third less canisters than others
13 may have thought.

14 And then that I'll just leave you with, because
15 Werner is here looking at me, and we can answer some
16 questions later.

17 This is a cartoon. The baseline envelope met the
18 requirements as set forth in the contract, and then the
19 Federal Office glass program became, "We know we can do much
20 better." We went about demonstrating--I like to use a hand;
21 my colleague John likes to use cauliflower, and you'll see
22 some of his presentations later. Mine, I went out and
23 demonstrated what were the tips of the fingers in terms of
24 performance overall, and now I'm filling in the webs between
25 the fingers so that the engineering and flowsheet people can

1 reconsider what feed vectors they send to the plant and know,
2 can I tolerate these different admixtures and can I make good
3 glass if I waste load it.

4 Sorry I've gone over (inaudible) back to Werner.

5 LUTZE: Thank you for the presentation.

6 Thank you for the four presentations, and we have
7 now, I would say, 20 minutes with the little bit of overrun
8 for discussion.

9 And let me just quickly mention that before the
10 meeting I had communicated with the panelists, these four and
11 the other ones, that I would encourage them to address
12 accomplishments in the past three years, accomplishments in
13 the next three years to be expected, challenges--technical,
14 political, and financial--if there are any, and to point out
15 where support would be necessary from R&D to support
16 engineering. And I have seen a lot of these issues being
17 addressed here very nicely, and that may help with the
18 discussion as well.

19 So I am opening the discussion for the Board. Rod.

20 EWING: So this panel deals with the technology; but
21 because Stéphane is on the panel, I want to jump ahead and
22 ask a question. So we have decades of, I think, very nice
23 substantive research in glass corrosion and various
24 conditions. So, just taking the glass as a primary barrier
25 to the release of radionuclides, what would be the best

1 geochemical and hydrologic environment for the disposal of
2 glass?

3 GIN: I think that after, as you said, decades of
4 research in glass, we can say that the basic mechanisms are
5 well understood. But the problem is, we don't know exactly
6 how these mechanisms are combined to give a rate to given
7 time and considering given environmental boundary conditions.
8 We know that the rate can evolve from a maximum value to a
9 minimum value. We also know what could be the final products
10 based on (inaudible) dynamics, but we don't know precisely at
11 which rate it goes from the maximum rate to the minimum rate,
12 and I would take time to go to the more stable compounds.

13 So we have models that assume some combination of
14 the basic mechanisms to give a rate, but the world community
15 of experts agree to say that the model must be improved to
16 have better provision, a better forecast of the rate,
17 depending on time and on chemical conditions.

18 Globally, I can answer to your question, saying
19 that we know that if you want a low rate--and that is a
20 requirement if we want to show that we have a good--we have
21 developed a good matrix for confining radioactivity for the
22 geological time scale. We know that a low water renewal
23 rate, a high concentration of glass former in the solution,
24 are some important conditions for allowing the rate to be
25 very, very slow, but--okay, I said that we know and we are

1 able to predict the maximum dissolution rate for a given
2 condition and given glass composition. But it doesn't look
3 very interesting for the geological disposal conditions,
4 because with a simple calculation you end up with glass
5 lifetime very short, typically one thousand years or
6 something like that. So it's not sufficient to demonstrate
7 that the glass is a good barrier.

8 To demonstrate that the glass is a good barrier, we
9 have to be in conditions where what we call the residue rate
10 can control the long-term rate, the long-term behavior of
11 glass. And for that we know now that it's better to have not
12 too alkine or too acidic groundwater and not to have a
13 renewal rate of the solution and, if possible, have a lot of
14 glass former like silicon in solution to help developing a
15 passivating layer.

16 EWING: Okay, thank you.

17 LUTZE: Thank you.

18 Yes, please?

19 CLARK: Sue Clark, Board. So I guess I wanted to try to
20 make sure I understood what I was seeing when I compare
21 between the different talks. I think I saw on Stéphane's
22 slide that the French experience you've produced a lot more
23 glass and a lot more radioactivity in that glass compared to
24 the kinds of things we were seeing at Savannah River where
25 maximum waste glass loading was about 38 percent. And so can

1 you comment?

2 GIN: We have not produced more in terms of amount of
3 glass. It's quite the same. The total figure I've mentioned
4 is about 7,000 tons of glass already produced in France and
5 similar number in the U.S., combining the different--summing
6 the different sites. But the amount of radioactivity
7 confined--because we have more fission products, because the
8 burnup of fuel are higher in commercial fuel than in defense
9 fuel, we have put in glass more radionuclides. But the waste
10 loading are not higher in France; they are lower. It's just
11 because it's more concentrated in fission products than here
12 because the burnup (inaudible) are different.

13 CLARK: Great. And so that was my other question. Does
14 waste loading equal radionuclides (inaudible)?

15 GIN: No, no, because (inaudible) this number of the
16 fuel burnup. That is very different from a commercial fuel.

17 KRUGER: Another big difference are the amounts of noble
18 metals. In commercial fuel there is a significant amount of
19 noble metals; in our defense fuels there aren't.

20 GIN: And I have not given the figure, but we've put
21 about three weight percent of noble metals in our glass.

22 LUTZE: I would like to raise a question to the panel
23 that came up during the previous talk, and that was: We have
24 learned so much about glass since 1980 when we formulated,
25 for instance, the waste acceptance product specifications and

1 so on. Wouldn't it be helpful if we raised the bar a little
2 bit and made the acceptance criteria a little stricter? And
3 I say that because of the acceptance criteria I used to model
4 the performance of the glass in the repository. So if we
5 offer a material that is worse for modeling than the one we
6 can make, then that is not very helpful for those who do the
7 modeling for the repository.

8 Could someone comment on it here?

9 KRUGER: I would simply state that--Albert Kruger, ORP.
10 That is a question that came up with the secretary's
11 presentation was taking credit. And I know Monica has a
12 large interest in actual looking at durability of the glasses
13 and taking credit for how durable they really are. In the
14 LA, when it was prepared, it was a performance assessment
15 that was based on a single test, the product consistency
16 test, the PCT test. And if your glass could perform better
17 than the environmental assessment glass in that test, that is
18 pretty much the sole criterion for performance.

19 Do the glasses actually perform better than that?
20 Yes, they do. The glasses with the high-weight waste loading
21 that I report, typically we don't approach the trip points,
22 if you will, for the EA glass by anything closer than an
23 order of magnitude. In those cases it's two orders of
24 magnitude. But, yet, there is a certain comfort in assuming
25 immediately availability of the nucleides in a performance

1 assessment to bound the risk to the public.

2 HAMEL: Yeah, if I can pop in on this question, I think,
3 you know, in terms of raising the criteria, I think the glass
4 is being challenged to perform at the level the glass is
5 being demonstrated at. Raising the criteria would, I think,
6 in my opinion, simply reduce the margin between your standard
7 and the performance. The performance is fairly high, as
8 we've seen from Albert's glasses. I don't know if raising
9 your criteria has any benefit to raising the performance of
10 the glass, as is the people that are working the glass such
11 as Albert, are striving for high waste loading and high
12 performance. I don't know if raising the standard gets you
13 any benefit, you know, to answer that question.

14 LUTZE: Okay. Yes, please.

15 FRANKEL: Jerry Frankel, Board. So I'm a little
16 confused, though, about this. So, at least for Yucca
17 Mountain, it's my understanding that the TSPA took no credit
18 for the waste form. And I think based on the assumption that
19 the environment that would make its way through the canister
20 would be extremely aggressive to the glass; is that correct?

21 KRUGER: In the Yucca Mountain model the canister was a
22 convenience for transportation. There was no credit given to
23 the availability of nuclides for diffusion through the
24 environment. So you had this stainless steel container, but
25 it in no way impeded diffusion of the material within the

1 canister.

2 FRANKEL: No, it was an Alloy 22 canister that was, you
3 know, where the credit was for the lifetime. But then once
4 that was perforated, the glass--there was no credit taken.
5 But--

6 SPEAKER: Perhaps Peter should answer that question.

7 SWIFT: Would you like me to?

8 LUTZE: Go ahead, please.

9 SWIFT: Peter Swift, Sandia National Labs. And I'm not
10 a glass person at all. I have worked in repository
11 performance, though. And, actually, Stéphane's answer
12 earlier was just great on that one, that if you're driven
13 to--and we are, of course, (inaudible) analysis considers the
14 range of glass performance. And if you're driven towards the
15 rapid degradation end of that, you end up with, as you said,
16 something on the order of a few thousand years for your glass
17 lifetime.

18 That's actually what happened on the Yucca Mountain
19 Project. We did take credit for the glass, but the best we
20 could do for the distribution that spanned--fractional
21 dissolution rate and surface area were the two important
22 variables in it. And the overall performance was dominated
23 by the tail of distributions that had the most rapid
24 degradation of the glass, lifetimes on the order of a few
25 thousand years.

1 So in a million-year analysis, that does, in fact,
2 look like no credit. But it was not because we chose to take
3 no credit. It's where the underlying information took us.

4 KRUGER: The glass that was used in those modeling, was
5 it ground glass?

6 SPEAKER: Huh?

7 KRUGER: The glass samples that were used in that
8 modeling were results of ground glass traditionally.

9 SWIFT: You'll have to get somebody else to answer that.
10 Go back to John Vienna (inaudible).

11 KRUGER: Right. And that's why I said with no credit,
12 because if you take a monolith and you grind it into
13 hundreds-of-square-meter-per-grams surface area, that's a
14 very different dissolution experiment than taking a monolith
15 and exposing it to the same conditions.

16 SWIFT: I don't actually know the answer to that
17 question. The work there was done by Jim Kinney and Bill
18 Ebert at Argonne.

19 FRANKEL: Just to clarify, you're saying that the
20 module--the waste form degradation module in the TSPA was
21 overly conservative? Is that the point that you're trying to
22 make, that we shouldn't be considering the possibility of
23 very fast degradation but the possibility of very slow
24 degradation? Can you clarify that?

25 KRUGER: I think that it's reasonable to understand what

1 the other end of the extreme is in terms of how will the
2 glass perform in that environment. But from the standpoint
3 of public safety, to say that it's degrading as rapidly as
4 possible, that's a very easy point to defend, that I've taken
5 the least performing aspect of my disposal facility, and
6 here's the consequence to the public. So, from that
7 standpoint, having the most rapid degradation is allowing you
8 to make those statements: I'm not exceeding clean drinking
9 water standards; I'm not going past certain other set points.

10 LUTZE: Rod.

11 EWING: Just to join the discussion, I think Peter would
12 agree with me that one of the lessons from the Yucca Mountain
13 Project is that realistic assessments of the performance of
14 the different barriers is really the way to go. Am I safe up
15 until that point?

16 SWIFT: Sure.

17 EWING: Okay, thank you. I'm sure we'll diverge
18 shortly, but--

19 SWIFT: Peter Swift, Sandia National Labs, again. A
20 realistic treatment of the full range of uncertainty. I
21 don't advocate trying to go for a single deterministic,
22 realistic result; rather, doing an analysis that incorporates
23 a realistic assessment of your uncertainty.

24 LUTZE: Yeah, it wasn't my intention to focus the
25 discussion on all of this here, so maybe if we have other

1 questions from--yes, please.

2 PEDDICORD: Lee Peddicord from the Board to Stéphane,
3 two questions. You mentioned that you shut down (inaudible)
4 at Marcoule. Do you intend to put in place some capability
5 then that--

6 GIN: In Marcoule?

7 PEDDICORD: Yes.

8 GIN: No, because in Marcoule we had three reactors for
9 the plutonium production and one reprocessing facility. And
10 all this stuff are stopped for a long time, and we are
11 cleaning up the site. And all the high-level waste arising
12 from the reprocessing facility have been vitrified, so it's
13 finished.

14 PEDDICORD: So--

15 GIN: And the remaining (inaudible) that could be
16 generated by the decommissioning of the plant will be sent to
17 la Hague for vitrification.

18 PEDDICORD: Then with respect to la Hague and as you
19 look at trends in higher burnup of the commercial fuel, is
20 that going to impact the way you are processing at la Hague
21 and as you develop (inaudible) capability there?

22 GIN: It impacts a little the glass composition. And
23 there is a request from the safety authority to demonstrate
24 the behavior of glass with respect to self-radiation
25 (inaudible). Because of the high amount of fission products

1 in minor and long-life minor (inaudible), we have to
2 demonstrate that the properties of the glass would be
3 maintained over time due to the self-radiation.

4 LUTZE: Yes, please.

5 FOUFOULA: Efi Foufoula, Board. I have a question for
6 Jonathan. So you emphasized (inaudible) future work the
7 growing need to provide flexibility (inaudible) windows and
8 (inaudible) to put this into perspective with the black cell
9 (inaudible). Is it fair to basically try to make sense of
10 the need for flexibility (inaudible)? That's just the
11 concept of (inaudible) flexibility.

12 BRICKER: I think, really, what I was trying to portray
13 there was something a little different. And, really, what I
14 was trying to address is the changes in compositions in the
15 feed streams that come through the facility, not necessarily
16 the components that make up the facility. So I was trying to
17 address something a little bit different there. We've seen a
18 lot of changes in just the different material that comes
19 through the facility, so we're looking at flowsheet changes
20 that don't necessarily impact or require us to change any
21 equipment out. So I was getting at something a little
22 different.

23 FOUFOULA: Sure, I understand. But (inaudible) because
24 this (inaudible) and what comes into the facility, it is
25 something that (inaudible) or accept that it will have some

1 possibility.

2 BRICKER: Yeah, I mean, the only thing I can comment
3 there is that, you know, we are--because we're an aging
4 facility, we are putting more focus on maintenance programs,
5 (inaudible) liability. But even in terms of--you know, we
6 mentioned the bubblers earlier. We had an option to either
7 change out the melter or retrofit the existing melter, and
8 that's what we did. So other than placing more focus on
9 those types of things--and there is something to be said for
10 our ability to use our decontamination cells and actually put
11 hands on the equipment. I think that's very important.

12 LUTZE: Any other questions? Please.

13 TURINSKY: Paul Turinsky from the Board. It seems that
14 Hanford is different in the sense that they used four
15 different reprocessing technologies over the years, which
16 gives you different waste streams. Is the idea of
17 reprocessing that, by the time it gets to the mixer, it all
18 looks the same? Or are the glasses--what's going into
19 basically vitrification going to be substantially different,
20 and are we going to have a range of glass performance then?

21 KRUGER: We will have--if the PCT is the single test
22 against which we measure performance, we will always have a
23 glass that exceeds that requirement. Though there were four
24 flowsheets for reprocessing fuel, there was certainly
25 absolutely no concept that I can discover in the manner in

1 which the wastes were managed in the farms. Things were
2 moved, things were sent to evaporators to make space, waste
3 put back in tanks, tank space availability. Yet, at the
4 other end of the spectrum, we enjoy a certain luxury in that
5 there were a certain number of tanks where they were filled
6 once from a specific facility, and then all the piping was
7 cut. Those are maybe five or ten of those small tanks.

8 But within the liquid, within the tank farms, the
9 underground tank storage, it's a cauldron, it's a witch's
10 brew of what's in there. Based on some of the estimates that
11 were originally given over history, if it didn't go out as a
12 product, then whatever was purchased is somewhere in one of
13 the tanks, in addition to rocks, two-by-fours, bricks,
14 cleanup efforts.

15 And so the glass models that we wish to develop
16 will allow the flexibility that, as waste is staged for
17 treatment, we know in advance roughly where we'll be
18 operating the facility, whether it'll be a high bismuth,
19 whether it's likely to have a high bismuth, an aluminum
20 component coming.

21 So the analytical scheme along the way allows for
22 staging waste months before it would ever be transferred,
23 determining whether it's an appropriate waste as it sits to
24 come to the plant, and then, as it moves out into HLW, as an
25 example, the three-, four-thousand-gallon batch that hits

1 that first vessel would have four samples taken. It would be
2 determined from those what needs to be added in terms of
3 glass formers. Then, once that's mixed, up to eight samples
4 are taken, glasses are fused, and you determine whether or
5 not you have an acceptable level of confidence that what the
6 rest of the batch will produce will be represented by those
7 eight--no more than eight samples.

8 So as we gain process experience, we can back off,
9 as long as we maintain at a very high level of confidence for
10 the process (inaudible) control.

11 LUTZE: Any other questions? Yes, please.

12 BRANTLEY: Sue Brantley, the Board. Again, to Albert, I
13 guess I was confused in your talk. You had accomplishments
14 so far, and then you talked about that. Were those
15 accomplishments in developing a model of glass that would
16 allow you to do this, or was it laboratory experiments to
17 making the glass? I didn't quite--

18 LUTZE: Predictive models.

19 KRUGER: Right. What we do is we begin with crucible
20 scale. And then as we demonstrate success in a crucible
21 scale, ultimately ends up at VSL with multi-day campaigns.
22 The sulfur work is an example where I showed we had high
23 sulfur with a number of the troubling--iron with chrome and
24 some of the other troubling cations. That was a metric ton
25 of feed as an initial experiment. That will then translate

1 into repetitive multi-day campaigns in the DM 1200 to confirm
2 what we've done.

3 Once all of those formulation and property
4 parameters are collected and verified and reproduced, along
5 with work that's done at PNNL, we then developed the glass
6 models. Part of our problem to date was, as an example,
7 there's a nepheline discriminator. The nepheline
8 discriminator served its function in order to get a plant at
9 Savannah River operating and operating safely in terms of
10 producing a compliant glass at the end.

11 But when you actually produce glasses that by this
12 discriminator would have been tossed out, you have to ask
13 why. And the question then becomes, because in the
14 discriminator those glasses had never been produced and put
15 into that model. And so now the effort that we're going
16 through is revisiting what glasses can be produced. And, as
17 an example, for the aluminum we're talking about a thousand
18 data points that are now being added back to the model, the
19 nepheline discriminators being challenged as a method of
20 determining what a good glass is, and it may, in fact, turn
21 out to be a different model for nepheline formation that
22 would give rise to a failing PCT.

23 And so that's the work that's currently going on
24 and will continue.

25 TURINSKY: What do you mean by a model? Is that a

1 (inaudible)--

2 KRUGER: A model means that--

3 TURINSKY: --or are we doing basic--

4 KRUGER: We're doing interpolation based on actually
5 producing glasses and defining a domain in which we've done
6 work. There is no extrapolation here. Extrapolation would
7 be--

8 TURINSKY: It's neither a physics-based model either,
9 though.

10 KRUGER: No, no. It's just where have you been and have
11 you measured properties that are acceptable processing and
12 then ultimate performance, right, no extrapolation, no
13 projections.

14 LUTZE: I would like to thank the speakers. We have
15 exceeded our discussion time, and we will continue to the
16 next part of our session. If the next three speakers would
17 please come up here. Thank you again.

18 (Pause.)

19 Well, we begin the second half of this morning's
20 session. We have three speakers and the same procedure. We
21 save the questions for the end of the presentations.

22 And our first presenter is now David Peeler. He is
23 the Senior Fellow Engineer at Savannah River National
24 Laboratory.

25 And I think you have a ceramic engineering

1 background, and you are in charge of everything that has to
2 do with the glass development at Savannah River. And so
3 please--

4 PEELER: Thank you, Werner. Thank you, Panel, for
5 having me today to speak to you on the glass formulation
6 efforts for the Defense Waste Processing Facility and some of
7 the accomplishments and some of the improvements we've made
8 over the past several years to improve not only melt rate,
9 waste loading, but open waste throughput.

10 As Werner mentioned, I am David Peeler, Savannah
11 River National Lab, head of the glass formulation team that
12 supports the Defense Waste Processing Facility with respect
13 to glass formulation, improving waste loading, and improving
14 melt rate. What I'd like to do is walk through a little bit.
15 There's been a lot of discussion on melt rate, waste loading,
16 and waste throughput; and it ultimately boils down to
17 reducing the mission life for the facilities, not only at
18 Savannah River, but also at Hanford.

19 When we talk about melt rate, we're actually
20 talking about processing faster; that is, we're both liquid-
21 fed systems. We want to convert that liquid feed into a
22 glass product as soon as possible. That increases--
23 increasing melt rate produces more canisters per year, your
24 output is more canisters per year, and that ultimately
25 translates into reducing production time or reducing mission

1 cost.

2 With respect to waste loading, you're making fewer
3 DWPF canisters; that is, you're putting more waste in each
4 can. Again, that translates into lower production time and
5 mission cost, as well as reducing the number of canisters
6 that ultimately go to permanent storage.

7 But ultimately there is a compromise between melt
8 rate and waste loading that we like to talk about in terms of
9 waste throughput. And waste throughput you can think of as
10 the amount of waste you're processing per unit time; that is,
11 you want to maximize waste throughput to process the maximum
12 amount of waste per unit time through the facility. And if
13 you're doing that, you're emptying tanks faster, and you're
14 closing your mission life.

15 And I'll demonstrate that here on this slide.
16 Sorry for the very small plot. Over here I'm plotting melt
17 rate or waste loading on the x-axis, and I'll talk about the
18 y-axis on the right y-axis over here, which is melt rate.
19 And, in general, before the implementation of the glass pump
20 or the bubblers that we've talked about previously, we had a
21 general trend that for any frit/sludge combination in DWPF,
22 we would actually see a decrease in melt rate as a function
23 of increasing waste loading. And this is over an operating
24 window, and that operating window is defined based on the
25 frit/sludge system, the waste loading range over which I

1 could process and meet all the processing criteria as well as
2 the product performance criteria that Carol Jantzen mentioned
3 earlier today.

4 So, again, as you increased waste loading, we
5 generally saw a decrease in melt rate. So then you ask--you
6 have to take a step back and ask yourself, okay, do I target
7 the maximum waste loading that I can where my melt rate is
8 slow, so that would generally push mission life out, but I'm
9 producing fewer canisters, but it may have a negative impact
10 on mission life; or do I come back at a lower waste loading,
11 produce cans faster, but make more canisters? And that was
12 kind of the boundary back in, I guess, the late '90s, early
13 2000s that we wrestled with.

14 What we found is--and this is just a schematic--
15 that, depending upon the shape of that curve, you actually
16 had an inflection point there where you could maximize waste
17 throughput; that is, you could process the maximum amount of
18 waste per unit time through the facility. And that's what
19 Jonathan was talking about in terms of an operation space.
20 This is the waste loading that they would target.

21 Now, with implementation of the glass pump and the
22 bubblers, in a sense what you've done is you've flattened
23 this curve out; that is, you've made the dependence of waste
24 loading and melt rate less dependent, if you will, and you've
25 flattened the curve out, so you can actually go to target a

1 higher waste loadings and still go to more higher--or
2 maximize waste throughput more effectively.

3 But ultimately what you will wind up with--even if
4 that curve was flat, you will ultimately wind up with a
5 processing or product performance constraint, the models
6 we've been talking about, that will ultimately limit waste
7 loading that you'll be able to achieve. And, as Albert was
8 talking about, one of the missions--and we'll talk about it
9 in a few minutes --is--one of the things that we need to look
10 for in the future is expanding these operating windows to
11 improve the waste loadings over these compositional ranges
12 that are coming down (inaudible) in terms of the facility
13 mission life.

14 Three things I've listed here in terms of physical
15 or chemical changes we've done to the facility for DWPF is
16 reducing the conservatism in the process control models.
17 Carol Jantzen talked a little bit this morning about this
18 would be a frit corner, a one-sludge composition, which may
19 be actinide removal process stream coming in. This may be
20 the actual sludge coming from Tank 40 or Tank 51. And
21 ultimately you can think of the waste compositions, the two
22 different waste compositions, the frit and the waste loading,
23 defining some multi-dimensional glass base.

24 Then you start implementing your process control
25 models, and this starts cutting off regions where it won't

1 allow you to go. So you ultimately wind up in this multi-
2 dimensional space with a little sphere in the middle, if you
3 will; and in terms of processing, you want that sphere to be
4 as large as possible. And Carol talked about the
5 implementation of these little lines in here. These are the
6 models; then we add some model uncertainty and some
7 measurement uncertainty to those models. Well, if we can
8 reduce the uncertainties in those models and that sphere
9 grows, we implement a new model, that sphere can grow, that
10 means we can get the higher waste loadings.

11 The other significant change has been this shift in
12 frit development strategy. When we originally started up, we
13 had a concept that one frit fits all. We had a frit, it was
14 frit-200, and basically it was designed to handle all of the
15 sludges coming downstream for the life of the facility and at
16 lower waste loadings. And when we got incentivized to look
17 at increasing throughput or waste loading or melt rate, we
18 transitioned from this global one-frit-fits-all concept to, I
19 think, a tailor concept where we specifically design a frit
20 for each sludge batch that Jonathan was talking about
21 earlier, that million-gallon tank. So we take advantage of
22 the waste form, the potential of that million-gallon tank,
23 design the frit specifically for that tank, and that allows
24 us to go to higher melt rates as well as higher waste
25 loadings.

1 Physical additions, we talked about those earlier.
2 Again, the glass pump implementation followed by the
3 implementation of the bubblers through Energy Solutions and
4 VSL, that has dramatically increased melt rate, again,
5 basically essentially flattening this response curve or the
6 curve here so we can target higher waste loadings, maximize
7 throughput, but ultimately we're going to be limited by some
8 process control model, whether it's performance through a
9 melter or a melter-related constraint or a product
10 performance constraint to how high on waste loadings that we
11 may be able to achieve.

12 With respect to accomplishments, I kind of took you
13 back historically to where we were in the beginning. This is
14 Sludge Batches 1A, 1B, and Sludge Batch 2 where we used this
15 global one-frit-fits-all concept frit. We were nominally
16 targeting waste loadings of around 28 percent waste loading.
17 During processing of Sludge Batch 2 we had extremely low melt
18 rates. We were asked to then design a new frit to increase
19 melt rate and potentially increase waste loading. We
20 designed Frit 320. Frit 320 was implemented along with the
21 new liquid (inaudible) model that Carol talked about earlier,
22 and we started processing Sludge Batch 2 at nominally 34
23 percent waste loading. And, again, that's the sixth waste
24 loading point increase for that particular sludge batch,
25 which is a significant--translates into a significant

1 reduction in the number of canisters for that particular
2 sludge batch.

3 And, as you can see down, we designed different
4 frits for different sludge batches, again taking advantage of
5 the waste form potential, designing the frit specific for
6 that sludge batch to improve melt rate, improve waste
7 loading, and ultimately improve waste throughput.

8 And we've targeted roughly 38 percent, 40 percent
9 waste loadings. The last two or three sludge batches we've
10 nominally targeted 36. If you look back at Jonathan
11 Bricker's slides for, I think, Sludge Batch 7B, I think the
12 overall nominal is around 38 percent waste loading. And as
13 Jonathan talked about Sludge Batch 8, the frit 803 that we've
14 designed, we should start processing that particular system
15 in May of this year.

16 Future challenges--and this is my last slide--
17 there's a lot of compositional swings in our current
18 flowsheet. Jonathan talked about the sludge processing
19 coming into one side, and we had the ARP stream coming into
20 the SRAT--sorry for the four-letter words there. And one
21 thing that the facility does is it basically has an on/off
22 switch; that is, it can process sludge only for some period
23 of time, then they'll get a slug of ARP product coming in.
24 And that stresses the glass formulation side from the
25 compositional swings that are pretty severe when we're trying

1 to design a single frit that can handle those compositional
2 swings.

3 And to give you some idea, the sodium
4 concentrations from a sludge-only to a coupled operations
5 flowsheet of about 2,000 gallons of ARP is about a six-to-
6 eight-weight-percent sodium oxide shift. So, again, trying
7 to design a frit that will handle that on/off switch, along
8 with (inaudible) 36, 38 percent waste loading, along with
9 trying to build in the variation that the facility has in
10 terms of hitting that waste loading, is really becoming a
11 challenge to us. So what we're trying to raise the little
12 red flag of is, you know, try to dampen out some of these
13 compositional swings that we're seeing and start becoming a
14 more continuous process through that coupled operations
15 flowsheet.

16 We also have updating models. Again, we talked
17 about that earlier in terms of expanding the compositional
18 range and over which we're going to process in the future.
19 It is an example now. Currently we are processing the
20 MCU/ARP stream that Jonathan talked about. And from the
21 glass side, we've put a limit on what volumes that they can
22 actually process through the facility. And that's based
23 strictly on a titanium concentration, that our models have
24 been developed and validated over a certain range of
25 titanium. They asked us to look at 7,000 gallons of MCU/ARP

1 coming into the facility. We turned around and said you
2 can't go above 2, because we don't have the data to take you
3 above the 2 to get you to where you need to process 7,000
4 gallons. So now we've got a throttle on the facility based
5 on the lack of data that we have in hand on how much they can
6 actually process of secondary waste.

7 So that's going to become of significant importance
8 on trying to get ahead of the facility and trying to get the
9 data necessary to update our models to allow this throttle to
10 be pushed through the floor instead of kind of pulling it
11 back and allowing the facility to operate where they need to
12 be.

13 I think Jonathan or Carol mentioned high-level
14 waste systems plan, and this is an annual plan that SR puts
15 together that kind of looks at the future mission of the
16 facility. What I see really vital to the success of the
17 facility is getting ahead of the curve; that is, we're really
18 good at getting DWPF--once we have a sludge batch, design the
19 frits, and getting it implemented.

20 What we lack a little bit, I think, is, again,
21 getting two years out, three years out ahead of the curve,
22 having influence over that high-level waste system plan, to
23 flag the issues that are going to come up downstream and
24 allowing us, the technical guys, to get the data we need to
25 update the models so they can meet their processing

1 expectations. That is where I think we really need to work.
2 That also has a player's role in the blending process as well
3 as the washing strategy the facility uses.

4 So, with that, I thank you.

5 LUTZE: Thank you, David.

6 We move on directly to the second speaker, Dr. Ian
7 Pegg. Dr. Pegg is Director of the Vitreous State Laboratory
8 at the Catholic University of America and Professor of
9 Physics. And I think all I need to say, if you talk about
10 glass, it doesn't take long and his name comes up. So he has
11 accomplished a lot in the area of glass vitrification over
12 the years and not only in the United States, also working
13 with Japan.

14 And I suggest you just go ahead and make your
15 presentation.

16 PEGG: Thank you, Werner. Thanks to everyone for
17 coming, and thanks to the Board for the invitation to present
18 today.

19 I am going to give a very brief overview of some of
20 the activities in the vitrification field that we're involved
21 in and then move on to talk on some of the issues related to
22 waste form development and implementation, most specifically
23 relating to the Hanford Waste Treatment Plant. So let me
24 move on to the slides.

25 So the Vitreous State Lab--forgive me if I say VSL

1 after I say the name fully--was established in 1968. We're
2 currently about 80 staff. And we have, as you've heard some
3 about, fairly extensive, large-scale pilot testing
4 facilities, which I'll say a few more words about, plus all
5 of the infrastructure for the materials development and
6 characterization that goes along with glass formulation and
7 waste form development.

8 A necessary aspect of work in this area is, of
9 course, the nuclear grade quality assurance programs, and
10 somewhat unusual for a university is not only NQA-1 program,
11 but also the repository, what was the Yucca Mountain program,
12 the DOE-333P program, that we have in place and frequently
13 audited.

14 Just in the outline, over the years we developed
15 the glass formulation that was implemented at the West Valley
16 Demonstration Project that was used to convert 660,000
17 gallons of high-level waste to 275 canisters of glass
18 successfully at West Valley; a program at the Savannah River
19 site, another 660,000 gallons of mixed low-level waste
20 converted into glass that was ultimately delisted, resulting
21 in significant cost reductions in the disposal.

22 We've been providing support to the Waste Treatment
23 Plant since 1996, and I'll say more about that (inaudible) in
24 the HLW and LAW formulation area; to the Japanese program at
25 Rokkasho, support since 2005. That mostly involves things

1 like mitigation of yellow phase salt formation, increasing
2 waste loadings, and management of the very high noble metals
3 that you find in these high burnup commercial reprocessing
4 wastes.

5 And, last but not least, but since about 2009,
6 support to DWPF, working with David and Jonathan and others.
7 An example there, we just finished the support for the
8 qualification of the Sludge Batch 8 glass composition that,
9 as David mentioned, will go into service in May this year.

10 So moving on then to look at the generic question
11 of glass waste form development. And, as you've probably
12 heard from some of the forgoing talks, this really is a
13 question of materials optimization subject to a set of
14 constraints. And those constraints are defined by a number
15 of factors for the particular application, not least of which
16 is the waste composition, its variability, the engineered
17 system that you're going to be using to treat that waste.
18 That's the melter type and its characteristics.

19 And, of course, the performance requirements on the
20 glass product. So one way of looking at this is here,
21 product quality, which for the Hanford high-level waste is
22 various kinds of standard leach tests, the product
23 consistency test that you've heard about, and then on the EPA
24 side the toxicity characteristic leaching procedure is also a
25 required test.

1 Processability, this relates to characteristics of
2 running this glass melt through the particular melter, which
3 has its own engineered characteristics. And so we get into
4 things like the viscosity of the melt at high temperature,
5 the electrical conductivity of the melt at high temperature,
6 because it's heated by conducting current through the melt
7 itself. And both of those properties you need to know the
8 temperature dependence and the composition dependence of
9 those properties. You need to understand the phase behavior
10 of the melt. Fundamentally, these melters are liquid phase
11 reactors, but glass has the tendency if pushed to create
12 crystalline phases; and excessive crystalline phase formation
13 can lead to issues such as clogging the melter, preventing
14 discharge. So you need to understand those relationships.

15 And then, last but not least--we've heard some
16 about this, but probably should hear more--these are economic
17 factors, and David touched on this a few minutes ago. And
18 under that heading are all of these aspects such as waste
19 loading; that is, how much waste is packed into each kilogram
20 of glass that's produced. Obviously the more densely you can
21 pack waste into the glass, the less glass you have to make to
22 work off the pile of waste. That's clearly an economic
23 advantage to packing more waste into the glass. That's what
24 we mean by waste loading.

25 Another factor that David also touched on is how

1 fast does that waste convert into glass, and there's a number
2 of factors that come into play there. The engineered melter
3 itself, that will have its own characteristics. But there's
4 things you can do with the glass and flowsheet chemistry to
5 affect those reaction rates. Fundamentally, these are
6 chemical reactions converting the feed into glass, and by
7 judicious choice of chemistries you can make that reaction go
8 faster or slower, so you have both flowsheet and engineered
9 factors to play with there.

10 There are other factors that come into play in
11 terms of the economics, things like materials compatibility
12 that affect the lifetime of the melter, corrosion of the
13 refractories, the electrodes, the bubbler tubes, the thermal
14 welds, etc., all of which have costs for replacement and
15 maintenance.

16 So, fundamentally then, we have this optimization
17 process of looking back over history, what data do we have,
18 what correlations do we have, design formulation subject to
19 these constraints. As a first pass, go out and make these
20 glasses, characterize them, collect data, improve the
21 relationships. This is the small scale, sometimes called the
22 crucible scale testing end of things.

23 But very quickly then you have to get into testing
24 in real live melters. And some of the unique capabilities
25 that we have at the Vitreous State Lab is not only the

1 largest test melter of its kind--this is a one-third pilot
2 plant for the Hanford high-level waste melter that we saw
3 some of yesterday--but also the largest array of such test
4 platforms, five operating platforms in place. And why so
5 many and why so large? Well, the problem basically at
6 Hanford is the melters we're talking about are the largest of
7 their kind. These are the West Valley melter sizes, the DWPF
8 melter sizes, so DWPF about 2.6 square meters. The LAW
9 melter is 10 square meters, and there are two of them; the
10 HLW melter 3.75 square meters, and there are two of those.
11 These are very, very large melters, pushing the limits of the
12 technology.

13 And the only real way to understand the performance
14 ahead of time is to look at the scaling relationships. And
15 so we have in place under one roof a factor of 60 scale-up.
16 And the program to support the WTP involved two pilot plants,
17 two one-third scale pilot plants, one for the Hanford high-
18 level waste melter and one for the Hanford low-activity waste
19 melter. As Albert said earlier, this melter was run very
20 successfully for five years, made 8 million pounds of glass,
21 and was decommissioned after the fact, taken apart, examined.
22 A lot of very useful performance information off that melter.
23 Fortunately, the high-level waste melter is still in place.
24 It's been running for over ten years now and is still being
25 actively used for testing to support the WTP. So--

1 SPEAKER: (Inaudible.)

2 PEGG: Five minutes. Okay. Well, then we won't get
3 very far. All right, speed it up.

4 So the message here is integration of the glass
5 formulation materials aspects with the engineered facility.

6 We've heard a lot about these melter bubblers. And
7 I apologize for the flashing here; that shouldn't be
8 happening. But, in a nutshell, these are computer models of,
9 actually, the Hanford high-level waste melter. This is the
10 refractory, the electrodes in the wall that Bill Hamel
11 mentioned. The bottom electrode was removed based on the
12 reduction of the temperature gradient due to the bubblers.
13 But, fundamentally, the glass forming materials and the waste
14 fed onto the top of this molten glass. And you have a
15 reaction of an interface, and the rate of that reaction is
16 limited by heat and mass transfer to this cold cap region.
17 And the unbubbled system, this is a viscous fluid. We rely
18 on natural convection, and that can be the rate limiting
19 process for converting waste to glass.

20 What we came up with some years ago--a subject of a
21 series of patents--is a bubbling technology that creates
22 active mixing, increases this heat and mass transfer process,
23 and the amazing this is, really, that just that simple
24 addition can give you up to five times increase in
25 throughput. And in a nuclear facility, if you can get five

1 times the throughput from the same footprint, that can
2 translate into huge capability advantages. We've heard a lot
3 about the implementation of the retrofit into DWPF. The
4 important thing here is, this technology is in the baseline
5 for both the Hanford high-level and low-activity waste
6 melters.

7 A few words on the Hanford challenges. And one way
8 to look at this--

9 SPEAKER: (Inaudible.)

10 PEGG: Okay, thank you. I'll go as fast as I can and
11 wrap this up.

12 And I wanted to contrast the challenges at Hanford,
13 really, with the challenges, for example, in commercial waste
14 processing vitrification, the likes of which are Sellafield
15 in the U.K. or at la Hague. And this really comes down to
16 scale and complexity. The scale at Hanford is just mind-
17 boggling, and the compositional complexity just bears no
18 comparison even to Savannah River and DWPF. As we've heard,
19 the whole history of the nuclear reprocessing flowsheet
20 development is out there in the tanks at Hanford. This shows
21 you the ranges of compositions just of some of the major
22 elements in some of the Hanford waste. And just look at
23 aluminum here, from 10 percent on an oxide basis up to some
24 70 percent. And this is after pretreatment to remove
25 aluminum.

1 So waste form development here is, it's all glass,
2 but we're really talking about very different kinds of
3 glasses, the borosilicate glasses. But, for example, the
4 high aluminum glasses, we're getting 26, 27 percent aluminum.
5 It's nothing like the glasses at DWPF or in Europe.

6 The other things to note are the reprocessing
7 waste, the waste that's coming at those vitrification
8 facilities, is coming off a well-controlled reprocessing
9 flowsheet. It's acid waste, very low solids. It's very well
10 controlled, tight composition, low volume. It's just a very,
11 very different problem. This is neutralized acid waste, huge
12 amounts of sodium added, precipitated solids. You have a
13 solids slurry management issue.

14 And just to roll this up into scale then, if, for
15 example, we took the WTP HLW melter (inaudible) capacity
16 after the first melter change-out to 7.5 metric tons per day,
17 if we use the Sellafield and the French hot wall induction
18 technology, those two melters would have to be replaced by 13
19 melters. If we went to the cold crucible melting, we'd be
20 talking about 6 parallel lines to get that capacity.

21 And the (inaudible) just gets ridiculous if you
22 look at LAW. The two LAW melters at Hanford, if you put in
23 the hot wall standard, la Hague melters, you'd be talking
24 about 50 parallel lines to do that job. If you put in the
25 cold crucible melters, 23 parallel lines. We stood next to

1 the LAW melter yesterday. If you can imagine taking that out
2 and putting in 23 parallel lines, with all of the support on
3 the (inaudible) end, all of the support on the off-gas end,
4 it's just not--could it be done? Maybe. But would you
5 really want to do it? I don't know.

6 Very, very quickly then, Werner asked that I touch
7 upon some of these discussion questions. These are some of
8 my thoughts on recent accomplishments, significant
9 accomplishments in the vitrification area. We've heard a lot
10 about the bubblers. Obviously that's close to my heart. I
11 think the installation of the cold crucible on one of the
12 lines at la Hague was a very significant accomplishment.
13 (Inaudible) is still the hot wall induction technology.

14 The completion of the vitrification program
15 (inaudible) the VEK program also very significant.

16 Hot commissioning of Rokkasho, it's had its
17 problems. They did make 190 canisters of glass, hoping to
18 start up at the end of this year.

19 And then these new advanced formulations that
20 address these very high concentrations of things like
21 aluminum, iron, bismuth, phosphorus, etc., I think, are very
22 important in terms of the economics.

23 And in the near future, just a few thoughts here on
24 some of the future direction. Perhaps I should stop there
25 and try and make it another time. So thank you.

1 LUTZE: Thank you, Ian, for your presentation and for
2 keeping in time.

3 That brings us to John Vienna, the last
4 presentation. So John is Chief Scientist in the Glass
5 Development group at Pacific Northwest Laboratory, and you're
6 a materials scientist, strong materials science background.
7 And you are not only involved in the glass, but in many other
8 aspects of the WTP and the Hanford Project.

9 VIENNA: Thank you, Werner.

10 I'm going to take a little bit of a different
11 approach from the previous two talks, and I am going to
12 summarize the research being performed and the aims of the
13 research for both the Office of Environmental Management and
14 the Office of Nuclear Energy that demonstrate the overlap
15 between those; and then I'll give some examples from
16 Environmental Management.

17 So the Office of Environmental Management is in
18 charge of managing the legacy defense wastes, in general, and
19 that's where most of the focus is here today. And there is a
20 significant effort in waste form development, primarily
21 funded out of the local office here, the Office of River
22 Protection, where Albert and Bill Hamel work and so forth.

23 One of the many missions of the Office of Nuclear
24 Energy is the development of sustainable nuclear fuel cycles;
25 that's performed under Monica's office here, the Fuel Cycle

1 Technology office. And both of these perform research. And
2 with these similarities in mission, there is overlap in the
3 waste form research being done, so let me start down the list
4 of research topics.

5 The first one that you've heard a lot about is
6 improving the waste loading of tank waste in glass. And
7 that's done--and I've got here that it's done to improve
8 economics. That's absolutely true, but, as Albert pointed
9 out, there is an additional purpose for that. The additional
10 purpose is to open up opportunities for other processing
11 actions. Right now the baseline is to pretreat everything
12 by, first, aluminum leaching and, secondly, oxidative
13 leaching to remove aluminum and to remove chrome. During
14 advanced waste loading and advanced glasses would allow us to
15 do less leaching, and it may also allow us to do things like
16 feed the HLW vitrification facility or the low-activity waste
17 vitrification facility directly from tank farms without first
18 requiring the need to go through pretreatment. And so this
19 is a very important aspect, and that's why it's gotten a lot
20 of focus in the previous talks.

21 The second one is understanding the melting
22 process, and that's being done--Albert talked an awful lot
23 about that and Ian also. There are a lot of coupled chemical
24 and physical processes that occur when you go from melter
25 feed to the melt, and understanding those processes is

1 critical to ensuring that the vitrification facilities run
2 appropriately at Hanford. With the wide variation in waste
3 composition, each of them melt at a little different rate.
4 And the numerous examples of process upsets that have
5 occurred, not just in the U.S. but also abroad from a melting
6 standpoint, what we're trying to do is fundamentally
7 understand that process so that we can predict it, we can
8 optimize the throughput, and we can avoid process upsets. So
9 these two are primarily ORP-focused.

10 We also want to understand the long-term
11 performance of glass, and we want to do that for a couple of
12 reasons, primarily so we can increase the disposal options.
13 This is a joint fuel cycle technology and EM-funded activity.
14 We had a lot of questions in all of the earlier talk sessions
15 about performance of glass. And what we're doing here is
16 we're studying the long-term performance of glass so that we
17 can look at other disposal options, so that we can remove
18 some of the conservatism in the current models and take
19 advantage--take better advantage of the inherent durability
20 of glass. So this is the first joint project between fuel
21 cycle technology and EM.

22 We are also developing advanced glass ceramics or
23 crystal tolerant glasses. We're doing that for Office of
24 River Protection and Nuclear Energy in order to increase the
25 waste loading. But also, in the case of fuel cycle

1 technology, we're looking at increasing the performance of
2 the waste form, trying to get a higher performing waste form.
3 We're both looking at advanced waste forms for technetium.
4 We're doing that at Hanford primarily to increase the
5 treatment options. If there are ways that we could remove
6 technetium from the low-level waste stream, treat it
7 separately, dispose it separately, there are a lot of
8 advantages to the process that way. We're doing it for fuel
9 cycle technologies and ORP primarily to improve the long-term
10 performance, putting this long-lived radionuclide,
11 technetium-99, into a very highly durable waste form.

12 In the nuclear energy side, we're looking at waste
13 forms for pyrochemical processing wastes. Those are required
14 to enable that technology to be used, and that technology has
15 an awful lot of advantages, particularly for metallic wastes,
16 metallic fuels, fast reactor fuels. We're developing waste
17 forms for gaseous fission products. That's also an enabling
18 technology. The rest of the world aren't capturing all of
19 the same gaseous fission products as we would in the U.S.
20 And we're finding coupled processes that can both capture the
21 radionuclide and immobilize it into a waste form; and they
22 have to be long-lived waste forms, particularly for the
23 iodine-129. And, finally, we're developing alternative high-
24 level waste forms for significantly improved performance for
25 nuclear energy.

1 So what I was going to do is give a few examples of
2 the Environmental Management; but before I did that, I want
3 to point that there is an awful lot of correlation and
4 collaboration between the two offices. There is no accident
5 that Ken Picha and Pete Lyons and Monica Regalbuto are all
6 sitting right next to each other. The two offices do work
7 very closely together. They collaborate in these areas and
8 others. You heard Ken talk about the salt processing. So
9 this is a very collaborative effort, and that's why I'm
10 presenting it in this way.

11 So these four examples I have slides for. I'll get
12 as far as I can through this. First, if we look at waste
13 loading for Hanford wastes, the first step is to determine
14 what (inaudible) of the glass is limited by what constraints.
15 And what we see here is 70 percent of the glass is limited
16 by--and this is high-level glass, high-level waste glass.
17 This is limited by high aluminum (inaudible), spinel
18 precipitation, and nepheline precipitation problems. We have
19 a sulfur limit that's about 10 percent--I'm having trouble
20 seeing that--phosphate limits at about 10 percent; sodium
21 limits are only 1 percent; and chrome limits are somewhere
22 around the 10 percent also.

23 So what we're doing is--this is a snapshot of
24 today. What we're doing is trying to push back those
25 frontiers. And we're doing that at a collaborative research

1 project that's being run by the Office of River Protection,
2 and it involves Catholic University and PNL and DOE. And one
3 example of where we've gotten is the nepheline models where
4 we have--here we show ternaries of sodium, alumina, and
5 silica submixture in the glass. And here the white hot
6 regions are the regions where there's a very high potential
7 for nepheline precipitation on cooling. And the green
8 regions, there is effectively almost no chance. And we've
9 got two ternaries. One of them has no boron, and the other
10 one has 15 percent boron. And you can see how both the
11 sodium, aluminum, silicon, and boron strongly affect that.

12 By implementing this simple model, we were able
13 to--to enable this model, we needed experimental data. So
14 we've been collecting experimental data, an awful lot of
15 experimental data, and then we fit it to composition. And by
16 implementing this model, we could increase the average
17 maximum alumina content in glass from about 20 percent to
18 about 28 weight percent; and that's a very significant impact
19 on both the amount of glass we would produce and the
20 potential flexibility to do less aluminum removal in the
21 retreatment process.

22 Likewise, we've been looking at crystal tolerant
23 glasses for Hanford. The ubiquitous crystal that we see all
24 the time is the spinel crystal. I think you've heard it
25 mentioned by both David and Ian. And it's quite a nice

1 crystal, actually. If you look at the crystal there, it's
2 regular shaped. This is an example of a spinel crystal that
3 precipitated from a glass melt. And, in and of itself, it's
4 not a problem. What becomes a problem is if it agglomerates.
5 And this is an optical micrograph of a spinel sludge. And if
6 this forms at the bottom of the pour-spout riser, it could
7 have fairly significant impacts to the ability to operate
8 that melter. You could plug the pour-spout and not be able
9 to initiate pouring.

10 And so what we're trying to do is develop a new set
11 of constraints, a new way of looking at the problem, where we
12 truly avoid the deleterious effects of this sludge formation
13 while not limiting the loading of glass as much as we did.
14 And we've got preliminary models for that, and there is a
15 poster on this later this afternoon, so I encourage you to
16 look at the poster for more details.

17 The next one is understanding the melting process.
18 A lot of complicated physical and chemical processes that are
19 all coupled occur right here in this cold cap region. You
20 have gases generated in the cold cap and in the melt that
21 interact with it; you have multiple liquid and solid phases
22 that interact; and so what we're trying to do is develop
23 models to fundamentally understand that.

24 One of our accomplishments is we've achieved a
25 one-dimensional model, and that model would basically be the

1 reaction model that would go on top of the heat mass charged
2 transport model that Ian showed in his diagrams where there
3 was a flow inside the melter. He used the cold cap as sort
4 of a static heat sink and mass source. This would replace
5 that set of boundary conditions with a real coupled
6 chemical/physical model, and we're doing a lot of testing to
7 parameterize that model. And there's two posters on this
8 this afternoon.

9 And the last one--and I did want to just maybe go
10 over by one minute, if that's okay, because there was a lot
11 of questions about glass corrosion early on in the session.
12 And one of the issues is that there is a very broad range of
13 predicted responses. What this is is the Yucca Mountain
14 license application model for glass corrosion. This is what
15 it basically predicts. As a function of pH, this is the log
16 half-life of glass. And you can see that for two different
17 temperatures--we have 25°C here and 100°C here--we have
18 several orders of magnitude difference of glass corrosion.

19 And, as Peter Swift said, we tend, by the way we
20 did the sampling in the Yucca Mountain license application,
21 to favor the low durability/high release curves much more.
22 And the reason why we have this is twofold. If you look at
23 the reasons, there's two basic reasons. If you take glass
24 and you put it in static or very slow-flowing water, what you
25 get as a function of time, the amount of glass released, is

1 an initial jump in the amount of glass followed by a residual
2 rate that Stéphane Gin talked an awful lot about. And there
3 is a potential for some systems for that rate, that Stage II
4 rate here, to jump back up again to Stage III.

5 So one of the significant events that caused these
6 low durabilities is that we didn't have the data or the
7 understanding to rule out this jump up in rate, this Stage
8 III potential. It only happens under certain conditions.
9 And then the other one was the surface area. We had to
10 estimate the surface area of glass; and as the rate is a
11 per-surface-area rate, every time you increase the surface
12 area, you increase the amount of glass corrosion.

13 And so we're attacking those and other problems now
14 as part of the international program on glass corrosion. And
15 what we hope to get out of it is a better understanding so we
16 can use this lower Stage II rate, avoid this jump up to Stage
17 III, and we hope to be able to open up the options for
18 disposal of glass into a range of different environments.
19 We're studying this glass corrosion not just for Yucca
20 Mountain or not just in deionized water, but as a function of
21 the disposal environment. And so we're hoping to open up
22 those options.

23 And that's all I have. Thank you.

24 Oh, and there's a poster on this also this
25 afternoon on the technical details.

1 LUTZE: Thank you, John.

2 So we can now proceed to the discussion part. We
3 have about 25 minutes of discussion. So are there any
4 questions? Yes.

5 PEDDICORD: Yes, a question to David. First of all,
6 your work with the frits was very interesting, and it looks
7 like your return on investment in terms of R&D (inaudible).

8 The question I wanted to--I'm sorry, Peddicord from
9 the Board. The question I wanted to ask then was: In
10 contrasting your experiences with the situation here at
11 Hanford and given the kind of mixtures that we heard from
12 Albert and Ian and so on, are these same strategies
13 applicable and usable in terms of tailoring the frits to
14 sludges and so on (inaudible) characterize what your
15 experience is.

16 PEELER: The short answer is yes. They do it a little
17 bit differently. They actually tailor the frit on the fly,
18 because they're using glass formers. So they can bring in a
19 batch, a smaller batch, get its analysis, and then calculate
20 the glass formers and their concentrations on the fly. So
21 they do not use a prefabricated frit like DWPF does, that
22 they will be doing optimization on the fly.

23 PEGG: Yeah, I think that's a very important point. The
24 question of flexibility has come up a number of times. And
25 because of the variability complexity of the Hanford waste,

1 the decision was made not to use the frit strategy--that is,
2 a premade frit--because the issue is you need to have that
3 frit made in advance, delivered. It's probably a three- to
4 four-month cycle to procure all that frit. So what actually
5 happens--and we saw some of it at the WTP--is there's a
6 series of eleven silos of raw glass-forming chemicals, which
7 you can dial in and essentially make the components that
8 would make that frit on the fly. And those ingredients are
9 dialed in based on the batch that comes in; it's analyzed;
10 you dial in essentially the frit you need on the spot. So
11 that's a key aspect of the flexibility to respond to the
12 compositional diversity at the WTP.

13 LUTZE: I would like to ask one question to everybody.
14 As I understand it, the technetium at Hanford and WTP, that
15 is not captured in the solidification process. It goes to
16 the IDF, and there it obviously poses a problem, because
17 that's one of the critical elements to be released there. So
18 do we have enough data and research done to understand the
19 behavior of technetium in the LAW and the WTP vitrification
20 plants? Anybody want to address this?

21 PEGG: I guess I can say something about that. So it's
22 perhaps a little surprising that it's only in the recent few
23 years that that question has been looked at with any scale in
24 terms of the performance of the unit operations in the WTP.
25 So some years ago we had an EMSP program, looking at some of

1 the more fundamental aspects of the incorporation and
2 structurally of the technetium into LAW glasses. But the
3 question of, for example, the decontamination factors for
4 each of the off-gas unit operations, what is the split
5 between technetium in the feed, the fraction that's retained
6 in the melt, versus how much goes to the off-gas (inaudible).
7 Those were really not run, but there was a program at work
8 the past three years or so where that testing was done. So
9 the LAW flowsheet, there are first-of-a-kind data on
10 decontamination factors across a range of LAW waste
11 compositions. The same information is not available for the
12 HLW side of the WTP flowsheet.

13 But I think the short answer is, some has been
14 done, but I think more would be very useful.

15 LUTZE: Thank you.

16 Any other questions? Yes, please.

17 FRANKEL: Jerry Frankel, the Board. So I had a question
18 earlier for Carol about the long-term performance testing.
19 But, John, you brought it up again, so maybe you can address
20 it. I'm interested in the acceleration factors in those
21 tests and how you know that you're not making hard-boiled
22 eggs.

23 VIENNA: That's an excellent question. The root of most
24 of the difficulty in studying glass corrosion--and it
25 actually becomes much more difficult when you go to ceramic

1 corrosion, as some of those are quite a bit more durable.
2 The amount of corrosion that you get is so little within
3 laboratory time scales that you have to do--in some cases,
4 you have to do the acceleration. And so we try to understand
5 the processes that we use to accelerate so that we can
6 back-calculate what it would be unaccelerated. And then we
7 use either man-made or natural analogs to help understand
8 whether we got that correct.

9 And we have a series of man-made glasses that we
10 know the conditions that they've been in. They've been in
11 the Mediterranean or Adriatic Sea, and they've been there for
12 2,000 years. And we know the temperature, we know the
13 composition of the solution, and by looking at these samples
14 we are gaining the understanding of the process rates. And
15 we are using those as benchmarks to make sure that what we do
16 to accelerate and how we reverse that process is true to the
17 real data at least out to about 2,000 years.

18 FRANKEL: So you're using temperature and acidity and--

19 VIENNA: We use temperature, we use surface area--are
20 our primary two aspects to accelerate.

21 LUTZE: Just to add to this, one cautioning, of course,
22 and that is, with the natural analogs, these are not
23 compositionally different glasses, so they're not
24 borosilicate glasses. So we have to make the assumption that
25 the mechanism is the same. But it's a good tool that

1 supports the research. Just a little warning.

2 BRANTLEY: Just on that point--Sue Brantley of the
3 Board--it's also very different to be buried in the
4 Mediterranean Sea as opposed to being a porous media in
5 vadose zone or something like that; right? I mean, there's a
6 big difference in terms of that.

7 VIENNA: Certainly, yes. But it is a known environment.
8 And so at least we have a touchstone that goes out for 2,000
9 years in the case of the man-made samples, and we use that to
10 compare to experiments that we do with the same glass
11 composition in our laboratory environments. But it's really
12 ultimately reactive transport models, populated reactive
13 transport models, that allow us to predict what the
14 performance of the glass will be in a disposal environment.

15 LUTZE: More questions? Rod.

16 EWING: Rod Ewing, Board. A question for John. So you
17 showed the very nice diagram of the change of the release
18 rate with glass as a function of time and raised the issue of
19 this third stage where suddenly the rate would go up. There
20 are many things that can cause that rate to increase, but
21 fundamentally these are changes in the boundary conditions;
22 that is, the flow rate changes or material--a new phase
23 begins to precipitate. And so, thinking about avoiding that
24 possibility, are there geochemical environments that you can
25 imagine for a repository that would be better or worse

1 because of this possibility?

2 VIENNA: Certainly, yes. I think that we saw, for
3 example, the Belgian disposal concept, which very much
4 promotes an accelerated rate because of the high-pH cement
5 pore water solution that would be in contact with glass.
6 That is a relatively harsh environment for glass to be
7 disposed of compared to tuff at Yucca Mountain or the
8 argillite in France or some of the other disposal concepts
9 we've looked at.

10 EWING: What environment would enhance the behavior of
11 glass?

12 VIENNA: A dry one is one potential; it's just one
13 potential.

14 LUTZE: Are there other questions?

15 I would like to ask one other question that would
16 have been to Stéphane Gin, but also to everybody who intends
17 to reprocess in the future. You showed this enormous amount
18 of activity vitrified, which also means that there was an
19 enormous amount of krypton somewhere released during the
20 reprocessing. Is there any active research going on to
21 consider what to do with the krypton as a waste form? I
22 mean, we all know (inaudible) rubidium, and rubidium is a
23 corrosive alkali element. And yet I think something could be
24 done with the krypton.

25 Stéphane? Is he still here?

1 GIN: I'm not sure about the studies that are conducted
2 in France for immobilizing krypton. I know that we are doing
3 some work on iodine, because it could be a request in the
4 future that iodine must be confined in durable matrix, and we
5 are starting corroboration with PNNL in this field. But,
6 yes, you said krypton and other mobile elements that are
7 released in the atmosphere or in sea water at the present
8 time. But I'm not sure--the question is more important for
9 krypton compared to iodine. So we are doing some work on
10 iodine. The krypton, I don't know.

11 VIENNA: I can answer what the U.S. is working on there.
12 The U.S. is looking at a range of options for krypton with
13 such a short half-life. Old (inaudible) fuel, the krypton
14 could be vented with minimal impacts; but for short
15 (inaudible) fuel, by federal regulation, it would have to be
16 captured, and it would have to be stored. The base option
17 that we're looking at is storing it in a compressed cylinder
18 in the presence of a metalorganic framework that helps to
19 decrease the pressure in the gas cylinder and potentially
20 immobilize the daughter product in a network so that it
21 doesn't reach the valves and the side walls of the canister
22 so much. But we've also got an option where we solidify it
23 by sputtering either in silicon carbide or in copper.

24 LUTZE: Thank you.

25 More questions? Well, we need more questions. We

1 have ten minutes.

2 Well, one thing to my mind, and that was, we talked
3 a lot about the bubblers and the beneficial effect of the
4 bubblers by the strong increase of the throughput, which
5 gives us a chance to complete the mission earlier, but there
6 are probably limits as to how much funds are available to do
7 so. Does the funding of the production of canisters go
8 step-in-step with the increase of the production rate? So
9 can you actually make at DWPF as many canisters as you like,
10 or is there a funding limit? Can somebody answer that
11 question?

12 PEELER: I'd refer to Jonathan on that, if he's still in
13 the audience.

14 LUTZE: Maybe Carol knows the answer to that. Carol?

15 JANTZEN: Sorry, Werner, I wasn't listening.

16 HERMAN: (Inaudible) canisters per year (inaudible)
17 include that in the budget. So there is--I'm sorry.

18 LUTZE: Come here, please, to the microphone.

19 HERMAN: Connie Herman from SRNL. From my understanding
20 of the DWPS budget allocation is they assume, with their
21 system plan David referred to, how many canisters they're
22 going to make per year, so there is a high end. They have
23 some flexibility in their budget to be able to accommodate--
24 if you're going to make more canisters, you would need more
25 frit, so a materials budget. So they would have to

1 accommodate that in their overall budget.

2 So they are planning--you know, if they're going to
3 make 200 cans and they get to a production rate of 300, so
4 the next year they would adjust that for that. So it is a
5 reallocation they do within their budget.

6 LUTZE: Okay, thank you.

7 Any other questions? Yes, please.

8 ZOBACK: Mary Lou Zoback, Board. This is a really naïve
9 question, but it's been bugging me since Carol's
10 presentation. The Savannah River melters are circular;
11 they're round. And West Valley and the plan for Hanford are
12 square. And my experience melting things in round pans--
13 pots--is that that's an efficient way to do it. So what are
14 the advantages of the square design for melters?

15 PEGG: I think the round melter concept came basically
16 around the tank-type design. The more square or rectangular
17 design, particularly as you get to larger and larger scales,
18 affords a fair amount of simplicity in the refractory
19 shaping. So these refractories are very dense, high-chromium
20 refractories, and straight, flat walls versus curved
21 surfaces, especially as the melter gets larger and larger, so
22 there is a fabrication and construction element involved.
23 But, beyond that, the West Valley experience did not suggest
24 that--and, for example, the (inaudible) experience did not
25 really suggest a significant difference between the round

1 shape and the corners, particularly--

2 ZOBACK: You don't get material stuck in the corners
3 (inaudible)?

4 PEGG: Much less so with the bubbled melters as well.
5 When you have the pool being even more active, it becomes
6 less and less of an issue.

7 ZOBACK: Okay, thanks.

8 LUTZE: Well, if there are no further questions, then we
9 conclude this morning session. And I would like to thank all
10 the contributors one more time. Thank you very much.

11 EWING: And just a few words before we all leave for
12 lunch. I wanted to thank Werner for running both panels. I
13 think it went very well.

14 We'll start promptly at 2:15. And we realize that
15 everyone will scatter for lunch. So, just to give you maybe
16 a time advantage, we have notified Anthony's that there may
17 be a lot of people coming, so they're ready for you should
18 you choose to eat at Anthony's. It's not an advertisement
19 for them, but it may be efficient.

20 So we'll see you at 2:15. Thank you.

21 (Whereupon, a lunch recess was taken.)

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AFTERNOON SESSION

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GEPHART: Welcome to this afternoon's meeting on comments by tribes, state, public organizations. We have asked the six folks that I will be introducing to share some of their views and, most important, technical issues associated with the (inaudible) of high-level waste and spent nuclear material that is stored at the Hanford site.

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The basic protocol to get us through the next hour and a half is, we have about fifteen minutes per individual, in which I will recommend ten minutes for presentation and about five minutes for discussion with the Board so there's an opportunity for the Board and the speakers to engage.

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By the way, my name is Roy Gephart, as Rod just noted. I am a consultant to the Board for this Hanford visit, and I retired ten months ago after nearly 40 years with the Hanford contractors and Pacific Northwest National Laboratory, retiring as a Chief Environmental Scientist with the Pacific Northwest National Laboratory.

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The only other items I have is, I passed around to the Board and to the staff copies of Russell Jim's discussion. The other talks for which there are thumbnail copies electronically, they are over here on our computer, and members of the Board and others will receive electronic copies of those. Not everyone will have PowerPoint

1 presentations.

2 So, with that, it is my pleasure to introduce
3 Russell Jim. Russell Jim is the Manager of the Yakama
4 Nation's Environmental Restoration and Waste Management
5 Program.

6 Russell, you can sit or you can stand, whatever
7 your pleasure is.

8 JIM: Good afternoon. Welcome to the ancient land of
9 the Yakamas. This area was the wintering ground for the
10 Yakama for millennia, based upon a slow geographical locale,
11 witnessed by a confluence of three rivers, the Snake, the
12 Columbia, and the Yakima. A little point of irony: Where we
13 are now is where my ancient relatives camped. The camps had
14 their families and extended families move in, and from here
15 they decided in the spring which way to go. They discussed
16 it all winter and decided to go to the usual and accustomed
17 places: Canada, Montana, Arizona, northern California, or
18 the coast. That was the lifestyle based upon not wanting to
19 deplete the resources by staying at one spot. And right now
20 the foods and medicines are coming out. Here they would be
21 near done. They would be gathering them as they went up
22 towards the alpine area.

23 Another bit of irony I heard this morning, the
24 bugle call that was summoning the people back into the room.
25 It sounded like the one they used to use and circle the

1 wagons.

2 So what I am going to address is the problems we're
3 having as the most affected people in this region, most
4 affected by the Manhattan Project. The Manhattan Project was
5 moved in here because they knew they needed the abundance of
6 cold water, the cheap electricity from Bonneville Dam. It
7 was an isolated wasteland, and the people were expendable.
8 That was a little upsetting when I read that in 1979. We are
9 the most affected people, but hardly any entity will put that
10 in black and white.

11 I have read the United State Constitution, in which
12 it states that the treaties are the law of the land. How
13 many in this room have read the Treaty of 1855 between the
14 Yakama Nation and the United States of America? One, two,
15 three, four, five. That's been the problem. Not many have
16 taken the time to read or understand, and so we consistently
17 have to address over and over the logic of the uninformed.

18 I have a written statement here, and I'll try to
19 finish up in the next five minutes.

20 "Thank you for the opportunity to speak to the
21 Review Board today. Today the Board is considering daily
22 activities concerning the vitrification of high-level
23 radioactive waste at Hanford for eventual disposal in a
24 repository. The Yakama Nation has concerns at Hanford, which
25 include this matter and which encompass issues far beyond

1 this matter as well. The Yakama concerns arise from our
2 contract with the United States, the Treaty of 1855, which
3 guarantees perpetual rights to fish, hunt, and gather our
4 traditional foods and medicines on open or unclaimed land,
5 including land at the Hanford site. The Yakama land at
6 Hanford was ceded to the United States in the treaty with the
7 provision that such rights would remain forever. The treaty
8 involved a grant of land and rights to the United States by
9 the Yakama Nation, not the reverse.

10 "For those of you who may question whether a treaty
11 issue is properly in the domain of your review activities, I
12 offer the following: The Board was created under the 1987
13 amendments to the Nuclear Waste Policy Act and as part of the
14 statutory framework for dealing with spent fuel and high-
15 level nuclear waste. The treaty ratified by Congress and
16 signed by the president is part of the statutory framework
17 for relations between sovereigns and is considered the
18 supreme law of the land.

19 "Absent an express act of Congress, to the
20 contrary, treaty rights at Hanford are fully intact and have
21 a direct bearing on the issue addressed today. Only the
22 Yakama government has the authority to express its unique
23 concerns regarding Nuclear Waste Policy Act implementation
24 and effects on its rights. We ask that you understand and
25 support full compliance with those treaty rights. Notably,

1 the Yakama Nation contributed to the parent language of the
2 Nuclear Waste Policy Act in 1982, which led to the inclusion
3 of affected tribal governments in the high-level waste
4 disposal review process.

5 "A distinct area of concern for the Yakama
6 government is potential reclassification of high-level
7 radioactive waste at Hanford. Eleven years ago the Yakama
8 Nation joined a federal lawsuit to prevent the Department of
9 Energy from reclassifying high-level waste. The District
10 Court sided with the Yakama, Washington State, and others.
11 On appeal it was ruled that the matter was not ripe for a
12 decision. Despite this legal uncertainty, I request that the
13 Board consider all the relevant factors which will affect
14 future disposal of this extremely toxic waste.

15 "During the high-level waste legal challenge, DOE
16 requested that Congress clarify the Nuclear Waste Policy Act.
17 This led to an unusual situation, to say the least, in which
18 Congress was provided--has provided the Secretary of Energy
19 with authority to reclassify high-level waste in South
20 Carolina and Idaho, but not at Hanford. The Yakama Nation,
21 Washington, and others oppose reclassification. Senator
22 Maria Cantwell's efforts prevented such redefinition
23 authority at Hanford in the 2005 Defense Authorization Act.
24 At the time these concerns were articulated by former
25 Governor Christine Gregoire, then Attorney General, in a

1 letter to Senator Cantwell on June 1, 2004, quote, 'The
2 problems associated with the storage and disposition of high-
3 level waste, however, require real solutions, not avoidance
4 through redefinition,' unquote.

5 "If this waste is reclassified, I am concerned that
6 DOE will attempt to dispose of significant volumes of high-
7 level nuclear waste at Hanford near the Columbia River rather
8 than at a geologic repository, as currently required. For
9 the Yakama Nation these concerns are undiminished today.

10 "In an August 7, 2012, letter to the EPA regional
11 administrator, the Yakama Nation cited possible violations of
12 the Nuclear Waste Policy Act at Hanford. Specifically, it
13 was documented that high-level radioactive waste was
14 historically stored at the 618-10 and 618-11 burial grounds
15 and that such retrieved waste is to be disposed of in the
16 landfill at Hanford. The waste (inaudible) criteria for
17 those landfills forbids emplacement of high-level waste. Of
18 course, the Nuclear Waste Policy Act forbids disposal of any
19 high-level waste in any landfill.

20 "I request that the Review Board examine the
21 situation in detail. A decision process in which high-level
22 waste exists only if the Department of Energy says it exists
23 is neither credible nor viable. Any violation of the Nuclear
24 Waste Policy Act will result in mistrust by the citizens of
25 this region, will undermine congressional intent, and will

1 pose a grave risk to human health and the environmental
2 protection.

3 "The situation cited here is of acute concern. On
4 the horizon is possible redefinition of high-level waste,
5 waste which was discharged or leaked from the single-shell
6 tanks, waste left in tanks as residuals, and waste separated
7 in the so-called low-activity waste, a term which has no
8 legal basis. The Review Board would do this region, the
9 nation's taxpayers, and Congress a great service by
10 highlighting the consequences of this ad hoc high-level waste
11 strategy, which defers to DOE on classification questions and
12 leaves the most important question unanswered: How much of
13 this high-level radioactive waste is planned to be left at
14 Hanford?

15 "I would like to bring forth another matter to the
16 Board, which involves a consequence of not cleaning up and
17 properly disposing of this toxic material. In 2002 the
18 Yakama Nation initiated the CERCLA Natural Resource Data
19 Assessment for the Hanford site by bringing a claim for
20 injury to resources in the 1100 area. Damages under CERCLA
21 arise from injury to natural resources caused by the release
22 of hazardous substances. The CERCLA was later expanded to
23 include hazardous releases from the entire Hanford site and
24 was joined by Washington, Oregon, the Umatilla Tribe, the Nez
25 Perce Tribe.

1 "A recent completed preliminary estimate of damages
2 calculated primary restoration costs in excess of 20 billion.
3 This estimate excludes injury arising from any high-level
4 waste left in the tanks, leaked from the tanks, or which may
5 be otherwise left at Hanford. The Yakama Nation would like
6 its treaty resources at Hanford to be restored. Compliance
7 with the 1855 Treaty requires such restoration.

8 "The treaty signatories did not contemplate the
9 exercise of treaty rights, which cause extraordinary health
10 effects and fatal cancers. Since I believe that your charter
11 is ultimately to protect humans and the environment, I ask
12 that you document for the record the potential impacts to our
13 tribal members and to our treaty rights from any high-level
14 waste which could be left at Hanford.

15 "The Yakama government has embarked on an effort to
16 research the spectrum of effects of exposure to these
17 dangerous toxins. I request that the Board consider the
18 unique pathways, exposure, and effects from high-level waste
19 to our genetically and culturally distinct people when you
20 are deliberating the disposition of this material and
21 document the impacts accordingly."

22 I thank you for your time, and I'll be glad to
23 answer any questions.

24 GEPHART: Are there any questions from the Board?

25 Questions from the staff?

1 Then we'll proceed with the second presentation.

2 Thank you, Russell, very much.

3 The second presentation is by Suzanne Dahl.

4 Suzanne is the Tank Waste Section Treatment Manager for the
5 Washington State Department of Ecology, one of the signers of
6 the Tri-Party Agreement.

7 Suzanne.

8 DAHL: Thank you, Roy.

9 And thank you, Russell, for going in front and
10 leading the way, as always.

11 On the part of the Department of Ecology in
12 Washington State, welcome to our state, and thank you for the
13 work that you do on important issues regarding high-level
14 waste. I know that other folks on the panel will discuss
15 many of the things--I had some opportunity with you
16 yesterday--thank you--and talked about a regulatory
17 framework, so I'm not going to go back over that. And I'm
18 going to try to concentrate on some specific questions that
19 had been asked when we were setting up for this meeting in
20 the last couple of weeks.

21 So, as you've heard undoubtedly through the
22 morning, that there is spent fuel at Hanford; there is high-
23 level waste. The high-level waste is in different forms.
24 I'm going to spend a little bit of my time talking about the
25 Waste Treatment Plant and where we need to go and a little

1 bit more answering the question of why is there a need to
2 glassify the immobilized low-activity waste at Hanford.

3 Hanford has 60 percent of the nation's high-level
4 waste. It's about 195 million curies, an amazing 190,000
5 tons of chemicals which are hazardous waste. Ten of our
6 single-shell tanks have been retrieved out of the 149. Six
7 single-shell tanks are currently leaking, and one of the
8 double-shell tanks are currently leaking. This is important
9 because, even though we've had 67 past leakers in the past,
10 we had at that point assumed they hadn't been leaking anymore
11 after the liquid had been removed from them. And so we
12 thought that for the moment that they were sound and not
13 leaking into the environment; to step off into this year,
14 into 2013, and have six tanks leaking to the environment and
15 a double-shell tank that's significantly compromised and
16 needs to be taken out of service is a significant issue for
17 the State of Washington.

18 The waste is managed, from the State's perspective,
19 under the Dangerous Waste or the RCRA regulations, and we
20 have both Tri-Party Agreement, Hansford Consent Order and
21 Agreement milestones for the cleanup of tank waste, and then
22 we also have a 2010 consent decree that's signed in front of
23 a judge.

24 You guys have seen some similar graph before. Most
25 of the waste from the tanks will--most of the volume will go

1 to the low-activity side. This is important for us. Most of
2 the chemicals will go to the low-activity side; about five
3 percent of the curies will go to the low-activity; 95 percent
4 of the curies will go to the high-level side, destined for
5 deep geologic repository, but that turns out to only be a
6 small fraction of the volume.

7 So, not to have a pun in your title of the slide,
8 but for us the path forward for tank waste is as clear as
9 glass. We need to maintain focus on building the five
10 facilities that make up the Waste Treatment Plant. And if
11 there's modifications that need to be made in the design,
12 then we need to take this time to make those modifications so
13 that we have a durable, workable set of facilities when we
14 get started. We need to prepare the facilities and
15 infrastructure that are needed to feed the waste from the
16 tank farms. And, also, if there are facilities or systems
17 that need to be put in place to make sure that the waste
18 coming from the tank farms to the Waste Treatment Plant is
19 compatible with the Waste Treatment Plant, then we need to
20 make sure that we're doing that now also.

21 We need to provide current and future safe storage
22 for the tanks waste while the treatment facility is being
23 completed and for the 30 years that it will take to treat all
24 the tank waste; and obviously this means moving as quickly as
25 possible waste from the single shells to the double shells.

1 And if that in cases means that we need additional capacity
2 to do it, additional tank capacity, then that's something
3 that must be considered.

4 And then, lastly, we need to construct other
5 support facilities which you can't run the whole Waste
6 Treatment Plant without. That includes a place to put the
7 high-level waste glass as it's being stored and waiting to go
8 to a deep geologic repository. That's a facility that needs
9 to be designed and constructed and funded. And then also, in
10 order to get all of our waste treated, we need to have the
11 additional low-activity systems on board that we need to get
12 the mission done within a 30-year time frame and not an
13 80-year time frame.

14 So, just to give you a little bit of history on the
15 immobilized low-activity waste at Hanford, in the mid-'90s,
16 while Savannah River was moving forward with its
17 vitrification facility, the Department of Energy asked the
18 State of Washington to delay our vitrification facility. We
19 had started it, there were some issues, and they asked us to
20 delay it. And, one, there just simply wasn't funding to do
21 both major vitrification facilities at the same time; and
22 there was the need to want to learn from how this facility
23 would work at Savannah River.

24 And so the trade that was made for that was, we
25 would take the delay. We thought it was going to be a

1 ten-year delay. We had no way of imagining it would become a
2 20-year delay. And what we got for that trade was the
3 commitment then to vitrify our low-activity waste, because
4 that was the issue at the time was the current studies of the
5 low-activity waste, which was going to be grouted at the
6 time, there were impacts to the groundwater beyond nitrate
7 concentrations that were unacceptable and beyond and other
8 constituents, technetium and iodine. And so that commitment
9 and change was made in the '90s, and we have incurred the
10 impact from that commitment in the delays, and so we continue
11 to keep our aim towards getting the low-activity waste
12 vitrified.

13 In 1996 the TWRS EIS then followed along with the
14 decision that both the high-level and the low-activity waste
15 would be vitrified. In '97, leading up to '97, from '93 to
16 '97, there was a series of interactions between Department of
17 Energy and the Nuclear Regulatory Commission. And they were
18 agreeing on criteria for how you could allow high-level waste
19 that had been separated, low-activity waste, to be disposed
20 of in a near-surface environment, because it's not low-level
21 waste. It's still technically high-level waste. You've just
22 removed enough of the fission products so that you can
23 dispose of it in a near-surface environment instead of in a
24 deep geologic repository. And there were several important
25 separation commitments that needed to be met to do that, and

1 that's the basis of our pretreatment facility. And then
2 also, in that discussion and in that commitment back and
3 forth between Department of Energy and NRC, was the idea that
4 the low-activity waste would be vitrified.

5 In 2003 the Department of Energy asked the
6 Department of Ecology to consider other options for the low-
7 activity waste with the idea that if there could be other
8 options that could be found that would be just as protective
9 to the environment, would we consider them. And we said,
10 okay, let us be part of the studies, let us work forward with
11 you, and there was always this promise that it would be
12 cheaper and faster. And we said, okay, we'll enter in with
13 you in looking at it and studying it.

14 And then that all culminated in a milestone in 2006
15 where they were to bring the data forward to show what had
16 met the bar, what waste forms would turn out to be, in fact,
17 cheaper and faster and then, in addition, be as good as
18 glass. And at that time none of the other waste forms proved
19 out to be as good as glass or to be protective enough to meet
20 drinking water standards.

21 In 2010 we had the settlement agreement that
22 resulted in the consent decree signed in front of a federal
23 judge. And in there we agreed to milestones that--there's
24 seven different milestones in there that talk about
25 supplemental treatment vitrification.

1 And then, lastly, in 2011 the final tank closure/
2 waste management EIS was issued, and they looked at different
3 options for supplemental treatment. None of them proved to
4 be as good as glass and also to be completely protective of
5 the different drinking water stands.

6 This is a graph out of the Environmental Impact
7 Statement, and the low-activity glass is the red line on the
8 bottom. The others are either grout or steam reforming or
9 bulk vitrification. And you can see the point of this graph
10 is that none of them proved to be as good as glass. And then
11 if you look at further data in the EIS, you'll find that both
12 grout, steam reforming, and--well, both grout and steam
13 reforming at some point in the future would violate the
14 drinking water standards for nitrate, chrome, iodine, and
15 technetium, and, I believe, uranium.

16 Some people will ask this question, so I thought
17 I'd put it on a slide of what are the different sites and
18 what do they do with their low-activity waste. And, as far
19 as how they're regulated, all the different sites that you've
20 heard about today are all regulated. Their high-level waste
21 is regulated under RCRA, except for when you get to Savannah
22 River's. The choice was made there to do it under the Clean
23 Water Act. And that has to do with what's the LDR treatment
24 standard, the land disposal restriction treatment standard,
25 associated with RCRA and the metals associated with high-

1 level waste. And the treatment standard for metals
2 associated with high-level waste is something called HL vit;
3 it's vitrification.

4 And so at Hanford the plan is for immobilized
5 low-activity waste to stay in the near surface in a landfill.
6 The current plan is to vitrify it, although there are some
7 other options being continued to look at. At Idaho--this is
8 a little bit incorrect--their high-level waste will go to a
9 deep geologic repository. Their other waste that's
10 associated with tank waste will go somewhere, and I have it
11 going to WIPP there. And that's probably not correct. But
12 it's written into a consent decree with Idaho that it's not
13 staying at Idaho. So the point is, they're not disposing of
14 their low-activity fraction or any of their fraction of their
15 high-level waste.

16 At Savannah River we've heard quite a bit of, it is
17 currently being disposed of in a near-surface environment in
18 the form of saltstone. And at West Valley, all of their
19 waste will eventually go off-site too. Their low-activity
20 waste has already gone off-site, and eventually their high-
21 level will go off-site.

22 So, really, it's just the difference between
23 Hanford and Savannah River. And there is some geologic
24 differences too, is that we have a great depth to our
25 groundwater. They're much closer to their groundwater. And

1 this may seem a little bit backwards, but what happens at
2 Hanford is, our really low infiltration rate and our slow
3 groundwater flow ends up concentrating underneath the
4 landfill. So if your contamination and you're coming down to
5 the groundwater, it ends up concentrating. So if you look at
6 a point of compliance next to the landfill or out a little
7 ways, you'll find that the concentration is greater if you
8 were to dispose of grout in our landfill than it is at
9 Savannah River. At Savannah River you have higher
10 infiltration rates, a much closer distance to the
11 groundwater, and a faster flowing groundwater regime, so it
12 tends to sort of dilute and move the waste along. And so
13 those are the key differences.

14 And so just a little bit in summary. For the past
15 15 years we've had a commitment to the assumptions that the
16 low-activity waste was going to be vitrified at Hanford.
17 This is important to us, because a lot of the--a significant
18 number of the mobile constituents are both the radiological
19 ones, and the chemical ones are in the low-activity waste.
20 To date, none of the various efforts to prove out other waste
21 forms have resulted in something that was protective of the
22 environment or as good as glass. There are some land
23 disposal restrictions coming out of RCRA that points to our
24 low-activity glass needing to be in a vitrified form. And
25 we've had recent and commitments going back to the '90s that

1 say that our low-activity waste needs to be vitrified.

2 So, in summary, Hanford's got a large risk volume
3 that already exists, all the cribs and canyons, the other
4 facilities, eventually the closed tank farms, the past
5 landfills, the current landfills. So we've got a risk burden
6 that's already here from waste that's not leaving the site.

7 In addition, there has been this long-standing
8 approach to pretreat the chemicals and some of the
9 radionuclides off of the tank waste and leave the chemicals
10 here and send the more highly concentrated high-level waste
11 to a deep geologic repository. And while that is a good
12 economic decision because of the great cost of disposing of
13 logs in a deep geologic repository, it puts an additional
14 risk burden on Hanford on top of our other environmental risk
15 burden that we already have. And so for that reason it's
16 been important and continues to be important for us that our
17 immobilized low-activity waste form be the most durable
18 dependent waste form that we can have, and that's why the
19 State has really settled on that being vitrification.

20 And then the last thing I wanted to say is, the
21 leaking tanks right now--if you just want to look at
22 observational science, the leaking tanks are telling us that
23 they can't wait decades upon decades for the waste to be
24 immobilized and that we need to move forward with the Waste
25 Treatment Plant; we need to fix the issues that are

1 associated with it; we need to move forward with it. And
2 then, in the meantime, we need to be looking for safe storage
3 of the tank waste.

4 GEPHART: Are there any questions for Suzanne?

5 SPEAKER: Hold on.

6 GEPHART: Hold on.

7 SPEAKER: Somebody is knocking the cord out that's
8 holding--you moved your chair back.

9 GEPHART: Let's try that again. Rod.

10 EWING: All right, thank you.

11 So in your presentation you define the low-activity
12 waste stream as technically high-level waste. And then,
13 drawing on Russell Jim's presentation, who has finally the
14 authority to declare that it's not high-level waste?

15 DAHL: The answer to that probably depends on who you
16 ask. If you ask the Department of Energy, I believe they
17 would probably say they have the final authority. I think in
18 the court ruling and in the State's interpretation of the
19 Nuclear Waste Policy Act, we believe that that's not a
20 decision that can solely be made by the Department of Energy.
21 And so that's why it was important for us that in the '93
22 through '97 discussions with NRC, that NRC was part of that
23 agreement and discussion in defining what could be called
24 low-activity waste.

25 EWING: Right. But that was, I think, again, in your

1 words, a consultation. And in my reading, the language was
2 very carefully qualified in terms of their consultation. So
3 is this a topic that would finally be settled in court some
4 years from now? Is that the path we're on?

5 DAHL: I'm not sure that that's where it gets settled.
6 I guess I--

7 EWING: I guess, more to the point, I'm wondering at the
8 wisdom of following this strategy without a clear definition
9 that this waste stream can be declared as not high-level
10 waste.

11 DAHL: The State has been happy with that NRC
12 consultation process that happened, and also a similar
13 process is allowed for in the Tri-Party Agreement on the tank
14 residuals when we get there. And we've been happy with that
15 consultation process. And as long as it's something that
16 resembles that, that seems to be something that the State can
17 stand behind. It's when the Department of Energy does it by
18 itself solely is when we have issues.

19 EWING: Okay, thank you.

20 GEPHART: Yes, sir.

21 PEDDICORD: Yes. Peddicord from the Board. So to help
22 me understand, is there then a distinction between the Yakama
23 Nation's solution outlined by Mr. Jim and the State of
24 Washington in terms of implementing the vitrification of low-
25 activity waste, and do you all see that as the

1 reclassification that you referred to? I'm trying to
2 understand if you're consistent on this position or there's a
3 distinction?

4 DAHL: Do you want to answer it, Russell?

5 JIM: I'll try a part of it. Part of the 2005
6 Reauthorization Act 3116 portion allows the reclassification
7 at Idaho and Savannah River but not at Hanford. And coupled
8 with that is the Department of Energy's 435.1; that's an
9 order, a DOE order, yet some feel that it's a law, and it is
10 not. And eventually they want to mesh 3116, 435.1 to
11 gradually mesh together to justify defining low-activity
12 waste, which, of course, you heard and probably know that
13 there is high-level waste mixed in with it to justify leaving
14 it here near the surface of the earth. That's our concern.

15 PEDDICORD: So the Yakama Nation has a different
16 position than the State of Washington?

17 DAHL: There is a nuance difference. In the court case
18 that Mr. Jim was talking about, they filed along with
19 others--the State of Washington entered in as a friend of the
20 court, meaning that we had information on both sides of the
21 case, and that tells you at least a little bit that there's a
22 difference there.

23 GEPHART: Sue.

24 CLARK: So, following along the same line of discussion
25 but in the context of high-level waste that is leaked, what

1 is the State's position and then also the Yakama Nation's
2 position about the material that's leaked from the waste
3 tank? Is it high-level waste? Is it not?

4 DAHL: The State has taken the position that the waste
5 that's leaked is still high-level waste, because the way that
6 you get through the process, whether it's talking about tank
7 residuals or whether it's talking an immobilized low-activity
8 waste form, the way you get to not being a high-level waste
9 that needs to go to a deep geologic repository is that you've
10 removed sufficient fission products, that you've gone through
11 an immobilization technique, and that you've looked at the
12 risk of it. Those were the three criteria that were laid out
13 in '93 with the NRC, and the other things in there, they're
14 consistent, at least in the State's mind, with information in
15 the Nuclear Waste Policy Act. And so waste that's leaked to
16 the ground, there is no magic separation technology or
17 immobilization that happened just because you leaked. So we
18 believe it's still high-level waste.

19 CLARK: And the Yakama Nation?

20 JIM: We go back to the source term issue in the Nuclear
21 Waste Policy Act; and, therefore, we do have more problems.
22 When I was told here last year that--when I questioned, "How
23 much are you going to leave in the tanks?" And they said,
24 "Well, just about an inch," not verifying that that inch
25 could be some of the most deadliest material. And to do so,

1 then I asked, "Aren't you concerned about the source term of
2 the Nuclear Waste Policy Act?" And there seemed to be a
3 misunderstanding. I was told then that the source term came
4 out of RCRA, which is not true. I helped write the Act.
5 Thank you.

6 BECKER: Steve Becker, Board. I'm trying to get my arms
7 around this issue as well. Suzanne, what is the formal
8 definition low-activity waste? Is there a certain
9 quantitative measure? How is it defined in a formal way?

10 DAHL: It's actually defined in those interactions back
11 and forth between '93 and '97 between the Department of
12 Energy and the NRC, and I can get you some of that
13 documentation that gets you to that. It talks about using
14 single-pass ion exchange to remove cesium and talks about
15 liquid solid separation on each batch of waste that's
16 processed, and it talks about--oh, I've lost the third one.
17 Oh, and then there were some specific tanks that had
18 transuranics in it, and it talks about doing a different
19 separation on those.

20 And then in that they also assumed a waste
21 performance, that obviously you couldn't dispose of it if it
22 wouldn't meet certain waste performance criteria. And the
23 waste performance that they assumed in this discussion back
24 and forth and in the calculations back and forth was
25 vitrification. And it was assumed that it was disposed of in

1 that disposal unit that you guys saw yesterday or something
2 closely located to that in the integrated disposal landfill.

3 BECKER: So is there a specific level of activity or a
4 cut-off or some sort of criterion for distinguishing what is
5 versus what isn't low-activity waste?

6 DAHL: There were number specific--and somebody will
7 have to help me because I don't have--I had it memorized and
8 they've left me, but on both the removal of cesium and
9 strontium, but then also those other criteria I named. And I
10 can get them to you.

11 GEPHART: Any other questions? Board? Staff?

12 I think, as you heard over the last few minutes,
13 this has been an extremely critical issue for the last 25
14 years that remains unresolved and that will determine
15 significant adjustments or continuing with Hanford strategy
16 as it is. So it's important. I really appreciate the
17 engagement.

18 Our next speaker, Ken Niles, is--Ken is with the
19 Oregon Department of Energy, Nuclear Safety Division. This
20 organization provides Oregon's oversight of Hanford
21 remediation. And whenever, throughout the year, I have
22 needed the answer to the question--What's happening south of
23 the Columbia River?--I call Ken up.

24 Ken.

25 NILES: Thank you, Roy.

1 Good afternoon, everyone. Thank you for the
2 opportunity to provide you with Oregon's perspective on
3 Hanford's high-level waste and spent nuclear fuel.

4 The State of Oregon has had about a 30-year history
5 of working on Hanford issues, since before cleanup and very
6 active during the cleanup process. I've been with the Agency
7 now working on Hanford issues for about 24 years, so we've
8 been a long time interested in these issues.

9 I'd like to spend just a couple of minutes of my
10 time kind of giving you a bit of a perspective on Oregon's
11 relationship with Hanford and our view, as you will, from
12 south of the Columbia River and south of the border. Georgia
13 is the only other state that really is similar to Oregon in
14 that they're an adjoining state to the Savannah River site,
15 although typically not as consistently active as Oregon has
16 been through the years.

17 So Oregon's primary concern is with Hanford as
18 potential contamination to the Columbia River, which flows
19 through the site and becomes the Oregon-Washington border all
20 the way to the Pacific Ocean. The Columbia River is often
21 referred to as the lifeblood of the region. Its water is
22 vital for drinking water, for irrigation, supports salmon
23 fisheries, and the Columbia River is a very popular place for
24 recreational activities, camping, boating, fishing, and some
25 of the best wind surfing site in the world. Any

1 contamination that's in the groundwater moving to the
2 Columbia River or contamination now that's in the soil moving
3 to the groundwater and eventually to the Columbia River is
4 what does concern us. So the tank issues, the groundwater
5 issues are high on our list of things that we're concerned
6 about.

7 Oregon is also a primary transportation corridor
8 into and out of Hanford. If there is ever highly vitrified
9 waste, if there is spent nuclear fuel that leaves the site
10 that goes to a geologic disposal site or to an interim
11 storage facility, it will likely travel through 200 miles of
12 northeast Oregon, which is also currently--although not at
13 the exact moment right now--but it is a transportation
14 corridor for transuranic waste going from Hanford to the
15 Waste Isolation Pilot Plant.

16 Oregon was very much involved in working with
17 western states through the Western Governors' Association and
18 with the U.S. Department of Energy in developing a
19 transportation safety program for WIPP shipments, and we're
20 also now currently in discussions with the U.S. Department of
21 Energy and other states around the country as we begin
22 preliminary transportation planning for what the Department
23 of Energy has targeted the opening of an interim storage
24 facility in 2021.

25 So that background out of the way, let me now

1 address the topic of this session, views on the most
2 important technical issues in terms of Hanford's high-level
3 waste and spent nuclear fuel. And, frankly, it's difficult
4 to get to that topic of disposal when we have so many
5 technical issues and obstacles right now that prevent us from
6 getting our waste into a form in which it could even be
7 considered for disposal.

8 Now, vitrification has been the target and it's
9 been the goal at Hanford for several decades, but we're not
10 there yet. I understand that this Board does not have
11 purview over the design and construction of the Waste
12 Treatment Plant, but when you look at those issues and you
13 look at the overall scope of disposal and disposal form at
14 Hanford, can't get there without talking about the urgency we
15 have in trying to resolve the problems of these facilities
16 and getting vitrification operational at the Hanford site.
17 That's our goal. And I will second Suzanne's comments that
18 there needs to be certainty that the eventual form of the
19 treated waste at Hanford does have a pathway into a geologic
20 disposal facility.

21 So let me talk about a couple of other waste
22 streams that concern us at Hanford, and some of this has been
23 discussed already by Mr. Jim and by Suzanne. But these waste
24 streams I'm going to talk about seem more and more to be
25 slipping under the radar at Hanford. The first is the waste

1 that is leaked from the tanks, which was just a discussion
2 point. In our view as well, this is high-level waste, at
3 least a million gallons estimated to have been leaked from or
4 intentionally spilled from single-shell tanks at Hanford.
5 And, as you follow the news, you know that Secretary Chu has
6 announced that there are least six actively leaking single-
7 shell tanks, so they're adding to the burden of waste in the
8 vadose zone. And though the amount that is being leaked
9 today is relatively small, we have no assurance that will be
10 the case two weeks from now, two years from now, two decades
11 from now.

12 The waste is spread throughout the vadose zone all
13 the way to the groundwater. There are twelve single-shell
14 tank farms at Hanford. They are shown in brown in that
15 graphic. Every one of those twelve single-shell tank farms
16 has at least one, and in most cases far more than one,
17 leaking single-shell tanks over the past many decades.

18 This is a one-volume, one-copy--I see some groans,
19 so people do know what this is. This is the final Tank
20 Closure Waste Management Environmental Impact Statement at
21 Hanford. We ended up getting three in the mail. One was,
22 believe me, sufficient. This document does detail potential
23 environmental and human health risk in the future caused by
24 waste that's in the vadose zone at Hanford, including leaked
25 tank waste. The Department of Energy has been fairly clear

1 that their intent is to leave most of this waste in the soil
2 where it is. They do express a strong preference for
3 landfill closure of the tank farms, although in the EIS, the
4 final one, they do acknowledge it may be necessary to remove
5 or immobilize some of the waste.

6 But, as we just had this discussion a few minutes
7 ago, leaving this waste in the soil does nothing to change
8 the fact it is, in our view and many others', high-level
9 waste. As Suzanne mentioned, the active leaking out of a
10 tank does not in any way remove key radionuclides, does not
11 result in any kind of immobilization of that. So we're
12 deeply concerned as well about leaving the leaked tank waste
13 in the subsurface.

14 Another waste stream of concern at Hanford is
15 nearly 2,000 capsules of cesium and strontium. Cesium and
16 strontium was removed from the Hanford tanks as long as 40
17 years ago to remove heat from the waste tanks at Hanford.
18 It's stored currently underwater in a water-filled basin just
19 adjacent to Hanford's B Plant. The canisters represent the
20 largest concentrated source of curies at the site, about 100
21 million or more.

22 In the past there's been a number of discussions
23 about what to do with these. There has been consideration,
24 can we direct disposal of these canisters into a geologic
25 repository. There's been talk about blending it in with the

1 high-level waste stream at the vitrification plant. In
2 recent years it seems there has been more and more discussion
3 by the Department of Energy about storage on site, allowing
4 the radioactivity to decay and then sometime in the future
5 shallow land disposal burial at Hanford.

6 Doing that, though, ignores the fact again--you
7 know, if you look at a process to reclassify waste, the
8 cesium and the strontium are some of the key radionuclides
9 you would pull out of that waste stream in order to
10 immobilize that in a geologic disposal facility. So this is,
11 you know, the actors that you would pull out. And we also
12 want to make sure that these eventually go to deep geologic
13 disposal.

14 One other waste stream let me mention briefly that
15 I'm not sure the Board has heard about. You would be hard-
16 pressed to find it in very many of the documents at Hanford
17 that talked about cleanup. Back in the 1980s DOE made some
18 vitrified glass logs for the West German government, and
19 these logs were intended to be a heat source in studies of a
20 deep geologic disposal facility in Germany. Those logs never
21 left the Hanford site. They're stored in casks in the 200
22 West area and have several million curies of radioactivity.
23 This is a waste stream as well that we want to see eventually
24 leave Hanford and go into a geologic disposal facility. We
25 believe it's high-level waste under our interpretation; we

1 believe it would be as well under many others.

2 Finally, I recognize that the Board has interest in
3 the spent nuclear fuel at Hanford. In our opinion and I
4 think many others, the spent nuclear fuel is stored safely at
5 the canister storage building and, we believe, can be stored
6 safely for several decades to come. Eventually, though, we
7 do want to see this waste as well leave the site and find a
8 home somewhere in a deep geologic disposal facility.

9 So thank you again for your time and the
10 opportunity to come and speak with you, and I'd be happy to
11 try and answer any questions. I'm glad those really tough
12 ones about reclassification went to Suzanne first.

13 PEDDICORD: Peddicord, Board. So, Mr. Niles, is the
14 list of the four topics the State of Oregon priority in terms
15 of the order to be addressed (inaudible)?

16 NILES: I did not put them in an order of priority. I
17 would have to think about it a moment. You know, we didn't
18 even address really the waste that's in the tanks, and that
19 would be our first priority of dealing with that 56 million
20 gallons of waste. But in terms of issues that I thought the
21 Board would find relevant, those are some of the issues that
22 we have here, yes.

23 FRANKEL: Frankel of the Board. So I am in complete
24 agreement with you about the leak plume under the tanks being
25 high-level waste and a concern, but I don't think we heard

1 any viable approach for dealing with it. Is there any
2 possible idea or concept for remediating those plumes that
3 are going all the way down to the water table? And it's a
4 huge thing.

5 NILES: It is a huge issue. And, you know, the issues I
6 gave here were mostly policy issues. That is certainly a
7 policy and a very big technical issue. It has only been in
8 recent years that the vadose zone at Hanford has really drawn
9 much in the way of attention and scientific study. Ernest
10 Moniz, as you know, the nominee as Secretary of Energy, back
11 when he was, I think, undersecretary of the Department of
12 Energy, referred to the vadose zone at Hanford as virgin
13 territory, unknown territory. We really didn't understand
14 what was going on there.

15 There was a focus a few years ago to begin shifting
16 and looking at alternatives for what to do, how to immobilize
17 or retrieve some of the waste. It kind of fell victim to
18 some of the funding cuts of a few years ago, and we're really
19 not very far along in terms of coming up with viable
20 solutions. We don't have a solution to offer to say this is
21 what you should do, only that this needs additional study
22 because we have concerns about this waste and it is high-
23 level waste.

24 GEPHART: Any other--yes.

25 ZOBACK: Mary Lou Zoback, the Board. This kind of came

1 up briefly yesterday, and it hasn't been talked about. Have
2 radionuclides gotten into the Columbia River? All I remember
3 is someone saying that there was ten times more uranium
4 coming from fertilizer than there was from Hanford.

5 NILES: There is radioactive material that's entering
6 the Columbia River. If you go back to the operational years
7 in the '40s, '50s, '60s, there were huge amounts of
8 radioactive materials getting into the river and could be
9 detected in shellfish up and down the Oregon and Washington
10 coastline. Since the shutdown of the last--

11 ZOBACK: Coastline?

12 NILES: Of the coastline all the way--

13 ZOBACK: Of the river or the ocean?

14 NILES: Of the ocean.

15 ZOBACK: Oh my gosh.

16 NILES: Of the ocean. There was that much volume of
17 material going into the river. When the last single-pass
18 reactor was shut down in 1971, that dramatically reduced the
19 volume of radioactive material going into river.

20 ZOBACK: Were they actually putting the effluent from
21 the power plant directly into the river?

22 NILES: Yes.

23 ZOBACK: Ah, geez.

24 NILES: It was a single-pass design that carried
25 (inaudible). So there are small amounts of radioactive

1 materials and chemicals entering the Columbia today.

2 ZOBACK: Even today.

3 NILES: It's pretty closely monitored. The State of
4 Oregon had an environmental monitoring program on the
5 Columbia River specifically for Hanford that went from the
6 mid-1960s till the mid-1990s and basically shut down because
7 at that point they'd been measuring zeroes year after year
8 after year. And a few years ago we did an analysis, separate
9 sampling, just to verify that assumption. So there is small
10 amounts going in. They are quickly diluted. The concern we
11 have is in the future if a whole lot more gets into the
12 river.

13 ZOBACK: I guess this is another question related, and
14 I'm not sure who it's for. I'm actually a seismologist, so a
15 lot of this is way beyond me, but the reason you have those
16 hills out there are from earthquakes in the past. And I
17 think there was discussion about buildings and their
18 structural safety, but I just think about all the underground
19 pipelines and the joins in all the pipelines, and a lot of
20 those pipelines are awful old now. And when they were
21 talking about pumping all the waste from all the tanks into
22 that treatment plant, I just thought about miles and miles of
23 underground pipelines. Has anyone looked at that from the
24 safety--from strong shaking and soft material which would
25 amplify this shaking?

1 NILES: Well, I would add that the concern we would have
2 as well is the integrity of the tanks themselves and that you
3 could have a dome collapse. You could have just a larger
4 damage to the tanks, which could really exacerbate the
5 problem with leakage. Beyond that, I am not a seismologist
6 and can't answer those.

7 ZOBACK: Suzanne, where is the State of Washington on
8 seismic concerns?

9 DAHL: They've done a couple of looks at the tank farms
10 as far as different ground motion events, and there is
11 definitely the concern that if you--I think the concern is
12 when you start thinking about the most probable earthquake or
13 the one that the Waste Treatment Plant is designed to, and
14 it's a one-in-two-thousand-year reoccurrence interval. But
15 if you had that type of ground motion at that plant, you
16 would also have that type of ground motion at the tanks and
17 pipelines that are now, you know, 40, 50, 60 years old. And
18 so our concern is, while we need a safe operating facility,
19 we also need to realize that the waste is currently sitting
20 in tanks that probably would have equal problems, if not
21 greater problems, withstanding the same event.

22 ZOBACK: And no option--if you needed excess storage
23 immediately, no option.

24 GEPHART: If we may go on here, since we're a little bit
25 tight on time, but any of the folks that are on these panels,

1 please see them afterwards during the break and ask questions
2 there. They'd be happy to be available.

3 Very quickly, our next speaker is Steve Hudson. He
4 is the newly elected--shall I say that?--newly elected
5 Chairman of the Hanford Advisory Board.

6 HUDSON: Thank you, Roy.

7 I should begin by saying, I am also from south of
8 the Columbia River. I am an Oregon resident as well. And
9 people, when they ask me what I'm doing these days after
10 retiring from teaching college English for 39 years and I say
11 I'm a member of the Hanford Advisory Board and now the Chair,
12 they look at me askew, because they know I'm a rhetorician
13 and a grammarian, although I do explain that as an
14 undergraduate I was a chemistry and math major, and my son is
15 a professor of physics at Penn State, and my daughter and her
16 husband are both environmental chemists at Cal State
17 Fullerton. So many of the topics today delight me even if I
18 don't really understand everything you're talking about.

19 I want to just give you very briefly and pretty
20 compactly a bit of insight into the Hanford Advisory Board,
21 because clearly the issues that the Hanford Advisory Board
22 deals with are well within the purview of this particular
23 committee.

24 For the Hanford Advisory Board, the most important
25 technical issues associated with the eventual disposal of

1 high-level radioactive waste and spent nuclear fuels are
2 those that fail to adequately address the health,
3 environmental, and economic conditions and needs of the
4 Hanford community. And when I speak of the Hanford
5 community, I talk about that community that extends all the
6 way from the Idaho border to the Pacific Ocean, from the
7 Canadian border to the California border. Because we all are
8 members of the Hanford Advisory Board, we represent those
9 communities of peoples, those interests.

10 In effect, the most important issues for the HAB,
11 for the Hanford Advisory Board, are those issues which allow
12 the best combination of actions to be taken, the best balance
13 of alternatives. Now, for clarity--and let's begin by noting
14 that because the Hanford Advisory Board's mission is to
15 provide the TPA agencies--the DOE, the EPA, the Washington
16 State Department of Ecology--with advice on major high-level
17 cleanup decisions, its response to issues, and its response
18 to issues that are raised by the TPA membership and by our
19 own stakeholders, we do not typically or even usually address
20 technical issues. We don't write about technical issues.

21 Now, that said, it does not mean that when we
22 discuss various kinds of alternatives that the HAB advice is
23 not shaped and molded by technical discussions, for certainly
24 such discussions do take place (inaudible) and say the Tank
25 Waste Committee, the River and Plateau Committee, the Health

1 and Safety Committee, as one might expect. But such topics
2 are also addressed from time to time in the Budget and
3 Contract Committee and in the Public Involvement Committee
4 where inevitably, after we have had a public meeting,
5 somebody will say, "How do you deal with a question like 'Why
6 don't you just do this?'" And we are trying to find answers
7 that would be credible for the constituents that we often
8 have to deal with.

9 And while I should also note that there are a
10 number of HAB members who would be comfortable in discussing
11 many of the topics you raised this morning--that is, talking
12 and discussing the chemical and structural roles of bismuth
13 borosilicate melts with regard to the vitrification of high-
14 level radioactive waste and similar topics--we certainly have
15 people on the HAB that are, in fact, able to do so.
16 Discussions such as that, however, rarely occur in advice.
17 And because they are not showing up in advice, I do not like
18 to speculate about what the HAB may or may not have said
19 about a particular kind of resolution.

20 In essence, I think it's important to realize that
21 the Hanford Advisory Board is less concerned with ranking the
22 technical complexities of high-level waste treatment
23 alternatives than with how those alternatives impact and
24 mirror core HAB values and principles. For example, when
25 writing advice, the Hanford Advisory Board will consider

1 whether or not a proposed technical solution protects the
2 broader environment and ensures that the solutions arrived at
3 do not harm anything during cleanup, that those proposed
4 technical solutions provide for the efficient vitrification
5 of Hanford tank waste and the disposal of the tank waste
6 safely and permanently, because, as you've heard from anybody
7 on this panel today again and again, vitrification of
8 Hanford's waste and the subsequent disposal of the vitrified
9 high-level waste in a deep geologic repository is one of the
10 most critical components of the Hanford cleanup process. And
11 if the technical solutions develop and deploy new technology,
12 the Hanford Advisory Board would expect that those
13 technologies must do so without impeding cleanup.

14 Now, of course, the Board will also address with
15 equal concern--and these are the kinds of issues that you're
16 probably not typically involved with--concerns about: Does
17 the solution protect worker safety and health? Does the
18 solution protect the Columbia River? Does the solution
19 protect and restore the groundwater? Does the solution
20 involve the public? Now, for the Hanford Advisory Board, we
21 devote a lot of time to making sure that the public has a
22 conduit to provide their ideas, their concerns to the
23 decision makers. And as what Todd Martin, a former chair of
24 the Hanford Advisory Board, advice allows the public to
25 inform decision makers about what they need to care about.

1 And essentially that is one of the main obligations of the
2 Hanford Advisory Board.

3 So we involve the public because that is good
4 practice; we involve the public because it is a good basic
5 prerequisite for environmental justice; and we involve the
6 public because it represents fair and democratic decision
7 making. That is, if we wish to have public support for the
8 decisions we're making, we have to make sure that they are
9 provided with information that is broad, inclusive, open, and
10 accessible.

11 Now, one of the things that I did when I first
12 became a HAB chair was to read through the various pieces of
13 advice, and over 20 years the Hanford Advisory Board has
14 produced 250-plus pieces of advice. And I have to admit to
15 you, when I read through the advice, I was far more taken
16 with the quality of the writing and the language and the
17 various ways of expressing particular issues had changed over
18 time than I was with the content. So when I had to read it
19 to prepare for this presentation today, I had to look at the
20 content. I didn't find that quite as interesting as my first
21 read.

22 So, in reading through those 20 years of advice,
23 what I noticed--and it's very important--one, the high degree
24 of consistency. From 1994 to 2013 the Hanford Advisory Board
25 has held essentially to the same concerns and issues that

1 they wish to be resolved. Over those years the changes have
2 been the early 1994 positions and using phrases like, "We
3 don't care what you do out there, just get on with it," kind
4 of the colloquial. Today, if I read the advice, it's far
5 more--it has a stronger sense of urgency; it has a strongly
6 focused sense of, "This is what needs to be done."

7 And I'll flip ahead, because I know we're really
8 short on time.

9 In essence, if you go through those pieces of
10 advice, clearly the principle and goals are familiar,
11 especially the need to address all waste streams; clearly the
12 need to address all processes, facilities, and products and a
13 need to secure vitrified high-level waste in a deep geologic
14 repository.

15 However, it should also be noted that, at least
16 Hanford Board advice about high-level waste, because of that
17 consistency, shows that this reflects not only something
18 about the nature of the Hanford Advisory Board, but it also
19 reflects something very, very much about our constituency,
20 about what the public is interested in and what the public
21 would like to see happen. And, as I said, one of the 1994
22 statements was to get on with it, to get the cleanup done,
23 while remaining--it is important, I think, that we remain
24 sufficiently patient to avoid embracing short-term technical
25 solutions which fail to produce quality, long-term results.

1 And, as I said in the beginning, the most
2 important--the most important technical issues for the HAB
3 are those issues which allow the best combination of actions
4 to be taken, the best balance (inaudible) to take place and
5 to, in fact, allow decision makers to get on with it.

6 And I thank you.

7 GEPHART: Thank you very much, Steve, for allowing me to
8 nudge you a little bit. You're an example to my next three
9 speakers. We're doing our best. I'm going to nudge folks a
10 little bit quicker, because we do have some other schedules
11 coming up--my apologies--so you'll feel a little bit of a
12 push.

13 Our next speaker is Pam Brown Larsen. She is
14 Executive Director for the Hanford Communities and member of
15 the Hanford Advisory Board. Pam.

16 LARSEN: Thank you, Roy.

17 I am going to skip through some of the prepared
18 comments, because I think that a lot of this has been
19 addressed already in your meeting. But I do want to welcome
20 you today on behalf of the Hanford Communities. It is an
21 association of the cities of Richland, Pasco, Kennewick,
22 Benton, and Franklin Counties and the port of Benton. Those
23 entities came together 19 years ago to work together on
24 Hanford issues, and I've been Executive Director for those 19
25 years. Also, to bring a sense of immediacy to the concerns

1 of people in this region, keep in mind, the City of Richland
2 draws its drinking water from the Columbia River just south
3 of the Hanford site, so we pay a lot of attention to what's
4 going on.

5 You have had discussions about the tanks, our 1,900
6 canisters of cesium and strontium, the 2,300 metric tons of
7 spent nuclear fuel, but I also want to draw to your attention
8 that on the Hanford site we have 570 tons of commercial spent
9 fuel at Energy Northwest in dry storage, and we have 124--and
10 that may not be a current number--decommissioned submarines.

11 Hanford is in effect an interim storage facility,
12 because we have no place to send our high-level waste that is
13 by law destined for a deep geologic repository. And I want
14 to point out that there are financial consequences to Hanford
15 because of the fact that there is no place to send this
16 material. Funding for guards and guns that protect spent
17 nuclear fuel comes out of the cleanup project, and this is a
18 significant amount of money each year, tens of millions of
19 dollars.

20 High-level glass that will be produced by our Waste
21 Treatment Plant must be stored until it can be shipped;
22 therefore, funding will have to be pulled from cleanup
23 efforts to design and build a facility to store this
24 material. We're calling it a temporary storage facility, but
25 it's being designed so that it can be expanded if needed,

1 because there is no place to send the high-level glass. One
2 of the consideration in the country now is whether interim
3 storage could include high-level vitrified glass, and we're
4 very hopeful that that would be a possibility. We believe
5 that there should be some compensation to our community and
6 others who are interim storage sites. And I'll go into this
7 in just a bit.

8 In regards to the path forward, we support the
9 recommendations made by the Blue Ribbon Commission to
10 establish a new entity dedicated to implementing the nation's
11 high-level waste storage program. We support the licensing,
12 construction, operation of a permanent geological repository
13 for high-level waste, and we are encouraged that there are
14 communities in this country that have asked to be considered.
15 We believe that the Nuclear Regulatory Commission should
16 complete the evaluation of the Yucca Mountain license
17 application and that public hearings should be held as
18 required by law.

19 In regards to the future of the Hanford site,
20 Hanford encompasses 586 square miles. Nearly 90 percent of
21 the site will be remediated in the next few years. However,
22 the highly radioactive tank waste will require decades of
23 work to solidify and encapsulate. Our community has
24 requested 1,640 acres of land just north of the City of
25 Richland to develop an energy park to attract new green

1 manufacturing jobs, and there is a strong interest in new
2 missions.

3 While we believe that there should be some quid pro
4 quo for communities hosting interim storage, we do not seek
5 monetary compensation as the nation struggles to reduce the
6 deficit. We do ask that Hanford be given consideration for
7 new future missions. And we believe that our future lies not
8 just with the Department of Energy's Environmental Management
9 program, which will be engaged for years in dealing with tank
10 waste, but with other DOE programs.

11 I would like to just briefly address some of the
12 questions that have been raised. There were questions raised
13 about the waste incident up to reprocessing, which is
14 referred to as 3116 of the Defense Authorization Act. When
15 we were approached about whether Hanford should be included
16 in that discussion, we pointed out that there is within the
17 Tri-Party Agreement a path forward for dealing with tank
18 waste, closing tanks, and closing tank farms that has been
19 vetted in this region and that we think that should be the
20 path forward. It was pointed out that because this material
21 is RCRA material, that that may not be adequate. And so we
22 have asked that there be something written to put into
23 federal law that validates the Tri-Party Agreement process
24 that we have already chosen.

25 Also, as has been pointed out, there are not just

1 tanks, but there are pipes that are potentially leaking; and
2 we concur with a statement that these leaking T-farm tanks
3 are the canary in the coal mine. There is going to be more.
4 And so we emphasize to those who have authority that we need
5 tank waste treatment as soon as possible. It's extremely
6 important to our region. And one of those alternatives that
7 is going to be considered is the possibility that some of the
8 T-farm tanks, based on process knowledge, meet the definition
9 of what material can go to the Waste Isolation Pilot Plant.
10 If that is acceptable to the residents of New Mexico, we
11 would really like to see that alternative pursuit.

12 And, finally, there was a question about concern
13 about leak tanks. I also want to point out to you that
14 during the years of operation at Hanford, the highly
15 radioactive waste did go to the tanks, but over 450 billion
16 gallons of liquid were poured into trenches, and a lot of
17 chemical and radionuclides are therefore in the ground
18 between the surface and the vadose zone. We're very pleased
19 that a pump-and-treat facility has been put on line, a
20 \$120 million facility that we benefitted from stimulus
21 funding. That facility is going to operate for many, many
22 years. We think that's the right thing to do to get those
23 contaminants captured before they move towards groundwater
24 and the river, but it is a significant problem for us in the
25 long-term.

1 Thank you very much. And I do have an issue paper
2 that I forgot to provide. And so, Roy, that's copies of
3 that. So I'd be happy to answer any questions from a local
4 government perspective.

5 GEPHART: If I may exercise moderator preference,
6 because of the time, if anyone would like to speak to Pam,
7 please do so afterwards. We're just going to sort of keep
8 moving on a little bit.

9 Our next-to-the-last speaker is Gary Petersen.
10 Gary is the Vice President for Hanford Programs under the
11 Tri-City Development Counsel.

12 PETERSEN: In the interest of time and safety, both
13 yours and mine, I'm not going to walk behind that
14 (inaudible).

15 Let me point out, too, in full disclosure, I'm
16 going to pass this around as I talk, because I want Al to
17 talk too. I'm not going to repeat the numbers, but in the
18 interest of full disclosure, I am one of the three people who
19 brought suit against NRC, DOE, and the President of the
20 United States relative to Yucca Mountain. We were later
21 joined, thank heaven, by both the Attorney General of
22 Washington State and also Savannah River. So I am a part of
23 that lawsuit. We hope the court will rule here shortly. And
24 I want to raise up and talk about this issue both personally
25 and on a broad basis, and I want to start by making it very

1 personal.

2 I was born in 1940. Those tanks started to be
3 built in 1945. And the Blue Ribbon Commission has now said
4 that there will be a place for this waste come 2046. At that
5 time I'll be 106 years old. My predecessor, Sam Volpentest,
6 lived to 101. I don't anticipate being that lucky. I also
7 want to point out that I take it personally because I live in
8 North Richland, and I drink the water from the Columbia
9 River, and I have since 1965. So when we talk about the
10 safety of the river and the radionuclides going into the
11 river, I take it very personally.

12 Let me raise up the issue, though. At Hanford, as
13 Pam said, we have all types of waste. We have commercial
14 waste, the 570 tons from Energy Northwest. We have the
15 Trojan Reactor. I'm passing around a picture of the
16 submarine reactors. Incidentally, in that stack there are a
17 variety of pictures. When we talk about 124 Navy reactor
18 cores, they are both submarine and cruiser missiles. And
19 very shortly over the next few years we're going to be
20 receiving eight reactors from the aircraft ship Enterprise.
21 And so we have it all. We also have a very small Washington
22 State mostly medical low-level waste repository out there as
23 well.

24 The issue in part for me is, I deal with Congress.
25 I deal with them on a regular basis. And I'm talking about

1 not just once a day or once a week, but very often. The
2 issue is that we don't have enough money to clean up all of
3 these weapons complex sites. That's not just here. That's
4 Savannah River, that's Oak Ridge, that's Paducah. Everywhere
5 you're concerned with, we have an issue with cleaning up all
6 those sites.

7 Senator Murray has stated very often that we don't
8 want to rob from one site and give to another, nor do we want
9 to rob from within a site and give to another. In other
10 words, there are three DOEs on this site. We've have twelve
11 DOE manages in the last eight years, and it's very hard to
12 keep a consistent, attentive program going when you have that
13 kind of turnover both at headquarters and here. And you have
14 three DOE site managers, all of whom report to the top.

15 And so one of the things I want to express is, we
16 have the law. Currently we think that we are breaking the
17 law with Yucca Mountain. We have the law. Russell Jim did
18 an excellent job of outlining where that law rests. What we
19 don't have is we don't have the funding to do it. And I know
20 what sequestration means; I know what the impacts are
21 currently. When we have 2,000 of our staff go on furlough
22 and another 300 to 600 laid off, there is no way that you can
23 meet the milestones that we've got in front of us unless
24 something is done.

25 So my request to the Board--and I'm trying to be

1 short--is that you pay attention to how we actually go
2 forward when money is in the way. And so I leave with you
3 the message: it's personal, I think those tanks were not
4 meant to live until 104 years old, and so we need help. We
5 need your assistance to move this whole program forward.

6 With that, I was short.

7 GEPHART: Gary, thank you very much. And I would give
8 my apology for those that I have been pushing just a little
9 bit on the time, but thank you for helping us.

10 Our last speaker before a quick break is Al Boldt.
11 Al Boldt is representing the non-profit public interest group
12 Hanford Challenge.

13 BOLDT: Well, the previous speakers have stolen almost
14 all of my thunder, so this ought to be pretty fast.

15 We have nine reactors, 100,000 tons, have processed
16 most of it, a thousand tons of spent fuel. 2,000 tons of
17 spent fuel, I want to say here, is what's critical. It's in
18 about two-foot-by-13-1/2-foot-long containers, multiple
19 canister overpacks, about half-inch stainless steel. Biggest
20 thing difference on these is, the contents are spent fuel,
21 uranium metal fuel clad and zirconium. Some of them are
22 hollow; the bulk of them are shattered, broke, in pieces
23 being stored in a basin for twenty years and contain uranium
24 hydride that was formed in the water. This field is
25 pyrophoric, and I'm just making that statement. The previous

1 Yucca Mountain (inaudible) contents will not be pylophoric.

2 This material will ship to your new repository
3 about 2048 schedule, either needs new criteria or requires
4 treatment to the old criteria. Out of that spent fuel is
5 also associated with it. In the basin we've got .2 cubic
6 yards of spent fuel fines. This is stuff that would pass
7 either an eighth-inch or quarter-inch mesh, really fine,
8 really bad pyrophoric, and other stuff that's diluted with
9 dirt.

10 The treatment of this has been deferred for a
11 number of years. They're going to do a technology selection
12 in 2015 just to decide what to do with it. It's to be
13 shipped to the--it's still out there on the banks of the
14 Columbia River. Supposed to ship it to T-plant, decide what
15 to do for treatment. This is still high-level waste. The
16 plan is that the sludge is lower-level waste, will be
17 classified as remote handled TRU and go to WIPP.

18 The 100,000 tons or 98,000 tons we processed went
19 through four reprocessing plants (inaudible) the process, and
20 the tank farms would go to uranium recovery operation from
21 the early plants. We took out strontium and cesium to get
22 the heat out so we could get more waste in the number of
23 tanks we had. We concentrated the supernate, got it hot, put
24 it back in the tanks, let it crystallize; so we have an extra
25 sludge, salt-cake, and supernate. In amongst this was a

1 number of tank transfers managing tank waste, so we got a
2 somewhat mixed inventory. 67 of them leaked. Mention we
3 have recovered the strontium/cesium.

4 Now, the EIS, that four-foot-tall stack of
5 documents that were shown, was issued December without
6 preferred alternatives for some of the critical tank waste
7 treatment operations, and no records of decisions obviously
8 were deferred also.

9 As you've seen, we retrieved the wastes, then the
10 "empty" tanks. Empty is--a number of cases studied--what I'm
11 saying is one percent of the residual in the tanks, one
12 percent of the tank contents, the tank farm infrastructure
13 that you mentioned, all of the pipelines that are buried over
14 the tank farm, miscellaneous underground tanks, and the
15 vadose zone are treated for disposal. The retrieved wastes
16 are separated into three waste classifications for treatment
17 and disposal.

18 Now, the disposal of our failed melters, which is
19 another high-level waste here, I projected twelve melters
20 based on our System Plan 6. The tank farm closure, which is
21 this combination of vadose zone contamination and stuff we
22 left behind, is about two percent of high-level waste
23 inventory.

24 Now we go down the rabbit hole further on how do we
25 change this from high-level waste. We go to the NRC, and

1 they make a determination that this residual inventory after
2 treatment is reclassified as "waste incidental to
3 reprocessing". I want to point out that high-level waste,
4 low-level waste, and waste incidental to reprocessing are
5 specific terms defined, and they're called the Federal
6 Register. Means something to a lawyer. So that's why we
7 call it low-activity waste. It's not a legal finding.
8 Low-level waste is also defined by source in the regulations.
9 The NRC makes this determination. There's three rules there.
10 Basically it's got to be safe to the environment.

11 For further discussion and enlightenment, there's
12 an attachment. The "Legal Bar Against Reclassifying High-
13 Level Waste" at Hanford is on this attachment. The
14 discussion you started entering into is by no means resolved
15 legally.

16 Our retrieved waste that we separate, again, there
17 was no preferred alternative and no records of decision. We
18 separate three fractions (inaudible) TRU after twenty tanks,
19 both contact and remote handled. NRC has to reclassify this
20 high-level waste. The vitrified high-level waste for
21 disposal at the National Repository, that's the easy one.
22 The low-activity waste we've been talking about, the
23 (inaudible) Waste Treatment Plant will process about 30
24 percent of that waste, the total waste. We didn't select an
25 alternative or the waste form that they want to go ahead

1 with. The current Waste Treatment Plant and the one-third of
2 the waste that's LAW has a provisional classification as
3 "waste incidental to reprocessing" by the NRC. That
4 provision is, if you change the waste form and a couple other
5 little nits in the provision, it says come back, it's no
6 longer LAW.

7 So the NRC's waste classification says it's glass
8 (inaudible). That's what their--

9 GEPHART: You have to be wrapping up pretty quickly.

10 BOLDT: Okay. As was mentioned before, we have some
11 other materials, the capsules, a couple thousand of them.
12 Probably not mentioned, the WESF Basin has had damage to the
13 concrete walls behind the liner. They're looking at
14 potentially putting the stuff in dry storage in the future;
15 in 2017 we'll decide what to do. The future repository has
16 to decide whether they're going to take this overpack
17 material that is a water soluble source, just another
18 criteria for the new repository.

19 We have the 34 German logs, and they're currently
20 in eight steel storage casks. These storage casks have
21 (inaudible) life of 40 years. We need to recertify or
22 replace them in 2037, eleven years before potential shipment
23 to the repository.

24 There is also a second attachment on here that goes
25 into the discussion of the LAW waste, the indecision on what

1 to do about alternate waste forms, just more detail of what
2 you've heard.

3 GEPHART: Thank you, Al.

4 All Board members, all staff, all that you have
5 seen on the screen, we have electronic copies, so those will
6 be forwarded to you so you have all the information that has
7 been provided.

8 And for the speakers, I thank you very much. If
9 there is anything else that you wanted the Board to see or to
10 read that you didn't present, feel free to give it to me,
11 give it to the gentlemen who are up here, and we will assure
12 you that the Board will receive that.

13 We have about ten minutes before our next session,
14 so coffee, tea, restrooms. Let's go. Thank you very much.

15 (Whereupon, the meeting was adjourned for a brief
16 recess.)

17 EWING: Please, if you could take your seats. In
18 particular, if you can find a panel member, bring them in.
19 For the balance of the afternoon, we're going to shift gears
20 a little bit, and the presentations will cover wider ground.

21 The first this afternoon is by Bill Boyle from the
22 Office of Nuclear Energy, Department of Energy. He'll be
23 giving us an update on the potential for direct disposal of
24 dry storage containers currently in service in nuclear power
25 plants.

1 BOYLE: Thank you for this opportunity. And as Chairman
2 Ewing just said, I'm glad there was the break right now,
3 because we're making a fundamental break from all the talks
4 that preceded today. I'm not going to talk about glass. I'm
5 not going to talk about DOE defense wastes or anything in
6 tanks. We're going to discuss commercial spent nuclear fuel
7 in dry storage at the power plants today. As two of the
8 speakers, Ms. Larsen and Mr. Petersen, said, there is an
9 example of that right up the river here that we didn't see
10 yesterday on the tour.

11 So what I'm going to talk about so that everybody--
12 we're all speaking the same language here, it's the direct
13 disposal of dual-purpose canisters. Well, what are the two
14 purposes of the dual-purpose canister? They have been
15 designed and licensed by the Nuclear Regulatory Commission
16 designed by vendors licensed by the NRC. The first purpose
17 is the storage of spent nuclear fuel, and the second purpose
18 is for the subsequent transportation of that spent nuclear
19 fuel.

20 Bear with me for a second while I read this slide.
21 The rest of my slides after this are all technical, but this
22 is a technical presentation that does not take into account
23 the contractual limitations under the standard contract.
24 Under the provisions of the standard contract, DOE does not
25 consider spent fuel in canisters to be an acceptable waste

1 form, absent a mutually-agreed-to contract modification. To
2 ensure the ability to transfer the spent fuel to the
3 government under the standard contract, the individual spent
4 fuel assemblies must be retrievable for packaging into a
5 DOE-supplied transportation cask.

6 So this is my way of telling people that there is a
7 contractual issue related to the direct disposal of these
8 dual-purpose canisters. I've alerted you to it, and I'm not
9 going to speak about it anymore. I'm not an attorney; I'm
10 not a contracting officer. It's just to let people know that
11 there is--the DPCs are entangled in a contract. The standard
12 contract that's referred to is the contract that the
13 government signed with each of the power utilities where the
14 utilities were charged the mill per kilowatt hour, give the
15 money to the government. That's the contract that's being
16 referred to. And the government's view, not only the
17 Department of Energy but the Department of Justice, is: Go
18 read the contract; we don't have to take the DPCs. So that's
19 an issue for others to look at.

20 What my group is looking at is: Let's pretend that
21 the contracts never existed. Let's just look at the existing
22 DPCs and ask ourselves, Can we dispose of them? They weren't
23 designed for disposal, but now we're looking at them after
24 the fact and asking ourselves, Could we do it? What are the
25 challenges?

1 So for the remainder of this talk I'm going to
2 focus on what is a dual-purpose canister, show some
3 information about them, get to the fundamental question of
4 can we dispose of them as is or do we need to take out the
5 spent fuel and repackage it, and then a brief description of
6 our ongoing research and development.

7 So this slide has various information and pictures
8 from one of the vendors of such storage systems in the United
9 States, NAC International, and this figure in the middle
10 tells us what we need to know. I refer to them as DPCs,
11 dual-purpose canisters. This diagram refers to it as a
12 transportation and storage canister, and it's this cylinder
13 in the middle that's steel. Here's another example. This
14 is a picture of that cylinder in the middle. Each of these
15 channels would get a spent fuel assembly, and then a lid is
16 put on this, sometimes bolted, more commonly welded. And
17 then that's put inside another sleeve, if you will, of
18 concrete, which you see down here.

19 So this is the dual-purpose canister. Here it's in
20 its storage mode. Eventually, when it comes time to
21 transport, this DPC comes out; it's put into a different
22 transportation cask; and then it's transported to wherever it
23 needs to go. What remains behind are all these concrete
24 cylinders. And, essentially, they should be clean. The DPCs
25 shouldn't leak; there should be no activation products.

1 These are big, these concrete shells, so big that they're
2 typically made on site. There is no factory that produces
3 these, which would be the general preference, you know, have
4 somebody set up and mass produce them. But they're so large,
5 they're typically produced right at the reactor sites.

6 So this was one of the vendors in one of the modes,
7 vertical. Another thing, the storage works for both
8 pressurized water reactor assemblies, PWRs, and also boiling
9 water reactor assemblies, BWRs. For those who aren't reactor
10 experts, the PWR assemblies are typically bigger. That's why
11 you end up--whether it's PWR or BWR, you get about the same
12 amount of material. So in terms of--it only takes 37 PWR
13 assemblies to equal 87 of the BWRs.

14 Here is an example of horizontal storage. You can
15 look down here at the right-hand side. The only horizontal
16 system is called NUHOMS®. It comes from another vendor,
17 Transnuclear/AREVA. But, similar to the one I showed on the
18 previous slide, they're very big. You can always look at the
19 humans in the photos and get that impression.

20 And, finally, there is a third vendor in the United
21 States. This diagram on the right was very similar to the
22 first one I showed with the DPC inside the concrete
23 oversleeve. This diagram here shows--this provides for
24 below-grade storage, helping to mitigate aircraft crash
25 hazard.

1 Now, this slide is courtesy of AREVA, one of the
2 vendors. I could show a similar slide for the other vendors.
3 The main point to get out of this slide that, as time has
4 gone by, the number of assemblies in the canisters has gotten
5 monotonically larger. And the reason for that is, the
6 customers for these dual-purpose canisters are the utilities.
7 And they want bigger ones, not littler one. And why do they
8 want bigger ones? We'll use the 32 as an example, because
9 it's not a prime number like 37. It's much more productive
10 for the utilities to load one 32-capacity DPC than to load
11 eight 4-assembly DPCs. It's just that much more efficient
12 for them. They have a preference for large DPCs. It's their
13 choice. That's what they buy. And, like I said, this is
14 specific to the AREVA products, but NAC International and
15 Holtec® would show a similar trend. They only get bigger
16 with time.

17 And it's this bigness that gets at the fundamental
18 question to dispose or repackage. It's the bigness. Big
19 things are more challenging to handle the transport than
20 little things. For a given heat output per kilogram of
21 material, the more kilograms you have, the more heat you have
22 coming out. And, similarly, with respect to criticality,
23 whatever amount of fissile material you have per kilogram of
24 material, the more kilograms you have, the more fissile
25 material you have. So it's the bigness of the DPCs that pose

1 the fundamental challenges with respect to disposal.

2 Now, this slide I phrased in terms of the pros and
3 cons of direct disposal. I could have turned it around and
4 said the pros and cons of repackaging, you know, you just
5 have to change the words around, but the issues are the same.
6 And, in short, the pros of direct disposal of the existing
7 DPCs is essentially you're close to done. All you have to do
8 is pick them up at the power plants, transport them to a
9 repository, put them in a waste package, you're done. You're
10 not opening them, you're not cutting them open, you're not
11 unbolting them, you're not handling them. All of that
12 handling and repackaging costs time, money. It would all be
13 done safely, but you also are running the risk of accidents,
14 you're pulling an assembly out, you drop it, and things like
15 that. So the big pro of direct disposal is, you don't do any
16 of that, and you're close to being done already.

17 I've already discussed the cons of direct disposal
18 of the existing dual-purpose canisters. It's essentially all
19 related to their large size, the amount of material they have
20 in them. They're thermally hot, they're harder to handle,
21 they require bigger underground openings, which can be more
22 challenging, if you will, and it also--for long-term disposal
23 we cannot analyze the criticality the same way the storage
24 and transportation people do because of the long time frame
25 involved. We have to do the analysis a different way, and so

1 we've got to do research and development on that.

2 So I'll deal first with the big size, and these
3 first three sub-bullets give you some metrics of how much
4 things weigh. For example, the largest waste package
5 considered anywhere in the world by far was the Yucca
6 Mountain Naval spent nuclear fuel, and that was 74 metric
7 tons. And if you look at what a DPC would weigh in its waste
8 package and a transfer cask as you're moving it underground,
9 it will weigh 150 metric tons.

10 Well, as this picture shows, cranes can lift heavy
11 objects, but the challenge with a repository, if you're
12 considering a vertical shaft, is actually the great depth.
13 If you're looking at 2,000 feet, the lengths of the ropes
14 themselves become--you have to lift the ropes as well as the
15 package. It's easy enough to go on line and go to
16 manufacturers of mine hoisting equipment, which they start
17 off with the opposite problem that a repository looks at. A
18 repository generally is looking at, the heavy weight starts
19 with all the ropes coiled up and then lets them down. Mines
20 start with, the ore is loaded at the bottom and they try to
21 lift it up. And if you go check the skip capacities of mine
22 hoisting equipment, it's nowhere near those numbers that I'm
23 showing there.

24 Now, these bullets say the Germans are looking at
25 it, that maybe they can solve getting it down a shaft, but I

1 don't know that they can retrieve. There's requirements for
2 retrieval but, not that I'm aware of, that retrieval must be
3 done the same way you did emplacement. But, having said
4 that, that's the most easily understood form of retrieval is
5 just undo what you did.

6 And vertical shafts, because of the great weight of
7 these items, would pose a bit of a problem. But the good
8 news is, there is potential ways around it. If you've got
9 enough property, a ramp at a shallow grade--one or two
10 percent works--this is the approach the French are taking for
11 their repository. They literally have the shafts start in
12 the next apartment; right? You know, they need a lot of
13 space, and so they started next door, and down they go. If
14 you're limited by space, you can corkscrew down in a shallow
15 ramp. So as long as there is no requirement to use shafts, I
16 think the large size perhaps people can work around. The
17 French also looked at, in addition, if you're challenged by
18 real estate and you can't do a shallow-grade ramp or
19 corkscrew, they even looked at a cogged funicular railway.

20 But other technical issues related to the big size
21 include--these aren't insurmountable, but everything else
22 being equal, large openings underground take a little more
23 work than small openings, and these would have to be bigger
24 openings. And also sealing big openings is perhaps more
25 challenging than sealing a small opening.

1 And here, for post-closure performance, we also
2 have issues with respect to the heat output. These have a
3 lot of material, they would be hot, but there are ways around
4 that. You can just wait, you can ventilate, but there are
5 ways to get around the thermal problems. And, as I mentioned
6 also, the fissile material amount, you cannot do criticality
7 calculations the way they were done for storage and
8 transportation. Because of the long time frame involved,
9 other things can happen.

10 And so we'll probably have to do work on what's
11 called burnup credit, which, for the non-reactor experts,
12 gets down to--it's related to the discussion this morning
13 where Chairman Ewing engaged with Peter Swift. Burnup credit
14 is: Don't make overly conservative assumptions about what
15 you're analyzing; analyze what you actually have and find out
16 what your challenges really are.

17 There have been presentations to the Board on this
18 topic. As a matter of fact, some of our original work on
19 this topic was encouraged by a former Board member, Andy
20 Kadak. We were doing systems studies, looking at smaller
21 packages, and he essentially asked, "Well, why don't you look
22 at the existing packages as well?" And so we took on the
23 challenge.

24 So this slide is just to show you covers of some
25 reports we're doing. In order of the challenges, the

1 technical challenges, people have look at this before all the
2 way back to the 1990s. The Office of Civilian Radioactive
3 Waste Management contractor issued a report on direct
4 disposal. The next M&O contractor, management and operating
5 contractor, they did as well. Electric Power Research
6 Institute has looked at it. And the consensus is, in a rank
7 ordering of the challenges, it's probably criticality,
8 thermal effects, operations in descending order.

9 There is another challenge here that's not related
10 to disposal, but it's more to storage. The concept when
11 these storage systems were developed was, they'd be loaded,
12 they'd be put in, they'd be removed, transported, and
13 disposed. But because we do not have a repository in the
14 United States, there is now discussions of, well, let's
15 remove them from the power plants, put them in interim
16 storage, perhaps transport them later, and it's all this
17 handling, the putting in and taking out, putting in, taking
18 out, transporting.

19 The certificates of compliance--for example, the
20 NUHOMS®, the horizontal systems issued by the Nuclear
21 Regulatory Commission--it states, "There shall be no undue
22 galling, gouging, or scratching," which the utilities
23 actually have to demonstrate before they're allowed to load
24 with spent fuel. But they demonstrated on a one-time basis
25 what we're talking about here is multiple times; and if it

1 got gouged, galled, or scratched, would it call into question
2 the ability to actually store it before you even got to
3 disposal. So that's other work we have to look at as well.

4 So my last slide, it's neither good nor bad. There
5 are pluses with direct disposal. There are definite
6 challenges. We're doing the ongoing R&D to provide
7 information to decision makers in case they want to go down
8 the path of direct disposal or not. That's my last slide.

9 EWING: Okay, thank you.

10 Questions from the Board? Yeah, Paul.

11 TURINSKY: Paul Turinsky, Board. Have you found
12 (inaudible)?

13 BOYLE: No. Historically, the one that--the technical
14 reason that caused people to say, "No, not right now," it was
15 the criticality. As you go back and you look at the reports
16 done by the TRW group in the '90s or the BSC group in the
17 2000s or even in the EPRI report, that's the biggest--that's
18 why I ranked it as the top. That's the one that--over the
19 course of geologic time, you know, they lose their shape,
20 they clump, water gets in. All of a sudden it becomes a more
21 difficult problem, particularly if you don't take account for
22 the burnup credit. You know, it's really there, if you will.

23 EWING: Mary Lou.

24 ZOBACK: Mary Lou Zoback, Board. Given the graph you
25 showed for the AREVA canisters with the--and you said all of

1 them have the trend to make them increasingly larger--and it
2 seems like your analysis right now, really looking at the
3 technical challenges for as big as they are now, should the
4 utilities be told to not use anything bigger than now? I
5 mean, why are we going to allow them to go down a path that
6 is not going to be--

7 BOYLE: Okay. It is subject to a contract; right? I
8 don't think--

9 ZOBACK: Well, I know, but we've got to solve the
10 problem.

11 BOYLE: I'll get there, I'll get there. I don't think
12 the government can go to the other party in a contract and
13 tell them what to do. The government could enter into good
14 faith negotiations and say, Hey, these are problematic for
15 us; what would it take for you to do four 8s? My colleague,
16 Jeff Williams, works for Monica as well, we do do studies on
17 that sort of thing, different systems, if you will. And we
18 are doing that work.

19 But to the best of my knowledge, no one has engaged
20 with the utilities yet to find out what would it take. My
21 supposition is, the government might have to offer them a
22 whole lot of money to do this. And the challenge becomes, in
23 order to load those eight 4s instead of one 32, you're using
24 their pool and their crane a lot more. They need to use that
25 crane and pool for power generation activities and the

1 running of the plant; and if it were to be used for other
2 purposes, it impacts their ability to run the plant.

3 And this was actually covered, if you will, in two
4 recent reports, one by Electric Power Research Institute and
5 one by the Government Accountability Office, in response to
6 suggestions, get the spent fuel out of the pools and into dry
7 storage. And both groups within the last year issued reports
8 that highlighted this issue that in order to do that, to get
9 it all out of the pools as much as you can and into dry
10 storage, you'd be using the crane and the pool for activities
11 other than the utility really wants to use them for.

12 So the government could go to them. I don't know
13 what the price would be, but it might be more than anybody is
14 willing to spend.

15 ZOBACK: But in the end, what the utilities want is for
16 this to go somewhere; right?

17 BOYLE: That is true. That is true.

18 EWING: Bill, I have a question. Rod Ewing, Board. You
19 listed in your pros and cons the impact on geologic disposal
20 perhaps restricting the options.

21 BOYLE: Yes.

22 EWING: But you didn't discuss it in the presentation
23 really, so what would be some of the restrictions?

24 BOYLE: Okay. For example, if your geology is such that
25 you want to rely upon a backfill, you'd have to look at would

1 the high heat output be bad for something like bentonite, and
2 there is a reason why those countries that are choosing to
3 use bentonite don't go with things this hot. Like, for
4 example, Sweden is disposing of spent fuel four assemblies at
5 a time, not 32 or 37. So it would be the heat effects on not
6 only the rock itself, you know, can the rock--does it have a
7 high enough thermal conductivity to get the heat away from
8 the item itself, because there might be thermal limits on the
9 packages themselves that you put down, but also the materials
10 that you're counting on to work, particularly if you have
11 backfill.

12 EWING: So backfill would be an issue. But jumping to
13 another type of geology, say salt, which has a high thermal
14 conductivity, so at first blush that's a positive--but if you
15 have a hot package weighing 50 to 100 tons in a material that
16 deforms plastically, wouldn't it sink?

17 BOYLE: People have looked at that. I know that is one
18 of the issues that people are concerned about with salt. I
19 think I've heard some of them say, based on their look at it,
20 that's probably not as big a worry, but it's something we
21 would have to look at. Each of the geologies, plus or minus,
22 may have their own considerations.

23 Like, take bedded salt, for example. If you can't
24 have a vertical shaft--well, I'll put it another way. If you
25 had a vertical shaft with horizontally-bedded salt, it's much

1 easier to think of the seal system for that penetration, it's
2 just a cylinder, right, a right circular cylinder. If you
3 spiral through it on a shallow ramp or you come through it on
4 a shallow ramp instead of a circle, you have an ellipse and
5 cross-section, it's a longer distance of the seal for a given
6 thickness of salt. So there are challenges, each of the
7 geologies.

8 EWING: All right. Other questions? Lee.

9 PEDDICORD: Bill, a couple things. As you've looked at
10 direct disposal of canisters like this, to what extent have
11 you looked at the issue (inaudible)?

12 BOYLE: I think some of them do. I think in the--

13 PEDDICORD: If you can count on it; right?

14 BOYLE: Right. And it's something that would have to be
15 looked at. I think people do know that. And, again,
16 whether--I don't know if any of the ones with failed fuel
17 currently got the certificate from the NRC, but it's
18 something people would have to look at.

19 PEDDICORD: I forgot to say, this is Peddicord from the
20 Board. So the question that comes to mind then as well, too,
21 in your list of pros and cons, if you can accept failed fuel
22 in such canisters (inaudible) NRC (inaudible) direct
23 disposal, that becomes a plus that you would not have to
24 repackage (inaudible) failed fuel. Another thing that some
25 of our staff colleagues have identified is, by doing the

1 direct disposal, reducing worker dose significantly. And you
2 might want to put that on your (inaudible).

3 BOYLE: Time, money, worker dose, even non-radiological
4 accidents. So, again, back to the pro column, you're
5 essentially done; you're not doing much more. And every time
6 you do more with things like this, there is dose involved and
7 also non-radiological hazards and time and money.

8 PEDDICORD: Also, on the criticality question, are there
9 not both internal and external--and I'm talking about
10 (inaudible) package strategies you could utilize to minimize
11 criticality (inaudible).

12 BOYLE: Oh, yes. And the Office of Civilian Radioactive
13 Waste Management, when it was working on the TADS--
14 transportation, aging, disposal system--there were to be
15 internal to the DPC, if you will, itself the TAD criticality
16 controls.

17 PEDDICORD: And, finally, one last question. Comparing
18 Slides 4 and 5 of your presentation, just to kind of help me
19 understand, in Slide 4 you show the vertical canister
20 (inaudible), which is the middle container, as you pointed
21 out, and you go on to the next slide, the NUHOMS®, where this
22 is being stored horizontally. But what is the kind of
23 shielding that allows the workers there to be that close to
24 the package containing that many elements?

25 BOYLE: I am pretty sure that the big object that we're

1 looking at there does not stay in. It's providing the
2 shielding. The spent fuel is actually inside that. That's a
3 transport device.

4 PEDDICORD: (Inaudible) package (inaudible).

5 BOYLE: Yes. And then they shove it in, and then it's
6 that concrete structure that provides the shielding.

7 EWING: Other questions? Staff?

8 All right, Bill, thank you very much.

9 So the last speaker for the day--and we've
10 certainly saved the best for last--is Peter Lyons, Assistant
11 Secretary for Nuclear Energy, Department of Energy. And he
12 will be giving us an overview of the administration's
13 response to the recommendations of the Blue Ribbon Commission
14 on America's Nuclear Future.

15 LYONS: Thanks, Rod. And it's good to interact with
16 your group again. I found it fascinating to listen to the
17 presentations today. But, as Bill noted in his talk, his
18 talk was pretty much a complete break with what you heard
19 earlier. Mine's going to be a complete break as well. At
20 least Bill's talk probably had the word "technical" in it, so
21 it matched your charter. I don't think my talk has that at
22 all. What I'll be doing is really giving you a policy talk,
23 and I'll apologize for the lack of technical. But that's the
24 way it will be.

25 I was asked to describe the administration's

1 strategy for basically the response to the Blue Ribbon
2 Commission. You've had other briefings in the past on the
3 BRC recommendations. I certainly don't want to talk through
4 this again. You're well aware that it starts with consent-
5 basing moves to the importance of the new organization and
6 talks about the importance of access to the waste fee along
7 with a number of other very important recommendations.

8 As you probably know, the outgoing Secretary,
9 Secretary Chu, spoke very positively about the Blue Ribbon
10 Commission, which he had chartered. You're probably also
11 well aware that the nominee for Secretary of Energy, Dr.
12 Moniz, was on the Blue Ribbon Commission; and he was asked in
13 his confirmation hearing just last week to comment on his
14 degree of support for the BRC. And I think it's fair to say
15 that it was off the charts. He indicated a very, very strong
16 interest in moving ahead with the recommendations of the Blue
17 Ribbon Commission.

18 After the BRC submitted its report, though, in
19 January of 2012, there was an effort within the
20 administration to essentially develop a response to that; and
21 that would then become the administration's statement of
22 their views--the overall administration's--this is multi-
23 agency--of their views on the BRC recommendations. And, as
24 such, it becomes a basis for administration discussions with
25 Congress as hopefully we move ahead with legislation and also

1 a very clear statement, I think, of the importance that the
2 administration places on moving ahead with steps on the back
3 end of the fuel cycle.

4 The administration's strategy dealt with three
5 large primary areas, which I show here. These are also not
6 dissimilar to what the BRC focused on. And, in general, with
7 very, very few exceptions, the administration's strategy
8 pretty much endorsed what the BRC said, sometimes with small
9 variations or very important variations or key points; but,
10 in general, very strong agreement.

11 The administration's strategy starts with the
12 importance of consent-based siting, recognizing that the
13 current path we've been on, which Secretary Chu and I have
14 defined as unworkable, was anything but a consent-based
15 approach. And if one looks around the world or within the
16 country at the current situation at WIPP, I think it's very,
17 very clear that a consent-based approach has a far higher
18 probability of success than what we have been on.

19 In terms of a system design, the BRC and the
20 strategy talks about key words like phased, adaptive, and
21 staged. I'll say more, but it talks about a pilot interim
22 storage facility, moving to a larger consolidated storage
23 facility, and the importance of geologic repository, of
24 course, as you're very, very well aware.

25 And then it goes into the importance of both the

1 governance and funding questions with endorsement--and this
2 is now from the administration--endorsement of a new
3 organization to take over these responsibilities as well as
4 the importance of providing mechanisms for funding. And I'll
5 be saying quite a bit more about this as we go ahead.

6 A couple of key points, which I'll try to
7 reemphasize on a few of the subsequent slides. For each of
8 those system elements--the pilot, the consolidated, and the
9 geologic repository--the strategy makes it clear that, as far
10 as the administration is concerned, subject to guidance from
11 Congress, we think that all of those facilities could be
12 considered for defense and civilian waste. And for some of
13 the discussion that you had earlier today, that ability to
14 potentially move defense waste early might become extremely
15 important.

16 In addition, the strategy makes clear that--at
17 least, again, from our perspective, subject to what Congress
18 eventually passes--we think all of these facilities could be
19 co-located. And at least we think that giving a preference
20 in the selection procedure to facilities that were willing to
21 be considered for co-location could be a very positive step
22 for the country.

23 In terms of the implementation that the strategy
24 lays out for the interim storage facilities, I think there's
25 been a lot of discussion of these different dates. We think

1 it's possible to assume--assuming we have legislation by
2 2014, we think we could have a pilot facility open in 2021
3 that would focus on servicing the shutdown reactors--
4 probably could take quite a bit more than that; that's only
5 about 3,000 metric tons--and would depend on how Congress
6 chooses to define what the parameters of a pilot might be.
7 But we think it would be possible by 2021, a larger facility
8 by 2025. So we're looking at dates that are reasonably soon
9 and, we think, can be real targets if we can get the
10 legislation fairly soon.

11 And that last bullet down there I already
12 emphasized, what we think is important to consider, servicing
13 both the environmental cleanup and defense sites as well as
14 the civilian.

15 You're well aware of the locations of the shutdown
16 reactors. There's already been comments about the Trojan
17 site somewhat south of here. Several other sites are south
18 of here, but then spread across the country. And there are
19 some hints that there will be additional sites. Kewaunee
20 will be shutting down May 7th; you probably have heard that.
21 Crystal River is probably shut down permanently now, Crystal
22 River, of course, down in Florida. So you could add a few
23 more sites here that may be in this category of shutdown
24 reactors in the not-too-distant future.

25 The strategy as it addresses geologic disposal goes

1 through what it believes is a reasonable time scale of the
2 operations of the geologic disposal, allowing very
3 substantial time to move through a consent-based process,
4 then, of course, along with a characterization design,
5 licensing steps. Maybe it would be possible to do this
6 faster, and that might depend on which particular sites might
7 come forward and express interest in this. I sometimes hear
8 people say, "Well, gee, you just delayed Yucca Mountain by 50
9 years." Well, as someone who has had a great deal of
10 experience in Nevada, you're well aware that 1998 was not
11 exactly a reality. And in my humble opinion, counting on any
12 date for Yucca Mountain is not very likely to lead to
13 success. And we can go into that if you want, but as someone
14 who grew up in Nevada and conducted--I directed the research
15 at Los Alamos on Yucca--I have some pretty strong feelings on
16 the degree of opposition in Nevada.

17 A comment that's made here is that, while the
18 strategy talks about one each of these various types of
19 facilities, we certainly recognize that when we talk about a
20 consent-based process, you may have restrictions,
21 requirements placed in the process of the consent process
22 that could require that you need more than one of some of
23 these facilities. The strategy talks about one, but it
24 certainly doesn't intend to preclude the possibility that
25 you'll need more.

1 I wanted to give you at least a flavor, because I
2 don't think you've been briefed on it, a fascinating study
3 that we asked Oak Ridge to lead. They involved a number of
4 other national laboratories. And essentially the question
5 was: If one looks at the existing inventory of used fuel
6 across the country today and if you had capabilities to
7 reprocess today or if you wanted to save it for reprocessing,
8 would that make sense to look at the current inventory and
9 ask whether disposal, saving it for research, or recycling it
10 makes the most sense?

11 And they looked into things--which I guess I was
12 aware, but the study certainly brought it home--that if you
13 look at the existing inventory of used fuel across the
14 country, it's a tremendous number of cats and dogs, all kinds
15 of enrichment levels, all kinds of burnups, all kinds of
16 configurations. And if you imagine trying to come up with
17 front ends to reprocessing systems that would handle all
18 those different mods, their conclusion was, it doesn't make
19 sense. Furthermore, you're generating 2,000 tons of the
20 stuff a year, so it's not like you're running out of it if
21 you choose to reprocess in the future.

22 Out of that came, again, what to me was a
23 fascinating conclusion. Their recommendation was that 98
24 percent of the current inventory should be viewed as moving
25 directly towards disposal, saving some for R&D that would

1 relate to Monica's programs on possible future closed cycles
2 and some, particularly the Naval fuel, that was viewed as
3 having a possible strategic interest to the nation in the
4 future. But, coming up with the statement that 98 percent of
5 what we have today, in their view, ought to simply be viewed
6 as moving as directly as possible towards disposal.

7 Going back then to the administration's strategy,
8 the implementation aspects of that, I've already noted
9 consent-based process figures very, very prominently--
10 absolutely no argument--across the administration that it has
11 to be a consent-based process, although the details of
12 exactly what's meant by a consent-based process probably are
13 going to be debated in Congress. And I'm leading up to the
14 point that there's many areas here that will be defined in
15 legislation that will be needed in order to move ahead in
16 this area.

17 But consent-basing, strong agreement; new
18 organization, extremely strong agreement. We did contract
19 with Rand Corporation and asked them to look at different
20 organizational structures that might be used for looking into
21 the future. Now, you would be quite correct in pointing out,
22 there have been multiple studies in the past looking at
23 alternative management structures as well, but this was
24 another new study from Rand and certainly a very, very
25 thoughtful study available on our website. I don't think

1 it's any great surprise that the bottom line was, there's no
2 one organizational type that is guaranteed of success. They
3 indicated that a government corporation or an independent
4 government agency appeared to have the characteristics for
5 success, but the features like the independence, the
6 leadership, the longevity of the leadership, the continuity
7 of the organization may be far more important than the
8 details of exactly how you structure that organization.

9 Funding also was considered in the strategy, but
10 I'm going to come back, assuming I haven't bored you too
11 much, with a few comments on what the administration proposed
12 in their budget last week as it was released that actually
13 put some teeth on this. But the administration's strategy
14 when it came out in January recognized the importance of
15 three different areas. One was to maintain some component of
16 the funding that was subject to ongoing appropriations. The
17 view there was that Congress will and should demand, as
18 should the administration, some annual oversight of the
19 progress being made by this new entity, and that by having
20 some level of appropriations, that would give some degree of
21 oversight to the Congress.

22 But it also recognized the importance of
23 reclassification of either the fees or the spending. Again,
24 you're probably very familiar--and I certainly don't want to
25 go into it, but you've got a situation now where the fees are

1 mandatory and the spending is discretionary. And those two
2 sides of the budget simply don't offset each other. And
3 that's where we have been for now probably a couple of
4 decades on this, and, if anything, it's getting worse over
5 time. You need to reclassify either the fees or the spending
6 in order to get them both on the same side of the ledger and
7 be able to talk about offsets. And then, finally, access to
8 the nuclear waste fund. And I'll come back in a minute and
9 show you how the administration addressed all of these points
10 in their budget proposal that came out last week.

11 I'm going to skip through several slides that go
12 into some of the ongoing R&D in the program that Monica and
13 Bill and Jeff direct. But on this one I did want to note
14 that, particularly in the area of storage R&D--because I
15 think this is of strong interest to you--that second bullet,
16 we've been working on a competitive basis to select teams to
17 come up with a demonstration to provide data on, frankly,
18 what happens on high-burnup fuel in storage. This also is
19 part of a multi-university program--happens to be led by
20 Texas A&M--where we have also asked that they study with
21 several partners to try to gain a better understanding of
22 what can happen--what are the degradation mechanisms that
23 could affect used fuel in long-term dry cask storage and, in
24 particular, to do it for high-burnup fuel.

25 What's noted here in that second bullet is, we're

1 in the final steps of the procurement process. Yes, we are.
2 And it was announced today that EPRI will move ahead--they
3 were selected to move ahead with a cost-shared program. This
4 will lead to highly instrumented casks. It will involve a
5 demonstration site. There will be a research plan that will
6 be developed as pretty much the next step in this. That will
7 go out for public comment, and we will certainly invite the
8 NWTRB to comment on the research plan; because as we look at
9 virtually any of these options looking into the future, there
10 is going to be a need for dry cask storage for significant
11 periods of time. Between the university program and this
12 actual set of measurements, we're hoping to provide you and
13 the country with far more information on what happens--
14 hopefully nothing--but what happens in dry cask storage and
15 how long can you reasonably consider using dry cask storage.

16 I think this is one I'll skip through fairly
17 quickly. It's already been commented today by some of the
18 other speakers that we are already involved in these strategy
19 endorses, that we move ahead with planning to the extent we
20 can--there's a limit to what we can do without legislation--
21 but starting to plan for aspects of the transportation
22 program. And that would include reactivating some of the
23 work with the state and regional programs to try to begin to
24 recreate some of the capabilities that were being nurtured
25 earlier on in this program.

1 In terms of transportation R&D, I think this is
2 another one that I'll skip over, but also within Monica's
3 program.

4 In disposal R&D, again, under Monica's program with
5 Bill and Jeff--Jeff Williams--we're continuing to try to
6 develop a greater understanding of alternative geologic
7 media. This certainly includes salt, where arguably the U.S.
8 is one of the leaders, maybe the leader along with Germany.
9 And I'll comment briefly on how, in addition to studies at
10 Sandia, we're crafting various international agreements to
11 try to get a better understanding of alternative geologic
12 media that might be proposed in the consent basis. Noting
13 here the interest in continuing work at WIPP, I have to say,
14 not from the standpoint of selecting WIPP, but from the
15 standpoint of using WIPP as a generic salt experimental bed
16 and continuing work, much of that is being done jointly with
17 EM.

18 Also, Monica's team is cranking up a borehole
19 disposal R&D program to try to look at that as a possibility
20 for the future.

21 In addition, I mentioned the work with the
22 international partners. We've certainly tried to nurture
23 this, and Monica has been taking a very strong lead in trying
24 to expand the different international cooperative ventures
25 that we have. Again, we have a great deal of expertise in

1 salt, and we do some of that jointly with the Gorleben in
2 Germany. But we recognize that, for example, if it's
3 granite, Sweden and Finland are doing outstanding work in
4 that area; France certainly; Switzerland to a lesser extent;
5 outstanding work in clay-based systems. So there is MOUs;
6 there is joint programs; we've reactivated collaboration with
7 all of those entities. We have an MOU with ANDRA. And I
8 think that's highly appropriate, given where we are in the
9 current program.

10 The strategy concludes with the recognition that
11 we've got to have legislation. You're well aware of the
12 limitations within the Nuclear Waste Policy Act, and
13 virtually everything I've described here is not going to
14 happen without changes in the Nuclear Waste Policy Act.

15 You're probably also aware that Senator Bingaman
16 last year worked to pull together a bill with a number of his
17 colleagues. He did introduce the bill, but in some of the
18 final negotiations he was the only sponsor of that bill.
19 That bill died in the last Congress. Now Senator Wyden, who
20 has taken on the leadership of the committee that Senator
21 Bingaman had with Energy and Natural Resources, as Senator
22 Bingaman retired, Senator Wyden took over that leadership.
23 Senator Wyden has spoken repeatedly about his efforts,
24 working, again, with several very influential colleagues who
25 hold key positions on the appropriate committees, is working

1 towards legislation. He has made it quite public that they
2 started with Bingaman's bill but have worked to identify some
3 of the sticking points that led to different concerns within
4 Bingaman's approach. He has indicated that they may be
5 within a month or two of providing that legislation for
6 review. I'm certainly eagerly awaiting that. And, just in
7 general, I'm very, very gratified to know that there is a
8 group of senators, who recognize this importance and led by
9 Dr. Wyden, are looking towards real progress in this area.

10 I wanted to switch ever so briefly to comments on
11 the budget. The budget was released last Wednesday. I'm not
12 going to talk through these numbers at all. They're in
13 there. Let me just note that, number one, it's an austere
14 budget, as you can tell by the bottom line. Number two, we
15 worked very hard to protect several key initiatives within
16 our program. One of those is small modular reactors, which
17 may or may not be of interest here, but certainly not to
18 discuss. The other is fuel cycle R&D--again, Monica's
19 program--where we are working diligently to try to maintain
20 funding to move ahead as we can with an existing legislation
21 on different aspects in preparation for moving ahead on the
22 Blue Ribbon Commission recommendations.

23 Again, I won't talk through this in any detail, but
24 you'll note the first bullet up there under Planned
25 Accomplishments is to continue activities that support the

1 administration's strategy. And then a number of other
2 aspects listed there; some of them I've already talked about.

3 But I wanted to close--at least I think that's my--
4 yeah, that's my last slide. I wanted to close with this
5 slide. Within the administration's budget were some very,
6 very specific proposals. I guess you could say there has
7 been some of the predictable responses, but there has also
8 been some bipartisan statements that recognize that what the
9 President has done here is to really try to put teeth on the
10 administration's strategy that he proposed.

11 So the budget--and this is laid out in multi-year--
12 the budget goes for ten years. And laid out within that ten
13 years is \$5.6 billion over those ten years to move ahead on
14 this program, to move ahead with the construction and
15 operation of the pilot interim site, as well as substantial
16 progress on the interim storage site and on the geologic
17 disposal. \$5.6 billion over ten years is a very substantial
18 number.

19 In terms of details of the funding, I gave you
20 earlier what the strategy recommended. That was in January.
21 Now what the President has done is to say, okay, I agree that
22 we ought to have annual appropriations in order to give
23 Congress and the administration some degree of annual
24 oversight of the progress. But the statement now is that
25 there will be ongoing discretionary approps up to 200

1 million, but above 200 million and reflected in the budget
2 now is to move to mandatory approps.

3 So, again, I described earlier how you've got to
4 either change the fees or the approps so you get them lined
5 up on the same side of the ledger so they can offset. So now
6 the proposal is to go to mandatory appropriations starting in
7 2017--and this is reflected in the ten-year budget--tapping
8 into the nuclear waste fund as you continue to move ahead
9 with program costs. This is an expensive business. That's
10 why the 5.6 billion. But if this proposal will be accepted
11 by Congress, this is now putting the administration on record
12 as supporting the ability to tap into the nuclear waste fund
13 with mandatory appropriations. That's a very big difference.

14 In addition, I'm missing one bullet up here, which
15 is also very important. You're, I think, well aware that
16 every year the utilities sue the Department because we didn't
17 take their waste. Well, okay, we didn't take their waste.
18 That's obvious. And every year they win. We have now paid
19 2.6 billion to the utilities, and this is growing by an
20 average of 400 million per year. But these awards are paid
21 out of the so-called "judgment fund" of the Treasury. That
22 judgment fund is not subject to OMB or congressional
23 oversight. That is a mandatory payment. And in the past
24 that has never been counted within the budget, at least the
25 budget looking forward. It's always put in the budget

1 looking backwards, but by then it's too late in some sense.

2 So what the administration is doing and have built
3 into their budget is a forecast of the future liability
4 payments in order to try to make the point to everyone that,
5 folks, this inaction is really costing us, it's costing the
6 taxpayers, and that those liability payments should be viewed
7 also as an offset towards progress into the future.

8 This may not sound like a big deal, but it is,
9 folks, to actually recognize that we're going to lose these
10 suits obviously, and we're going to plan on losing them.
11 Count the liabilities then as an offset looking into the
12 future. So another extremely important point.

13 And then the very last element down there, which
14 should also interest you greatly, one of the BRC
15 recommendations was to point out that Yucca Mountain is being
16 done under site-specific disposal regulations. And the BRC
17 pointed out, that's not a really good way of looking into the
18 future, folks, and then the EPA needs to move ahead with
19 generic disposal regulations. So the President's budget not
20 only funds, it provides the authority to the EPA to move out
21 with the development of generic standards.

22 And I'm guessing that last one resonates with some
23 of you.

24 I didn't bring a watch up here. I hope this isn't
25 too late.

1 EWING: No, you're fine.

2 LYONS: In any case, thank you, and I'll take questions
3 if you still have any time.

4 EWING: So, Pete, thank you very much for a wonderfully
5 informative presentation.

6 And we do have time for questions, so, first,
7 questions from the Board? Yes, Lee.

8 PEDDICORD: Peddicord from the Board. Dr. Lyons, I
9 agree. Thank you for a very interesting (inaudible)
10 information. Two questions come to mind. What is now the
11 administration's position in (inaudible) this new entity
12 (inaudible), however it's defined, in terms of it (inaudible)
13 an R&D mission towards disposal, and what might be the
14 breakdown between R&D (inaudible) the new entities, assuming
15 it has this responsibility, and what you're currently doing
16 in the fuel cycle R&D (inaudible)?

17 LYONS: The BRC recommended and the strategy concurs
18 that the new entity should focus on the disposal mission but
19 that the research on separations, possible closed fuel
20 cycles--that's R&D--that should stay within the Department of
21 Energy at last for the foreseeable future. And Monica's
22 program, in all of its elements, is working towards providing
23 information to future decision makers as to whether at some
24 point in the future the country may wish to move towards a
25 closed cycle. But certainly the agreement is, the first

1 thing we better do is demonstrate that we can move ahead with
2 a repository system, which, of course, would be required,
3 closed cycle or not.

4 PEDDICORD: Let me drill down just a little bit more
5 into the question. In spinning off what we were discussing
6 with Bill Boyle (inaudible) possibility in this issue of
7 (inaudible) fuel and so on, my (inaudible) requires some
8 research and technology development. Is that going to be
9 something then that (inaudible) because this is very much
10 related to the disposal mission (inaudible)?

11 LYONS: Congress may provide us guidance on that. At
12 least my view is that, to the extent R&D is required specific
13 to the disposal mission, it would be done within the new
14 organization.

15 PEDDICORD: A second question, if I may as well, too.
16 Now that (inaudible) under the consent-based approps, you
17 have communities stepping forward to express interest. So
18 until new legislation is in place, how does that progress?
19 Is it something your office handles, considers, responds to;
20 or are these (inaudible) until (inaudible) some activity is
21 taken by Congress?

22 LYONS: Well, a number of organizations across the
23 country are talking with us of their interest, and certainly
24 a number of them have very strong interest in finding ways
25 for us to move ahead with site-specific funding. Under the

1 Nuclear Waste Policy Act, at least most interpretations would
2 say I could probably move ahead on consolidated storage and
3 go partway; certainly couldn't move towards actual
4 construction or operation.

5 Again, my view is that it's quite a mistake to do
6 anything site-specific until we have legislation. We have
7 right now quite a split between, I think, the House and the
8 Senate on this. And I am concerned that if I undertook site-
9 specific activities, I would be exacerbating a divide between
10 the two houses. My view is that we're far better off to
11 wait. Even though I could move a little way, I can't move
12 very far. I think we're better off to wait until we have a
13 new legislative framework that takes the place of the Nuclear
14 Waste Policy Act and makes it clear what the congressional
15 intent is. I think we're far more likely to have a
16 successful outcome that way.

17 This is certainly debatable, Lee. I'm in debates
18 frequently on this. That's at least my view is we need to
19 wait, and we need to see what the legislation is and then
20 move out.

21 PEDDICORD: Thank you.

22 EWING: Other questions from the Board? Staff?

23 LESLIE: Bret Leslie, Staff. Dr. Lyons, I have a
24 question in terms of the consent-based approach. The
25 framework for interactions between states and affected units

1 of local government is set forth in the Nuclear Waste Policy
2 Act. Do you foresee that the consent-based approach is
3 specified in legislation, or is it kind of a generic
4 statement we should use a consent-based approach? And if
5 it's the latter, how do you determine what those principles
6 are to define what consent-based would mean?

7 LYONS: Well, first, yes, Congress may tell us. On the
8 other hand, there has also been a lot of studies in the past,
9 looking at what would go into a consent basis. We would
10 certainly draw on that. We would try to find opportunities
11 for public input on how this would be constructed. Although
12 we are--well, I feel very strongly that we should not do
13 site-specific work.

14 We're quite interested in working with, let's say,
15 national organizations that are interested in helping to
16 define what some of these parameters might be, for example,
17 ECA, NARUC. Those would be examples--there's several
18 others--of organizations that I'm more than happy to work
19 with to help us at least provide ideas and suggestions on
20 what might go into a consent basis, along with several other
21 aspects where I think they can provide important information.
22 But I think it's at least completely obvious that it has to
23 include states, tribes, and the communities.

24 And, frankly, we can point to some rather
25 substantial examples within this country where it's had

1 strong local support and no state support and has fallen flat
2 on its face. So I have no desire to repeat that, and I want
3 to move ahead. To me, that takes all three of those
4 entities.

5 EWING: More questions? Staff? Board? Yes, Jerry.

6 FRANKEL: This is Jerry Frankel of the Board. I'm new
7 to the Board and don't know much about politics, and I should
8 indicate that what I know is from the able tutelage of the
9 staff who are here. So thank you for that.

10 But I just don't have a lot of confidence in our
11 legislature coming together to pass a bill on this soon. So
12 where does it leave us if they can't come to an agreement any
13 time soon and you want to start with some planning? Is there
14 an alternative in case the law isn't changed soon?

15 LYONS: I don't see the alternative. I fully agree with
16 the Secretary, past Secretary, that Yucca Mountain is not
17 workable. I've spent a good fraction of my life working on
18 Yucca Mountain, and I'd like to see some progress. And, to
19 me, that progress means admitting that it's unworkable and
20 looking into the future.

21 In the Senate we have Senators Murkowski,
22 Alexander, Feinstein, led by Wyden; so you've got the
23 chairman and ranking member of the Appropriations Committee
24 and the Energy and Natural Resources Committee. That is an
25 extraordinarily powerful combination. My most earnest hope

1 is that legislation, if it can come forth, supported by those
2 four, would be very powerful and quite likely to move ahead
3 in the Senate. I think the House, from what I can see, will
4 be quite a challenge. From anything I've seen, I see the
5 Senate as leading on this, ecstatic to have House leadership
6 as well; but at least I'm not aware of groups of
7 representatives working together on this. But a bill
8 originating in the Senate, if it can be passed by the Senate,
9 presents some opportunities for perhaps consideration by the
10 House.

11 EWING: Last call. Questions?

12 Pete, again, thank you very much for your time and
13 the presentation. It's very helpful.

14 LYONS: Thanks to all of you for what you're doing.

15 EWING: So we're at the end of the day, but we have
16 still an important part of the program, that is, public
17 comment. I have one person listed, Robert Smith, and I'll
18 ask if there are others after Smith.

19 Robert Smith.

20 SMITH: My name is Richard Smith.

21 EWING: Oh, sorry.

22 SMITH: By way of introduction, I am retired from the
23 Pacific Northwest Laboratory, worked there for about 40 years
24 in the nuclear area, including the initial (inaudible)
25 storage program where I was involved in the design and

1 planning for the system performance of that facility. I have
2 also been a member of the Hanford Advisory Board for the past
3 ten years, but I want to be careful to say that what I'm
4 going to say this afternoon should not be construed as any
5 consensus opinion of the Board. It's my own personal view,
6 and you can take it for what it's worth.

7 We've heard a lot about high-level waste today.
8 And one of the problems I've had for years is that the
9 current definitions of high-level waste don't really speak to
10 protection of human health and the environment. It's
11 somewhat of an arbitrary definition (inaudible) came from
12 there--it's high-level waste--and without any
13 characterization of what that means, what that stuff is, and
14 how you might protect the environment from it.

15 So basically what I'm suggesting is that I think it
16 would be worthwhile to fall back and say, well, what would be
17 a good, solid, science-based approach to defining what high-
18 level waste is and thereby being able to better establish
19 what the criteria for treatment and disposal are. The
20 current definition would allow you to classify as high-level
21 waste stuff that was really low-level waste and have to treat
22 it and dispose of it in the very expensive fashion as
23 associated with high-level waste.

24 So I would like to see your group, or anybody else
25 for that matter, to suggest to DOE that they engage the

1 national academies to develop this question: What makes
2 sense for a good technical, operable definition of high-level
3 and low-level waste? I've searched the literature to
4 (inaudible), and I can't find the technical bases I was
5 looking for. So if all you do is say, well, why don't you
6 get a committee, a national (inaudible) look at this
7 question, that would be a major step forward.

8 The present definitions have led us into all kinds
9 of sort of strange machinations, like the creation of the
10 "waste incidental to processing" thing, and other questions
11 about, well, a lot of this stuff in the tanks may be TRU; how
12 do you deal with that? What is it? Is it really TRU, or is
13 it really high-level waste, or is it something else? And how
14 do you package and treat that stuff, and where can you
15 dispose of it? I'm just trying to erase some thinking on
16 this area, because I think it has given us--the present
17 definitions have given us poor service in the recent years.

18 That's all I had to say. Thank you.

19 EWING: All right, thank you.

20 Any other comments on (inaudible) section?

21 Carol, I know you--

22 JANTZEN: If there is somebody else behind me, they can
23 go first.

24 EWING: No, you're next.

25 JANTZEN: Okay.

1 EWING: Identify yourself again.

2 JANTZEN: I'm Carol Jantzen from Savannah River National
3 Lab. I just wanted to correct two statements that I heard
4 this morning that were incorrect. One was the response to
5 Mary Lou Zoback's question about square melters versus round
6 melters. We did not choose a round melter because it was the
7 same shape as a waste tank. We chose it because we did
8 convection modeling. You remember in my talk I spoke about
9 the fact that melters that started in 1994, 1995, which was
10 us and West Valley, you know, nobody had done this before.
11 So we depended on natural convection. We hadn't come up with
12 the airlift coffee pot thing, and we hadn't come up with the
13 bubbler designs yet, so we were depending on natural
14 convection.

15 There were many, many studies done, theoretical
16 studies, of how fluids behaved, how the natural convection
17 behaved in a round object versus a square object, and the
18 natural convection was better in a round melter.

19 We also had--the way the electrodes were, there
20 were four electrodes about halfway up the side or down--
21 whether you're coming from the top or the bottom--that cross-
22 fired, so there were four around the circumference at regular
23 distances so that the electrodes could properly cross-fire
24 and maximize those natural convection currents. Also, at
25 that time, with natural convection, you found crystals

1 forming in the corners of the melters, because the corners
2 were colder and they were more stagnant. And so until the
3 bubbler technology came along, a round melter was preferred
4 to a square or a rectangular shape. And to get around that,
5 West Valley made theirs an inverted prism; okay?

6 So that's the story of the square versus the round
7 peg.

8 Now, the other comment was about the total system
9 performance license application. It was said that it was
10 based on the PCT test. It was not based on the PCT test.
11 All of the different durability tests that I showed you up
12 there, which included monolith tests, which included single-
13 pass blow-through tests, which included thermodynamic
14 modeling, was all included in the Yucca Mountain license
15 application modeling.

16 So I just wanted to set that record straight.

17 EWING: Okay, thank you.

18 In a moment I'm going to cut you all off.

19 PEGG: I certainly want to defer to Carol's memory
20 better than mine on that subject. When I said "a tank", I
21 didn't mean a waste tank. What I've been told is that
22 Dupont, who was the site contractor at the time, charged with
23 designing a first-of-a-kind entity, this waste glass melter,
24 knew how to design tank-based designs. That was a natural
25 choice for this melter shape.

1 One thing I forgot to mention this morning, though,
2 is an important factor in the WTP melters. When we go from
3 the West Valley, the Savannah River scale, to a ten-square-
4 meter scale, one thing you have to remember is thermal
5 expansion. The LAW melter we saw yesterday physically grows
6 from room temperature to the operating temperature by about
7 this much. What that means is you have to allow for active
8 thermal expansion within all of the refractories and all of
9 the components inside that shell. That's much easier to do
10 if you have a square entity with a sliding wall. And these
11 melters actually have internal to the cavity a sliding wall
12 with a jack bolt system; and as the temperature rises, as you
13 bring it up to operating temperature, those jack bolts are
14 pulled back, and that wall actually moves to allow this
15 things to grow. There's a size limit at which that becomes
16 necessary. A small enough melter you can put in simply a
17 crushable refractory that'll take up that slack. But once
18 you get large enough, this much is hard to design around with
19 the ceramic refractory.

20 So just thought I'd make that addition.

21 EWING: Okay, thank you.

22 Last comment. Roberto.

23 PABALAN: I've got a question, not a comment.

24 EWING: Identify yourself.

25 PABALAN: My name is Roberto Pabalan from Southwest

1 Research Institute. This morning Bill Hamel talked about
2 lessons learned from West Valley applied to the Waste
3 Treatment Plan. Given that the Hanford waste streams are
4 more complicated and also the waste volumes certainly are
5 much larger, what lessons learned from the DWPF have been
6 applied to the Waste Treatment Plant? That's one question.

7 EWING: To whom?

8 PABALAN: Bill Hamel or somebody from Hanford.

9 EWING: Actually, if you look at some of the
10 presentations, there were some specific examples (inaudible).
11 Well, Ian.

12 PEGG: I'm sorry, I missed the question.

13 EWING: Lessons applied from Savannah River to Hanford.

14 PEGG: Once again?

15 PABALAN: What lessons learned from the DWPF were
16 applied to the Waste Treatment Plant?

17 PEGG: Yes, I'm sure there is a lot better people that
18 could speak to this, but, for example, when Bechtel took over
19 the contract in March of 2001, a lot of the former staff from
20 Savannah River and West Valley were part of that team. So
21 there were a lot of lessons transferred in terms of operating
22 issues, off-gas pluggages, how to respond to them. Most of
23 the melter design really is much closer to the West Valley
24 experience than the Savannah River experience. It's the same
25 basic technology.

1 But, for example, the melter shape, the way the
2 glass is discharged, it's not a vacuum discharge like DWPF,
3 but it is an airlift discharge exactly like West Valley. If
4 you look at the off-gas treatment system, the first stage of
5 operation there is something called a submerged BAT (?)
6 scrubber. It's exactly like what was used at West Valley.
7 There were lessons learned from West Valley that were
8 implemented into a next generation of submerged BAT scrubber.
9 So, in fact, there's a whole list of lessons learned from,
10 actually, both facilities, West Valley and DWPF, that were
11 incorporated into the WTP design and flowsheet.

12 Another example I mentioned this morning is the
13 glass forming chemical system. In Germany and France, in the
14 U.K., they use premade glass as the source of the chemicals
15 to combine with the waste to make a viable glass product.
16 That's what's also done at DWPF. But, as I mentioned, the
17 compositional variability made that challenging at WTP. What
18 was selected was what had been used at West Valley, which is
19 raw chemical additives, which you can tune on the fly.

20 So I think the short answer is: I'm sure not every
21 possible lesson was learned, but you can certainly make a
22 very long list of lessons that were learned from previous
23 facilities in this country and abroad that were implemented
24 at the WTP.

25 VIENNA: Just a quick comment.

1 EWING: You have to come forward.

2 And let me emphasize, Roberto, we're looking for
3 comments, not questions, because we're not sure who's here to
4 respond to questions.

5 VIENNA: John Vienna from PNL. There is a DOE report
6 on lessons learned from vitrification--I believe it's 1999--
7 that has the lessons from Savannah River, from West Valley,
8 and actually some international experience also. All of that
9 was taken into account during the design of Waste Treatment
10 Plant melters.

11 PABALAN: Okay, that's it.

12 EWING: Other comments? All right, I'd like to thank
13 the audience for staying through today, particularly the
14 speakers, by making their presentations and contributing to
15 the discussion, even without being reminded. (Inaudible)
16 direct you to the poster session, new for our meetings. I
17 think that you can buy a glass of wine or a beer and then
18 bring it to the poster session (inaudible). I've had the
19 pleasure of discussing (inaudible).

20 So thank you all, and we'll see you in the poster
21 session in a moment.

22 Board members and staff, a reminder, 7:30 at
23 Anthony's Event Center.

24 SPEAKER: No, Anthony's is the restaurant.

25 EWING: No, it's Anthony's Event Center.

1 (Whereupon, the meeting was adjourned.)

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