



U.S. DEPARTMENT OF
ENERGY

Nuclear Energy

Coupled Model for Thermal-Hydrological- Chemical Processes in a High-Level Radioactive Waste Repository in Salt

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U.S. Nuclear Waste Technical Review Board Meeting

Albuquerque, NM

March 19, 2014

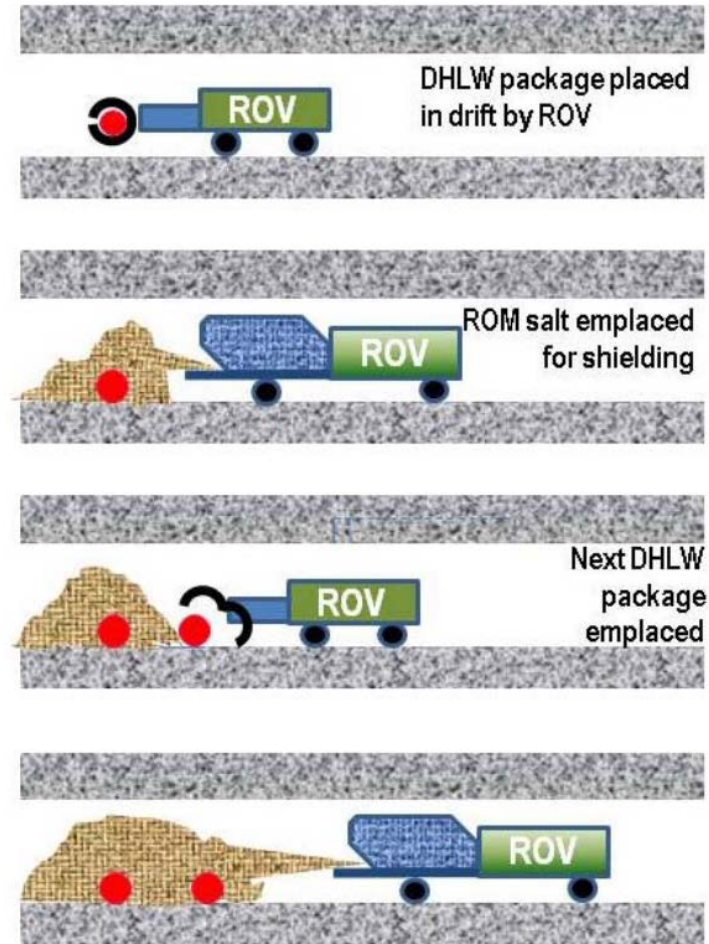
Outline

- 1) In-drift concept for Defense High Level Waste
- 2) Waste composition by thermal load
- 3) Background on salt and heat pipes
- 4) Simulator description : FEHM developed at LANL
- 5) Code validation example
- 6) Simulations with heat only
- 7) Addition of water and water vapor transport
- 8) Processes added to couple water and chemistry
- 9) Results of fully coupled Thermo/Hydro/Chem simulations

In-drift waste emplacement strategy

**Simple lower
cost method.
Backfill is readily
available in salt
formations**

Hardin et al., FCRD-UFD-2012-000219



DOE/CBFO-12-3485

Los Alamos Team Members

- Dylan Harp – Simulation Expert
- Amy Jordan – PhD Student
- George Zyvoloski – Code Development
- Terry Miller – Mesh Generation
- Hakim Boukhalfa – Chemistry
- Florie Caporuscio – Chemistry
- Bruce Robinson – Project Coordination





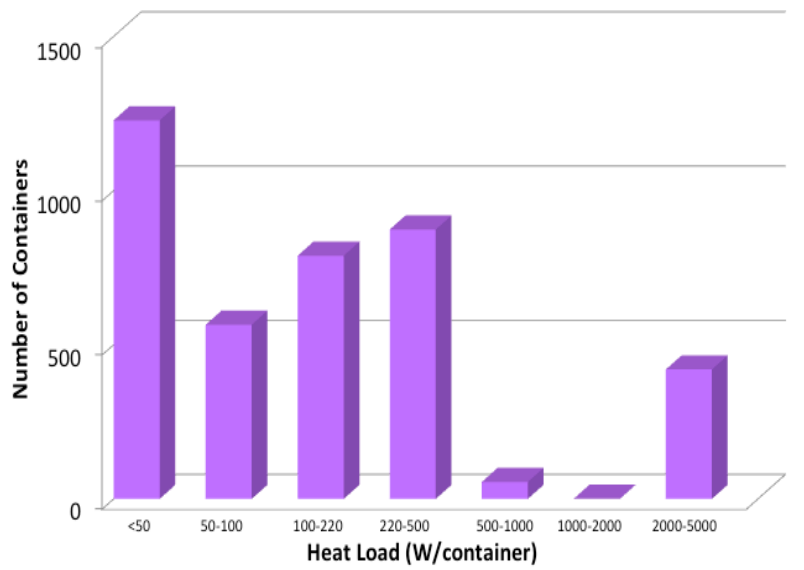
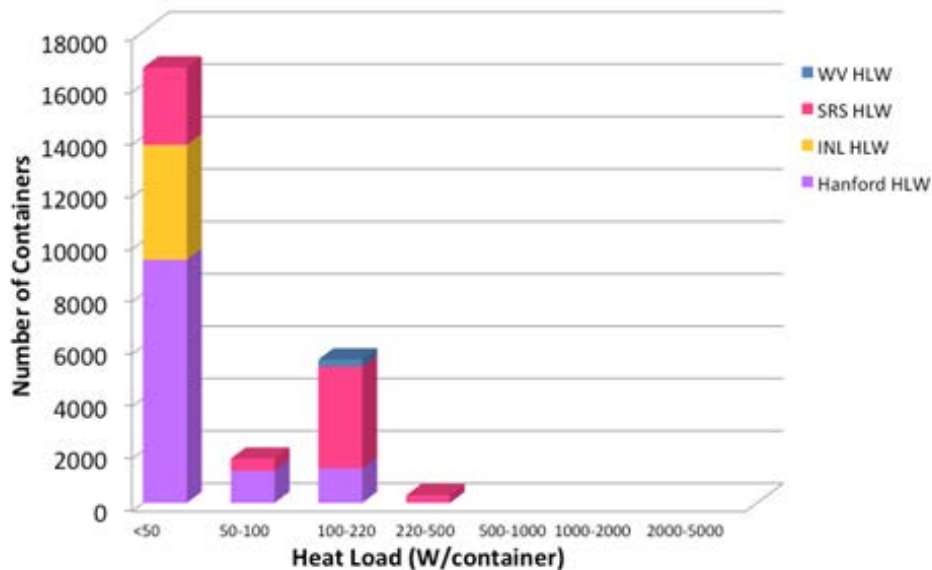
Distribution of Heat Loads: DOE-Managed Waste

More than 90% of is less than 220W

Defense waste heat loads are much lower than commercial SNF heat loads planned for Yucca Mountain (6200 – 8800W/canister)

High-Level Waste

DOE Spent Nuclear Fuel



Carter, J.T., A.J. Luptak, J. Gastelum, C. Stockman, A. Miller. 2012. Fuel Cycle Potential Waste Inventory for Disposition. DOE Office of Nuclear Energy Report FCR&D-USED-2010-000031, Rev 5.

Background

■ Bedded salt has favorable characteristics for heat-generating waste disposal:

- Self-sealing rheology
- Very low permeability in its intact/final states
- High thermal conductivity

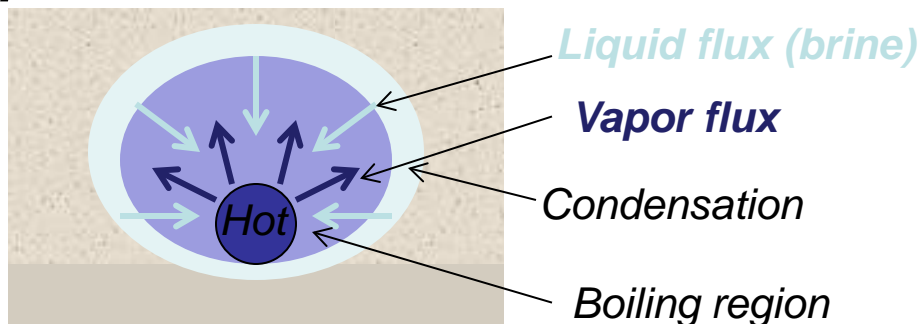
■ Past heater tests in salt provide data for basic model validation and salt material properties

- Evidence of **heat pipe** activity around a 130°C heater



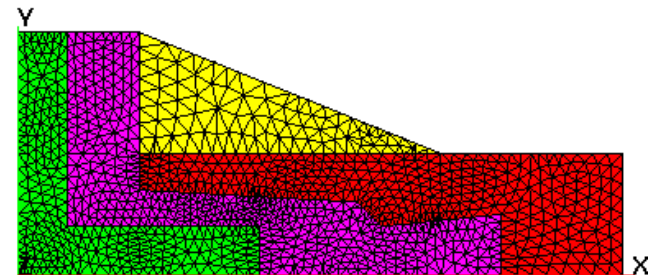
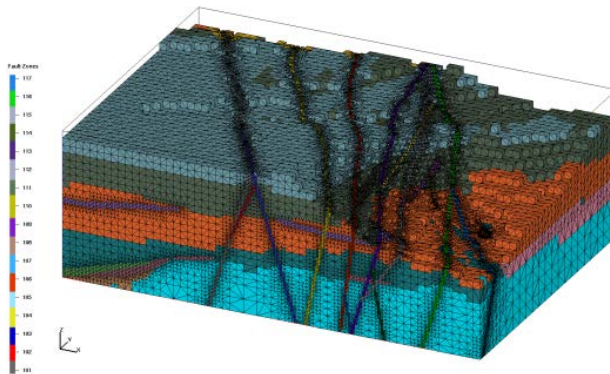
From Brady et al. (2013).

Heat pipe:



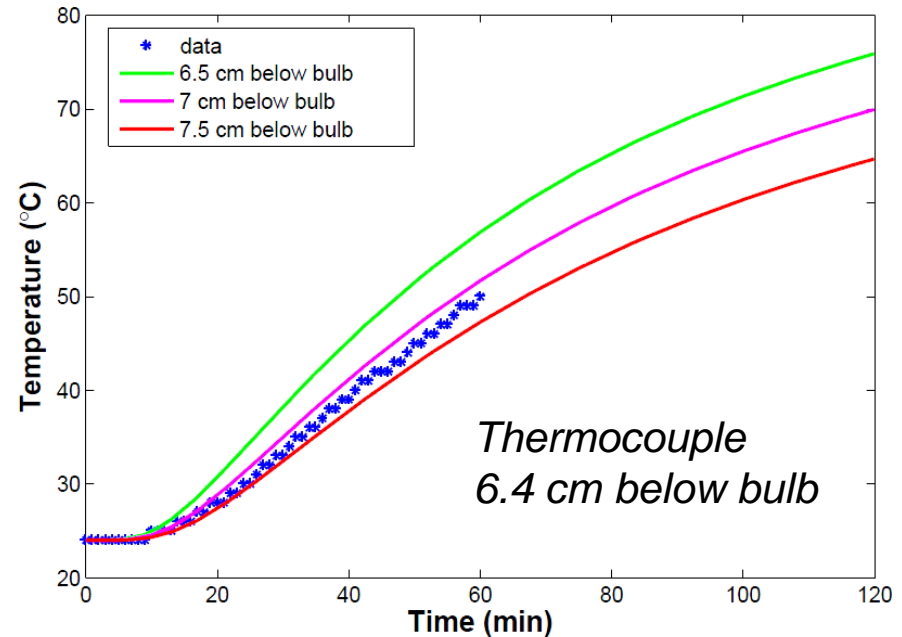
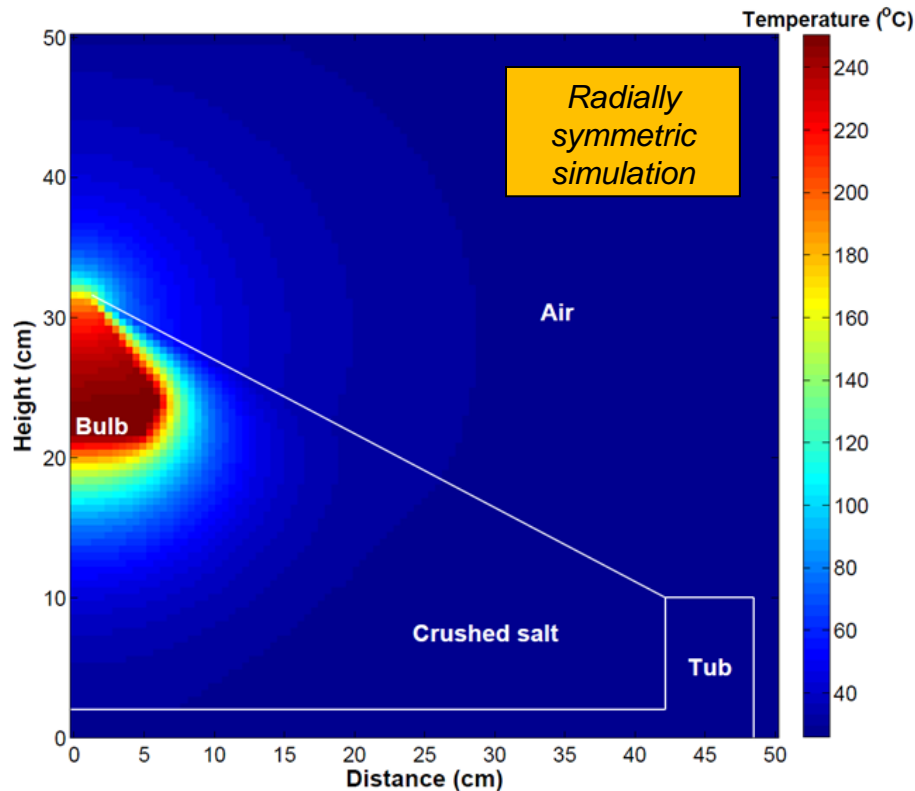
Simulator Description

- **FEHM developed at Los Alamos 30+ years** fehm.lanl.gov
- **Used for 150+ peer reviewed articles**
fehm.lanl.gov/pdfs/FEHM_references_list.pdf
- **Fully coupled thermal, mechanical, chemical, multiphase (gas, water vapor, water, rock)**
- **Uses LaGriT: Powerful 3-D grid generation tool**



Code Validation Sample

■ Pile of crushed salt:



● Purpose:

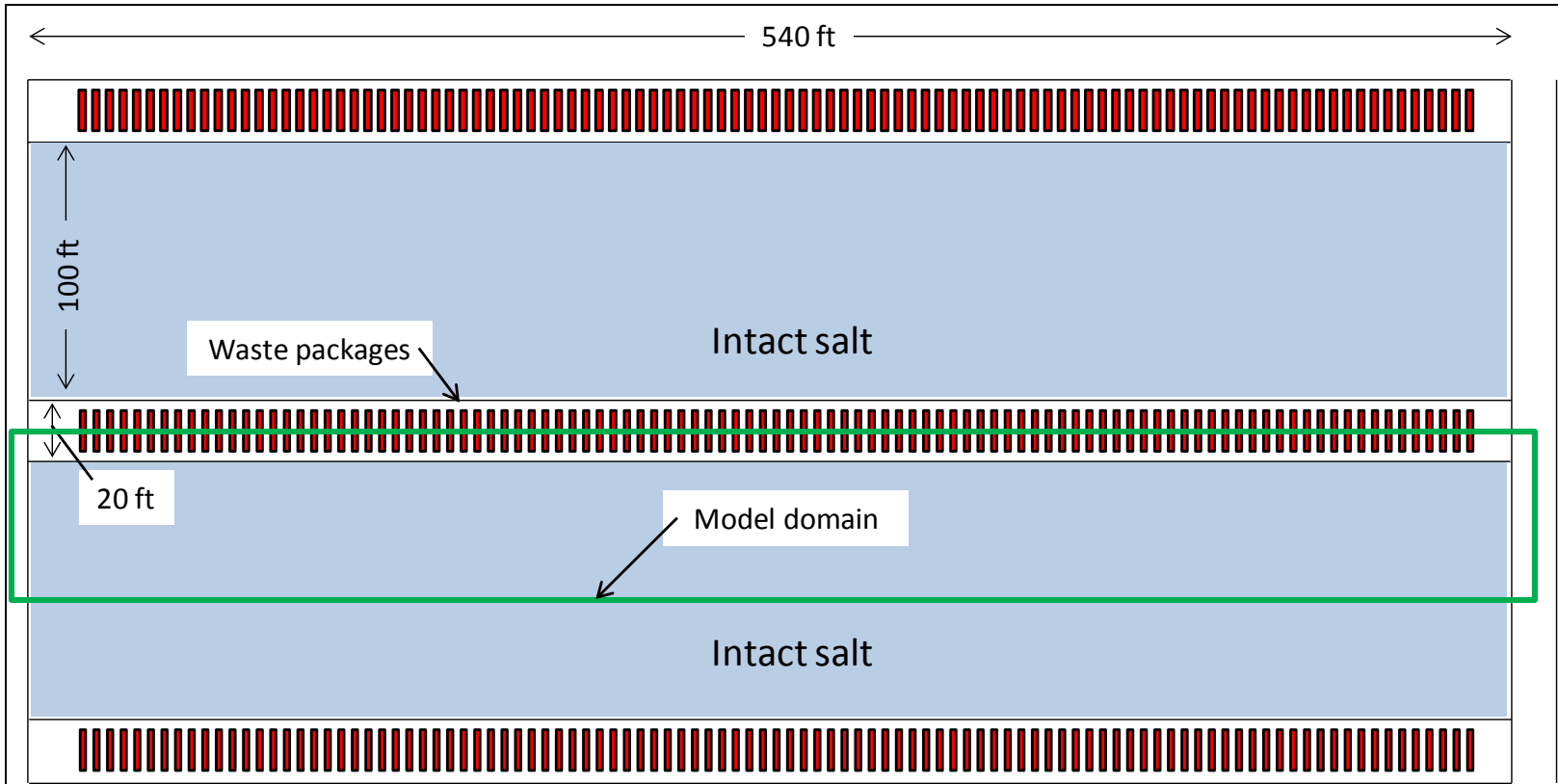
- Test crushed salt (RoM) thermal model:

$\square = \text{porosity}$

$$k_{T-WIPP}(\phi) = 1.08(-270\phi^4 + 370\phi^3 - 136\phi^2 + 1.5\phi + 5)$$

Thermal only simulation examples

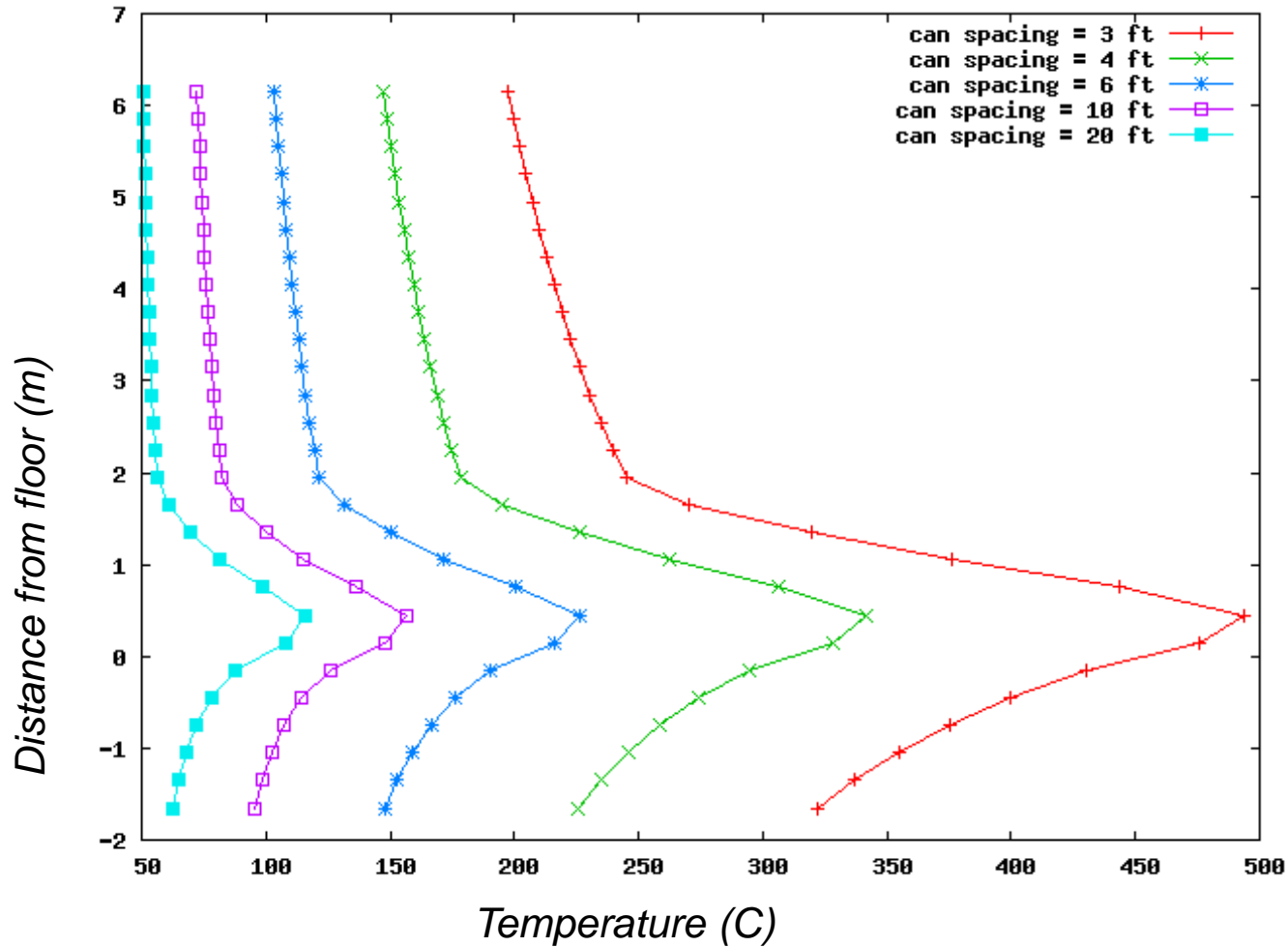
Map view of a potential salt repository : in-drift style



Reflection boundaries are used to reduce model domain

Thermal only simulation examples

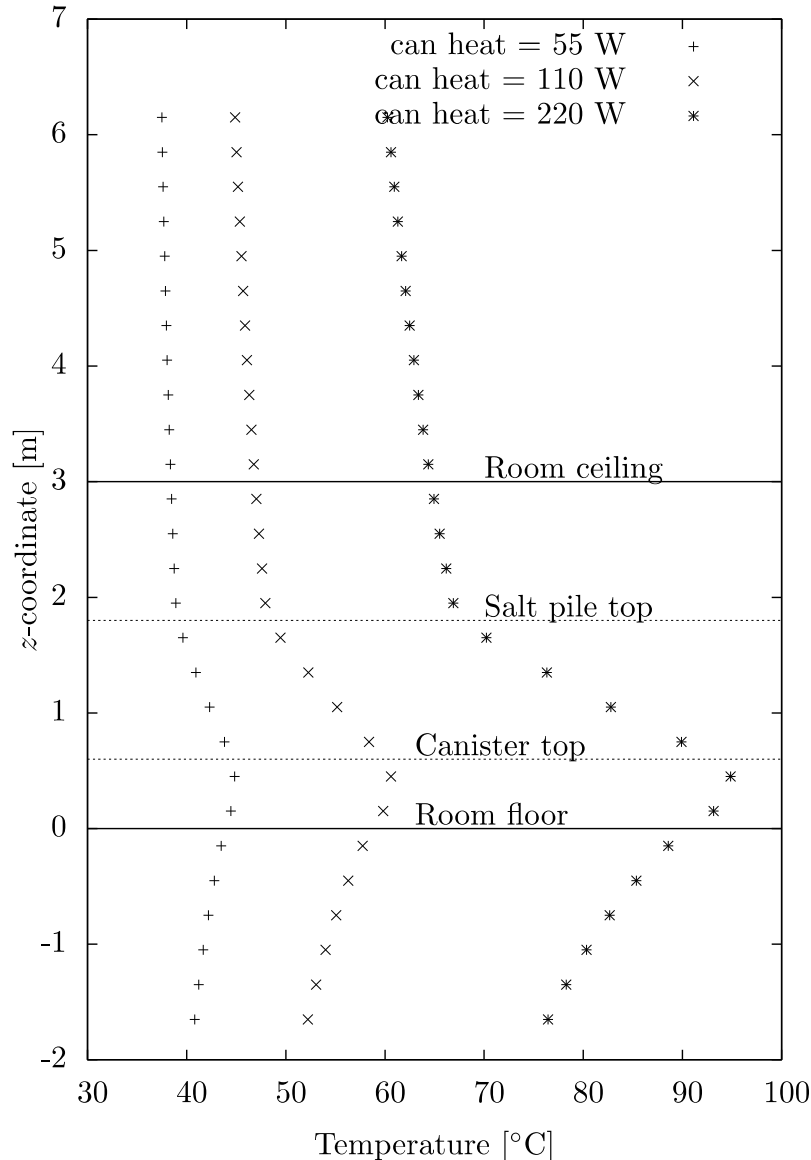
Vertical temperature for different 1000W canister spacing



2014 Harp et al., Modeling of High-Level Nuclear Waste Disposal in a Salt Repository, accepted, Nuclear Technology.

Thermal only simulation examples

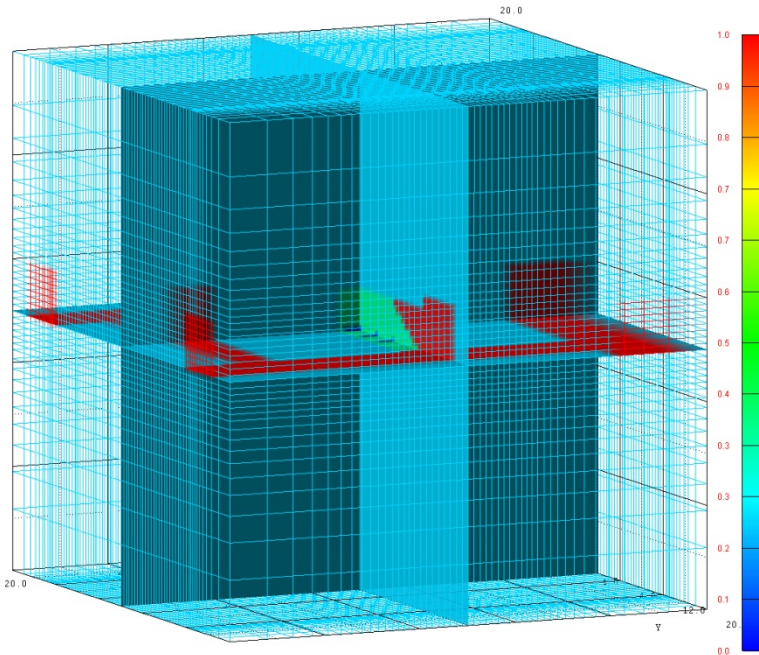
Vertical temperature for canisters spaced 0.9 m apart



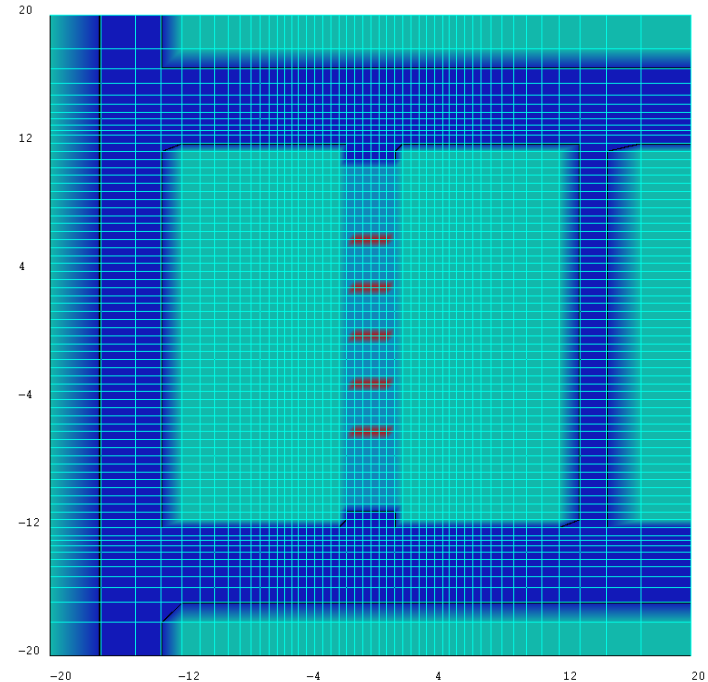
220W Canisters reach 95C

2014 Harp, D.R., P.H. Stauffer, P.K. Mishra, D.G. Levitt, B.A. Robinson, Modeling of High-Level Nuclear Waste Disposal in a Salt Repository, accepted, Nuclear Technology.

*Many of the remaining simulations are for a set of
5 canisters lying in-drift on the floor*



*3-D model domain with red
access tunnels and green
backfill. Intact salt is cyan.*



*Map view at the drift floor,
canisters are red*

Comparison of Thermal only VS Thermal + water + water vapor

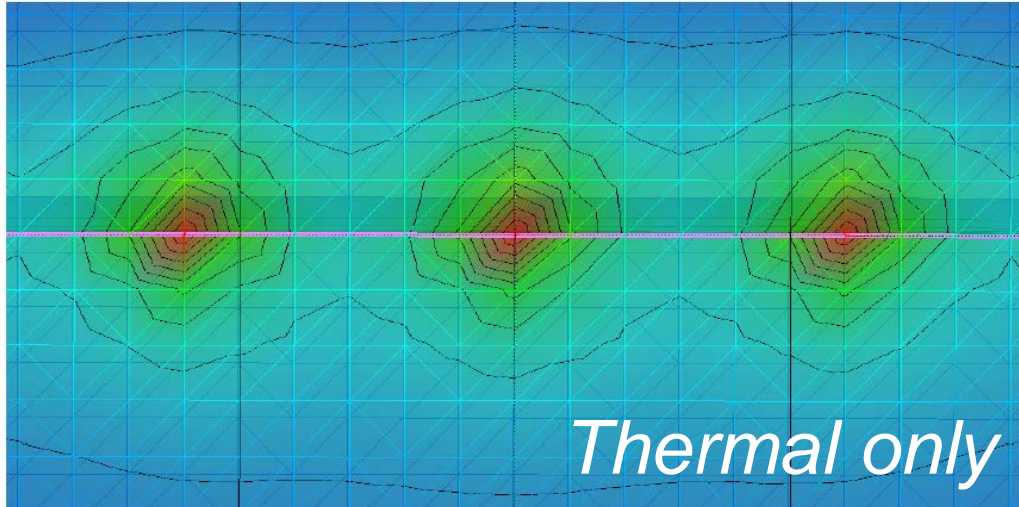
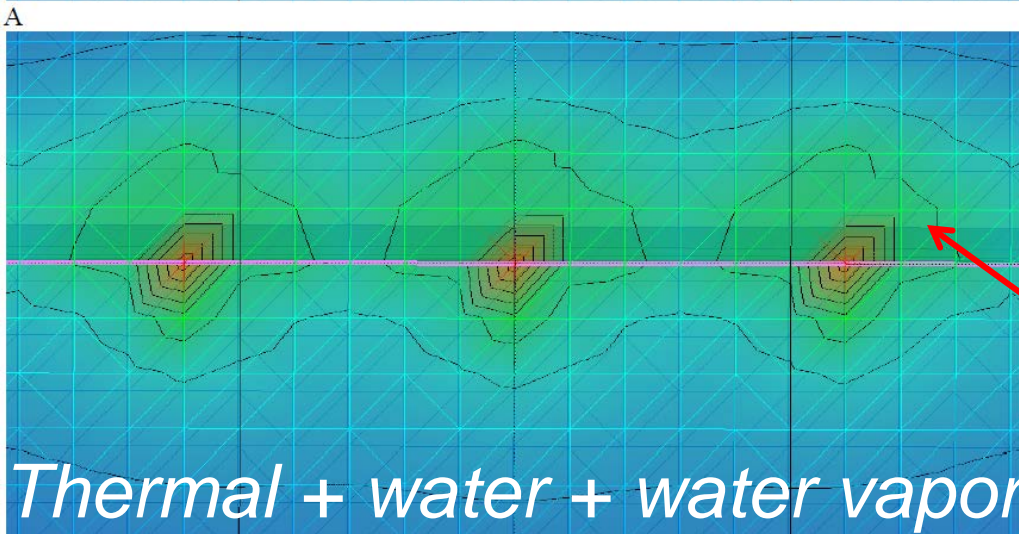


Image is zoomed in on three
of the five heaters

Heat load = 1500W/canister

Time = 730 days after heating
begins.

Canisters spacing = 1 m.



Isothermal region indicative of
heat pipe



30 49 69 88 107 127 146 165 184 204 223

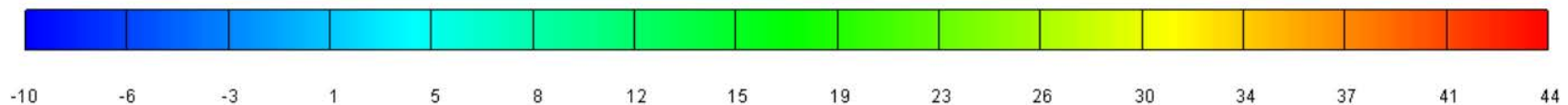
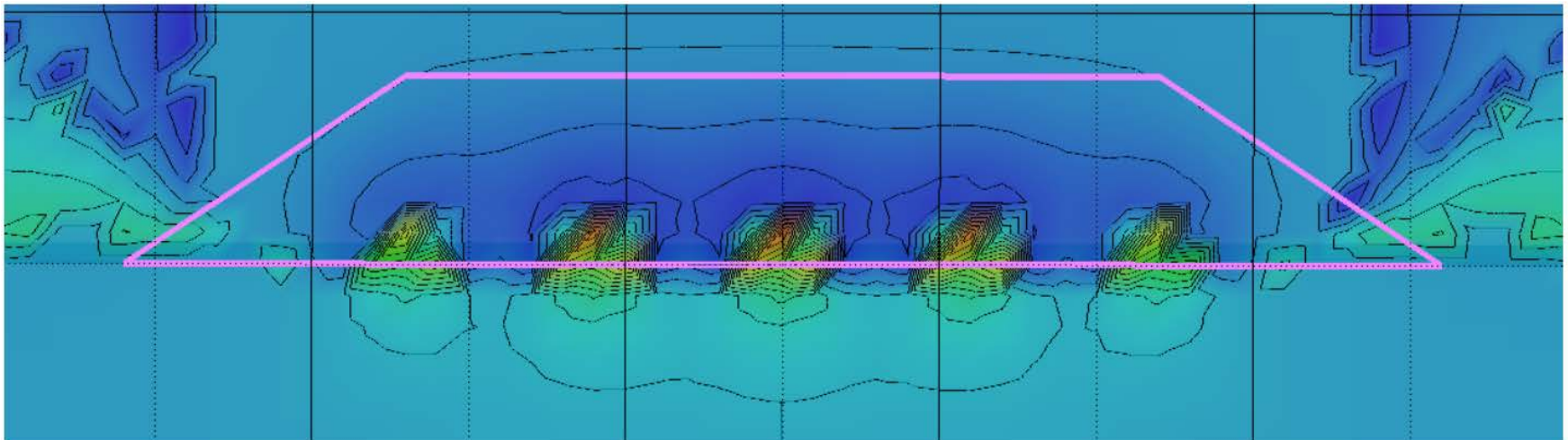
Temperature C

Temperature Difference Image

Thermal only – (Thermal + water + water vapor)

Heat load = 1500W/canister
Time = 730 days after heating begins.
Canisters spacing = 1 m.

Vapor/liquid heat pipe is 44C cooler in the heaters



Delta Temperature C



Thermo Hydrological Chemical Simulations require many coupled processes with feedbacks

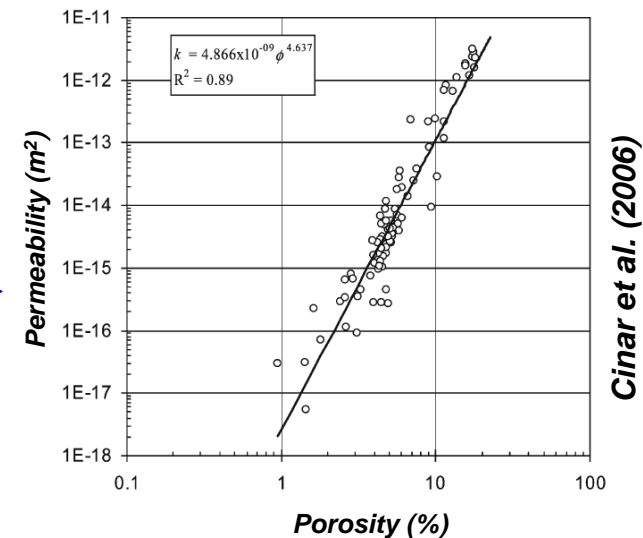
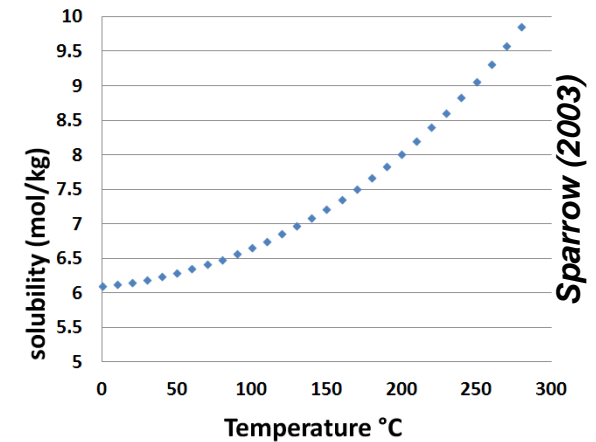
- **Changes in porosity lead to changes in:**
 - permeability
 - thermal conductivity and heat capacity
 - vapor diffusion coefficient
- **Changes in temperature lead to changes in:**
 - thermal conductivity
 - salt solubility
 - water vapor pressure
 - brine viscosity



Salt specific algorithms in FEHM for Thermo Hydrological Chemical Simulations

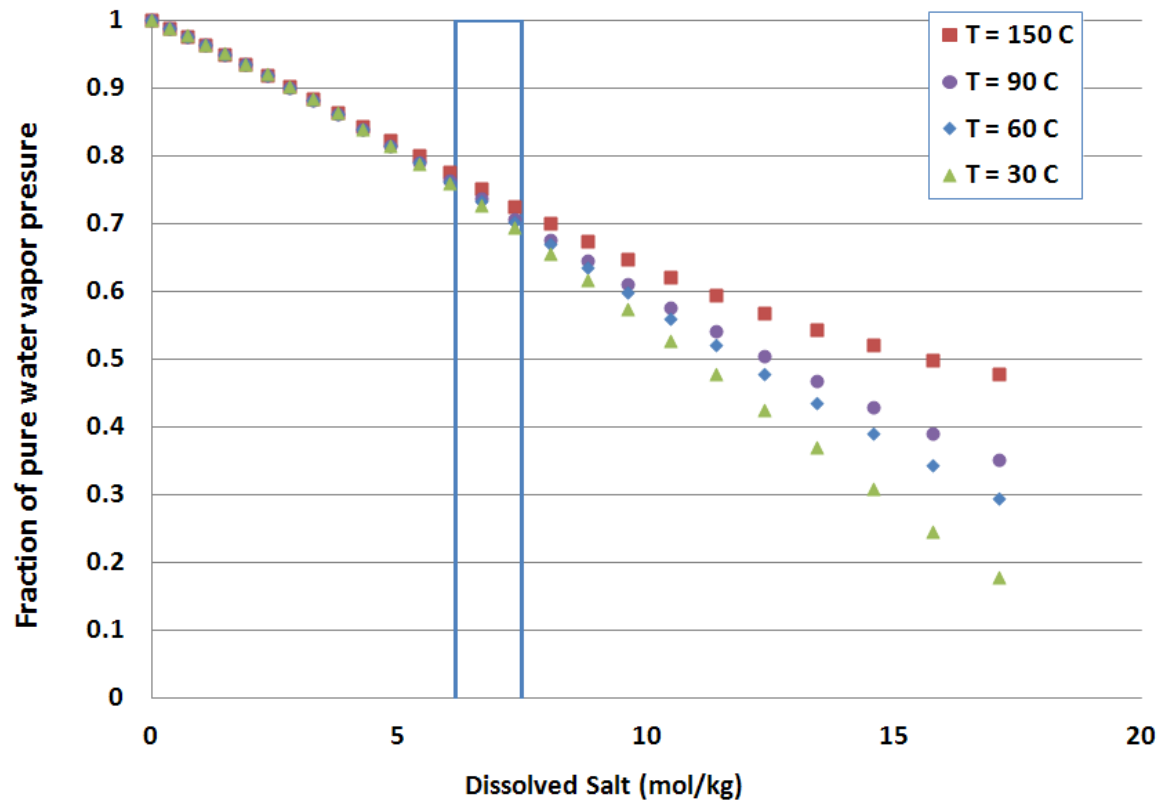
- Thermal Conductivity of Salt as a Function of Porosity and Temperature
- Salt solubility as a function of temperature →
- Precipitation/Dissolution of Salt
- Water vapor diffusion coefficient as a function of pressure, temperature, and porosity
- Capillary pressure relationships
- Permeability-Porosity Relationship for RoM Salt →

2013 Stauffer, P.H., et al., Coupled model for heat and water transport in a high level waste repository in salt, FCRD-UFD- 2013-000206 Los Alamos National Laboratory Document LA-UR 13-27584



Salt specific algorithms in FEHM for Thermo Hydrological Chemical Simulations

■ Vapor Pressure of Water as a function of Aqueous Sodium Chloride Concentration and Temperature



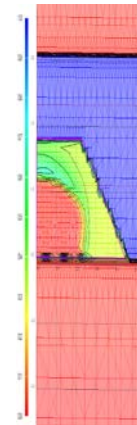
The blue vertical lines span the region of interest for most of our simulations

Sparrow (2003)

Salt specific algorithms in FEHM for Thermo Hydrological Chemical Simulations

- Heat transfer across the air gap (radiation + convection)
- Clay Dehydration
- New diagnostic output (water vapor pressure, vapor diffusion coefficient, permeability, porosity, thermal conductivity)

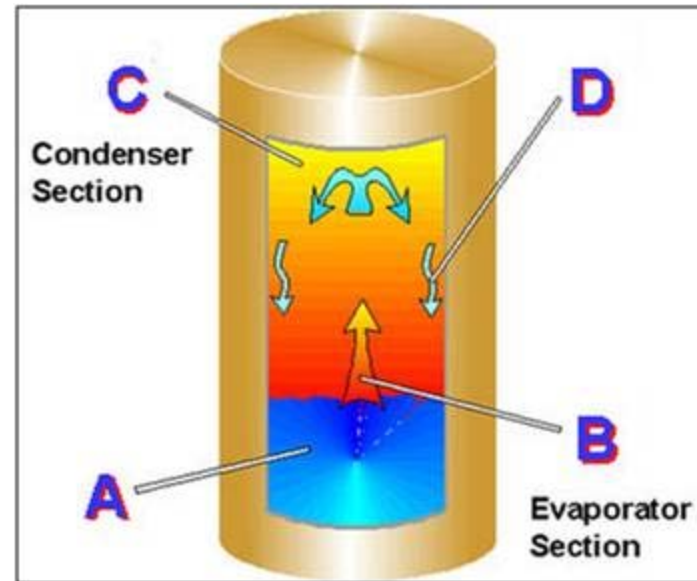
	Node	perm (m2)	porosity	Kx W/(m K)	Pwv (MPa)	D* _{wv} (m2/s)	ps_delta_rxn	
Top	117495	0.10000E-20	0.10000E-02	5.3361	0.31557E-02	0.10629E-12	0.0000	Intact Salt
	102233	0.10000E-18	0.10000E-01	5.0982	0.64491E-02	0.87686E-09	0.29872E-11	Intact Salt
	85866	0.10000E-10	0.99900	14.000	0.80089E-02	0.28997E-04	0.0000	AIR
	70134	0.27929E-11	0.48402	0.57114	0.19505E-01	0.19459E-04	0.39003E-05	Crushed Salt
	54963	0.15400E-17	0.89400E-02	4.0922	0.10769	0.79177E-05	0.0000	Crushed Salt
	43160	0.88841E-17	0.13046E-01	3.9657	0.10906	0.93052E-05	0.0000	Crushed Salt
	39022	0.10000E-20	0.10000E-04	1.1000	0.11204	0.88788E-06	0.0000	Waste Canister
	34337	0.10000E-20	0.10000E-04	1.1000	0.11307	0.90078E-06	0.0000	Waste Canister
	15980	0.10000E-20	0.10000E-04	1.1000	0.84381E-01	0.86175E-06	0.0000	Waste Canister
	10667	0.10000E-18	0.10000E-01	4.2656	0.64085E-01	0.39347E-05	-0.91248E-07	Intact Salt
Bottom	3612	0.10000E-18	0.10000E-01	4.3988	0.44300E-01	0.23626E-05	0.11403E-07	Intact Salt
	2639	0.10000E-20	0.10000E-02	4.8613	0.12512E-01	0.56724E-08	0.0000	Intact Salt
	21	0.10000E-20	0.10000E-02	5.3361	0.31560E-02	0.24695E-13	0.0000	Intact Salt



Example diagnostic output for a vertical line of nodes

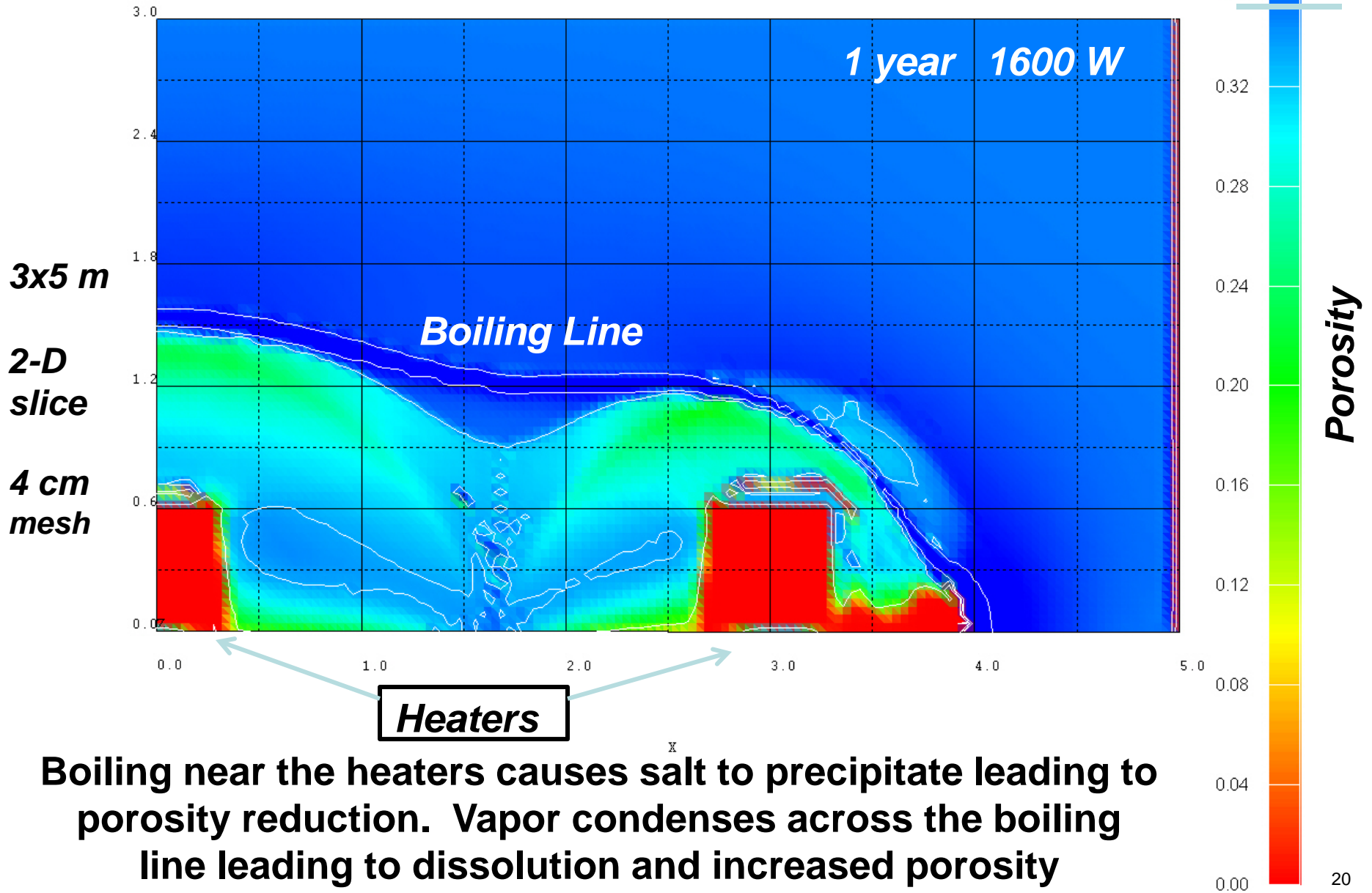
Generic Heat Pipe Explanation

- **Liquid at A**
- **Vaporizes at B**
- **Condenses at C**
- **And D, flows back as liquid to A.**

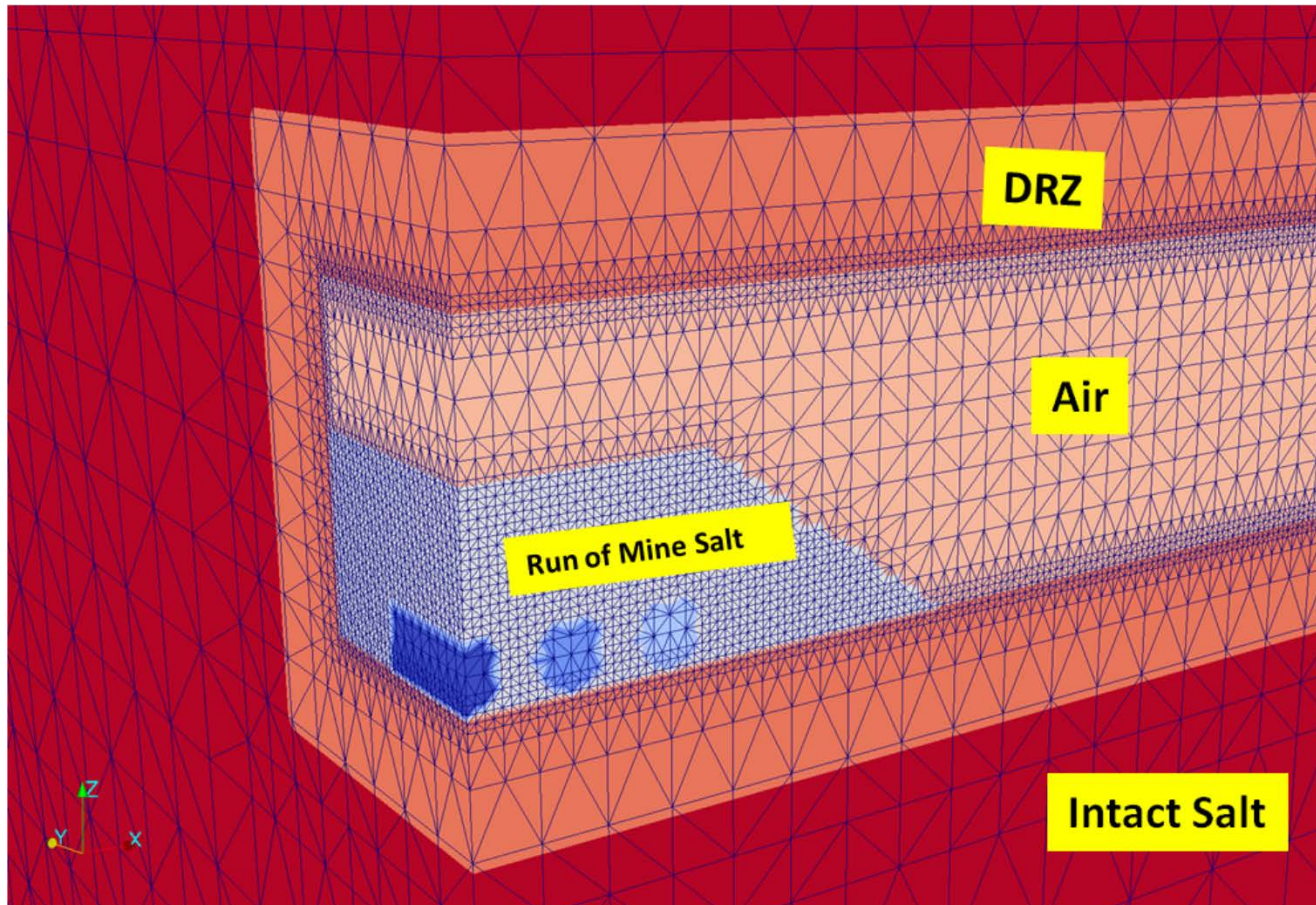


Heat pipes lead to isothermal regions where phase change is absorbing energy

High resolution Thermo Hydrological Chemical Run of Mine salt covering hot waste packages



New mesh to get more resolution for coupled Thermal Hydrological Chemical Simulations



2 Reflection boundaries are used to reduce mesh size (1/4 space)

Range of parameters used in the simulations

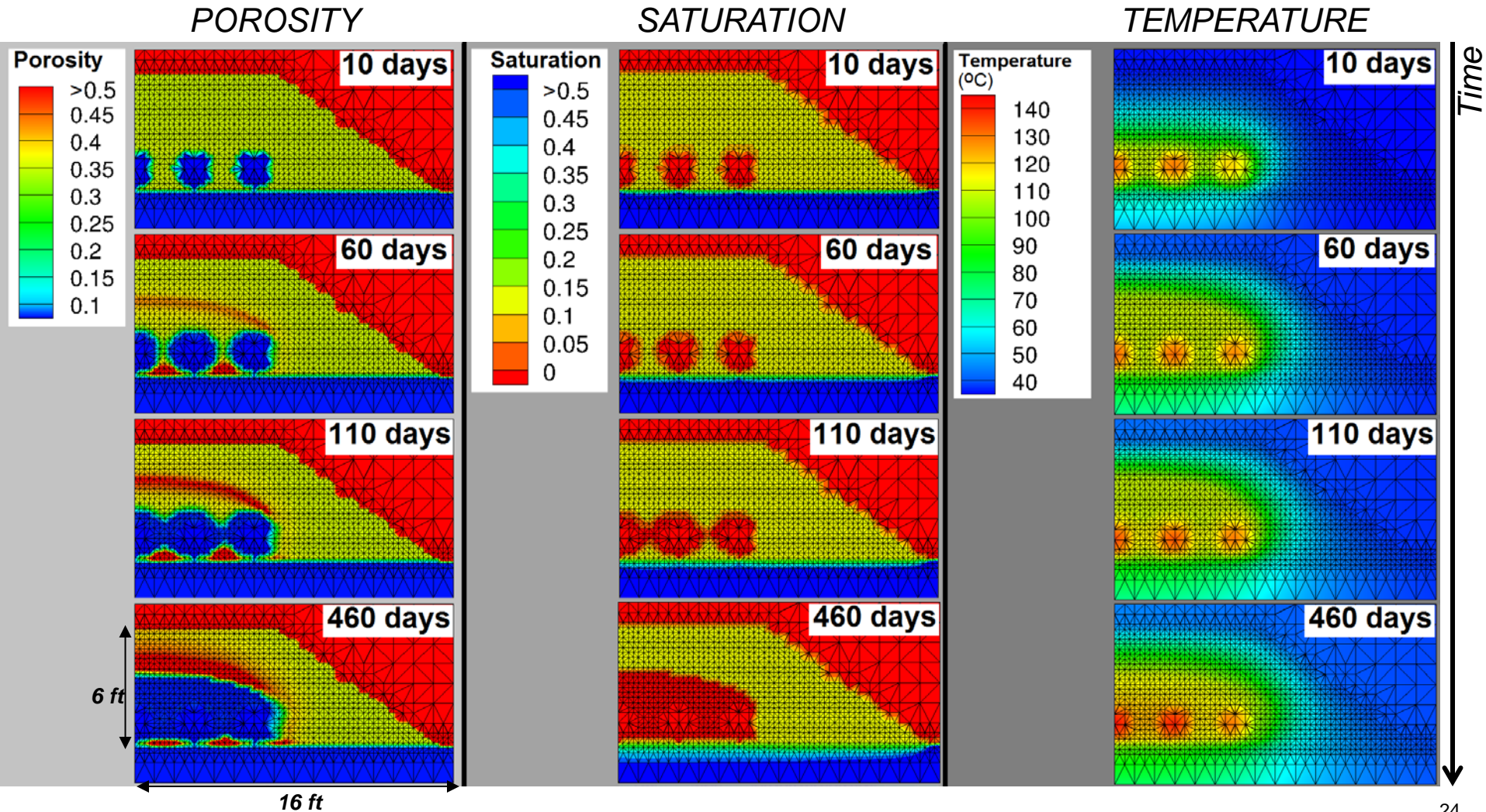
Parameter	Natural Range	Simulated Range
Backfill saturation	0.01 – 0.05+	0.01 – 0.1
Backfill porosity	0.3 - 0.4	0.35
Clay content	0 – 15%+	0 – 10%
Background temperature	15 – 30 C	30 C

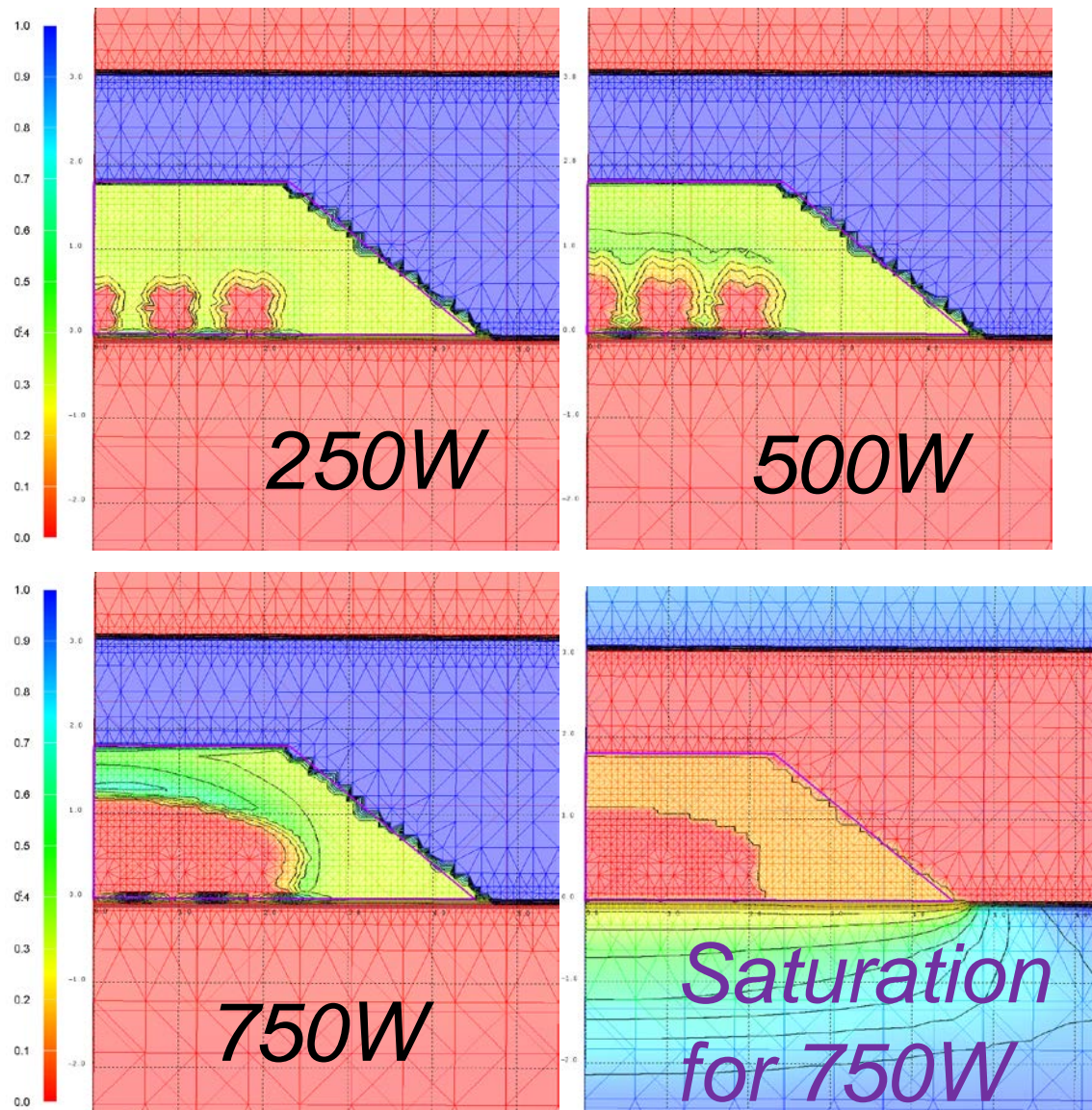
Results

**Fully Coupled
Thermo Hydrological Chemical
simulations
at the drift scale**

Thermo/Hydro/Chem Modeling: Results

Simulation parameters: Heater temperature (750 W), initial saturation in backfill ($S = 10\%$), maximum capillary suction at zero saturation ($P_{cap,max} = 0.5$ MPa), clay fraction (none), residual water saturation ($S_r = 0.1$)

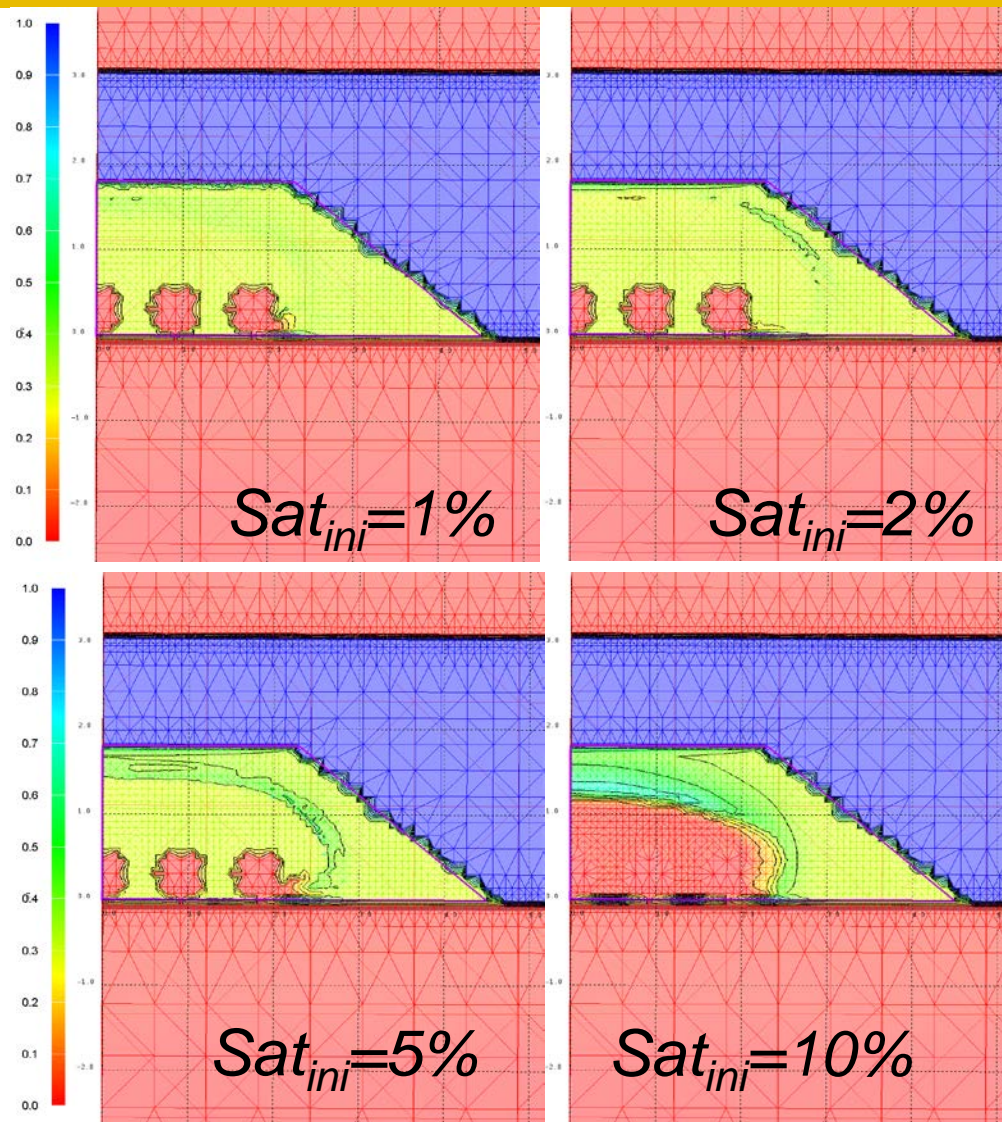




**Porosity
changes more
with
higher
heat loads**

**More heat pipe
at higher
temperatures**

**Time = 2 years
Sat_{ini} = 10%**



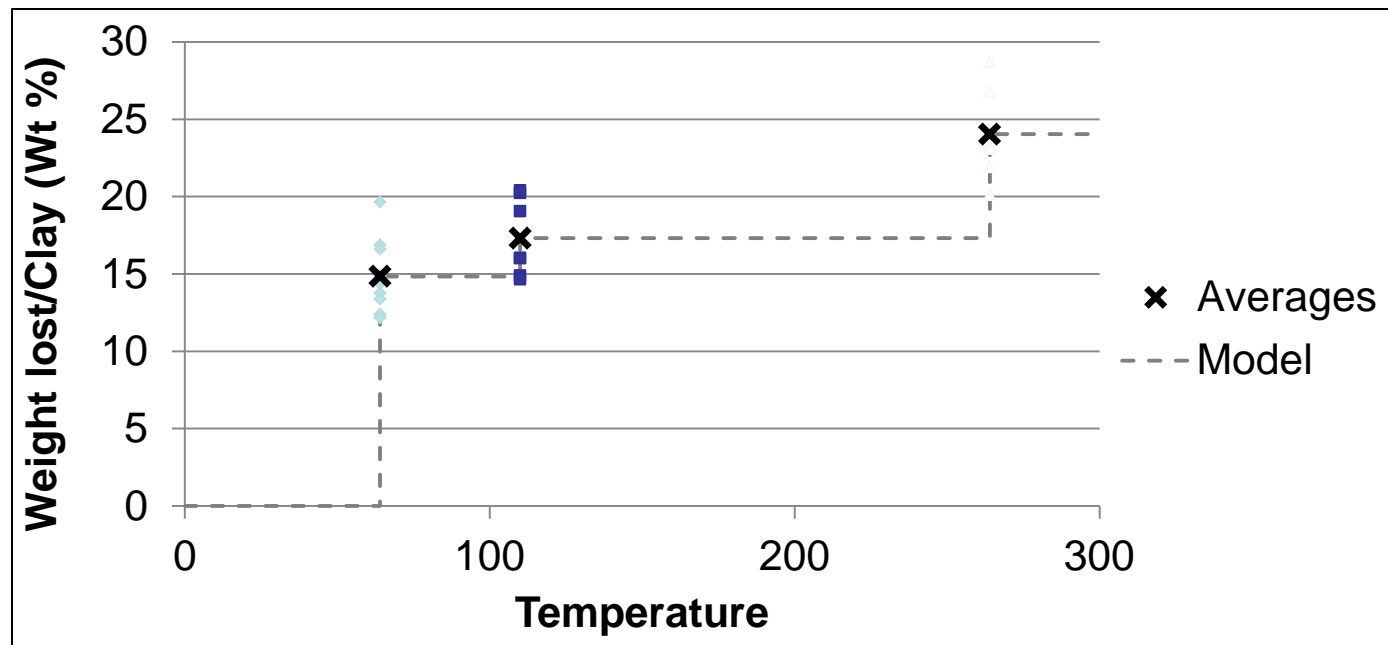
***Porosity
changes more
with
higher
Initial saturation
in the run of
mine salt
backfill***

***More heat pipe
in a wetter
system***

***All at 750W
Time = 2 years***

Clay Dehydration

- WIPP salt is impure – contains clays and other minerals
- In laboratory experiments with run-of-mine (RoM) salt, water release from clays was observed at discrete temperatures:



Data from
Hakim
Boukhalfa

- Mass of water produced at 64°C at node i based on the fraction of clay (f_c), porosity, density of rock, and volume of the cell:

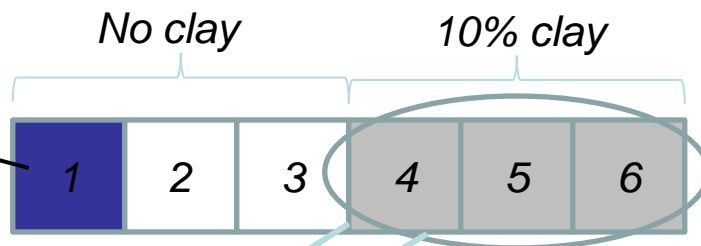
$$M_w = 0.148 f_c (1 - \phi_i) \rho_r V_{cell}$$



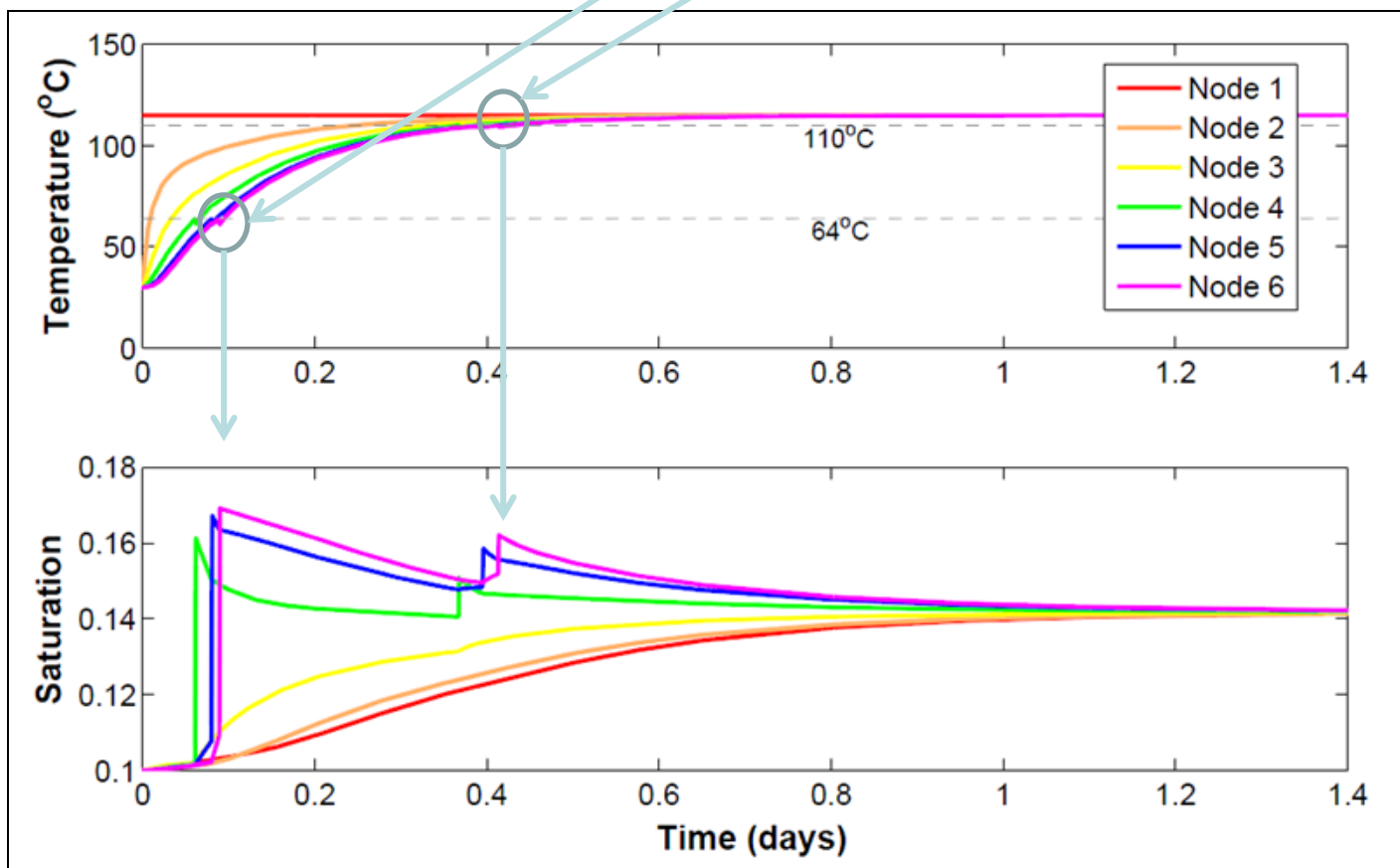
Clay Dehydration: Test Problem

6 nodes

Boundary condition:
115°C



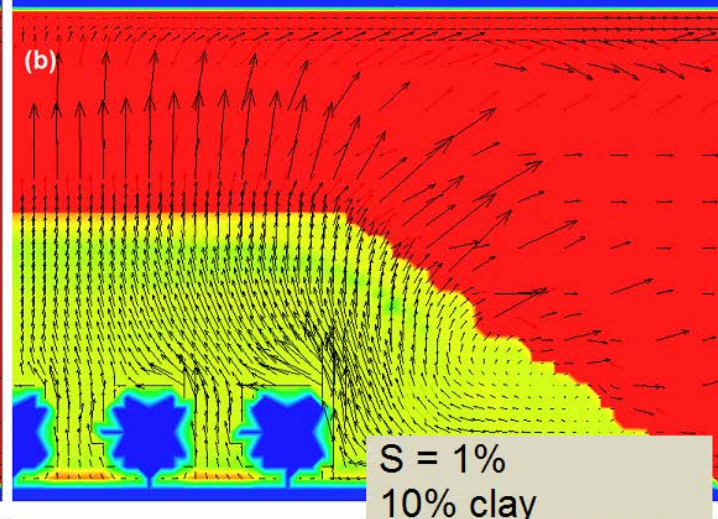
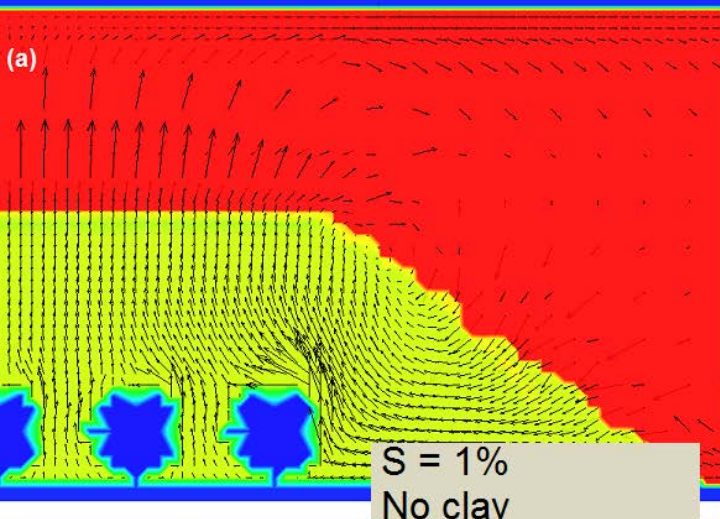
Initial condition: Node 1
= 115°C
Nodes 2 – 6 = 30°C



Clay Dehydration Modeling: Results

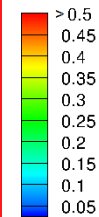
No clay

10% clay



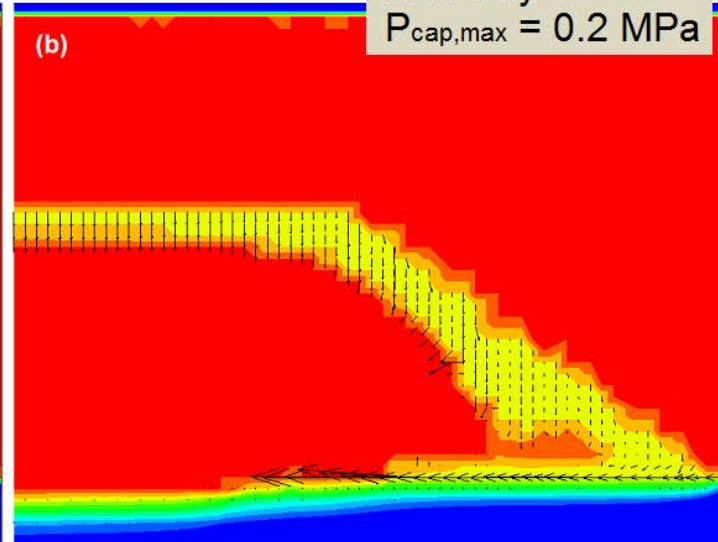
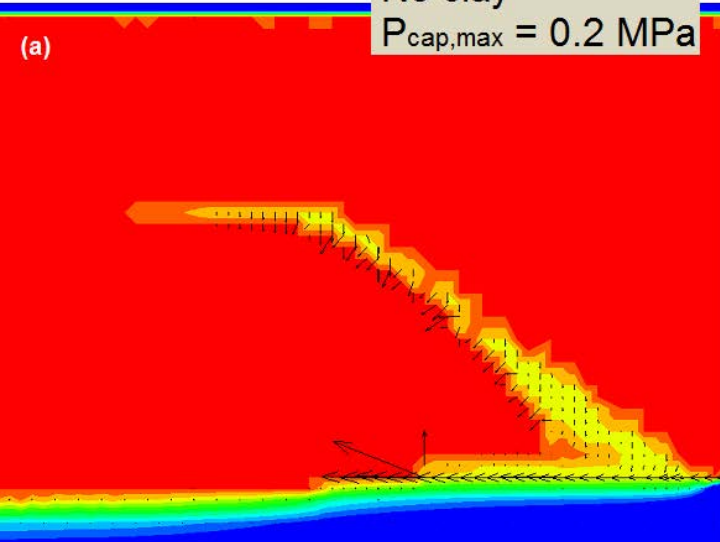
Results at 460 days

Porosity

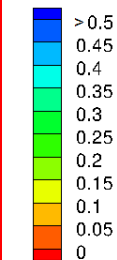


Vapor flux (m/s)

→ = 3.2×10^{-7} m/s



Saturation



Liquid Flux (m/s)

→ = 1×10^{-8} m/s

Conclusions

- **Including water and water vapor in simulations leads to**
 - Not much change in low energy cases (<250W per canister)
 - Heat pipes in some higher energy cases (>250W per canister)
 - Lower temperatures near the canisters
 - Salt mass transfer toward the canisters
 - Increased thermal conductivity near the canisters
 - Heat pipe development is positively correlated with:
 - Initial backfill saturation
 - Backfill capillary suction
 - Clay content in the backfill
 - Water mobility at low saturation
 - Water movement into the backfill from the damaged rock zone

Future Work

■ Experimental validation

- Heat pipe generation in Run of Mine salt backfill
- Retention characteristics of Run of Mine salt backfill
- Drift scale testing at WIPP

■ Inclusion of isotopic tracers in the simulations

■ Inclusion of evaporation

- Barometric pumping
- Pressure flow through the underground
 - Seasonal humidity and pressure differences
 - Bulkhead impacts
 - Damaged rock zone impacts